# Environmental Health, Public Safety, and Social Impacts Associated with Transportation Accidents Involving Hazardous Substances

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#### 16. Abstract

Accidents involving chemicals or radioactive materials represent a significant threat to the environment, public health and safety, and community well-being. In an increasingly complex and interconnected world, no community is immune from the threat posed by environmental accidents and contamination. Even communities far removed from industrial production or storage facilities can still be at risk from accidents associated with the transport of hazardous materials. While a variety of studies have been conducted on aspects of major transportation accidents, few have attempted to examine both environmental and community aspects of the problem. In contrast, this report takes an integrated approach to hazardous transportation accidents by considering environmental, safety, economic, and psychosocial issues. The purpose of the project is to (1) quantify transportation-related accidents involving hazardous materials in the state, and (2) identify key longer-term environmental health, public safety, and social impacts that are often overlooked after major transportation-related hazardous materials accidents.

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#### **Executive Summary**

Accidents involving chemicals or radioactive materials represent a significant threat to the environment, public health and safety, and community well-being. In an increasingly complex and interconnected world, no community is immune from the threat posed by environmental accidents and contamination. Even communities far removed from industrial production or storage facilities can still be at risk from accidents associated with the transport of hazardous materials. While a variety of studies have been conducted on aspects of major transportation accidents, few have attempted to examine both environmental and community aspects of the problem. In contrast, this report takes an integrated approach to hazardous transportation accidents by considering environmental, safety, economic, and psychosocial issues. The purpose of the project is to (1) quantify transportation-related accidents involving hazardous materials in the state, and (2) identify key longer-term environmental health, public safety, and social impacts that are often overlooked after major transportation-related hazardous materials accidents.

The project was comprised of four main tasks: consultation with key stakeholders; summarizing and analyzing representative transportation-related accidents involving hazardous materials that have occurred in Alabama since 1990; presentation of simplified chemical transport and fate models; and presentation of information to help anticipate important social, psychological and related community impacts that can occur after major transportation-related hazardous materials accidents.

Three case studies of transportation accidents involving hazardous materials are presented. The first, which took place near Dunsmuir, CA, in 1991, involved a train derailment that spilled a large quantity of the pesticide metam sodium. The second case study, a truck accident on Interstate-65 in Alabama, was far smaller and far less serious than the Dunsmuir case. It is noteworthy, however, because it illustrates how an accident involving even a very small quantity of hazardous material can produce significant problems. The third case study presented is of a massive gasoline pipeline break and resulting explosion that occurred in 1999 in Bellingham, WA. All three of these case studies present extensive discussions of community impacts, along with descriptions of the physical problems that occurred during the accidents.

Alabama hazardous material transportation related accident information was collected and analyzed using data from the National Response Center. The purpose of this task was to identify the most common hazardous materials lost, where the accidents occurred, and which media was affected. This information was used to present procedures that can be used to predict the movement and dispersion of the lost material.

More than 1700 transportation related accidents involving hazardous materials occurred in Alabama during the past ten years, involving a large number of different materials, although many petroleum hydrocarbons were the most common hazardous material lost. Of the 226 reported accidents in 1998, there were 20 deaths and 27 injuries. In addition, four accidents caused property damage, two accidents resulted in evacuations, and nine accidents resulted in road closures. The locations with the most frequent spills are the USS Alabama Battleship and the hazardous waste landfill at Emelle, probably due to diligent reporting by the site operators. Additional locations of frequent spills include several sites where chemicals are transferred from marine craft to land vehicles, such as trains and trucks.

The report presents several procedures to predict the fate and transport of spilled hazardous materials. The initial discussion is a general procedure that stresses downwind toxic and explosive hazards, summarized from a recent EPA manual and is applicable for a wide range of hazardous materials. Two detailed examples are also presented describing problems associated with spills of petroleum hydrocarbons, by far the most common material lost in Alabama transportation accidents, and losses of ammonia, an example of a toxic gas.

Major transportation accidents involving hazardous materials can produce profound economic, social and psychological impacts in affected communities. These impacts can be both widespread and long lasting. Details from the Bellingham pipeline explosion are presented, along with a more general discussion of the economic, social and psychological effects of hazardous transportation accidents. Current scientific research is reviewed, examples are provided, and implications are considered.

Recommendations and conclusions are presented which are intended to illustrate the types of community impacts that can occur and steps that can be taken to enhance preparedness and response capabilities. The report also contains extensive appendices which present detailed information of Alabama accidents for the past ten years, and properties of hazardous materials that are needed for the calculation of expected exposure conditions.



"Workers transfer drums of hazardous material from the overturned truck into a van" (July 24, 1998). (Copyright Photo by *The Birmingham News*, 2000. All rights reserved. *Reprinted with permission*).

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We wish to express our thanks to the stakeholders from state agencies and other organizations, who kindly agreed to meet with us and share their insights and experiences. Finally, we are grateful to the anonymous reviewers at UTCA who provided us with valuable feedback on our research proposal and this final report.

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#### **Section 1. Introduction**

#### **Project Rationale**

Accidents involving chemicals or radioactive materials represent a significant threat to the environment, public health and safety, and community well-being. In an increasingly complex and interconnected world, no community is immune from the threat posed by environmental accidents and contamination. Even communities far removed from industrial production or storage facilities can still be at risk from accidents associated with the transport of hazardous materials. In the U.S., a staggering 4 billion tons of hazardous materials are moved each year via highways, railroads and other transportation routes (Lillibridge 1997; Quarantelli 1993).

Fortunately, the majority of transportation accidents involving hazardous materials are small and relatively easily managed. However, when major transportation accidents involving hazardous materials do occur, serious environmental health, safety and social problems can result. Indeed, depending on the nature and circumstances of an accident, some impacts can be both widespread and long-lasting.

While a variety of studies have been conducted on aspects of major transportation accidents, few have attempted to examine both environmental and community aspects of the problem. In contrast, this report takes an integrated approach to hazardous transportation accidents by considering environmental, safety, economic, and psychosocial issues. The approach combines the insights and experience of several disciplines, including civil and environmental engineering, public health, and social and behavioral science.

The project focuses on the medium and longer-term impacts of transportation-related accidents involving hazardous materials. Rather than addressing the already well-explored topic of immediate emergency response and cleanup activities, this project deals with issues specifically related to contingency planning and post-emergency response. More specifically, the purpose of the project is to (1) quantify transportation-related accidents involving hazardous materials in the state, and (2) identify key longer-term environmental health, public safety, and social impacts that are often overlooked after major transportation-related hazardous materials accidents.

The project addresses the University Transportation Center for Alabama's (UTCA) priority on safety issues. Furthermore, the high priority topic of technology transfer is also addressed because an upper division/graduate class is being developed on environmental modeling for contingency planning utilizing the material presented in this research report. This class will be one of four graduate-level classes related to disaster management at UAB. The others are Natural Disaster Policy, Complex Disasters (in the School of Public Health) and an interdisciplinary course on Environmental Disasters (see Becker 2000). In addition, information from this report will also be used in Environmental Management classes at UAB. Finally, material from the project can also be presented in a condensed format as a short course as part of other technology transfer projects funded by UTCA.

#### Methodology

The project was comprised of four main tasks: consultation with key stakeholders; summarizing and analyzing representative transportation-related accidents involving hazardous materials that have occurred in Alabama since 1990; presentation of simplified chemical transport and fate models; and presentation of information to identify and mitigate potential long-term adverse community impacts.

Stakeholder Meetings: Formal stakeholder meetings were held with staff from a variety of agencies and organizations that have a role to play planning for, or responding to, accidental hazardous releases. This included the Alabama Department of Transportation, the Alabama Department of Environmental Management, the Alabama Department of Public Safety, and others. In addition, informal discussions were held with personnel from the

Alabama Department of Public Health, the Red Cross, and local emergency responders. Information from the stakeholder meetings was used to identify issues needing coverage in the report.

Diversity, Frequency, and Magnitude of Transportation Accidents Involving Hazardous Materials: For this task, the major types of transportation-related accidents involving hazardous materials in Alabama were quantified and described. The major source of information was the National Response Center's (NRC) nation-wide database on oil and hazardous materials spills. From this database, all transportation accident information for Alabama since 1990 was summarized. Data analyses were conducted so as to measure frequency of accidents by severity (volume of chemical spilled and number of accidents involving a particular chemical) and by location. Public records of several newspapers in the state were also reviewed (especially the Birmingham News and Post Herald, the Huntsville Times, the Anniston Star, the Mobile Register, the Montgomery Advertiser, plus the Gadsden and Dothan newspapers) to compile case histories of several representative transportation-related accidents. However, because many of these accidents were only reported in one issue of the paper, a complete case study for Alabama was only prepared for one transportation-related accident, the acrylonitrile spill on Interstate 65 in 1994. Additional case studies were also prepared for several notable national and international transportation accidents (a gasoline pipeline explosion in Bellingham, Washington; a train derailment in Dunsmuir, California; and a train derailment in M issasauga, Canada). These additional accidents were examined to provide additional information about local response scenarios and potential long-term social impacts of major transportation-related accidents that involved hazardous materials.

Simplified Chemical Transport and Fate Models: Hazardous materials that may be involved in transportation-related accidents are highly varied in their characteristics and potential amounts that may be lost during an accident. In addition, site conditions where an accident occurs can have significant effects on the behavior of the released materials. The results of the database analysis were used to determine the categories of potentially problem-causing chemicals frequently spilled in the state (such as petroleum hydrocarbons, ammonia, and chlorine). Transport and fate estimation procedures for several classes of chemical compounds, using methods given by EPA 1999, Thomann and Mueller 1987; and Turner 1993; were used to produce generic (a some specific) exposure procedures in this report. This approach has frequently been used during the preparation of contingency plans (as required for the Coast Guard National Response Center and Federal Regional Contingency Plan regulations) for complex chemical facilities where numerous chemicals may be involved. In fact, several examples taken from oil spill and ammonia contingency plans and environmental impact reports, are included as case studies. These general procedures, in addition to the specific procedures for petroleum hydrocarbons and ammonia, should cover the majority of accident conditions that have been encountered in the state.

The steps involved in predicting potential exposures to hazardous materials involved in transportation-related accidents are generally as follows:

- 1. Identify materials lost, location (land or water), amount lost, and loss rate (and volume).
- 2. Predict likely combinations of materials that may be involved in individual accidents that may increase the seriousness of the incident.
- 3. Predict the fate of the spilled material (air or water media)
- 4. Estimate downwind atmospheric and downstream water concentrations.

Identification of Potential Longer-Term Community Impacts of Major Transportation Accidents: Firefighters, police officers and other first responders have accumulated considerable experience in identifying and managing the immediate effects of transportation-related hazardous material incidents. Well-established protocols are in use, and training is conducted on a regular basis. However, because there is far less experience dealing with longer-term impacts, these effects can easily be overlooked. The project's fourth task, therefore, was to provide information to help anticipate important social, psychological and related community impacts that can occur after major transportation-related hazardous materials accidents. To do so, this report drew upon information from the three above-noted tasks, plus recent social science and public health studies. The two-fold aim was to enhance university-based training related to transportation accidents in the state and contribute to the state's planning, preparedness and response process.

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# Section 2. Transportation Accidents Involving Hazardous Materials: Two Case Studies

In this section, two case studies of transportation accidents involving hazardous materials are presented. The first, which took place near Dunsmuir, California in 1991, involved a train derailment that spilled a large quantity of the pesticide metam sodium. The second case study, a truck accident on Interstate-65 in Alabama, was far smaller and far less serious than the Dunsmuir case. It is noteworthy, however, because it illustrates how an accident involving even a very small quantity of hazardous material can produce significant problems.

#### Case Study: Train Derailment near Dunsmuir, California, July 14, 1991

The town of Dunsmuir, California lies near the base of Mt. Shasta along the Sacramento River. The town itself is sits close to the river, and people from all over come to fish for wild trout there. As U.S. Representative C. Christopher Cox noted, "tourism and fishing in particular have been vital to the town's economy." At the same time, Dunsmuir is also a railroad town, with many of its citizens having worked for Southern Pacific through the years.

At approximately 9:40 pm on July 14, 1991, a 6000-foot long train operated by Southern Pacific Railroad derailed outside of Dunsmu ir. The train had 4 diesel electric locomotives and 97 cars, 86 of which were empty. A car containing metam sodium landed partially inverted in the water, sending some 19,000 gallons of the chemical into the Sacramento River. Developed during World War Two, metam sodium is a herbicide that is used as a soil fumigant. When it interacts with water, it breaks down quickly into several byproducts, including methylisothiocyanate (MITC), methylamine and hydrogen sulfide. These breakdown products immediately begin to be released as a gas and are respiratory irritants. Indeed, according to Dr. Lynn R. Goldman, Acting Chief of the Office of Environmental and Occupational Epidemiology within the California Department of Health Services, MITC has some similarities to MIC, the chemical that caused serious respiratory effects in victims of the 1984 Bhopal, India chemical disaster. "MITC is very similar in structure to MIC; it has similar toxicological effects, although it has different potency."

Early the next morning, the environmental damage wrought by the spill was clearly in evidence. There were dead fish in the river and the foliage above the river was beginning to wither. As Howard Sarasohn, Deputy Director of the California Department of Fish and Game explained:

... the damage caused by the spill took a number of different forms. As the plume of airborne contaminants moved down the river, all plants and animals in its path were exposed, as were all life form ins in the river as the waterborne plume moved down it. We observed that virtually all of the plants and animals in the river were killed instantly: Fish, algae, plankton, insects, and other organisms. It literally sterilized the stream. Many of the effects were visible in the form of dying fish and, of course, the foliage began to turn brown and fall off.

In addition, according to statements by Southern Pacific, a report of an odor and burning, teary eyes came in early that morning from Dunsmuir, as did word of the tail end of a light yellow green plume being spotted about a half mile south of Southern Pacific's Dunsmuir yard office. By noon, the California Highway Patrol closed a major highway that runs adjacent to the Sacramento River after complaints of discomfort from fumes. A mandatory evacuation of Dunsmuir had also been ordered by the City Manager, but this was downgraded to a voluntary evacuation about an hour later.

This combination -- mandatory highway closing and voluntary evacuation of the town - was to be viewed angrily by some area residents. In testimony before Congress, Kristi Osborn from Concerned Citizens of Dunsmuir said the following:

Most people, if notified at all, were told that evacuation was voluntary and definitely not necessary. This included some pregnant women and senior citizens with preexisting health conditions. Traffic on the freeway was stopped and rerouted, but if you were local, it was perfectly safe to be here. After the freeways was reopened, travelers were told to drive through Dunsmuir without stopping, and they were told not to use their air conditioners or vents and keep their windows shut tight. It was safe for us to live here, but it was not safe for motorists to breathe while driving through. When we complained about the double standard, the people traveling through were no longer warned. We had hoped instead for some concern over the townspeople.

There was also controversy over the quality of information that was available. Dr. Lynn Goldman from the California Department of Health Services complained that inadequacies in available information hampered efforts by public health officials to protect the public:

In the first place, metam sodium was not contained in the emergency response manual that is compiled by the Department of Transportation.... Second, the material safety data sheet (MSDS) that is available in almost every workplace is largely inadequate. Lack of information about long-term effects and releases of the substances at high levels and poor quality assurance are the major shortcomings. So, even though an MSDS was quickly available, the information provided was inadequate. Third, because metam sodium is a pesticide, much of the detailed data about its toxicity are considered to be "trade secrets."

Information related to birth defects was of particular concern:

In this case, public health agencies did not have prompt access to very important information related to birth defect hazards (neural tube defects) of the metam sodium and possibly of MITC as well. The data summaries had been prepared by the regulators at the EPA and within the state of California did not include this information. To be sure we had all the information that was available, we sent a toxicologist into the locked room at the California Department of Pesticide Registration in order to dredge through an enormous shelf of dense technical documents. As soon as we were able to evaluate the information, we shared it with the public. Unfortunately, this was a few weeks after the spill occurred, so that we were not able to use it to inform the public during the spill. We were able to warn the public about the possibility of neural tube defects if a woman had been exposed during the first few weeks of pregnancy. There is a blood test called the AFP that detects this type of birth defect during the early part of pregnancy. But... we learned that three women who were pregnant in the area have suffered adverse reproductive effects: two had premature births and one had a child that was still born. Were these problems caused by the spill? We may never know. But any parent who is placed in this situation will naturally suspect this as a cause for their misfortune.

The lack of complete and timely health information left some residents disillusioned and angry. As citizen group leader Kristi Osborn put it, "When can we trust our public health officials? They have destroyed their credibility, and there is no way to take our fear away."

A preliminary evaluation of the spills health effects by the California Department of Health Services (Goldman) noted the following impacts:

During the week after the spill, 6 persons were admitted to the hospital for illnesses most likely related to spill by-products.

Three others, a person with chronic lung disease and two persons with asthma were admitted for worsening of their prior medical problems.

Three others were admitted for new problems, one with nausea, vomiting and dizziness and a second with pneumonia. The last was a worker who had helped with the initial response and was admitted to the hospital for an unusual cardiac arrhythmia.

Many more minor illnesses were observed in the aftermath of the spill. A review of emergency room records between July 15 and July 31 found a total of 252 visits, compared to 8 visits the first three weeks of August. The most common symptoms that occurred were nausea (51%), headache (44%), eye irritation (40%), throat irritation (26%), dizziness (23%), vomiting (22%), and shortness of breath (21%).

In addition, workers who were brought in to clean up the spill in and near the river on July 21 and 22 developed unusual skin rashes on the feet and ankles, despite the fact that levels of contamination were thought to be extremely low.

Finally, Dr. Lynn Goldman also expressed concern about the psychosocial impacts of the accident:

The community may be experiencing considerable stress, as a result of the spill, the relocation, the uncertainties that they have had to experience. This can cause symptoms during the immediate period but can also have significant long term medical consequences.

Later studies would should that such concerns were well-founded, with residents affected by the spill showing a range of psychosocial impacts. (These are discussed in Section Five.)

Southern Pacific has taken steps to help the community of Dunsmuir recover from the chemical spill. Among other things, the company

- Offered to fund the re-stocking of the river and assist with logistics.
- Opened a community assistance office in Dunsmuir and opened two claims offices, one in Dunsmuir and one at Lake Head, and have settled over 500 claims.
- Paid for over 500 physical examinations in a community of 2100 people.
- Begun paying a bill totaling \$1,400,000 submitted by government agencies for their emergency response costs.

At the time, they had paid about \$2 million on the cleanup and for individual and community assistance. They were also working with the community of Dunsmuir on a public relations campaign to get tourists back to their area, including promotional train trips for Southern Pacific employees and others, the proceeds of which have gone to the restoration efforts within the community. In addition, they agreed to pay the startup costs of a computer database and library to have all current and future information about the spill and its aftermath.

There are varying views within the community about the short-term and long-term effects of the accident. Dr. William Baker, an area physician expressed the view that "the long term effects of exposure will be very minimal." And Ron Martin, a member of the Dunsmuir Chamber of Commerce, called on the EPA to "give our air and water a clean bill of health and publicize it." Martin zeroed in on the media and the need to restore the town's tarnished image:

The air is still fresh and the water is still the best on earth. People are not dying in Dunsmuir due to our air and water. In general, they are very healthy and have a very delightful town to visit and reside in. Our economy had suffered a severe blow due to inaccurate and negative media coverage. What we need is our town to be made whole

In the view of Kristi Osborn of Concerned Citizens of Dunsmuir, making the town whole would be difficult. In the aftermath of the accident, said Osborn, the town was split.

Tourism and fishing in particular, have been vital to the town's economy. The town is built around the river, physically, economically, and emotionally. However, Dunsmuir is also a railroad town. Train memorabilia is everywhere. Generations of families have made their livings with Southern Pacific. Now, sadly the community is divided, and it is difficult for some to choose sides.

The effects of the spill, said Osborn, were profound: "There are hundreds of people still sick in a town with a population of considerably less than 3000. I'd call that a 'significant' number. We didn't cause this disaster, but we are paying for it with our everyday lives." Furthermore, Osborn did not expect the lingering impact of the spill to go away anytime soon. The "biggest concern is, in 5 years, how will our health be? Or in 10 years?" Concluded Osborn: "We all want to forget the spill, but we, as people who have been forced to live in the midst of the disaster, have changed. The spill affects our lives daily and will for a very long time."

Note: This case study was based upon materials from *Train Derailments and Toxic Spills*, A Hearing before the Government Activities and Transportation Subcommittee of the Committee on Government Operations of the House of Representatives, One Hundred and Second Congress, First Session, October 3, 1991, Washington, DC: U.S. Government Printing Office, 1992.

## Case Study: "The Big One." A Rural Community Responds to a Transportation Accident Involving a Hazardous Material, Interstate-65, Monday, February 7, 1994

A March 8, 2000 story in the *Birmingham News* noted that "One in every 20 tractor-trailer rigs traveling through Birmingham contains hazardous cargo, according to a survey conducted for the Jefferson County Emergency Management Agency." The potential for an accident on the highways exists when so many hazardous materials carriers pass through a major metropolitan area. Birmingham has a hazardous materials response unit. However, many small communities do not, and the question becomes "what happens when an accident happens in the jurisdiction of a small community?" The community of Warrior, Alabama found out on February 7, 1994.

The spill occurred near the Warrior-Robbins exit on Interstate-65 about 20 miles north of Birmingham, Alabama. About 4:15 am, firemen from the Warrior City (pop. 3357) volunteer fire department responded to the call involving a tanker truck that had overturned on the interstate median (*Birmingham News*, February 7, 1994), apparently caused when the driver of the truck lost control of the vehicle (*Birmingham News*, February 8, 1994) due to trying to avoid a cinder block in the road (*Birmingham News*, February 9, 1994). A later investigation by the Alabama State Police report that the driver lost control of the truck when he fell asleep, although the driver and the trucking company both deny the police report (*Birmingham News*, February 23, 1994). The firefighters removed the two injured men from the vehicle, discovered that the truck was carrying a hazardous material, and then pulled back and established a perimeter (unidentified firefighter, personal communication). The truck, a tanker from Miller Transporters Inc. of Jackson, Mississippi, was carrying a load of acrylonitrile (also known as 2-propenenitrile or vinyl cyanide), a toxic substance used in the making of acrylic fibers (*Birmingham News*, February 8, 1994).

Although the tanker was carrying approximately 6,000 gallons of acrylonitrile (*Birmingham Post-Herald*, February 8, 1994a), only about 1 gallon of this substance was released as a result of the accident (*Birmingham News*, February 10, 1994), since the tanker leaked but did not rupture in the accident. As the firemen looked it up in their "yellow book," they knew that they would need some help with this situation because although some of the firemen had gone through hazardous materials training, they did not have the appropriate equipment, both for personal protection and for actual cleanup, that was needed. They had responded to the accident and pulled the injured persons wearing only their regular turn-out gear (unidentified firefighter, personal communication). The guidelines from the "orange book" (and the International Safety Card on acrylonitrile) state that the acrylonitrile is colorless or pale yellow liquid with a pungent odor. The vapor is heavier than area, i.e., it can travel along the ground, and vapor/air mixtures may be explosive. The substance decomposes on heating producing toxic fumes including nitrogen oxides, and hydrogen cyanide. It violently with strong oxidants and strong bases, causing a fire and explosion hazard. The recommendation is that immediate area should be evacuated. Cleanup should include collecting leaking liquid in covered containers and absorb any remaining liquid in sand or an inert absorbent. Acrylonitrile should not be washed into the sewer system; it is toxic to aquatic organisms. The concern with the location of this accident was that "there are storm drains in the median that run directly into an unnamed tributary of Cane Creek," said James

Davidson of the Alabama Department of Environmental Management (*Birmingham Post-Herald*, February 8, 1994a).

This chemical is the 39<sup>th</sup> highest volume chemical produced in the United States. According to Catherine Lamar, spokesperson for the Alabama Department of Environmental Management, acrylonitrile is in a category with those chemicals classified as "poisonous or fatal if inhaled, swallowed or absorbed through the skin. Contact may cause burns to skin and eyes" (*Birmingham News*, February 7, 1994). According to the International Safety Card information, acrylonitrile can enter the body through inhalation, ingestion, and skin absorption [occupational exposure limits: TLV 2 ppm vapor, 4.3 mg/m³ by skin]. Inhalation can be expected to cause headaches, dizziness, nausea, vomiting, tremors and uncoordinated movements. Non-fatal exposure is treated with fresh air and rest. The symptoms of ingestion include, in addition to the nausea and headaches, abdominal pain and shortness of mouth. Treatment of ingestion is through drinking a slurry of activated charcoal and inducing vomiting. Long-term effects of exposure of non-lethal, short-term exposure may be on the liver and central nervous system, and medical observation is recommended. Long-term or repeated exposure may cause dermatitis if exposure is through the skin, and acrylonitrile is a probable carcinogen. Periodic medical follow-up is recommended by the International Safety Card.

The Warrior City volunteer fire department, with the help of the Warrior city police and the Jefferson County Sheriff's Department, established a perimeter of one-half mile around the accident site and evacuated about 100 persons (initial reports were of 200 evacuated) from area homes and businesses in the perimeter area by going door to door (*Birmingham News*, February 8, 1994). The Jefferson County Sheriff's department and the Alabama state troopers were called in to handle traffic control as four miles of both the northbound and southbound lanes of Interstate 65 were closed to traffic. At least 60,000 cars were re-routed through the small town of Warrior along U.S. Highway 31 between the time of the accident and 1 p.m., and an unknown number were to follow before the interstate was re-opened to traffic at 7:30 p.m. Willis Graves, a Warrior resident who lives along Hwy 31 spent much of the day watching the long line of traffic backed up in front of his house and blocking him from leaving his driveway, thankful that he "wasn't planning on doing much today anyway." Re-routed drivers spent an average of four hours navigating the detour (*Birmingham News*, February 8, 1994). Warrior public schools were dismissed approximately forty-five minutes early on Monday due to the traffic. "The traffic was moving at such a slow pace, it would be night before some of the children got home," according to William Leatherwood, acting Warrior Police Chief (*Birmingham Post-Herald*, February 8, 1994a).

Once the perimeter was established and the traffic situation under control, the volunteer firemen called upon the local Emergency Management Agency (EMA) and the Alabama Department of Emergency Management (ADEM) for assistance. The Occupation Safety and Health Agency (OSHA) also became involved, as did Emergency Response Specialists, a private firm specializing in clean-up that was hired by the transportation company (unidentified firefighter, personal communication). Food was brought in and there was much discussion of how to proceed (necessary?). Clean-up began about three hours from the time of the accident and took about 12 hours to complete. A crew from Emergency Response Specialists had to transfer the remainder of the load from the tanker truck before it could be righted and moved. Once the tanker was away from the scene, the crews removed the visibly-contaminated soil from the median (*Birmingham Post-Herald*, February 8, 1994a). Tests of the soil surrounding the accident site were taken both by Emergency Response Specialists and ADEM. Preliminary results of these tests showed only minimal contamination (16 ppm at one sample site and 0.094 ppm at a second site), according to Lisa Moore, president of Environmental Response Specialists (*Birmingham News*, February 9, 1994). Workers returned to the site on Monday, February 14, 1994 and removed the top 12 inches of soil from the area surrounding the spill because it was contaminated by diesel fuel that was also spilled in the accident (*Birmingham News*, February 8, 1994).

Two men were pulled from the truck and taken to Carraway Methodist Medical Center in Birmingham where they were treated for minor cuts and released (Birmingham Post-Herald, February 8, 1994a). At least 12 firefighters, state police officers, and other emergency workers were treated at the scene or at Carraway (*Birmingham News*, February 8, 1994). The original responders as well as the other volunteer fire personnel who helped in this situation were encouraged to go to the hospital by emergency management personnel (unidentified firefighter, personal communication). One firefighter from the Kimberly, Alabama, fire department reported that they "could smell the

chemical all around us. There were guys getting headaches. Some of them said they could taste it." Another firefighter reported tightness in his chest. All were given blood tests and released. The results of these blood tests showed that 11 of these firefighters suffered some inability to oxygenate blood, potentially as a result of inhaling the acrylonitrile. One firefighter's wife reported that her husband's blood work showed an oxygen level of about seventy-five percent of normal levels. However, a spokesperson for Miller Transporters, Inc., said that "such a small leak wouldn't be enough to harm the suits or the firefighters. He suggested heat exhaustion may have caused their symptoms" (*Birmingham News*, February 11, 1994).

As can be seen in the reports from the *Birmingham Post-Herald* (February 8, 1994b), the spill and resulting evacuation affected the area residents. "It was not a normal day for 94-year old Henry Montcrief. He was having breakfast with his brother-in-law when a police officer knocked on his door. 'We did not even finish breakfast. I had to drive eight or nine miles around and it is usually just a mile.' The brother-in-law, C.M. Hunter said the news of a chemical spill made him nervous. 'I was just afraid of a gas of some kind. I just wanted to get away as quick as I could." Lt. Carl Johnson described the meeting that he had with a young mother who was trying to return to her apartment in the restricted area. "I told her that everyone was being evacuated to Warrior City Hall or the community center. And she started crying and saying, 'But I have to get home. My baby is wet.' People get upset when you do anything to disturb their sense of security."

The first concern after the incident was that the firefighters' gear was contaminated. "Until Warrior can be assured the suits are safe, firefighters won't use the gear, said Clay Neely, the fire department's adviser. "We can't send someone into a fire with a question mark" (*Birmingham News*, February 11, 1994). The spokesperson for Emergency Response Specialists said that no evidence was found that the gear would have absorbed the acrylonitrile, and that even if contamination was found, the gear could be treated and reused (*Birmingham News*, February 10, 1994). Tests were performed on all of the gear by Emergency Response Specialists and six firefighter suits were replaced as a result of the incident (unidentified firefighter, personal communication). Two lawsuits were filed after the incident. The city of Warrior filed a \$21,000 claim to have the transportation company replace the other eighteen sets of firefighter suits that the city feared were contaminated. "Firefighters fear that clothing exposed to the extremely flammable chemical will ignite when exposed to a fire," according to Brad Fuller, the deputy fire chief of Warrior. The Kimberly fire department, a second responder to this accident, had twelve of its firefighters' suits replaced by its insurance company, who was then planning to pursue reimb ursement from the trucking company (*Birmingham News*, March 17, 1994).

The city of Warrior also sued for lost tax revenue as a result of the accident, alleging that the accident caused the interstate to be closed and took business, and therefore tax revenue, from businesses along the interstate. The owner of the T & G Family Restaurant said, "It (chemical spill) has hurt my business. All I got were restroom customers today" (*Birmingham Post-Herald*, February 8, 1994b). The owner of a package store forced to close estimated that he lost \$8,000 in gasoline sales on the day of the spill. A local building supply company estimated that it lost at least \$4,000 (*Birmingham News*, March 17, 1994).

There was minimal impact of the spill on the fire department itself. No firemen quit the department following the incident, nor was there an increase in interest in becoming a member of the department from the larger community. There was, however, an increase in desire for further training among members of the department as a result of the accident. A dozen or more are now "technicians" in the fire department and have more training than the regular fire fighters, especially in the area of hazardous material management. At the time of the accident there were three "technicians" with this training. The department has made an effort to gain more training, but there is still no hazardous material gear for them to use, because it is too expensive for this rural town to purchase. The department has people trained to use the hazmat gear, they just do not have the equipment or the money (Fire Chief Tommy Hale, personal communication)

One result of the spill has been that the fire department has received more training in dealing with situations involving hazardous chemicals, but they have no more hazardous materials equipment than at the time of the accident. They would still be forced to respond to hazmat calls in only their regular turn-out gear, as they have nothing better. Even though the town of Warrior is only 20 minutes away from Birmingham, the town was responsible for dealing with the accident with minimal help from surrounding areas.

In the small town of Warrior, where this accident is still referred to in the fire station as "the big one," some fear one day another tanker truck will lose control on the interstate that passes about a mile from the downtown. Another day in which they will get the call for which they are still unprepared, for in the words of their current chief Tommy Hale, his voice filled with frustration, "we have the training, we just don't have the equipment to deal with this" (Hale, personal communication).

In the state of Alabama, acrylonitrile is transported in larger quantities than seen in this accident on the waterways. Just over one year after the Warrior accident, a tank barge carrying 903,000 gallons of acrylonitrile ran aground in the Tenn-Tom Waterway about three miles above the Bevill Lock at Pickensville. Fortunately, no material was released to the environment in this incident. The lessons from Warrior should, however, cause concern in many small communities, such as Pickensville, that may be forced to deal with a major transportation-related chemical emergency (*Birmingham News*, March 13, 1995).



Figure 2-1. "Firefighters in golf cart look on from safe distance as workers in protective clothing load spilled chemical into a tanker from an overturned truck on Interstate 65" (Feb. 8, 1994) (Copyright Photo by *The Birmingham News*, 2000. All rights reserved. *Reprinted with permission*).

#### **References for Section Two**

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# Section 3. Analysis of Transportation-Related Chemical Spill Data for Alabama

This report section summarizes the information collected and analyzed from the National Response Center involving transportation related accidents occurring in Alabama. The purpose of this task was to identify the most common hazardous materials lost, where the accidents occurred, and which media was affected. This information is used in the following section which describes methodologies that can be used to predict the movement and dispersion of the lost material.

This database is very comprehensive and includes all spills and accidents reported to local authorities and to the Coast Guard. It therefore incorporates many accidents that are of no interest to this project (such as sewage overflows and offshore marine operations). This project task included the following activities: separating the Alabama records, purging reports of no interest to the project, sort by transportation mode and location, sort by material, and sort by material lost.

Major features of the state's transportation network include:

- five major interstate highways and an extensive network of surface highways,
- the second longest inland waterway system in the nation and a deep-water port that is the nation's 12<sup>th</sup> busiest,
- five Class I railroads,
- eight commercial airports and 91 general aviation facilities,
- Alabama has almost 95,000 miles of roadways and motorist travel approximately 50 billion miles along them every year,
- the Port of Mobile serves 1,100 vessels annually, generating 66,000 truck movements and 119,000 train movements to and from the facility,
- there are over 5,200 miles of railroad track mileage in Alabama and that Birmingham is a major Southeastern hub.

With the large amount of transportation activity in the state, it is not surprising that more than 1700 transportation related accidents involving hazardous materials occurred in Alabama during the past ten years. These accidents have involved a large number of different materials, although petroleum hydrocarbon compounds were the most common hazardous material lost.

#### Methodology

This project phase consisted of collecting information on transportation accidents in Alabama that involved hazardous materials from the databases available from the National Response Center (NRC). The NRC's "primary function is to serve as the national point of contact for reporting all oil, chemical, radiological, biological, and etiological discharges into the environment anywhere in the United States and its territories" (<a href="http://www.nrc.uscg.mil/nrcback.html">http://www.nrc.uscg.mil/nrcback.html</a>, December 20, 2000). The NRC forwards these reports to the appropriate federal agencies, including the Department of Transportation, the Department of the Interior, the Department of Defense, the Department of Health and Human Services, the Federal Emergency Management Agency, the Environmental Protection Agency, the Nuclear Regulatory Commission, and the Federal Railroad Administration. The NRC is operated by the U.S. Coast Guard as part of the National Oil and Hazardous Substances Pollution Contingency Plan. Although the main intention of this database is to record losses of hazardous materials, many other materials have also been reported and included in the database by local law enforcement officials, environmental regulators, and shipping companies.

The database maintained by the NRC is accessible through the website <a href="http://www.nrc.uscg.mil/">http://www.nrc.uscg.mil/</a>. At the time of the this project, the databases covered the years 1990 through 1999. The NRC makes the information available in four files per calendar year. The first file describes the incident itself; the second, a description of the material(s) involved; the third, information on any railroad trains involved in the incident; and the fourth, information on derailed railroad cars. For this project, the four files for each year were combined, using the NRC Incident Report Number, into a single spreadsheet for all accidents that occurred in the state of Alabama during this ten year period. These spreadsheets were then culled for transportation-related incidents, and finally combined into one spreadsheet that describes the incidents reported for the decade of interest, and is presented in Appendix A of this report.

#### **Results**

Table 3-1 shows some of the hazardous materials that have been lost during transportation related accidents recently in Alabama. By far, the most common (and the largest) materials spilled are petroleum oils and fuels (fueloil, crude oil, kerosene, gasoline and diesel fuel). Ammonia material spills were also common. Numerous other toxicants and hazardous materials were also reported. Table 3-2 lists the locations of the 226 reported 1998 Alabama transportation-related accidents and the media directly affected. Of course, many of the land-based accidents affected other media through evaporation (to air) and runoff (to water). In the past 10 years, more than 1700 transportation related accidents have occurred in Alabama involving hazardous materials.

Table 3-1. Partial List of Materials Reported Spilled During Recent Alabama Transportation-Related Accidents

Ammonium	Ammonia,	Ammonium	Arsenic	Butadiene	Chlorine	Caustic Soda	Ethylene
Hydroxide	Anhydrous	Nitrate Solution				Solution	Glycol
Gasoline	Hydrogen Peroxide	Kerosene	Methyl Mercaptan	Yellow Paint	Asbestos	Mercury	Lindane
Sewage	Oil: Diesel	Oil, Fuel: No. 5	Hydraulic Oil	Oil: Crude	Oil, Fuel: No. 2-D	Oil, Transformer	Refrigerant Gases
Sulfuric Acid	Sulfur Dioxide	Sodium Hydroxide	Sulfur Oxide	Triethylene Glycol	Toluene	Turpentine	P-Xylene

Table 3-2. Locations of Reported 1998 Alabama Transportation-Related Accidents

Location and Media Directly Affected	Percentage of 1998 Alabama Transportation-Related Accidents
Highw ays	27%
Railroads	30%
Pipelines	1%
Marine terminals	43%
Land	33%
Water	52%
Air	2%
Unknown	14%

The reported 1998 Alabama transportation-related accidents also resulted in numerous immediate problems to people and property, and disruptions to the transportation systems. However, longer-term problems are not addressed by these accident statistics. Of the 226 reported accidents in 1998, there were 20 deaths and 27 injuries. In addition, four accidents caused property damage, two accidents resulted in evacuations, and nine accidents resulted in road closures.

Of special interest for this project was the frequency of accidents and the amount of the different materials spilled, the hazards of the chemicals spilled, and the accident locations. The spreadsheets generated in this part of the project (Appendix A) are organized according to the format that the information is received from the NRC. The information available includes the following:

- date and time of the accident.
- the location of the incident,
- the suspected responsible party (including contact information),
- the cause of the accident,
- a description of the accident
- a description of the environmental medium affected
- numbers of deaths, injuries and evacuation,
- a description (including volumes) of the chemicals spilled, and
- information on any train cars that derailed in the accident.

In some cases, the volume of chemical spilled was not known at the time of the report. The NRC information lists this lack of information as a "0" volume under the "Quantity Spilled" column. When conducting the additional analyses of the database, these 'potentially-unknown' quantities were retained, as these accidents, especially those involving petroleum products, are a significant fraction of the transportation-related accidents in Alabama. The information that was not retained in the additional analyses was the oil-sheen entries because no uniform representation of volume spilled could be developed based on the information reported to the NRC.

Table 3-3 is a summary of the largest quantities of hazardous material lost for each mode of transportation considered. The accidents occurring at "fixed" locations are generally loading operations and are not associated with building or storage tank disasters. The marine operations include shipping accidents and leaks and underwater pipeline leaks and breaks inland waterways, while the off-shore locations are mostly associated with accidents at drilling and well platforms. These data clearly shows that the most frequently spilled chemicals in Alabama are the petroleum products. In addition to these ethylene glycol (antifreeze) is also commonly spilled. This would be expected in an accident in which the radiator of a vehicle is damaged when the engine is damaged. These data also emphasize the variety of transportation modes (marine, highway, etc.) where these spills occur. Many different hazardous substances can be lost during transportation accidents, in addition to the most common oil and fuel spills. Many of the most hazardous substances were only associated with one or a very few incidents in the ten years of study, and only relatively small quantities of material were lost. Highly hazardous ammonium nitrate, ammonia, molten aluminum, sodium hydroxide, and different acids, were all lost in Alabama during this period.

Tables 3-4 through 3-12 are separated by location of the accidents (highways, railroads, pipelines, etc.) and also includes information, where available, from the National Fire Protection Association (NFPA) regarding the hazards associated with a particular chemical. It is primarily available for organic chemicals. The mode of transport with the fewest overall number of accidents is the air, i.e., airplane crashes. However, large quantities of pesticides (especially malathion) was lost to the environment during 13 crashes of crop dusting planes during this ten year period. The largest single accident was a crude oil spill of about 2,000,000 gallons at a marine terminal (the *T/V R*. *Hal Dean* ran aground in the Pensagoula Ship Channel on Jan 2, 1991, releasing 2,000,000 gallons of crude oil). The largest spills are associated with marine operations (ship casualties being the largest, by far), followed by highway and railroad accidents, and then pipeline accidents. In many cases, just a few accidents accounted for the majority of the spill volume for many substances.

The tables in Appendix B show the locations of the most frequent accidents. The locations with the most frequent spills are the USS Alabama Battleship and the hazardous waste landfill at Emelle. Additional locations of frequent spills include several sites where chemicals are transferred from marine craft to land vehicles, such as trains and trucks. At many of these sites, the quantities spilled are small in each incident. However, it may be anticipated that the frequent spills in one area may cause longer-lasting environmental impacts.

Table 3-3. Largest Spill Quantities Lost for each Major Transportation Mode Examined (1990 – 1999 Alabama Transportation Accidents)

Transportation Mode	Most Common	2 <sup>nd</sup> Ranked	3 <sup>rd</sup> Ranked	4 <sup>th</sup> Ranked
Aircraft accidents	Jet fuel (1330 gals/13 incidents)	Malathion (404 gals/13 incidents)		
Fixed locations	Hydrocarbons (fuel oil, gasoline, crude oil, diesel oil, hydraulic oil, kerosene, asphalt, transformer oil, and creosote) (82,901 gals/250 incidents)	Chromic acid/phosphoric acid (24,000 gal/1 incident)	Coal (12,000 lbs/1 incident)	Sodium hydroxide (5,000 lbs/2 incidents)
Highway accidents	Hydrocarbons (diesel oil, road tar, gasoline, fuel oil, asphalt, LPG, jet fuel, hydraulic oil, and creosote) (184,281 gals/225 incidents)	Poultry fat (49,720 lbs/2 incidents)	Ammonium nitrate and fuel oil (30,000 lbs/1 incident)	Molten aluminum (20,000 lbs/1 incident)
Marine operations	Hydrocarbons (crude oil, diesel oil, fuel oil, asphalt, motor oil, lubricating oil, waste oil, hydraulic oil, gasoline, jet fuel, and lubricating mud) (2,024,569 gals/584 incidents)	Sodium hydroxide (1,000 lbs/1 incident)	Bromine (900 lbs/1 incident)	Adiponitrile (640 lbs/1 incident)
Off-shore locations	Hydrocarbons (lubricating mud, drilling mud, diesel oil, hydraulic oil, crude oil, motor oil, fuel oil) (1188 gals/62 incidents)			
Pipelines	Hydrocarbons (fuel oil, crude oil, diesel oil, and gasoline) (14,166 gals/26 incidents)	Paraxylene (1,000 gals/1 incident)	Salt water (60 gals/1 incident)	Triethylene glycol (35 gals/1 incident)
Railroad and highway crossings	Hydrocarbons (diesel oil, fuel oil, and motor oil) (8,558 gals/13 incidents)	Formaldehyde solution (1 gal/1 incident)		
Railroad accidents	Coal (934,800 lbs/10 incidents)	Plastic pellets (262,500 lbs/2 incidents)	Hydrocarbons (petroleum oil, asphalt, diesel oil, creosote, lubricating oil, and hydraulic oil) (72,959 gals/108 incidents)	Limestone (3,000 lbs/2 incidents)
Unknown locations	Hydrocarbons (gasoline, fuel oil, diesel oil, hydraulic oil, and asphalt) (2,861 gals/191 incidents)	Sodium hydroxide (5 gals/1 incident)	Ethylene glycol (5 gals/1 incident)	

Table 3-4. Summary of Chemicals Spilled by Transportation Mode (air craft accidents)

Name of Material	Number of Incidents	Total Quantity Spilled	Unit of Measure	Range of Quantity Spilled	Unit of Measure	Median Quantity Spilled	Unit of Measure	Health	Flammability	Reactivity	Special
JET FUEL: JP-4	3	700	gal	3-692	gal	5	gal	1	3	0	
MALATHION	13	404	gal	9-62	gal	30	gal				
JET FUEL: JP-8	2	225	gal	25-200	gal	113	gal				
JET FUEL: JP-5 (KEROSENE, HEAVY)	4	205	gal	0-100	gal	53	gal	0	2	0	
JET FUEL	1	100	gal	100	gal	100	gal	0	2	0	
JET A FUEL	1	70	gal	70	gal	70	gal	0	2	0	
AVGAS	1	30	gal	30	gal	30	gal	1	3	0	
DIMILIN 2F	1	15	gal	15	gal	15	gal				
JET A FUEL	1	0	gal	0	gal	0	gal	0	2	0	
BRAVO	1	0	gal	0	gal	0	gal				

Table 3-5. Summary of Chemicals Spilled by Transportation Mode ("fixed" locations, usually transfer stations)

Table 3-3. Summary of Chemicals Spi		- anoportation	1 110000 ( 11		io, acaany			1	1		1
Name of Material	Number of Incidents	Total Quantity Spilled	Unit of Measure	Range of Quantity Spilled	Unit of Measure	Median Quantity Spilled	Unit of Measure	Health	Flammability	Reactivity	Special
CHROMIC ACID/PHOSPHORIC ACID	1	24000	gal	24000	gal	24000	gal	3	0	1	OX
OIL, FUEL: NO. 6	10	19363	gal	0-12000	gal	5	gal	0	2	0	
JET FUEL: JP-8	3	16075	gal	75-9000	gal	7000	gal				
OIL: CRUDE	1	15277	gal	0-12000	gal	20	gal	0	2	0	
COAL	1	12000	lbs	12000	lbs	12000	lbs				
PRODUCED WATER	1	12000	gal	12000	gal	12000	gal				
OIL: DIESEL	63	9636	gal	0-2500	gal	5	gal	0	2	0	
COAL TAR PITCH	1	8000	gal	8000	gal	8000	gal	0	1	0	
GASOLINE: AUTOMOTIVE (UNLEADED)	18	6346	gal	0-4697	gal	28	gal	1	3	0	
PARAXYLENE	1	6000	gal	6000	gal	6000	gal	2	3	0	
SODIUM HYDROXIDE	2	5000	lbs	0-5000	lbs	2600	lbs	3	0	1	
ALUMINUM SULFATE	1	4725	gal	4725	gal	4725	gal				
GASOLINE: AUTOMOTIVE (4.23 G Pb/GAL)	17	4433	gal	0-2000	gal	5	gal	1	3	0	
POTASSIUM HYDROXIDE	4	3700	gal	0-1500	gal	1100	gal	3	0	1	
WAX EMULSION	1	3568	lbs	3568	lbs	3568	lbs				
HYDROXYL AMMONIUM SULFATE SOLUTION (30%)	1	3500	gal	3500	gal	3500	gal	3	0	0	
NAPHTHA: VM & P (75% NAPHTHA)	1	3400	gal	3400	gal	3400	gal	1	3	0	
CHROMATED COPPER ARSENATE	1	3200	gal	200-3000	gal	1600	gal				
NITRIC ACID	1	3000	gal	3000	gal	3000	gal	3	0	0	OX
METHYL MERCAPTAN	4	2566	lbs	145-1510	lbs	456	lbs	4	4	0	
SULFURIC ACID	12	2119	gal	0-800	gal	8	gal	3	0	2	Water
OIL, MISC: MOTOR	28	2028	gal	0-1000	gal	0	gal	0	2	0	
INCINERATOR ASH	1	2000	lbs	2000	lbs	2000	lbs				
TENNECO T500-100	1	1500	lbs	1500	lbs	1500	lbs				
OIL, FUEL: NO. 2-D	23	1460	gal	0-400	gal	0.5	gal	0	2	0	
METHYLENE CHLORIDE	1	1391	gal	1391	gal	1391	gal	2	1	0	
HYDRAULIC OIL	33	1381	gal	0-600	gal	1	gal	0	2	0	
ETHYL ACETATE	1	1332	gal	1332	gal	1332	gal	1	3	0	

Table 3-5. Summary of Chemicals Spilled by Transportation Mode ("fixed" locations, usually transfer stations) (continued)

Name of Material	Number of Incidents	Total Quantity Spilled	Unit of Measure	Range of Quantity Spilled	Unit of Measure	Median Quantity Spilled	Unit of Measure	Health	Flammability	Reactivity	Special
OIL, FUEL: NO. 2	9	1225	gal	0-700	gal	2	gal	0	2	0	
UNKNOWN FUEL OIL	1	1000	gal	1000	gal	1000	gal	0	2	0	
WASH WATER WITH A TRACE OF OIL	1	1000	gal	1000	gal	1000	gal				
OIL, MISC: LUBRICATING	19	840	gal	0-500	gal	0.25	gal	0	2	0	
HYDROGEN SULFIDE	2	811	lbs	0-811	lbs	406	lbs	4	4	0	
WASTE SLUDGE	1	750	lbs	750	lbs	750	lbs				
KEROSENE	2	705	gal	5-700	gal	353	gal	0	2	0	
JET-A	1	400	gal	400	gal	400	gal	0	2	0	
ORTHOXYLENE	1	360	gal	0	gal	0	gal	2	3	0	
WATER BASE YELLOW INK	1	300	gal	300	gal	300	gal				
JET A FUEL	2	275	gal	100-175	gal	138	gal	0	2	0	
DIMETHYL SULFIDE	4	211	gal	0.5-146	gal	32	gal	1	4	0	
STYRENE/BUTADIENE LATEX	1	200	gal	200	gal	200	gal	2	3	2	
WATER BASED ASPHALT	1	200	gal	200	gal	200	gal				
AMMONIA, ANHYDROUS	4	200	lbs	0-200	lbs	100	lbs	3	1	0	
JET FUEL: JP-4	7	200	gal	0-100	gal	20	gal	1	3	0	
FINISH	1	170	gal	170	gal	170	gal				
OIL, MISC: TRANSFORMER	6	162	gal	0-50	gal	35	gal	0	2	0	
MIXTURE OF CRUDE OIL AND DIESEL	1	160	gal	160	gal	160	gal				
OIL, MISC: MINERAL	3	135	gal	10-125	gal	25	gal	0	2	0	
ETHANOL, 2-2-BUTOXYETHOXY	1	133	gal	133	gal	133	gal	1	2	0	
MERCAPTAN	1	120	lbs	120	lbs	120	lbs				
CREOSOTE, COAL TAR	1	100	gal	100	gal	100	gal	2	2	0	
PROPIONITRILE	1	100	lbs	100	lbs	100	lbs	4	3	1	
HAZARDOUS WASTE SOLID/NOS/F006	1	100	lbs	100	lbs	100	lbs				
PROPANOL	1	100	gal	100	gal	100	gal				
SLUDGE	1	100	gal	100	gal	100	gal				
NITRIC ACID	1	92	gal	92	gal	92	gal	3	0	0	OX

Table 3-5. Summary of Chemicals Spilled by Transportation Mode ("fixed" locations, usually transfer stations) (continued)

Table 3-5. Summary of Chemicals Spi		anoportation	· mode ( ii		is, asaany			illacaj	1		1
Name of Material	Number of Incidents	Total Quantity Spilled	Unit of Measure	Range of Quantity Spilled	Unit of Measure	Median Quantity Spilled	Unit of Measure	Health	Flammability	Reactivity	Special
OTHER OIL (UNKNOWN)	7	81	gal	0-80	gal	0	gal	0	2	0	
WASTE OIL	19	67	gal	0-40	gal	0	gal	0	2	0	
N-BUTYL MERCAPTAN	1	50	gal	50	gal	50	gal				
WASH WATER MIXED WITH HYDROGEN SULFITE	1	50	gal	50	gal	50	gal				
WASTE OIL/LUBRICANTS - POSS. CON	6	50	gal	0-20	gal	8	gal	0	2	0	
LEAN AMINE (CAS 105599)	1	45	gal	45	gal	45	gal				
EPOXY CURE	1	42	gal	42	gal	42	gal				
OTHER OIL (REFINERY SLUDGE)	1	40	gal	40	gal	40	gal	0	2	0	
BOILER FLY ASH	1	40	lbs	40	lbs	40	lbs				
CORROSION INHIBITOR OIL	1	40	gal	40	gal	40	gal				
CUTTERS STOCK	1	40	gal	40	gal	40	gal				
ASPHALT	2	40	gal	5-35	gal	20	gal	0	1	0	
BENZO(A)PYRENE	1	30	gal	30	gal	30	gal				
DRILLING FLUID	1	20	gal	20	gal	20	gal				
TINUVIN TARS	1	15	lbs	15	lbs	15	lbs				
ST20	1	13	gal	13	gal	13	gal				
JET FUEL: JP-5 (KEROSENE, HEAVY)	1	10	gal	10	gal	10	gal	0	2	0	
OTHER OIL(LIGHT FUEL OIL)	1	10	gal	10	gal	10	gal	0	2	0	
ROOFING TAR	1	10	gal	10	gal	10	gal	1	2	0	
NITROGEN DIOXIDE	1	10	lbs	10	lbs	10	lbs	3	0	0	OX
CACODYLIC ACID	1	10	gal	10	gal	10	gal				
CONTAMINATED GROUND WATER	1	10	gal	10	gal	10	gal				
WASTE SOLID NOS 9 AND NA3077	1	10	gal	10	gal	10	gal				
BENZENE	2	8.7	gal	0-8.7	gal	4	gal	2	3	0	
MINERAL SPIRITS	2	8	gal	0-8	gal	4	gal	0	2	0	
OIL, EDIBLE: SOYA BEAN	2	8	gal	3-5	gal	4	gal	0	1	0	
ASPHALT/DIESEL FUEL MIXTURE	1	6	gal	6	gal	6	gal	0	3	0	

Table 3-5. Summary of Chemicals Spilled by Transportation Mode ("fixed" locations, usually transfer stations) (continued)

Table 3-5. Summary of Chemicals Spi		Tarisportation	i wode ( ii		is, usually			illueuj	1		
Name of Material	Number of Incidents	Total Quantity Spilled	Unit of Measure	Range of Quantity Spilled	Unit of Measure	Median Quantity Spilled	Unit of Measure	Health	Flammability	Reactivity	Special
OIL:DIESEL (FUEL OIL NO. 5)	3	6	gal	1-5	gal	3	gal	0	2	0	
UNKNOWN OIL	34	6	gal	0-6	gal	0	gal	0	2	0	
AMMONIUM NITRATE	1	5	gal	5	gal	5	gal	0	0	3	OX
HYDRO TREATED GAS OIL	1	5	gal	5	gal	5	gal	0	2	0	
REFINED CHEMICAL OIL	1	5	gal	5	gal	5	gal	0	2	0	
FUEL WASTE	1	5	gal	5	gal	5	gal				
CHLOROFORM	1	3.2	gal	3.2	gal	3.2	gal	2	0	0	
OIL:DIESEL (BUNKER C)	1	3	gal	3	gal	3	gal	0	2	0	
D008	1	3	gal	3	gal	3	gal				
LEACHATE (F039 WASTE CODE)	1	3	gal	3	gal	3	gal				
PENTACHLOROPHENOL	2	3	gal	0-3	gal	2	gal	3	0	0	
GASOLINE: CASINGHEAD	1	2.5	gal	2.5	gal	2.5	gal	1	4	0	
CHLORINE	2	2.3	gal	0-2.3	gal	1.2	gal	4	0	0	OX
OTHER OIL, ROLLING OIL	1	2	gal	2	gal	2	gal	0	2	0	
BILGE SLOPS	1	2	gal	2	gal	2	gal				
CHEMICAL WASTE PRODUCTS	1	2	gal	2	gal	2	gal				
FO32 HAZARDOUS WASTE	1	2	gal	2	gal	2	gal				
NOS 9 MA3077	1	2	gal	2	gal	2	gal				
PROPIONIC ACID	1	2	gal	2	gal	2	gal				
ASPHALT	1	1	gal	1	gal	1	gal	0	1	0	
FEED STOCK OIL	1	1	gal	1	gal	1	gal	0	2	0	
WASTE OIL SLUDGE	1	1	gal	1	gal	1	gal	0	2	0	
M-XYLENE	1	1	gal	1	gal	1	gal	2	3	0	
XYLENE (O-, M-, P-, & MIXTURES)	1	1	gal	1	gal	1	gal	2	3	0	
CREOSOTE	1	0.94	gal	0.94	gal	0.94	gal	2	2	0	
ETHYLENE GLYCOL	16	0.83	gal	0-0.83	gal	0.05	gal	1	1	0	
NO 6 OIL WITH DIESEL MIXED IN	1	0.25	gal	0.25	gal	0.25	gal	0	2	0	

Table 3-5. Summary of Chemicals Spilled by Transportation Mode ("fixed" locations, usually transfer stations) (continued)

Table 3-3. Summary of Chemicals Sp		ransportation	i wode ( ii		is, asaany		10113) (0011	illucuj			
Name of Material	Number of Incidents	Total Quantity Spilled	Unit of Measure	Range of Quantity Spilled	Unit of Measure	Median Quantity Spilled	Unit of Measure	Health	Flammability	Reactivity	Special
ASPHALT (PRIMER)	1	0	gal	0	gal	0	gal	0	1	0	
HEAT TRANSFER OIL	1	0	gal	0	gal	0	gal	0	2	0	
LUBE GREASE	1	0	gal	0	gal	0	gal	0	1	0	
MARINE DIESEL	1	0	gal	0	gal	0	gal	0	2	0	
OIL BASED PAINT	1	0	gal	0	gal	0	gal	0	2	0	
POLYPROPYLENE	1	0	gal	0	gal	0	gal	0	1	0	
TRANSMISSION FLUID	1	0	gal	0	gal	0	gal	0	2	0	
ACETONE	1	0	gal	0	gal	0	gal	1	3	0	
ANTIFREEZE	1	0	gal	0	gal	0	gal	1	1	0	
METHYL ALCOHOL	1	0	gal	0	gal	0	gal	1	3	0	
PROPANE	1	0	gal	0	gal	0	gal	1	4	0	
CAUSTIC SODA SOLUTION	1	0	gal	0	gal	0	gal	3	0	1	
SULFUR DIOXIDE	1	0	gal	0	gal	0	gal	3	0	0	
ACRYLONITRILE	1	0	gal	0	gal	0	gal	4	3	2	
ASBESTOS	1	0	gal	0	gal	0	gal				
COBALT BROMIDE (OUS)	1	0	gal	0	gal	0	gal				
CONTAMINATED SOIL	1	0	gal	0	gal	0	gal				
LATEX	1	0	gal	0	gal	0	gal				
LEAD	1	0	gal	0	gal	0	gal				
LIQUOR, BLACK	1	0	gal	0	gal	0	gal				
MALATHION	1	0	gal	0	gal	0	gal				
MATERIAL OUT OF TANK TRUCK	1	0	gal	0	gal	0	gal				
MTBE	1	0	gal	0	gal	0	gal				
PAINT REMOVER	1	0	gal	0	gal	0	gal				
PAINT THINNER	1	0	gal	0	gal	0	gal				
POISON	1	0	gal	0	gal	0	gal				

Table 3-5. Summary of Chemicals Spilled by Transportation Mode ("fixed" locations, usually transfer stations) (continued)

Name of Material	Number of Incidents	Total Quantity Spilled	Unit of Measure	Range of Quantity Spilled	Unit of Measure	Median Quantity Spilled	Unit of Measure	Health	Flammability	Reactivity	Special
	incluents	'						Health	Tiaminability	Reactivity	Special
REFRIGERANT GASES	1	0	gal	0	gal	0	gal				
SUBSTANCE FROM INSIDE THE DYNAMITE STICK	1	0	gal	0	gal	0	gal				
TIRES	1	0	gal	0	gal	0	gal				
TIRES, ASBESTOS, PAINT CANS, ETC.	1	0	gal	0	gal	0	gal				
TIRES, SHINGLES, SHEET ROCK	1	0	gal	0	gal	0	gal				
TRANSMISSION FLUID	1	0	gal	0	gal	0	gal				
UNKNOWN MATERIAL	1	0	gal	0	gal	0	gal				
UNKNOWN TYPE CORROSIVE	1	0	gal	0	gal	0	gal				
WASH WATER	1	0	gal	0	gal	0	gal				
WASTE PAINT	1	0	gal	0	gal	0	gal				
BURNED TIRES	2	0	gal	0	gal	0	gal				
PAINT	3	0	gal	0	gal	0	gal				
BATTERY ACID	4	0	gal	0	gal	0	gal	3	0	2	Water
FREON	7	0	gal	0	gal	0	gal				

Table 3-6. Summary of Chemicals Spilled by Transportation Mode (highway locations)

Name of Material	Number of Incidents	Total Quantity Spilled	Unit of Measure	Range of Quantity Spilled	Unit of Measure	Median Quantity Spilled	Unit of Measure	Health	Flammability	Reactivity	Special
OIL: DIESEL	93	88260	gal	0-78000	gal	50	gal	0	2	0	
TAR (ROAD)	1	55560	lbs	55560	lbs	55560	lbs	1	2	0	
POULTRY FAT	2	49720	lbs	0-49720	lbs	24860	lbs				
PGP	1	42000	lbs	42000	lbs	42000	lbs				
AMMONIUM NITRATE & 6% FUEL OIL	1	30000	lbs	30000	lbs	30000	lbs	0	0	3	OX
HAZARDOUS WASTE SOLID	1	30000	lbs	30000	lbs	30000	lbs				
ANIMAL FAT	1	22000	lbs	22000	lbs	22000	lbs				
MOLTEN ALUMINUM	1	20000	lbs	20000	lbs	20000	lbs				
GASOLINE: AUTOMOTIVE (UNLEADED)	17	13906	gal	0-8000	gal	15	gal	1	3	0	
PRODUCED WATER	1	10000	gal	10000	gal	10000	gal				
PROPIONIC ACID	1	10000	lbs	10000	lbs	10000	lbs				
OIL, FUEL: NO. 2-D	53	6647	gal	0-2500	gal	60	gal	0	2	0	
HYDROCHLORIC ACID	10	5529	gal	0-4200	gal	50	gal				
COAL TAR PITCH	1	5000	lbs	5000	lbs	5000	lbs	0	1	0	
POULTRY BLOOD	1	5000	gal	5000	gal	5000	gal				
GASOLINE: AUTOMOTIVE (4.23 G Pb/GAL)	13	4926	gal	0-3500	gal	100	gal	1	3	0	
POLYCHLORINATED BIPHENYLS	5	4542	gal	0-4536	gal	0.09	gal	2	1	0	
FERROUS CHLORIDE	1	4500	gal	4500	gal	4500	gal				
ASPHALT	3	4030	gal	0-4000	gal	30	gal	0	1	0	
20-0-20 FERTILIZER; GRANULAR	1	4000	lbs	4000	lbs	4000	lbs				
SODIUM HYPOCHLORITE	4	3800	gal	0-3800	gal	0	gal				
WATERPROOFING RESIN - E-Z-REZ #710	1	3275	lbs	3275	lbs	3275	lbs				
ANILINE	2	2897	gal	150-2347	gal	1448	gal	3	2	0	
METHYL ETHYL KETONE	1	2500	gal	2500	gal	2500	gal	1	3	0	
FERRIC SULFATE	1	2500	gal	2500	gal	2500	gal				
BATTERY RECYCLING WASTE	1	2300	lbs	2300	lbs	2300	lbs	3	0	2	Water
OIL, FUEL: NO. 2	5	2120	gal	0-2000	gal	30	gal	0	2	0	
LIQUEFIED PETROLEUM GAS	2	1600	gal	0-1600	gal	800	gal	1	4	0	

Table 3-6. Summary of Chemicals Spilled by Transportation Mode (highway locations) (continued)

rable of or callinary of chemicals opi					(00						
Name of Material	Number of Incidents	Total Quantity Spilled	Unit of Measure	Range of Quantity Spilled	Unit of Measure	Median Quantity Spilled	Unit of Measure	Health	Flammability	Reactivity	Special
KARMEX (DIURON SOLID NOS)	1	1500	gal	1500	gal	1500	gal			-	
JET FUEL: JP-4	4	1395	gal	15-1000	gal	190	gal	1	3	0	
MONOCHLOROACETIC ACID	1	1320	gal	1320	gal	1320	gal				
OIL, MISC: MOTOR	36	1138	gal	0-1000	gal	0	gal	0	2	0	
CARBON DIOXIDE	1	1000	lbs	1000	lbs	1000	lbs				
AMMONIA, ANHYDROUS	3	1000	gal	0-1000	gal	0	gal	3	1	0	
JET FUEL: JP-5 (KEROSENE, HEAVY)	2	850	gal	400-450	gal	425	gal	0	2	0	
FERTILIZER	1	800	gal	800	gal	800	gal				
SULFUR	2	800	gal	100-700	gal	400	gal	2	1	0	
SODIUM HYDROSULFIDE SOLUTION	1	600	gal	600	gal	600	gal				
STYRENE (35%)	1	510	gal	510	gal	510	gal	2	3	2	
PCB (CONTAMINATED SOIL)	1	500	lbs	500	lbs	500	lbs	2	1	0	
D006 HAZARDOUS WASTE SOLID	1	500	lbs	500	lbs	500	lbs				
COPPER CHLORIDE DIHYDRATE	1	496	lbs	496	lbs	496	lbs				
HYDRAULIC OIL	19	477	gal	0-200	gal	5	gal	0	2	0	
HAZARDOUS WASTE	2	415	gal	15-400	gal	208	gal				
AMMONIUM NITRATE	1	300	gal	300	gal	300	gal	0	0	3	OX
SULFUR (MOLTEN)	1	300	gal	300	gal	300	gal	2	1	0	
PAINT	2	300	gal	0-300	gal	150	gal				
CREOSOTE	3	204	gal	30-104	gal	70	gal	2	2	0	
SULFURIC ACID	5	201	gal	0-186	gal	4	gal	3	0	2	Water
OIL, MISC: COAL TAR & WATER	1	200	gal	200	gal	200	gal	0	2	0	
NTX (75% METHYLENE CHLORIDE, FORMIC ACID	1	200	gal	200	gal	200	gal	2	1	0	
HAZ WASTE SOLID NOS(CONTAINS LEAD OXIDE)	1	200	lbs	200	lbs	200	lbs				
LIQUID ALLUM	1	200	gal	200	gal	200	gal				
JET FUEL: JP-8	4	193	gal	33-60	gal	50	gal				

Table 3-6. Summary of Chemicals Spilled by Transportation Mode (highway locations) (continued)

Name of Material	Number of Incidents	Total Quantity Spilled	Unit of Measure	Range of Quantity Spilled	Unit of Measure	Median Quantity Spilled	Unit of Measure	Health	Flammability	Reactivity	Special
AMMONIUM NITRATE SOLUTION	2	143	gal	17-125	gal	71	gal	0	0	3	OX
OIL, MISC: LUBRICATING	6	108	gal	0-100	gal	1.5	gal	0	2	0	
OIL, FUEL: NO. 1-D	1	100	gal	100	gal	100	gal	0	2	0	
J2 FUEL	1	100	gal	100	gal	100	gal	1	3	0	
PERCHLOROETHYLENE	1	100	gal	100	gal	100	gal	2	0	0	
CAUSTIC ALKALI LIQUID NOS	1	100	gal	100	gal	100	gal				
THIOPHENOL RESIDUE	1	100	gal	100	gal	100	gal				
CHLOROFORM	3	85	lbs	0-68	lbs	17	lbs	2	0	0	
WASTE FLAMABLE LIQUID	2	70	gal	20-50	gal	35	gal				
NAPHTHA: SOLVENT	2	56	gal	6-56	gal	31	gal	1	3	0	
FLAMMABLE LIQUID - ALIPHATIC HYDROCARBON	1	55	gal	55	gal	55	gal				
D001 FLAMABLE LIQUID	1	50	gal	50	gal	50	gal				
MERCURY CONTAMINATED WASTE WATER	1	50	gal	50	gal	50	gal				
NITRIC ACID	2	50	gal	0-50	gal	25	gal	3	0	0	OX
BATTERY PLANT TRASH	1	40	lbs	40	lbs	40	lbs	3	0	2	Water
BENZENE	2	33	gal	30-Mar	gal	17	gal	2	3	0	
FLAMMABLE WASTE LIQUID(NOS)	1	30	gal	30	gal	30	gal				
ORGANOPHOSPHOROUS PESTICIDES	1	30	lbs	30	lbs	30	lbs				
PROPANE GAS	1	25	gal	25	gal	25	gal	1	4	0	
METHYLENE CHLORIDE	1	25	gal	25	gal	25	gal	2	1	0	
ALKALINE CORROSIVE MATERIAL	1	25	gal	25	gal	25	gal				
BARIUM (FILTER CAKE)	1	25	gal	25	gal	25	gal	3	0	3	OX
FUEL WASTE	1	25	gal	25	gal	25	gal				
TOLUENE	1	22	gal	22	gal	22	gal	2	3	0	
ACETONITRILE	1	20	gal	20	gal	20	gal	2	3	0	
CONTAMINATED GROUND WATER	1	20	gal	20	gal	20	gal				

Table 3-6. Summary of Chemicals Spilled by Transportation Mode (highway locations) (continued)

Name of Material	Number of Incidents	Total Quantity Spilled	Unit of Measure	Range of Quantity Spilled	Unit of Measure	Median Quantity Spilled	Unit of Measure	Health	Flammability	Reactivity	Special
K0088: PIPELINE DEBRIS/OTHER EQUIPMENT PARTS	1	20	gal	20	gal	20	gal		,	·	
METHYLENE BISTHIOCYANATE	1	20	gal	20	gal	20	gal				
MIXED/WASTE SOLVENTS - POSS. CON	1	20	gal	20	gal	20	gal				
WASTE WATER TREATMENT SLUDGE	1	20	gal	20	gal	20	gal				
ETHYLENE GLYCOL	23	18.4	gal	0-3	gal	0.11	gal	1	1	0	
FUELS WASTE	1	15	gal	15	gal	15	gal				
WASTE CODE FO39 (LEACHATE)	1	15	gal	15	gal	15	gal				
OIL, MISC: MINERAL	3	15	gal	0-15	gal	0	gal	0	2	0	
BUTADIENE	1	13	gal	13	gal	13	gal	2	4	2	
OIL, MISC: RESIN	1	10	gal	10	gal	10	gal	0	2	0	
DIMETHYL-N-BUTYLAMINE	1	10	gal	10	gal	10	gal	2	3	0	
CREOSOTE CONTAMINATED SOIL AND DEBRIS	1	10	gal	10	gal	10	gal				
HAZARDOUS LIQUID WASTES(FO34)	1	10	lbs	10	lbs	10	lbs				
LEAD BATTERY LIQUID	1	10	gal	10	gal	10	gal				
LINDANE	1	10	gal	10	gal	10	gal				
SWEEPER TRASH (D007)	1	10	gal	10	gal	10	gal				
WASTE MATERIAL DOO1, DOO4 F001, F004	1	10	gal	10	gal	10	gal				
MINERAL SPIRITS	2	5.5	gal	0.5-5	gal	2.75	gal	0	2	0	
INCINERATOR DEBRIS	2	5.5	gal	0.5-5	gal	2.75	gal				
MINERAL SPIRITS	1	5	gal	5	gal	5	gal	0	2	0	
TRANSMISSION OIL	1	5	gal	5	gal	5	gal	0	2	0	
D004	1	5	gal	5	gal	5	gal				
D006, D007, D009, D018	1	5	gal	5	gal	5	gal				
F039 LEACHATE	1	5	gal	5	gal	5	gal				
LEACHATE F039	1	5	gal	5	gal	5	gal				

Table 3-6. Summary of Chemicals Spilled by Transportation Mode (highway locations) (continued)

Name of Material	Number of Incidents	Total Quantity Spilled	Unit of Measure	Range of Quantity Spilled	Unit of Measure	Median Quantity Spilled	Unit of Measure	Health	Flammability	Reactivity	Special
LEACHATE/F039, F001, F004, F005, U051, U076, U159	1	5	gal	5	gal	5	gal		-	-	
RCRA INCINERATOR ASH	1	5	gal	5	gal	5	gal				
WASTE ALKALINE	1	5	gal	5	gal	5	gal				
POLYOXYPROPYLENEDIAMINE	1	4.5	gal	4.5	gal	4.5	gal				
INCINERATOR ASH	1	4	gal	4	gal	4	gal				
WASTE DERIVED FUELS	1	3	gal	3	gal	3	gal	0	2	0	
DOO8 HAZARDOUS SOLID WASTE	1	3	gal	3	gal	3	gal				
DOO8, FOO6	1	3	gal	3	gal	3	gal				
INCINERATOR ASH	1	3	lbs	3	lbs	3	lbs				
OIL, MISC: TRANSFORMER	2	3	gal	0-3	gal	1.5	gal	0	2	0	
D008 RCRA WASTE (LEAD)	2	3	gal	1-2	gal	1.5	gal				
TRANSMISSION FLUID	2	2.5	gal	0.5-2	gal	1.25	gal	0	2	0	
ANTI-FREEZE	1	2	gal	2	gal	2	gal	1	1	0	
CREOSOTE, COAL TAR	1	2	gal	2	gal	2	gal	2	2	0	
BLAST FURNACE SLAG	1	2	gal	2	gal	2	gal				
LEAD, UNKNOWN TYPE	1	2	gal	2	gal	2	gal				
LIQUID LEAD	1	2	gal	2	gal	2	gal				
MILADHON- D	1	2	gal	2	gal	2	gal				
POLYALKYLAMINE	1	2	gal	2	gal	2	gal				
WASTE LIQUID	1	2	gal	2	gal	2	gal				
WATER CONTAINING FLY ASH	1	1.2	gal	1.2	gal	1.2	gal				
PETROLEUM NAPHTHA	1	1	gal	1	gal	1	gal	1	3	0	
80 % PHOSPHORIC ACID	1	1	gal	1	gal	1	gal	3	0	0	
ARSENIC (RQ OF 1LB)	1	1	lbs	1	lbs	1	lbs	3	0	0	
2-BUTOXY ETHANOL; GLYCOL ETHERS	1	1	gal	1	gal	1	gal	2	2	1	
BAG HOUSE DUST, D006,D008	1	1	gal	1	gal	1	gal				
D007(PAINT FILTERS)	1	1	gal	1	gal	1	gal				

Table 3-6. Summary of Chemicals Spilled by Transportation Mode (highway locations) (continued)

Name of Material	Number of Incidents	Total Quantity Spilled	Unit of Measure	Range of Quantity Spilled	Unit of Measure	Median Quantity Spilled	Unit of Measure	Health	Flammability	Reactivity	Special
DOO8 WASTE FUEL	1	1	gal	1	gal	1	gal				
HAZARDOUS WASTE: D001, F003, F005, U056	1	1	gal	1	gal	1	gal				
HAZARDOUS WASTE: U120, U156, U188	1	1	gal	1	gal	1	gal				
SULFUR TRIOXIDE	1	1	lbs	1	lbs	1	lbs				
COPPER CHROMIUM ARSENIC	1	0.75	gal	0.75	gal	0.75	gal				
HEXACHLOROBUTADIENE	1	0.5	gal	0.5	gal	0.5	gal	2	1	1	
D008 HAZARDOUS WASTE	1	0.5	gal	0.5	gal	0.5	gal				
WASTE D008	1	0.5	gal	0.5	gal	0.5	gal				
OTHER OIL	3	0.38	gal	0-0.25	gal	0.13	gal	0	2	0	
DOO4, DOO8, D009, D0011, D0019	1	0.25	lbs	0.25	lbs	0.25	lbs				
FREON	3	0.14	gal	0-0.14	gal	0	gal				
ETHYL ACRYLATE	1	0.13	gal	0.13	gal	0.13	gal	2	3	2	
OIL BASED PAINT	1	0	gal	0	gal	0	gal	0	2	0	
ENGINE STARTING FLUID	1	0	gal	0	gal	0	gal	1	3	0	
ETHYL ETHER	1	0	gal	0	gal	0	gal	1	4	1	
GASOLINE ADDITIVE	1	0	gal	0	gal	0	gal	1	3	0	
METHYL ALCOHOL	1	0	gal	0	gal	0	gal	1	3	0	
TAR BASE	1	0	gal	0	gal	0	gal	1	3	0	
CUMENE	1	0	gal	0	gal	0	gal	2	3	1	
DICHLOROMETHANE	1	0	gal	0	gal	0	gal	2	1	0	
BATTERY ACID	1	0	gal	0	gal	0	gal	3	0	2	Water
NITRIC ACID (70% OR LESS)	1	0	gal	0	gal	0	gal	3	0	0	OX
PENTACHLOROPHENOL	1	0	gal	0	gal	0	gal	3	0	0	
SULFUR DIOXIDE	1	0	gal	0	gal	0	gal	3	0	0	
ACRYLONITRILE	1	0	gal	0	gal	0	gal	4	3	2	
ALUMINUM PHOSPHIDE PESTICIDE	1	0	gal	0	gal	0	gal	4	4	2	Water
ALCOHOL	1	0	gal	0	gal	0	gal				

Table 3-6. Summary of Chemicals Spilled by Transportation Mode (highway locations) (continued)

Name of Material	Number of Incidents	Total Quantity Spilled	Unit of Measure	Range of Quantity Spilled	Unit of Measure	Median Quantity Spilled	Unit of Measure	Health	Flammability	Reactivity	Special
ALUMINUM SULFATE	1	0	gal	0	gal	0	gal				
BIOWASTES	1	0	gal	0	gal	0	gal				
BUTANOL	1	0	gal	0	gal	0	gal	2	3	1	
CADMIUM	1	0	gal	0	gal	0	gal				
CHROMIUM	1	0	gal	0	gal	0	gal				
COAL	1	0	gal	0	gal	0	gal				
DURSBAN	1	0	gal	0	gal	0	gal				
FERROUS OXIDE	1	0	gal	0	gal	0	gal				
GRANULAR FERTILIZER	1	0	gal	0	gal	0	gal				
GRANULAR NITROGEN	1	0	gal	0	gal	0	gal				
LEAD	1	0	gal	0	gal	0	gal				
RADIOACTIVE MATERIAL	1	0	gal	0	gal	0	gal				
RADIOACTIVE MATERIAL NOS	1	0	gal	0	gal	0	gal				
REFRIGERANT GASES	1	0	gal	0	gal	0	gal				
STRONTIUM CHROMATE	1	0	gal	0	gal	0	gal				
UNKNOWN HERBICIDES	1	0	gal	0	gal	0	gal				
CHLORINE	2	0	gal	0	gal	0	gal	4	0	0	OX
BLACK LIQUOR	2	0	gal	0	gal	0	gal				
GREEN LIQUOR (CORROSIVE)	2	0	gal	0	gal	0	gal				
RAW SEWAGE	2	0	gal	0	gal	0	gal				
WASTE OIL	5	0	gal	0	gal	0	gal	0	2	0	
UNKNOWN OIL	6	0	gal	0	gal	0	gal	0	2	0	

Table 3-7. Summary of Chemicals Spilled by Transportation Mode (marine operations)

Name of Material	Number of Incidents	Total Quantity Spilled	Unit of Measure	Range of Quantity Spilled	Unit of Measure	Median Quantity Spilled	Unit of Measure	Health	Flammability	Reactivity	Special
OIL: CRUDE	22	2007824	gal	0-2000000	gal	6	gal	0	2	0	
OIL:DIESEL	161	6458	gal	0-2000	gal	1	gal	0	2	0	
SODIUM HYPOCHLORITE (15% OR LESS)	1	3500	gal	3500	gal	3500	gal				
OIL, FUEL: NO. 2-D	67	2385	gal	0-500	gal	5	gal	0	2	0	
OIL, FUEL: NO. 6	39	1768	gal	0-1200	gal	0.8	gal	0	2	0	
ASPHALT	8	1104	gal	0-400	gal	50	gal	0	1	0	
OIL, MISC: MOTOR	18	1012	gal	0-605	gal	2	gal	0	2	0	
SODIUM HYDROXIDE	1	1000	lbs	1000	lbs	1000	lbs	3	0	1	
BROMINE	1	900	lbs	900	lbs	900	lbs	3	0	0	OX
ADIPONITRILE	1	640	lbs	640	lbs	640	lbs	2	2	1	
OIL, MISC: LUBRICATING	30	575	gal	0-200	gal	2.4	gal	0	2	0	
WASTE OIL AND WATER MIXTURE	16	533	gal	0-500	gal	1	gal	0	2	0	
WASTE OIL	2	502	gal	2-500	gal	251	gal	0	2	0	
HYDRAULIC OIL	54	480	gal	0-100	gal	2.25	gal	0	2	0	
BILGE SLOPS	7	380	gal	0-300	gal	0.13	gal				
OTHER OIL(IFO 180 FUEL OIL)	2	205	gal	80-125	gal	103	gal	0	2	0	
HYDRAULIC FLUID (BIO-DEGRADABLE)	1	200	gal	200	gal	200	gal	0	2	0	
IFO380 (BLEND OF DIESEL AND NO. 6 OIL)	1	200	gal	200	gal	200	gal				
UNKNOWN OIL	73	194	gal	0-55	gal	0	gal	0	2	0	
GASOLINE: AUTOMOTIVE (4.23 G Pb/GAL)	16	193	gal	0-40	gal	2	gal	1	3	0	
WASTE OIL	19	185	gal	0-30	gal	1.5	gal	0	2	0	
AFFF FOAM	1	180	gal	180	gal	180	gal				
GASOLINE: AUTOMOTIVE (UNLEADED)	10	170	gal	0-120	gal	0.06	gal	1	3	0	
OIL, FUEL: NO. 2	9	164	gal	0-150	gal	0.5	gal	0	2	0	
JET FUEL: JP-8	12	142	gal	0-40	gal	5	gal				
UNKNOWN OIL(DRILLING MUD)	1	100	gal	100	gal	100	gal	0	2	0	
OTHER OIL	16	93	gal	0-45	gal	1.5	gal	0	2	0	
OIL:DIESEL (BUNKER C)	7	82	gal	0.13-80	gal	0.5	gal	0	2	0	

Table 3-7. Summary of Chemicals Spilled by Transportation Mode (marine operations) (continued)

Name of Material	Number of Incidents	Total Quantity Spilled	Unit of Measure	Range of Quantity Spilled	Unit of Measure	Median Quantity Spilled	Unit of Measure	Health	Flammability	Reactivity	Special
MINERAL BASED DRILLING MUD	1	80	gal	80	gal	80	gal				
DE-GUMMED SOYBEAN OIL	1	75	gal	75	gal	75	gal	0	1	0	
OIL, FUEL: NO. 5	38	67	gal	0-50	gal	0.13	gal	0	2	0	
BUNKER-C/NO. 5 FUEL OIL	1	50	gal	50	gal	50	gal	0	2	0	
OTHER OIL-WASTE AND DIESEL	1	50	gal	50	gal	50	gal	0	2	0	
FUEL OIL	2	50	gal	50	gal	50	gal	0	2	0	
BILGE MATERIAL	2	45	gal	5-40	gal	23	gal				
OTHER OIL THERMAL OIL	1	40	gal	40	gal	40	gal	0	2	0	
HEAVY OLEFIN FEED	2	35	gal	0-35	gal	17.5	gal				
THERMAL HEATING OIL (VEGETABLE BASED)	1	25	gal	25	gal	25	gal	0	1	0	
NAPHTHA: SOLVENT	1	25	gal	25	gal	25	gal	1	3	0	
PAINT	2	21	gal	1-20	gal	10.5	gal				
OIL:DIESEL (BUNKER C, FUEL OIL 5)	4	21	gal	0-20	gal	0.25	gal	0	2	0	
OILY WASTE	1	20	gal	20	gal	20	gal	0	2	0	
BILGE OIL	1	20	gal	20	gal	20	gal				
ENGINE OIL	2	20	gal	0-20	gal	10	gal	0	2	0	
UNSPECIFIED JET FUEL	2	20	gal	5-15	gal	10	gal	0	2	0	
EMULSIFIED OIL	1	15	gal	15	gal	15	gal	0	2	0	
BENZYL CHLORIDE	1	15	gal	15	gal	15	gal	3	2	1	
CARBON MONOXIDE	1	12.3	lbs	12.3	lbs	12.3	lbs	3	4	0	
OIL, MISC: COAL TAR	1	10	gal	10	gal	10	gal	0	2	0	
OTHER OIL:ASPHAULT	1	10	gal	10	gal	10	gal	0	2	0	
STYRENE	1	10	gal	10	gal	10	gal	2	3	2	
SULFURIC ACID	1	10	gal	10	gal	10	gal	3	0	2	Water
FERRIC CHLORIDE	1	10	gal	10	gal	10	gal				
OIL-BASED MUD	1	3	gal	3	gal	3	gal	0	2	0	
PACKIKG OIL RESIDUE	1	3	gal	3	gal	3	gal	0	2	0	

Table 3-7. Summary of Chemicals Spilled by Transportation Mode (marine operations) (continued)

Name of Material	Number of Incidents	Total Quantity Spilled		Range of Quantity Spilled	Unit of Measure	Median Quantity Spilled	Unit of Measure	Health	Flammability	Reactivity	Special
ACRYLONITRATE	1	3	gal	3	gal	3	gal				
HEAVY AROMATIC HYDROCARBINS	1	3	gal	3	gal	3	gal				
ETHYLENE GLYCOL	1	2.3	gal	2.3	gal	2.3	gal	1	1	0	
VACUUM GAS OIL	1	2	gal	2	gal	2	gal	0	2	0	
VIRGIN GAS OIL	1	2	gal	2	gal	2	gal	0	2	0	
OIL, MISC: TRANSMISSION	1	1.1	gal	1.1	gal	1.1	gal	0	2	0	
CRUDE SOYBEAN OIL	1	1	gal	1	gal	1	gal				
MIXTURE OF HYDROCARBONS	1	1	gal	1	gal	1	gal				
MIXTURE OF ODS AND OSX	1	1	gal	1	gal	1	gal				
OIL, FUEL: NO. 1-D	1	0.5	gal	0.5	gal	0.5	gal	0	2	0	
GEAR OIL	1	0.25	gal	0.25	gal	0.25	gal	0	2	0	
OIL, MISC: NEATSFOOT	1	0.13	gal	0.13	gal	0.13	gal	0	2	0	
CHEMICAL AND DIESEL COMBINATION	1	0.11	gal	0.11	gal	0.11	gal				
PETROLEUM BASED PAINT	1	0.04	gal	0.04	gal	0.04	gal	0	2	0	
OIL, 90% FUEL: NO. 6,10% DIESEL FUEL	1	0	gal	0	gal	0	gal	0	2	0	
OIL: DIESEL AND BILGE SLOPE	1	0	gal	0	gal	0	gal	0	2	0	
OTHER OIL BILGE OIL	1	0	gal	0	gal	0	gal	0	2	0	
OTHER OIL GAS OIL	1	0	gal	0	gal	0	gal	0	2	0	
REFINED CORN OIL	1	0	gal	0	gal	0	gal	0	1	0	
UNKNOWN OIL (VACUUM GAS OIL)	1	0	gal	0	gal	0	gal	0	2	0	
VARIOUS KINDS OF OILS	1	0	gal	0	gal	0	gal	0	2	0	
METHYL ETHYL KETONE	1	0	gal	0	gal	0	gal	1	3	0	
NATURAL GAS	1	0	gal	0	gal	0	gal	1	4	0	
AMMONIA, ANHYDROUS	1	0	gal	0	gal	0	gal	3	1	0	
NITROGEN OXIDE	1	0	gal	0	gal	0	gal	3	0	0	OX
LEAD BASED PAINT	1	0	gal	0	gal	0	gal				
SEWAGE	1	0	gal	0	gal	0	gal				

Table 3-7. Summary of Chemicals Spilled by Transportation Mode (marine operations) (continued)

Name of Material	Number of Incidents	Total Quantity Spilled	Unit of Measure	Range of Quantity Spilled	Unit of Measure	Median Quantity Spilled	Unit of Measure	Health	Flammability	Reactivity	Special
STRONG SMELL OF OIL: DIESEL	1	0	gal	0	gal	0	gal		-	-	
VINYL PAINT	1	0	gal	0	gal	0	gal				
WASTE PAINT	1	0	gal	0	gal	0	gal				
ASPHALT BLENDING STOCKS: ROOFERS	2	0	gal	0	gal	0	gal	0	3	0	
KEROSENE	2	0	gal	0	gal	0	gal	0	2	0	
UNKNOWN MATERIAL	2	0	gal	0	gal	0	gal				
WASTE AND SEWAGE WATER	1	55	gal	55	gal	55	gal				
UNKNOWN OIL	1	0	gal	0	gal	0	gal	0	2	0	
UNKNOWN MATERIAL	2	0	gal	0	gal	0	gal				

Table 3-8. Summary of Chemicals Spilled by Transportation Mode (off-shore locations)

Name of Material	Number of Incidents	Total Quantity	Unit of Measure	Range of Quantity Spilled	Unit of Measure	Median Quantity Spilled	Unit of Measure	Health	Flammability	Reactivity	Special
SHELL SOL 71 OIL	1	400	gal	400	gal	400	gal	0	2	0	
OIL BASED LIQUID MUD	4	326	gal	20-126	gal	90	gal	0	2	0	
OIL, MISC: LUBRICATING	2	150	gal	0.5-150	gal	75	gal	0	2	0	
OIL BASED MUD	1	120	gal	120	gal	120	gal	0	2	0	
OIL: DIESEL	10	101	gal	0-55	gal	0.2	gal	0	2	0	
HYDRAULIC OIL	10	72.5	gal	0-20	gal	1	gal	0	2	0	
CRUDE OIL	3	14	gal	1-12	gal	1	gal				
OIL, MISC: MOTOR	3	2.5	gal	0.5-0.99	gal	0.99	gal	0	2	0	
OIL, FUEL: NO. 2-D	3	0.59	gal	0-0.59	gal	0	gal	0	2	0	
UNKNOWN OIL	24	0.58	gal	0-0.5	gal	0	gal	0	2	0	
WASTE OIL	1	0.1	gal	0.1	gal	0.1	gal	0	2	0	

Table 3-9. Summary of Chemicals Spilled by Transportation Mode (pipelines)

Name of Material	Number of Incidents	Total Quantity Spilled	Unit of Measure	Range of Quantity Spilled	Unit of Measure	Median Quantity Spilled	Unit of Measure	Health	Flammability	Reactivity	Special
OIL, FUEL: NO. 2-D	4	9101	gal	0-9000	gal	51	gal	0	2	0	
CRUDE OIL	9	4750	gal	0-1680	gal	200	gal				
PARAXYLENE	1	1000	gal	1000	gal	1000	gal	2	3	0	
OIL: DIESEL	2	150	gal	0-150	gal	75	gal	0	2	0	
GASOLINE: AUTOMOTIVE	3	150	gal	0-100	gal	75	gal	1	3	0	
SALT WATER	1	60	gal	60	gal	60	gal				
TRIETHYLENE GLYCOL	1	35	gal	35	gal	35	gal	0	1	0	
OIL:DIESEL (BUNKER FUEL)	1	15	gal	15	gal	15	gal	0	2	0	
PROPANE	1	0	gal	0	gal	0	gal	1	4	0	
NATURAL GAS	6	0	gal	0	gal	0	gal	1	4	0	

Table 3-10. Summary of Chemicals Spilled by Transportation Mode (railroad and highway crossings)

Name of Material	Number of Incidents	Total Quantity	Unit of Measure	Range of Quantity Spilled	Unit of Measure	Median Quantity Spilled	Unit of Measure	Health	Flammability	Reactivity	Special
OIL: DIESEL	8	7321	gal	0-4000	gal	160	gal	0	2	0	
OIL, FUEL: NO. 2-D	4	1112	gal	2-1000	gal	55	gal	0	2	0	
OIL, MISC: MOTOR	1	125	gal	125	gal	125	gal	0	2	0	
FORMALDEHYDE SOLUTION	1	1	gal	1	gal	1	gal	2	2	0	

Table 3-11. Summary of Chemicals Spilled by Transportation Mode (railroads)

Table 3-11. Sullillary of Chemicals Sp	ilica by	Transportation	יון טמטווווול	amoudoj							
Name of Material	Number of Incidents	Total Quantity Spilled	Unit of Measure	Range of Quantity Spilled	Unit of Measure	Median Quantity Spilled	Unit of Measure	Health	Flammability	Reactivity	Special
COAL	10	934800	lbs	0-640000	lbs	10000	lbs				- op consi
PLASTIC PELLETS	2	262500	lbs	500-262000	lbs	131000	lbs				
PETROLEUM OIL	1	23000	gal	23000	gal	23000	gal	1	3	0	
ASPHALT	3	20000	gal	0-20000	gal	0	gal	0	1	0	
OIL: DIESEL	43	18103	gal	0-2600	gal	30	gal	0	2	0	
SULFURIC ACID	0	15490	gal	0-10000	gal	1.5	gal	3	0	2	Water
OIL, FUEL: NO. 2-D	23	9534	gal	1-3000	gal	60	gal	0	2	0	
LIMESTONE	2	3000	lbs	1000-2000	lbs	1500	lbs				
AMMONIA, ANHYDROUS	8	2125	lbs	0-1500	lbs	63	lbs	3	1	0	
ETHYL PHOSPHONOTHIOIC DICHLORIDE	1	2000	lbs	2000	lbs	2000	lbs				
CHLOROFORM	1	1250	lbs	1250	lbs	1250	lbs	2	0	0	
CHARCOAL	1	1000	lbs	1000	lbs	1000	lbs				
PHOSPHORIC ACID	6	986	gal	0.1-983	gal	0.38	gal	3	0	0	
PENTANE	1	700	gal	700	gal	700	gal	1	4	0	
CREOSOTE	3	604	gal	4-500	gal	100	gal	2	2	0	
HYDROCHLORIC ACID	11	346	gal	0-330	gal	0.5	gal				
OIL, MISC: LUBRICATING	20	335	gal	0.25-50	gal	4	gal	0	2	0	
OTHER OIL: PARAFIN SOLVENT	1	300	gal	300	gal	300	gal	0	2	0	
ETHYLENE GLYCOL	3	215	gal	0-215	gal	15	gal	1	1	0	
HOMINY FEED	1	200	lbs	200	lbs	200	lbs				
HYDRAULIC OIL	9	176.3	gal	0-50	gal	20	gal	0	2	0	
TURPENTINE	3	151	gal	0.06-149	gal	1.5	gal	1	3	0	
OLEUM	1	150	gal	150	gal	150	gal				
BENZENE	2	132	gal	42-90	gal	66	gal	2	3	0	
TURPENTINE METHYL MERCAPTAN	2	130	gal	30-100	gal	65	gal				
SPENT POT LINER FROM ALUMINUM REDUCTION	1	100	lbs	100	lbs	100	lbs				
AMMONIUM NITRATE	1	65	gal	65	gal	65	gal	0	0	3	OX
XYLENE (O-, M-, P-, & MIXTURES)	1	55	gal	55	gal	55	gal	2	3	0	

Table 3-11. Summary of Chemicals Spilled by Transportation Mode (railroads) (continued)

Table 3-11. Summary of Chemicals S	Jilica by	Transportation	ii) buode (ii	am oaas) (co	ittiiiacaj						
Name of Material	Number of Incidents	Total Quantity Spilled	Unit of Measure	Range of Quantity Spilled	Unit of Measure	Median Quantity Spilled	Unit of Measure	Health	Flammability	Reactivity	Special
OIL, FUEL: NO. 2	2	50	gal	0-50	gal	25	gal	0	2	0	
NITROGEN FERTILIZER SOLUTION	4	43	gal	0.25-40	gal	1.5	gal				
CHLORINE	8	41.1	lbs	0-33	lbs	0	lbs	4	0	0	OX
COTTONSEED OIL, FATTY ACID	1	40	gal	40	gal	40	gal	0	1	0	
STYRENE	2	31	gal	1-31	gal	16	gal	2	3	2	
SODIUM HYDROXIDE	8	25.6	gal	0-10	gal	1.5	gal	3	0	1	
TRIMETHYL HEXAMETHYLENE DIAMINE	1	25	gal	25	gal	25	gal				
JET FUEL: JP-4	1	20	gal	20	gal	20	gal	1	3	0	
UNKNOWN OIL	3	20	gal	0-20	gal	0	gal	0	2	0	
BIPHENYL	1	14	gal	14	gal	14	gal	2	1	0	
CALCINIDE ALUMINUM ORE	1	10	gal	10	gal	10	gal	0	1	1	
DIPHENYL OXIDE	1	10	gal	10	gal	10	gal	1	1	0	
BLAZE MASTER POWDER	1	10	lbs	10	lbs	10	lbs				
ACRYLONITRILE PROPIONITIRICRMENRAL	1	6	gal	6	gal	6	gal	4	3	2	
TEREPHTHALIC ACID	2	5.2	gal	0.2-5	gal	2.6	gal	0	1	0	
CARBON DIOXIDE	5	5.13	gal	0-5	gal	0	gal				
CARALUMINA CALCINED	1	5	gal	5	gal	5	gal	0	1	1	
LIQUIFIED PETROLEUM GAS	1	5	gal	5	gal	5	gal	1	4	0	
ARSENIC ACID SOLUTION (95-97% WATER)	1	5	gal	5	gal	5	gal	3	0	0	
POLYCHLORINATE	1	5	gal	5	gal	5	gal				
P-XYLENE	2	5	gal	0-5	gal	2.5	gal	2	3	0	
OCTYL MERCAPTAN	2	4	gal	1-3	gal	2	gal				
BUTADIENE	2	3.87	gal	3.87	gal	3.87	gal	2	4	2	
POTASSIUM HYDROXIDE	3	3.1	gal	0.06-2	gal	1	gal	3	0	1	
OIL, EDIBLE: SOYA BEAN	1	3	gal	3	gal	3	gal	0	1	0	
OXANONE (OSB-OIL STRIPPER)	1	3	gal	3	gal	3	gal	0	2	0	

Table 3-11. Summary of Chemicals Spilled by Transportation Mode (railroads) (continued)

Table 5 11. Callinary of Official Cals Of		Transportation Mode (railroads) (continued)											
Name of Material	Number of Incidents	Total Quantity Spilled	Unit of Measure	Range of Quantity Spilled	Unit of Measure	Median Quantity Spilled	Unit of Measure	Health	Flammability	Reactivity	Special		
FLAMMABLE LIQUID (VITROPYLAMINE)	1	3	gal	3	gal	3	gal						
GLYCOL ETHER	1	3	gal	3	gal	3	gal						
HAZARDOUS WASTE SOLIDS	1	3	lbs	3	lbs	3	lbs						
OILY WATER MIXTURE	1	2	gal	2	gal	2	gal	0	2	0			
BUTYL ACETATE	1	2	gal	2	gal	2	gal	1	3	0			
GASOLINE: AUTOMOTIVE (UNLEADED)	1	2	gal	1	gal	1	gal	1	3	0			
POLYVINYL CHLORIDE	1	2	lbs	2	lbs	2	lbs						
OCTYL MERCAPTAS	1	1.5	gal	1.5	gal	1.5	gal						
OIL: CRUDE	3	1.25	gal	0.12-1	gal	0.13	gal	0	2	0			
ACETIC ACID, GLACIAL	2	1.13	gal	0.13-1	gal	0.57	gal	3	2	0			
ISOPROPYLAMINE	2	1.12	gal	0.12-1	gal	0.56	gal	3	4	0			
AMMONIUM NITRATE, LIQUID	1	1	gal	1	gal	1	gal	0	0	3	OX		
OTHER OIL	1	1	gal	1	gal	1	gal	0	2	0			
LIQUIFIED PETROLEUM GAS	1	1	gal	1	gal	1	gal	1	4	0			
CAUSTIC POTASH SOLUTION	1	1	gal	1	gal	1	gal	3	0	1			
POTASSIUM HYDROXIDE	1	1	gal	1	gal	1	gal	3	0	1			
SULFUR DIOXIDE	1	1	lbs	1	lbs	1	lbs	3	0	0			
CRUDE SULFATE TURPENTINE	1	1	gal	1	gal	1	gal						
FERRIC SULFATE	1	1	gal	1	gal	1	gal						
METHYLENE DIPHENYL DIISOCYANATE	1	1	gal	1	gal	1	gal						
PULPMILL LIQUIDS	1	1	gal	1	gal	1	gal						
TOLUENE 2,4-DIISOCYANATE	1	0.99	gal	0.99	gal	0.99	gal	3	1	3	Water		
PROPIONITRILE	1	0.99	lbs	0.99	lbs	0.99	lbs	4	3	1			
OTHER OIL (LUBE OIL)	1	0.5	gal	0.5	gal	0.5	gal	0	2	0			
OTHER OIL (TRANSMISSION OIL)	1	0.5	gal	0.5	gal	0.5	gal	0	2	0			
PETROLEUM NAPHTHA	1	0.5	gal	0.5	gal	0.5	gal	1	3	0			
HEXANETHYLENEDIAMINE	1	0.5	gal	0.5	gal	0.5	gal						

Table 3-11. Summary of Chemicals Spilled by Transportation Mode (railroads) (continued)

Table 3-11. Sulfilliary of Chefficals Spilled by		riansportation	iii wioae (ii	am caus, (co							
Name of Material	Number of Incidents	Total Quantity Spilled	Unit of Measure	Range of Quantity Spilled	Unit of Measure	Median Quantity Spilled	Unit of Measure	Health	Flammability	Reactivity	Special
WASTE FLAMMABLE LIQUID	1	0.5	gal	0.5	gal	0.5	gal				
CAUSTIC SODA SOLUTION	4	0.38	gal	0-0.25	gal	0.06	gal	3	0	1	
OTHER OIL (CONDENSATE)	1	0.25	gal	0.25	gal	0.25	gal	0	2	0	
SODIUM CHLORATE	1	0.25	gal	0.25	gal	0.25	gal	1	0	2	OX
DIISOBUTYLAMINE	1	0.25	gal	0.25	gal	0.25	gal	3	3	0	
SODIUM CHLORITE	1	0.13	gal	0.13	gal	0.13	gal	1	0	1	OX
BATTERY ACID	1	0.13	gal	0.13	gal	0.13	gal	3	0	2	Water
FLAMMABLE ALCOHOL	1	0.13	gal	0.13	gal	0.13	gal				
OIL, MISC: MOTOR	2	0.13	gal	0-0.13	gal	0.06	gal	0	2	0	
ISO-BUTYRALDEHYDE	1	0.1	gal	0.1	gal	0.1	gal	3	3	2	
CYCLOATXANOL	1	0.01	gal	0.01	gal	0.01	gal				
WASTE OIL	1	0	gal	0	gal	0	gal	0	2	0	
ACETONE	1	0	gal	0	gal	0	gal	1	3	0	
CYCLOHEXANONE	1	0	gal	0	gal	0	gal	1	2	0	
LIQUEFIED PETROLEUM GAS	1	0	gal	0	gal	0	gal	1	4	0	
METHYL ACETOACETATE	1	0	gal	0	gal	0	gal	2	2	0	
PARACYMENE XLYENE	1	0	gal	0	gal	0	gal	2	3	0	
SULFUR	1	0	gal	0	gal	0	gal	2	1	0	
VINYL CHLORIDE	1	0	gal	0	gal	0	gal	2	4	2	
ETHYLENEDIAMINE	1	0	gal	0	gal	0	gal	3	2	0	
PHENOL	1	0	gal	0	gal	0	gal	4	2	0	
AMMONIA FERTILIZER SOLUTION	1	0	gal	0	gal	0	gal				
CARBON BLACK	1	0	gal	0	gal	0	gal				
CARBUTADIENES, INHIBITED	1	0	gal	0	gal	0	gal				
COPPER CHLORIDE (IC) (10%)	1	0	gal	0	gal	0	gal				
FLUOROSULFONIC ACID	1	0	gal	0	gal	0	gal				

Table 3-11. Summary of Chemicals Spilled by Transportation Mode (railroads) (continued)

Name of Material	Number of Incidents	Total Quantity	Unit of Measure	Range of Quantity Spilled	Unit of Measure	Median Quantity Spilled	Unit of Measure	Health	Flammability	Reactivity	Special
HEXAFLOUROPROPYLENE	1	0	gal	0	gal	0	gal				
HYDROFLUOSILICIC ACID	1	0	gal	0	gal	0	gal				
SEWAGE	1	0	gal	0	gal	0	gal				
TALLOW	1	0	gal	0	gal	0	gal				
UNKNOWN MATERIAL	1	0	gal	0	gal	0	gal				

Table 3-12. Summary of Chemicals Spilled by Transportation Mode (unknown locations)

rable 5 12. Callillary of Officialicals op	i i ali spoi tatic	ransportation wode (unknown locations)									
Name of Material	Number of Incidents	Total Quantity Spilled	Unit of Measure	Range of Quantity Spilled	Unit of Measure	Median Quantity Spilled	Unit of Measure	Health	Flammability	Reactivity	Special
GASOLINE: AUTOMOTIVE (UNLEADED)	5	2500	gal	0-2500	gal	0	gal	1	3	0	
UNKNOWN OIL	142	255	gal	0-100	gal	0	gal	0	2	0	
OIL, FUEL: NO. 2-D	5	40	gal	0-40	gal	0	gal	0	2	0	
OIL: DIESEL	25	40	gal	0-40	gal	0	gal	0	2	0	
HYDRAULIC OIL	1	20	gal	20	gal	20	gal	0	2	0	
ASPHALT	1	6	gal	6	gal	6	gal	0	1	0	
SODIUM HYDROXIDE	1	5	gal	5	gal	5	gal	3	0	1	
ETHYLENE GLYCOL	3	5	gal	0-5	gal	0	gal	1	1	0	
OIL, MISC: TRANSFORMER	1	0	gal	0	gal	0	gal	0	2	0	
OIL: CRUDE	1	0	gal	0	gal	0	gal	0	2	0	
OTHER OIL (TAR BALLS)	1	0	gal	0	gal	0	gal	0	2	0	
TURBINE OIL	1	0	gal	0	gal	0	gal	0	2	0	
UNKNOWN SHEEN	1	0	gal	0	gal	0	gal	0	2	0	
GASOLINE: AUTOMOTIVE (4.23 G Pb/GAL)	1	0	gal	0	gal	0	gal	1	3	0	
BENZENE	1	0	gal	0	gal	0	gal	2	3	0	
A MIXTURE OF DIFFERENT COLOR PAINT	1	0	gal	0	gal	0	gal				
INSECTICIDE	1	0	gal	0	gal	0	gal				
RED PAINT LIKE MATERIAL	1	0	gal	0	gal	0	gal				
WASTE OIL	2	0	gal	0	gal	0	gal	0	2	0	
YELLOW LATEX PAINT	2	0	gal	0	gal	0	gal				
OIL, MISC: MOTOR	4	0	gal	0	gal	0	gal	0	2	0	

## **Section 4. Environmental Fate and Transport Modeling**

This report section presents several procedures to predict the fate and transport of spilled hazardous materials. The initial discussion is a general discussion that stresses downwind toxic and explosive hazards. These procedures, summarized from a recent EPA manual, is applicable for a wide range of hazardous materials. Specific characteristics for all regulated hazardous materials are also included in the appendices to enable the efficient use of these procedures. A discussion is also provided that considers mixtures of materials and how these mixtures may be more hazardous than individual material losses.

Based on the information presented previously in Section 3, two detailed examples are presented describing problems associated with spills of petroleum hydrocarbons, by far the most common material lost in Alabama transportation accidents, and ammonia, a very toxic material. Specific procedures are given for calculating the spread and transport of oil slicks, and a numerical example is shown. In addition, a detailed example is presented for predicting both air and water problems associated with ammonia spills. This example represents procedures for toxic and buoyant materials for which specific methods have been developed, based on actual field studies. These procedures enable the calculation of the magnitude of potential exposures to these hazardous materials.

# **Evaluation of Toxic and Explosive Atmospheric Conditions Associated with Transportation Accidents involving Hazardous Materials**

Much of the material in this report section is summarized from the recent EPA (1999) guidance document *Risk Management Program Guidance for Offsite Consequence Analysis*. This referenced EPA report provides guidance on how to conduct the offsite consequence analyses for Risk Management Programs required under the Clean Air Act, Section 112(r)(7). This act directed the EPA to issue regulations requiring facilities that handle, manufacture, store, or use large quantities of very hazardous chemicals to prepare and implement programs to prevent the accidental release of those chemicals and to mitigate the consequences of any releases that do occur. EPA issued 40 CFR 68 on June 20, 1996. This regulation requires these facilities to prepare a risk management system, including analyses of potential toxic and explosive conditions if such material is lost to the environment. The summarized material presented in this section refers to the worst-case scenario procedures included in the guidance document. This summarized material is not a substitute for the complete report for regulated facilities, of course, but is presented here as a currently accepted evaluation procedure that is suitable for evaluating transportation accidents involving hazardous materials. The results obtained using these methods are expected to be conservative (i.e., they will generally, but not always, overestimate the distance to toxic and explosive endpoints).

## Steps for Performing Analyses

## **Worst-Case Analysis for Toxic Gases**

To conduct worst-case analyses for toxic gases, including toxic gases liquefied by pressurization:

- **Step 1:** Determine worst-case scenario. Identify the toxic gas, quantity, and worst-case release scenario.
- Step 2: Determine release rate. Estimate the release rate for the toxic gas.
- **Step 3:** <u>Determine distance to endpoint</u>. Estimate the worst-case consequence distance based on the release rate and toxic endpoint. Select the appropriate table based on the density of the released substance, the topography of the site (urban or rural), and the duration of the release.

## **Worst-Case Analysis for Toxic Liquids**

To conduct worst-case analyses for toxic substances that are liquids at ambient conditions or for toxic gases that are liquefied by refrigeration alone:

- Step 1: Determine worst-case scenario. Identify the toxic liquid, quantity, and worst-case release scenario.
- **Step 2:** <u>Determine release rate</u>. Estimate the volatilization rate for the toxic liquid and the duration of the release:
- **Step 3:** Determine distance to endpoint. Estimate the worst-case consequence distance based on the release rate and toxic endpoint. Select the appropriate reference table based on the density of the released substance, the topography of your site (rural or urban), and the duration of the release. Estimate distance to the endpoint from the appropriate table.

## **Worst-Case Analysis for Flammable Substances**

To conduct worst-case analyses for all regulated flammable substances (i.e., gases and liquids):

- **Step 1:** <u>Determine worst-case scenario</u>. Identify the appropriate flammable substance, quantity, and worst-ase scenario.
- **Step 2**: <u>Determine distance to endpoint</u>. Estimate the distance to the required overpressure endpoint of 1 psi for a vapor cloud explosion of the flammable substance. Estimate the distance to the endpoint from the quantity released.

## **Determining Worst-Case Scenarios**

A worst-case release is defined as:

- The release of the largest quantity of a substance from a vessel or process line failure, and
- The release that results in the greatest distance to the endpoint for the regulated toxic or flammable substance.

This procedure assumes meteorological conditions for the worst-case scenario of atmospheric stability class F (stable atmosphere) and wind speed 1.5 meters per second (3.4 miles per hour). Ambient air temperature is assumed to be  $25 \,^{\circ}$ C (77  $^{\circ}$ F).

The procedure provides two choices for topography, urban and rural. EPA (40 CFR 68.22(e)) has defined urban as many obstacles in the immediate area, where obstacles include buildings or trees. Rural, by EPA's definition, means there are no buildings in the immediate area, and the terrain is generally flat and unobstructed. Thus, if the site is located in an area with few buildings or other obstructions (e.g., hills, trees), open (rural) conditions should be assumed. If the site is in an area with many obstructions, even if it is in a remote location that would not usually be considered urban, urban conditions should be assumed.

Toxic gases include all regulated toxic substances that are gases at ambient temperature (25 °C, 77 °F), with the exception of gases liquefied by refrigeration under atmospheric pressure and released into diked areas. For the worst-case consequence analysis, it is assumed that a gaseous release of the total quantity occurs in 10 minutes. Gases liquefied by refrigeration alone that would form a pool one centimeter or less in depth upon release must be modeled as gases. (Modeling indicates that pools one centimeter or less deep formed by gases liquefied by refrigeration would completely evaporate in 10 minutes or less, giving a release rate that is equal to or greater than the worst-case release rate for a gaseous release. In this case, therefore, it is appropriate to treat these substances as gases for the worst-case analysis.) Table C-1 lists the endpoint for each toxic gas. These endpoints are used for air dispersion modeling to estimate the consequence distance and are considered critical levels of the contaminants.

For toxic liquids, it is assumed that the total quantity in a vessel is spilled. This procedure assumes that the spill takes place onto a flat, non-absorbing surface. For toxic liquids carried in pipelines, the quantity that might be released from the pipeline is assumed to form a pool. The total quantity spilled is assumed to spread

instantaneously to a depth of one centimeter (0.033 foot or 0.39 inch). The release rate to air is estimated as the rate of evaporation from the pool. Table C-2 lists the endpoint for air dispersion modeling for each regulated toxic liquid (the endpoints are specified in 40 CFR part 68, Appendix A, and are considered to be critical levels of the contaminants).

For all regulated flammable substances, it is assumed that the worst-case release results in a vapor cloud containing the total quantity of the substance that could be released from a vessel or pipeline. This procedure assumes that the vapor cloud detonates using a TNT-equivalent method that assumes a 10 percent yield factor. The procedure uses an endpoint for a vapor cloud explosion as an overpressure of 1 pound per square inch (psi). This endpoint is the threshold for potential serious injuries to people as a result of property damage caused by an explosion (e.g., injuries from flying glass from shattered windows or falling debris from damaged houses).

## Release Rates for Toxic Substances

The following describes simple methods for estimating release rates for toxic substances for the worst-case scenario. Simple release rate equations are provided, and factors to be used in these equations are given (in Tables C-1, C-2, and C-3 for each regulated substance. These estimated release rates are used to estimate dispersion distances to the toxic endpoint for regulated toxic gases and liquids in the next section.

#### **Release Rates for Toxic Gases**

Hazardous substances that are gases at ambient temperature (25 °C, 77 °F) should be considered gases for these analysis, with the exception of gases liquefied by refrigeration at atmospheric pressure. Gases liquefied under pressure should be treated as gases. Gases liquefied by refrigeration that would form a pool one centimeter (0.033 foot) or less in depth should also be treated as gases. The evaporation rate from such a pool would be equal to or greater than the rate for a toxic gas, which is assumed to be released over 10 minutes; therefore, treating liquefied refrigerated gases as gases rather than liquids in such cases is reasonable.

<u>Unmitigated Releases of Toxic Gas</u>. If no passive mitigation system is in place (dikes or other containments), which would be the normal situation for most transportation accidents, the release rate is simple the largest amount of material that would be lost, divided by a 10-minute period.

As an example, if a tank contains 2,500 pounds of diborane gas, the release rate (QR) is:

QR = 2,500 pounds/10 minutes = 250 pounds per minute

Releases of Liquefied Refrigerated Toxic Gas in Diked Area. If a toxic gas that is liquefied by refrigeration alone is released into an area where it will be contained by dikes to form a pool more than one centimeter (0.033 foot) in depth, the worst-case analysis assumes evaporation from a liquid pool at the boiling point of the liquid. If the gas liquefied by refrigeration would form a pool one centimeter (0.033 foot) or less in depth, the previous 10 minute assumption for complete evaporation is used. If released in a diked area, first compare the diked area to the maximum area of the pool that could be formed.

The following equation can be used to estimate the maximum size of the pool:

$$A = QS \times DF$$
 Equation 1

where:  $A = Maximum pool area (ft^2), for a depth of one cm$ 

*QS* = Quantity released (lbs)

DF = Density factor (as shown in Tables C-1 and C-2)

If the pool formed by the released liquid would be smaller than the diked area, assume a 10-minute gaseous release, and estimate the release rate as described previously. If the dikes prevent the liquid from spreading out to form a pool of maximum size (one centimeter in depth), use the following equation:

$$QR = 1.4 \times LFB \times A$$

Equation 2

where: QR = Release rate (lbs/min)

*LFB* = Liquid Factor Boiling for hazardous gases liquefied by refrigeration alone,

or use LFA, Liquid Factor Ambient, for hazardous liquids at ambient

temperature (Tables C-1 and C-2)

 $A = Diked area (ft^2)$ 

1.4 = Wind speed factor =  $(1.5)^{0.78}$ , where 1.5 meters per second (3.4 miles per

hour) is the wind speed for the worst case

After the release rate is estimated, estimate the duration of the vapor release from the pool in the diked area (the time it will take for the pool to evaporate completely) by dividing the total quantity spilled by the release rate. The duration of the release for chlorine or sulfur dioxide, liquefied by refrigeration alone, is not needed for the analyses for critical distances.

## **Example for Mitigated Release of Gases Liquefied by Refrigeration (Chlorine)**

A refrigerated tank contains 50,000 pounds of liquid chlorine at ambient pressure. A diked area around the chlorine tank is 275 ft<sup>2</sup> and is sufficient to hold all of the spilled liquid chlorine. Once the liquid spills into the dike, it is then assumed to evaporate at its boiling point (-29 °F). The evaporation rate at the boiling point is determined from equation 2. For this calculation, the wind speed is assumed to be 1.5 meters per second and the wind speed factor is 1.4, LFB for chlorine (from Table C-1) is 0.19, and A is 275 ft<sup>2</sup>. The release rate is:

$$QR = 1.4 \times 0.19 \times 275 = 73 \text{ pounds per minute}$$

The duration of the release does not need to be considered for chlorine.

#### **Release Rates for Toxic Liquids**

For the worst-case analysis, the release rate to air for toxic liquids is assumed to be the rate of evaporation from the pool formed by the released liquid. Assume the total quantity in a vessel or the maximum quantity from ruptured pipes is released into the pool. Passive mitigation measures (e.g., dikes) may be considered in determining the area of the pool and the release rate. To estimate the critical distance using this method, the evaporate duration (the duration of the release) and the release rate need to be known.

The calculation methods provided here apply to substances that are liquids under ambient conditions or gases liquefied by refrigeration alone that are released to form pools deeper than one centimeter. Gases liquefied under other conditions (under pressure or a combination of pressure and refrigeration) or gases liquefied by refrigeration alone that would form pools one centimeter or less in depth upon release are treated as gas releases, rather than liquid releases, as described above.

Releases of Toxic Liquids from Pipes. To consider a liquid release from a broken pipe, estimate the maximum quantity that could be released assuming that the pipe is full of liquid. The time needed to stop liquid pumping also needs to be calculated as part of the release. The quantity in the pipe (in pounds) is the volume released divided by the Density Factor (DF) times 0.033. (DF values are listed in Table C-2. Density in pounds per cubic foot is equal to 1/(DF times 0.033).) Assume the estimated quantity (in pounds) is released into a pool and use the method and equations described below to determine the evaporation rate of the liquid from the pool.

<u>Unmitigated Releases of Toxic Liquids.</u> If no passive mitigation measures are in place, the liquid is assumed to form a pool one centimeter (0.39 inch or 0.033 foot) deep instantaneously. You may calculate the release rate to air from the pool (the evaporation rate) as discussed below for releases at ambient or elevated temperature.

If the liquid is always at ambient temperature, find the Liquid Factor Ambient (LFA) and the Density Factor (DF) in Table C-2. The LFA and DF apply to liquids at 25  $^{\circ}$ C. Calculate the release rate of the liquid at 25  $^{\circ}$ C from the following equation:

 $OR = OS \times 1.4 \times LFA \times DF$ 

Equation 3

where: *QR* Release rate (pounds per minute)

QS.

Quantity released (pounds) Wind speed factor =  $(1.5)^{0.78}$ , where 1.5 meters per second (3.4) 1.4

miles per hour) is the wind speed for the worst case

LFA Liquid Factor Ambient =

DF**Density Factor** 

## Example for an Unmitigated Liquid Release at Ambient Temperature (Acrylonitrile)

A tank contains 20,000 pounds of acrylonitrile at ambient temperature. The total quantity in the tank is spilled onto the ground in an undiked area, forming a pool, Assume the pool spreads out to a depth of one centimeter. The release rate from the pool (QR) is calculated from Equation 3. For the calculation, the wind speed is assumed to be 1.5 meters per second and the wind speed factor is 1.4. From Table C-2, the LFA for acrylonitrile is 0.018 and DF is 0.61. Then:

 $QR = 20,000 \times 1.4 \times 0.018 \times 0.61 = 307$  pounds per minute

The duration of the release would therefore be:

t = 20,000 pounds/307 pounds per minute = 65 minutes

If the liquid is at an elevated temperature (above 50 °C or at or close to the boiling point), find the Liquid Factor Boiling (LFB) and the Density Factor (DF) in Table C-2. If the temperature is above 50 °C, or the liquid is at or close to its boiling point, calculate the release rate of the liquid from the following equation:

> $QR = QS \times 1.4 \times LFB \times DF$ Equation 4

where: *QR* Release rate (pounds per minute)

QS

Quantity released (pounds) Wind speed factor =  $(1.5)^{0.78}$ , where 1.5 meters per second (3.4) 1.4

miles per hour) is the wind speed for the worst case

LFB Liquid Factor Boiling =

**Density Factor** DF

## Example of an Unmitigated Release at Elevated Temperature (Acrylonitrile)

A tank contains 20,000 pounds of acrylonitrile at an elevated temperature. The total quantity in the tank is spilled onto the ground in an undiked area, forming a pool. Assume the pool spreads out to a depth of one centimeter. The release rate from the pool is calculated from Equation 4. For the calculation, the wind speed factor for 1.5 meters per second is 1.4. From Table C-2, the LFB for acrylonitrile is 0.11 and the DF is 0.61. Then:

 $QR = 20,000 \times 1.4 \times 0.11 \times 0.61 = 1,880$  pounds per minute

The duration of the release would therefore be:

t = 20,000 pounds/1880 pounds per minute = 11 minutes

Mixtures Containing Toxic Liquids. If the partial pressure of the hazardous substance in the mixture is known, it is possible to estimate an evaporation rate. In this case, estimate a pool size for the entire quantity of the mixture, for an unmitigated release. If the density of the mixture is known, use it in estimating the pool size; otherwise, assume the density is the same as the pure regulated substance (in most cases, this assumption is unlikely to have a large effect on the results).

## **Example of a Mixture Containing Toxic Liquid (Acrylonitrile)**

A tank contains 50,000 pounds of a mixture of acrylonitrile (a hazardous substance) and N,N-dimethylformamide (not regulated). The weight of each of the components of the mixture is known (acrylonitrile = 20,000 pounds; N,N-dimethylformamide = 30,000 pounds.) The molecular weight of acrylonitrile, from Table C-2, is 53.06, and the molecular weight of N,N-dimethylformamide is 73.09. Using Equation 5, calculate the mole fraction of acrylonitrile in the solution as follows:

$$X_r = \frac{\left(\frac{W_r}{MW_r}\right)}{\sum_{i=1}^{n} \left(\frac{W_i}{MW_i}\right)}$$
 Equation 5

where:

 $X_r$  = Mole fraction of the hazardous substance

 $W_r$  = Weight of the hazardous substance

 $MW_r$  = Molecular weight of the hazardous substance  $W_i$  = Weight of each component of the mixture

 $MW_i$  = Molecular weight of each component of the mixture

n = Number of components of the mixture

$$X_r = \frac{(20,000/53.06)}{(20,000/53.06) + (30,000/73.09)}$$

$$X_r = \frac{377}{377 + 410}$$

$$X_r = 0.48$$

Estimate the partial vapor pressure of acrylonitrile as follows (using the vapor pressure of acrylonitrile in pure form at  $25 \,^{\circ}$  C,  $108 \, \text{mm}$  Hg, from Table C-2):

$$VP_m = 0.48 \times 108 = 51.8 \text{ mm Hg}$$

Before calculating evaporation rate for acrylonitrile in the mixture, the surface area of the pool formed by the entire quantity of the mixture is needed. The quantity released is 50,000 pounds and the Density Factor for acrylonitrile is 0.61 in Table C-2; therefore:

$$A = 50,000 \text{ lbs } \times 0.61 = 30,500 \text{ square feet}$$

Now calculate the evaporation rate for acrylonitrile in the mixture from Equation 6 using the  $VP_m$  and A calculated above:

$$QR = \frac{0.0035 \times U^{0.78} \times MW^{2/3} \times A \times VP}{T}$$
 Equation 6

where:

QR = Evaporation rate (lbs/min)

U = Wind speed (m/sec)

MW = Molecular weight (Table C-2)

A = Surface area of pool formed by the entire quantity of the mixture (ft<sup>2</sup>)

 $VP = Vapor pressure (mm Hg) (VP_m)$ 

T = Temperature (°K), °C plus 273 (298 for 25°C)

$$QR = \frac{0.0035 \times 1.0 \times (53.06)^{2/3} \times 30,500 \times 51.8}{298}$$

QR = 262 pounds per minute

#### Release Rates for Common Water Solutions of Toxic Substances and for Oleum

This section presents a simple method of estimating the release rate from spills of water solutions of several substances. Oleum (a solution of sulfur trioxide in sulfuric acid) also is discussed in this section.

The vapor pressure and evaporation rate of a substance in a solution depends on its concentration in the solution. If a concentrated water solution containing a volatile toxic substance is spilled, the toxic substance initially will evaporate more quickly than water from the spilled solution, and the vapor pressure and evaporation rate will decrease as the concentration of the toxic substance in the solution decreases. At much lower concentrations, water may evaporate more quickly than the toxic substance. There is one concentration at which the composition of the solution does not change as evaporation occurs. For most situations of interest, the concentration exceeds this concentration, and the toxic substance evaporates more quickly than water.

For estimating release rates from solutions, this procedure uses liquid factors (ambient) for several common water solutions at several concentrations that take into account the decrease in evaporation rate with decreasing concentration. Table C-3 provides LFA and DF values for several concentrations of ammonia, formaldehyde, hydrochloric acid, hydrofluoric acid, and nitric acid in water solution. Factors for oleum are also included in this table. These factors may be used to estimate an average release rate for the hazardous substances from a pool formed by a spill of solution. Liquid factors are provided for two different wind speeds, because the wind speed affects the rate of evaporation.

For the worst case, the factor for a wind speed of 1.5 meters per second (3.4 miles per hour) should be used. Consider only the first 10 minutes of the release for solutions under ambient conditions in estimating the critical distance, as the toxic component in a solution evaporates fastest during the first few minutes of a spill when its concentration is highest. Although the toxic substance will continue to evaporate from the pool after 10 minutes, the rate of evaporation is so much lower that it can safely be ignored in estimating the critical distance. Release rates are estimated as follows:

Ambient Temperature. If the solution is at ambient temperature, the LFA at 1.5 meters per second (3.4 miles per hour) and DF for the solution are obtained from Table C-3. Follow the instructions for liquids presented above to estimate the release rate of the hazardous substance in solution. Use the total quantity of the solution as the quantity released (QS) in carrying out the calculation of release rate.

## Example for Calculating the Evaporation Rate for a Water Solution of Hydrochloric Acid at Ambient Temperature

A tank contains 50,000 pounds of 37 percent hydrochloric acid solution, at ambient temperature. For the worst-case analysis, assume the entire contents of the tank are released, forming a pool. The release occurs in a diked area of 9,000 square feet. From Table C-3, the Density Factor (DF) for 37 percent hydrochloric acid is 0.42. From Equation 1, the maximum area of the pool would be 50,000 lbs times 0.42, or 21,000 square feet.

The diked area is smaller; therefore, the diked area should be used in the evaporation rate (release rate) calculation, using Equation 2. For the calculation, the pool area (9,000 square feet) and the Liquid Factor Ambient (LFA) for 37 percent hydrochloric acid are needed; also assume a wind speed of 1.5 meters per second, so the wind speed factor is 1.4. From Table C-3, the LFA is 0.0085. From Equation 2, the release rate (QR) of hydrogen chloride from the pool is:

 $QR = 1.4 \times 9,000 \times 0.0085 = 107$  pounds per minute

## Estimation of Worst-Case Distance to Toxic Endpoint

This procedure provides graphs (Figures 4-1 to 4-8) giving worst-case distances for neutrally buoyant gases and vapors and for dense gases and vapors for both rural (open) and urban (obstructed) areas. Neutrally buoyant gases and vapors have approximately the same density as air, and dense gases and vapors are heavier than air. Neutrally buoyant and dense gases are dispersed in different ways when they are released. These generic figures can be used to estimate distances using the specified toxic endpoint for each substance and the estimated release rate to air. In addition to the generic figures, chemical-specific figures are provided for ammonia, chlorine, and sulfur dioxide. These chemical-specific figures were developed based on modeling carried out for industry-specific guidance documents. All the figures were developed assuming a wind speed of 1.5 meters per second (3.4 miles per hour) and F stability. To use the figures, the worst-case release rates estimated as described in the previous sections are needed. For liquid pool evaporation, the duration of the release is also needed. In addition, the appropriate toxic endpoint and whether the gas or vapor is neutrally buoyant or dense is also needed, and are given in Tables C-1, C-2 and C-3.

## Regulated Toxic Substances Other than Ammonia, Chlorine, and Sulfur Dioxide

- Find the toxic endpoint for the substance in Table C-1 for toxic gases or Table C-2 for toxic liquids.
- Determine whether the figure for neutrally buoyant or dense gases and vapors is appropriate from Appendix Table C-1 for toxic gases or Table C-2 for toxic liquids. A toxic gas that is lighter than air may behave as a dense gas upon release if it is liquefied under pressure, because the released gas may be mixed with liquid droplets, or if it is cold.
- Determine whether the figure for rural or urban conditions is appropriate.
- -- Use the rural figure if your site is in an open area with few obstructions.
- -- Use the urban figure if your site is in an urban or obstructed area. The urban figures are appropriate if there are many obstructions in the area, even if it is in a remote location, not in a city.
- Determine whether the 10-minute figure or the 60-minute figure is appropriate.
- -- Always use the 10-minute figure for worst-case releases of toxic gases.
- -- Always use the 10-minute figure for worst-case releases of common water solutions and oleum from evaporating pools, for both ambient and elevated temperatures.
- -- If you estimated the release duration for an evaporating toxic liquid pool to be 10 minutes or less, use the 10-minute figure.
- -- If you estimated the release duration for an evaporating toxic liquid pool to be more than 10 minutes, use the 60-minute figure.

Neutrally Buoyant Gases or Vapors. If Tables C-1 or C-2 indicate the gas or vapor should be considered neutrally buoyant, and other factors would not cause the gas or vapor to behave as a dense gas, divide the estimated release rate (pounds per minute) by the toxic endpoint (milligrams per liter). Find the calculated release rate/toxic endpoint ratio on the x-axis of the figures (Figures 4-1, 4-2, 4-3, or 4-4), then find the corresponding distance to the y-axis (see example below).

## **Example for a Gas Release of Diborane**

The estimated release rate for diborane gas is 250 pounds per minute. From Table C-1, the toxic endpoint for diborane is 0.0011 mg/L, and it is a neutrally buoyant gas. The facility and the surrounding area have many buildings, pieces of equipment, and other obstructions; therefore, assume urban conditions. The appropriate data is therefore shown on Figure 4-3, for a 10-minute release of a neutrally buoyant gas in an urban area.

The release rate divided by toxic endpoint for this example is 250/0.0011 = 230,000.

From Figure 4-3, this value corresponds to a critical distance of about 8 miles.

<u>Dense Gases or Vapors</u>. If Table C-1 or C-2 or consideration of other relevant factors indicates that the substance should be considered a dense gas or vapor (heavier than air), find the critical distance from the appropriate figure (Figure 4-5, 4-6, 4-7, or 4-8) as follows;

- -- Select the curve on the figure that is closest to the toxic endpoint of the substance.
- -- Find the release rate closest to the release rate estimated for the substance on the x-axis of the figure.
- -- Determine the corresponding critical distance on the y-axis.

## Example for a Release of Ethylene Oxide, a Dense Gas

A tank contains 10,000 pounds of ethylene oxide, which is a gas under ambient conditions. Assuming the total quantity in the tank is released over a 10-minute period, the release rate (QR) is:

QR = 10,000 pounds/10 minutes = 1,000 pounds per minute

From Table C-1, the toxic endpoint for ethylene oxide is 0.09 mg/L, and the appropriate figure is for a dense gas. The facility is in an open, rural area with few obstructions; therefore, use the figure for rural areas.

Using Figure 4-5 for 10-minute releases of dense gases in rural areas, the toxic endpoint of 0.09 mg/L is closer to 0.1 than 0.075 mg/L. For a release rate of 1,000 pounds per minute, the distance to 0.1 mg/L is about 3.5 miles

## Example for Liquid Evaporation from a Pool of Acrylonitrile

The estimated evaporation rate is 307 pounds per minute for acrylonitrile from a pool formed by the release of 20,000 pounds into an undiked area. The estimated time for evaporation of the pool as 65 minutes. From Table C-2, the toxic endpoint for acrylonitrile is 0.076 mg/L, and the appropriate figure for a worst-case release of acrylonitrile is the dense gas figure. The facility is in an urban area, so the appropriate Figure 4-8 is used for a 60-minute releases of dense gases in urban areas.

From Figure 4-8, the toxic endpoint closest to 0.076 mg/L is 0.075 mg/L. The worst-case critical distance is therefore about 3 miles corresponding to the release rate of 307 pounds per minute.

Ammonia, Chlorine, or Sulfur Dioxide. Use the appropriate chemical-specific figure for the substance. If ammonia is liquefied by refrigeration alone, use Figure 4-10, even if the duration of the release is greater than 10 minutes. If chlorine or sulfur dioxide is liquefied by refrigeration alone, use the chemical-specific reference figure, even if the duration of the release is greater than 10 minutes. Use the rural curve on the figure if the site is in an open area with few obstructions, otherwise use the urban curve if the site is in an urban or obstructed area. The urban curve is appropriate if there are many obstructions in the area, even if it is in a remote location, not in a city.

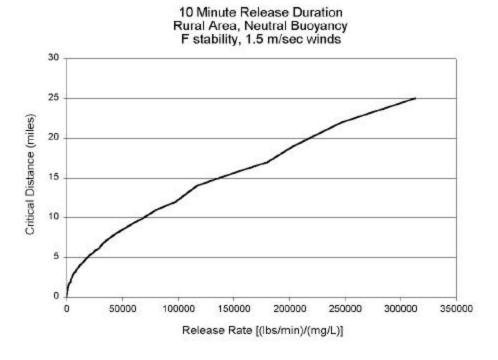


Figure 4-1. Neutrally buoyant gas in rural area, 10 minute release.

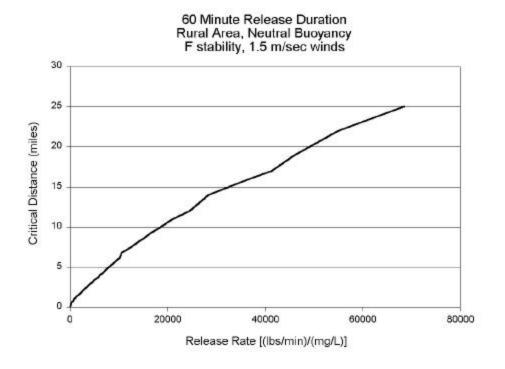


Figure 4-2. Neutrally buoyant gas in rural area, 60 minute release.

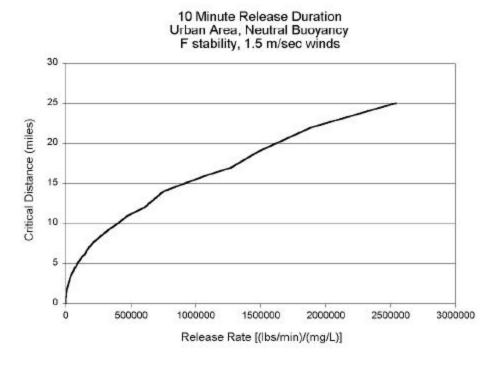


Figure 4-3. Neutrally buoyant gas in urban area, 10 minute release.

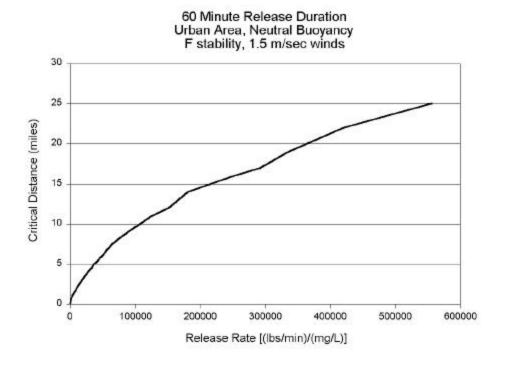


Figure 4-4. Neutrally buoyant gas in urban area, 60 minute release.

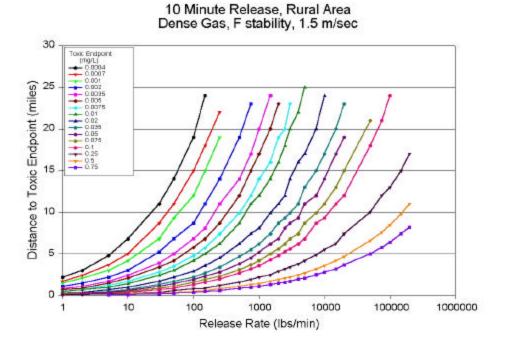


Figure 4-5. Dense gas in rural area, 10 minute release.

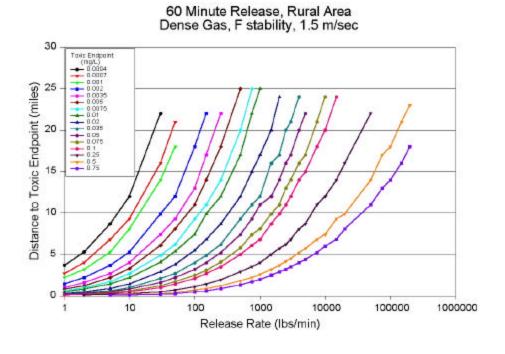


Figure 4-6. Dense gas in rural area, 60 minute release.

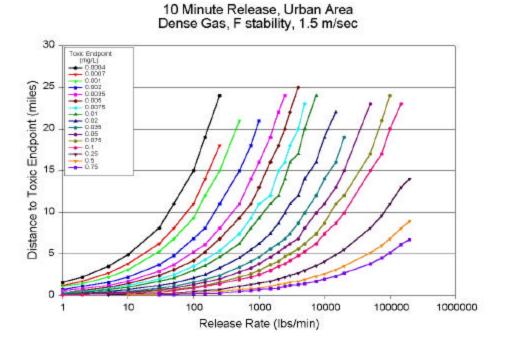


Figure 4-7. Dense gas in urban area, 10 minute release.

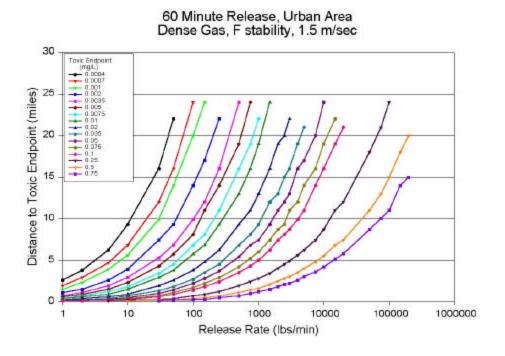


Figure 4-8. Dense gas in urban area, 60 minute release.

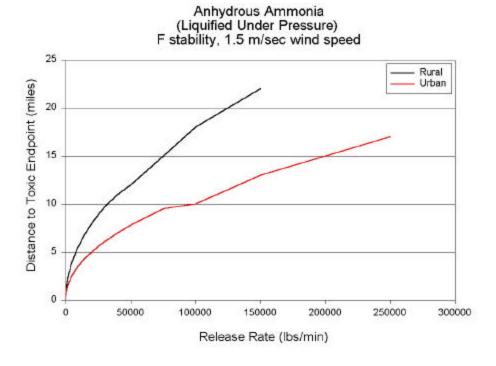


Figure 4-9. Anhydrous ammonia (liquefied under pressure) release.

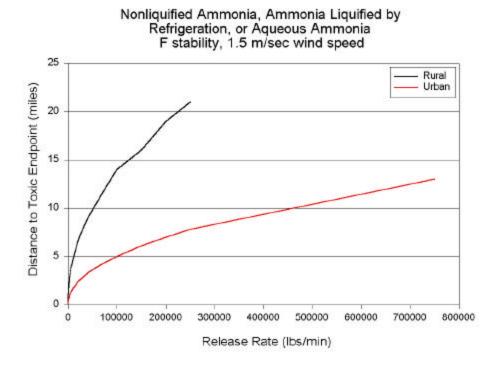


Figure 4-10. Anhydrous ammonia (non-liquefied, or liquefied by refrigeration, or aqueous ammonia) release.

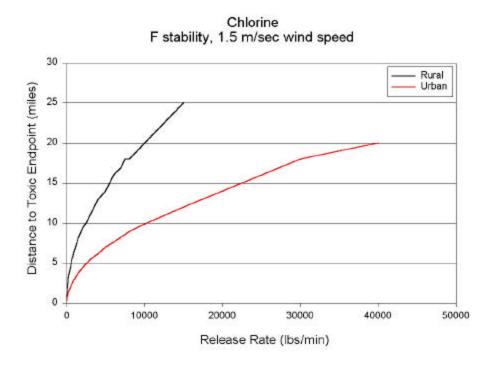


Figure 4-11. Chlorine release.

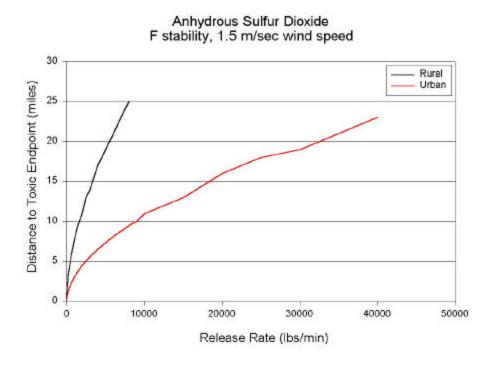


Figure 4-12. Anhydrous sulfur dioxide release.

## Estimation of Distance to Overpressure Endpoint for Flammable Substances

For the worst-case scenario involving a release of flammable gases and volatile flammable liquids, assume that the total quantity of the flammable substance forms a vapor cloud within the upper and lower flammability limits and the cloud detonates. As a conservative worst-case assumption, this procedure assumes that 10 percent of the flammable vapor in the cloud participates in the explosion. This procedure estimates the distance to an overpressure level of 1 pound per square inch (psi) resulting from the explosion of the vapor cloud. An overpressure of 1 psi may cause partial demolition of houses, which can result in serious injuries to people, and shattering of glass windows, which may cause skin laceration from flying glass. This section presents a simple method for estimating the area (distance from the explosion) potentially affected by a vapor cloud explosion of a hazardous substance. This procedure is based on a TNT-equivalent model.

#### Flammable Substances Not in Mixtures

For the worst-case analysis of a flammable substance that is not in a mixture with other substances, estimate the consequence distance for a given quantity of a regulated flammable substance using Table 4-1. This table provides distances to 1 psi overpressure for vapor cloud explosions of quantities from 500 to 2,000,000 pounds. An alternative is to calculate the worst-case distance for flammable substances using the heat of combustion of the flammable substance and the following equations.

Critical distances to an overpressure level of 1 pound per square inch (psi) may be determined using the following equation, which is based on the TNT-equivalency method:

$$D_{mi} = 0.0081 \times \left(0.1 \times W_{lb} \times \frac{HC_f}{HC_{TNT}}\right)^{1/3}$$
 Equation 7

where:

 $D_{mi}$ Distance to overpressure of 1 psi (miles)  $W_{lb} = HC_f =$ Weight of flammable substance (pounds)

Heat of combustion of flammable substance (kilojoules per kilogram), from Table

 $HC_{TNT} =$ Heat of explosion of trinitrotoluene (TNT) (4,680 kilojoules per kilogram)

## **Example for a Vapor Cloud Explosion of Propane**

A tank contains 50,000 pounds of propane. From Table 4-1, the critical distance to 1 psi overpressure is 0.3 miles for this quantity of propane. Alternatively, it is possible to directly calculate the distance to 1 psi using Equation 7:

$$D = 0.0081 \text{ x } [0.1 \text{ x } 50,000 \text{ x } (46,333/4,680)]^{1/3}$$

D = 0.3 miles

#### Flammable Mixtures

For a mixture of flammable substances, it is possible to estimate the heat of combustion of the mixture from the heats of combustion of the components of the mixture using Equation 8 and then use Equation 7 to determine the vapor cloud explosion distance. The heat of combustion of the mixture may be estimated as follows:

$$HC_{m} = \frac{W_{x}}{W_{m}} \times HC_{x} + \frac{W_{y}}{W_{m}} \times HC_{y}$$
 Equation 8

where:

 $HC_m$  = Heat of combustion of mixture (kilojoules per kilogram)

 $W_r$  = Weight of component "X" in mixture (kilograms or pounds/2.2)

 $W_m$  = Total weight of mixture (kilograms or pounds/2.2)

 $HC_x$  = Heat of combustion of component "X" (kilojoules per kilogram), from Table D-1

 $W_y$  = Weight of component "Y" in mixture (kilograms or pounds/2.2)  $HC_y$  = Heat of combustion of component "Y" (kilojoules per kilogram)

#### Example for Calculating Heat of Combustion of Mixture for Vapor Cloud Explosion Analysis

A mixture contains 8,000 pounds of ethylene (the reactant) and 2,000 pounds of isobutane (a catalyst carrier). To carry out the worst-case analysis, estimate the heat of combustion of the mixture from the heats of combustion of the components of the mixture. (Ethylene heat of combustion = 47,145 kilojoules per kilogram; isobutane heat of combustion = 45,576). Using Equation 8:

$$HC_{m} = \frac{[(8,000/2.2)\times47,145]}{(10,000/2.2)} + \frac{[(2,000/2.2)\times45,576]}{(10,000/2.2)}$$

$$HC_m = (37,716) + (9,115)$$

$$HC_m = 46,831$$
 kilojoules per kilogram

Now use the calculated heat of combustion for the mixture in Equation 7 to calculate the distance to 1 psi overpressure for vapor cloud explosion.

$$D = 0.0081 \text{ x} [0.1 \text{ x} 10,000 \text{ x} (46,831/4,680)]^{1/3}$$

$$D = 0.2$$
 miles

Table 4-1. Distance to Overpressure of 1.0 psi for Vapor Cloud Explosions of 500 - 2,000,000 Pounds of Regulated Flammable Substances Based on TNT Equivalent Method, 10 Percent Yield Factor (EPA 1999)

	in Cloud (pounds)	500	2,000	5,000	10,000	20,000	50,000	100,000	200,000	500,000	1,000,000	2,000,000
CAS No.	Chemical Name	Distance (	miles) to	1 psi Ov	erpressure							
75-07-0	Acetaldehyde	0.05	0.08	0.1	0.1	0.2	0.2	0.3	0.4	0.5	0.7	0.8
74-86-2	Acetylene	0.07	0.1	0.1	0.2	0.2	0.3	0.4	0.5	0.7	0.8	1.0
598-73-2	Bromotrifluoroethylene	0.02	0.04	0.05	0.06	0.08	0.1	0.1	0.2	0.2	0.3	0.4
106-99-0	1,3-Butadiene	0.06	0.1	0.1	0.2	0.2	0.3	0.4	0.5	0.6	0.8	1.0
106-97-8	Butane	0.06	0.1	0.1	0.2	0.2	0.3	0.4	0.5	0.6	0.8	1.0
25167-67-3	Butene	0.06	0.1	0.1	0.2	0.2	0.3	0.4	0.5	0.6	0.8	1.0
590-18-1	2-Butene-cis	0.06	0.1	0.1	0.2	0.2	0.3	0.4	0.5	0.6	0.8	1.0
624-64-6	2-Butene-trans	0.06	0.1	0.1	0.2	0.2	0.3	0.4	0.5	0.6	0.8	1.0
106-98-9	1-Butene	0.06	0.1	0.1	0.2	0.2	0.3	0.4	0.5	0.6	0.8	1.0
107-01-7	2-Butene	0.06	0.1	0.1	0.2	0.2	0.3	0.4	0.5	0.6	0.8	1.0
463-58-1	Carbon oxysulfide	0.04	0.06	0.08	0.1	0.1	0.2	0.2	0.3	0.4	0.5	0.6
7791-21-1	Chlorine monoxide	0.02	0.03	0.04	0.05	0.06	0.08	0.1	0.1	0.2	0.2	0.3
590-21-6	1-Chloropropylene	0.05	0.08	0.1	0.1	0.2	0.2	0.3	0.4	0.5	0.6	0.8
557-98-2	2-Chloropropylene	0.05	0.08	0.1	0.1	0.2	0.2	0.3	0.4	0.5	0.6	0.8
460-19-5	Cyanogen	0.05	0.08	0.1	0.1	0.2	0.2	0.3	0.4	0.5	0.6	0.8
75-19-4	Cyclopropane	0.06	0.1	0.1	0.2	0.2	0.3	0.4	0.5	0.6	0.8	1.0
4109-96-0	Dichlorosilane	0.04	0.06	0.08	0.1	0.1	0.2	0.2	0.3	0.4	0.5	0.6
75-37-6	Difluoroethane	0.04	0.06	0.09	0.1	0.1	0.2	0.2	0.3	0.4	0.5	0.6
124-40-3	Dimethylamine	0.06	0.09	0.1	0.2	0.2	0.3	0.3	0.4	0.6	0.7	0.9
463-82-1	2 2-Dimethylpropane	0.06	0.1	0.1	0.2	0.2	0.3	0.4	0.5	0.6	0.8	1.0
74-84-0	Ethane	0.06	0.1	0.1	0.2	0.2	0.3	0.4	0.5	0.6	0.8	1.0
107-00-6	Ethyl acetylene	0.06	0.1	0.1	0.2	0.2	0.3	0.4	0.5	0.6	0.8	1.0
75-04-7	Ethylamine	0.06	0.09	0.1	0.2	0.2	0.3	0.3	0.4	0.6	0.7	0.9
75-00-3	Ethyl chloride	0.05	0.08	0.1	0.1	0.2	0.2	0.3	0.4	0.5	0.6	0.8
74-85-1	Ethylene	0.06	0.1	0.1	0.2	0.2	0.3	0.4	0.5	0.7	0.8	1.0
60-29-7	Ethyl ether	0.06	0.09	0.1	0.2	0.2	0.3	0.3	0.4	0.6	0.7	0.9
75-08-1	Ethyl mercaptan	0.05	0.09	0.1	0.2	0.2	0.2	0.3	0.4	0.5	0.7	0.9
109-95-5	Ethyl nitrite	0.05	0.07	0.1	0.1	0.2	0.2	0.3	0.3	0.5	0.6	0.7
1333-74-0	Hydrogen	0.09	0.1	0.2	0.2	0.3	0.4	0.5	0.6	0.9	1.1	1.4
75-28-5	Isobutane	0.06	0.1	0.1	0.2	0.2	0.3	0.4	0.5	0.6	0.8	1.0
78-78-4	Isopentane	0.06	0.1	0.1	0.2	0.2	0.3	0.4	0.5	0.6	0.8	1.0
78-79-5	Isoprene	0.06	0.1	0.1	0.2	0.2	0.3	0.4	0.5	0.6	0.8	1.0
75-31-0	Isopropylamine	0.06	0.09	0.1	0.2	0.2	0.3	0.3	0.4	0.6	0.7	0.9
75-29-6	Isopropyl chloride		0.08	0.1	0.1	0.2	0.2	0.3	0.4	0.5	0.6	0.8
	Methane		0.1	0.1	0.2	0.2	0.3	0.4	0.5	0.7	8.0	1.0
	Methylamine		0.09	0.1	0.2	0.2	0.3	0.3	0.4	0.6	0.7	0.9
563-45-1	3-Methyl-1-butene		0.1	0.1	0.2	0.2	0.3	0.4	0.5	0.6	0.8	1.0
	2-Methyl-1-butene		0.1	0.1	0.2	0.2	0.3	0.4	0.5	0.6	0.8	1.0
115-10-6	Methyl ether	0.05	0.09	0.1	0.1	0.2	0.3	0.3	0.4	0.5	0.7	0.9

Table 4-1. Distance to Overpressure of 1.0 psi for Vapor Cloud Explosions of 500 - 2,000,000 Pounds of Regulated Flammable Substances Based on TNT Equivalent Method, 10 Percent Yield Factor (EPA 1999) (continued)

Quantity	in Cloud (pounds)	500	2,000	5,000	10,000	20,000	50,000	100,000	200,000	500,000	1,000,000	2,000,000
CAS No.	Chemical Name	Distance (	miles) to	1 psi Ov	erpressure	)		-				
107-31-3	Methyl formate	0.04	0.07	0.1	0.1	0.2	0.2	0.3	0.3	0.4	0.6	0.7
115-11-7	2-Methylpropene	0.06	0.1	0.1	0.2	0.2	0.3	0.4	0.5	0.6	0.8	1.0
504-60-9	1 3-Pentadiene	0.06	0.1	0.1	0.2	0.2	0.3	0.4	0.5	0.6	0.8	1.0
109-66-0	Pentane	0.06	0.1	0.1	0.2	0.2	0.3	0.4	0.5	0.6	0.8	1.0
109-67-1	1-Pentene	0.06	0.1	0.1	0.2	0.2	0.3	0.4	0.5	0.6	0.8	1.0
646-04-8	2-Pentene, (E) -	0.06	0.1	0.1	0.2	0.2	0.3	0.4	0.5	0.6	0.8	1.0
627-20-3	2-Pentene, (Z)-	0.06	0.1	0.1	0.2	0.2	0.3	0.4	0.5	0.6	0.8	1.0
463-49-0	Propadiene	0.06	0.1	0.1	0.2	0.2	0.3	0.4	0.5	0.6	0.8	1.0
74-98-6	Propane	0.06	0.1	0.1	0.2	0.2	0.3	0.4	0.5	0.6	0.8	1.0
115-07-1	Propylene	0.06	0.1	0.1	0.2	0.2	0.3	0.4	0.5	0.6	0.8	1.0
74-99-7	Propyne	0.06	0.1	0.1	0.2	0.2	0.3	0.4	0.5	0.6	0.8	1.0
7803-62-5	Silane	0.06	0.1	0.1	0.2	0.2	0.3	0.4	0.5	0.6	0.8	1.0
116-14-3	Tetrafluoroethylene	0.02	0.03	0.04	0.05	0.07	0.09	0.1	0.1	0.2	0.2	0.3
75-76-3	Tetramethylsilane	0.06	0.1	0.1	0.2	0.2	0.3	0.4	0.5	0.6	0.8	1.0
10025-78-2	Trichlorosilane	0.03	0.04	0.06	0.08	0.1	0.1	0.2	0.2	0.3	0.4	0.4
79-38-9	Trifluorochloroethylene	0.02	0.03	0.05	0.06	0.07	0.1	0.1	0.2	0.2	0.3	0.3
75-50-3	Trimethylamine	0.06	0.1	0.1	0.2	0.2	0.3	0.4	0.4	0.6	0.8	1.0
689-97-4	Vinyl acetylene	0.06	0.1	0.1	0.2	0.2	0.3	0.4	0.5	0.6	0.8	1.0
75-01-4	Vinyl chloride	0.05	0.08	0.1	0.1	0.2	0.2	0.3	0.4	0.5	0.6	0.8
109-92-2	Vinyl ethyl ether	0.06	0.09	0.1	0.2	0.2	0.3	0.3	0.4	0.6	0.7	0.9
75-02-5	Vinyl fluoride	0.02	0.04	0.05	0.06	0.08	0.1	0.1	0.2	0.2	0.3	0.4
75-35-4	Vinylidene chloride	0.04	0.06	0.08	0.1	0.1	0.2	0.2	0.3	0.4	0.5	0.6
75-38-7	Vinylidene fluoride	0.04	0.06	0.09	0.1	0.1	0.2	0.2	0.3	0.4	0.5	0.6
107-25-5	Vinyl methyl ether	0.06	0.09	0.1	0.2	0.2	0.3	0.3	0.4	0.6	0.7	0.9

## Spills of Mixtures of Hazardous Chemicals During Transportation Accidents

Spills involving more than one type of chemical are possible during some transportation accidents, especially involving railroads. Most other transportation modes (chemical tank trucks and pipelines) would only involve single hazardous materials. However, some trucks may be carrying smaller amounts of several different materials. Under certain conditions and combinations, multi-component spills may be dangerously reactive or generate hazardous by-products. The following discussion is an example evaluation for binary mixtures of some materials.

A example list of chemicals is shown on Table 4-2 by reactivity group. Table 4-3 displays these groups in the form of a matrix which indicates the potential for unsafe conditions if chemicals from any two groups may mix. Extreme caution would need to be taken to prevent accidental mixing of chemicals belonging to groups for which an "X" appears. Regulations restrict the transportation of large amounts of chemicals that may mix forming extremely hazardous conditions, but errors do occur. The accidental mixing of reactive groups could, in certain instances, result in violent and hazardous chemical reactions. The generation of toxic gases, the heating, overflow and rupture of storage tanks, and fire and explosion are possible consequences of such reactions.

The following discussion also gives a general overview of what products and conditions could be produced by the reaction of any potentially hazardous combinations of chemicals from two different groups. An extensive variety of combinations are possible when considering the reactions of broad groups of chemicals. Even though combinations of certain groups can be considered potentially hazardous, there may exist individual combinations which produce no unsafe conditions. Conversely, some combinations which are generally not considered hazardous might very well be if unusual circumstances occur. Combinations of more than two groups would be much more complex, but if any two reactive groups are involved, it should be assumed that hazardous conditions would be likely.

#### **Table 4-2. Reactivity Groups for Selected Chemicals**

**Inorganic Acids** 

Boric acid Chromic acid\* Fluoboric acid Hydrochloric acid

Hydrofluoric acid (Anhydrous) Hydrofluoric acid (Aqueous)

Nitric acid\*

Sulfur dioxide (Anhydrous)

Phosphoric acid Sulfuric acid\* (Oleum) Sulfur trioxide (Anhydrous)

Organic Acids
Acetic acid

Caustics

Sodium hydroxide

Soda ash

**Halogenated Compounds** 

Transformer oils
Silicon tetrafluoride

Petroleum Oils

Diesel fuel

Table 4-3. Chemical Compatibility

\*Compound may also be considered a strong oxidant.

<u>Ammonia</u>

Ammonia (Anhydrous) Ammonium hydroxide

Sulfur, Molten Sulfur liquid

**Inorganic Salts** 

Alum

Ammonium fluoride Calcium sulfate Fluorspar

Strong Oxidant Hydrogen Peroxide Potassium dichromate

Potassium dichromate
Potassium permanganate
Sodium bichromate

<u>Metals</u>

Arsenic precipitate

Bauxite Metal Oxides

1	Inorganic Acids	1										
2	Organic Acids	X	2									
3	Caustics	X	X	3								
4	Halogenated Compounds	X		X	4		_					
5	Petroleum Oils					5		_				
6	Ammonia	X	X				6		_			
7	Sulfur, Molten					X		7				
8	Inorganic Salt								8			
9	Strong Oxidant		X			X				9		
10	Metal Oxides	X									10	
11	Metals	X		X								Ī

<sup>&</sup>quot;X" represents a potentially hazardous combination.

## Reaction Products of Combinations of Potentially Hazardous Reactivity Groups

<u>Inorganic Acids + Organic Acids</u>

1. Vapor Products in the Presence of Water

- Depending on the heat generated by the reaction, fumes from the component acids may be given off. The reaction may form volatiles giving off ketones, aldehydes, and esters.
- 2. Solid or Liquid Products in the Presence of Water
- Possible formation of precipitates.
- 3. Vapor Products Without Water
- Same as with the presence of water.
- 4. Solid or Liquid Products Without Water
- Char or charcoal products may form depending on the circumstances.

#### <u>Inorganic Acids + Caustics</u>

- 1. Vapor Products in the Presence of Water
- The main products of this reaction are heat and salts. Component acid fumes may be given as a result of the heat involved.
- 2. Solid or Liquid Products in the Presence of Water
- No significant products are expected to occur from this reaction.
- 3. Vapor Products Without Water
- Water vapor, carbon dioxide, and possibly acid fumes will be produced.
- 4. Solid or Liquid Products Without Water
- A crusty mass of salt precipitates is expected to form with the possibility of acid and precipitate splatter.

#### <u>Inorganic Acids + Halogenated Compounds</u>

- 1. Vapor Products in the Presence of Water
- Water vapor and carbon dioxide will be produced along with the possibility of halogens and nitrous oxides being emitted.
- 2. Solid or Liquid Products in the Presence of Water
- This reaction could produce either or both solid and liquid products depending on the components.
- 3. Vapor Products without Water
- Basically the same products as those formed in the presence of water, only in larger quantities
- 4. Solid or Liquid Products Without Water
- Miscellaneous tars are expected to result from this reaction.

#### <u>Inorganic Acids + Ammonia</u>

- 1. Vapor Products in the Presence of Water
- Vapor emissions from components only, no vapor products are expected.
- 2. Solid or Liquid Products in the Presence of Water
- Depending on the concentrations of the components, ammonia salt precipitates are likely to occur.
- 3. Vapor Products Without Water
- No significant vapor products are expected to occur in this reaction.
- 4. Solid or Liquid Products Without Water
- Particulates of ammonium halides would be generated from this reaction.

#### <u>Inorganic Acids + Metal Oxides</u>

1. The same products as listed in the Inorganic Acids + Caustics reaction are expected to form in this reaction, but the reaction will be less violent.

#### <u>Inorganic Acids + Metals</u>

- 1. Vapor Products in the Presence of Water
- Hydrogen and water vapors will be produced from this reaction, violent splattering may also occur.
- 2. Solid or Liquid Products in the Presence of Water
- Various solids are likely to be precipitated out depending on the acid involved.
- 3. Vapor Products Without Water
- Highly toxic arsinces and stybines would result from arsenic precipitate combining with inorganic acids.
- 4. Solid or Liquid Products Without Water
- Same as with water except that larger quantities of solids will be produced.

#### Organic Acids + Caustics

- 1. Vapor Products in the Presence of Water
- Vapor products from this reaction will be primarily odors resulting from the formation of soaps. Phenol derivatives might also occur as vapors.
- 2. Solid of Liquid Products in the Presence of Water
- Solid products will occur in the form of various, insoluble materials and soaps.
- 3. Vapor Products Without Water
- Mainly soap vapors and gases.
- 4. Solid or Liquid Products Without Water
- Same as with water.

#### Organic Acids + Ammonia

- 1. Vapor Products in the Presence of Water
- These would be vapors of the components and the various reaction products.
- 2. Solid or Liquid Products in the Presence of Water
- The components are soluble with little or no precipitates.
- 3. Vapor Products Without Water
- Vapors are the same as those with water except in larger quantities.
- 4. Solid or Liquid Products Without Water
- Ammonium acetate and salts present in a gum-like substance.

## Organic Acids + Oxidants

- 1. Vapor Products in the Presence of Water
- This reaction will produce a myriad of vapor products which could include gases such as formaldehyde and methane.
- 2. Liquid or Solid Products in the Presence of Water
- Possibly some solid products will form.
- 3. Vapor Products Without Water

- This reaction will produce more vapor products than if water was present. Water vapor would be given off explosively along with carbon dioxide.
- 4. Liquid or Solid Products Without Water
- Possible formation of solids, more so than with water.

#### <u>Caustics + Halogenated Compounds</u>

- 1. Vapor Products in the Presence of Water
- Vaporous halogens can be expected to be given off by this reaction.
- 2. Solid or Liquid Products in the Presence of Water
- Very little, if any, solids are likely to be produced in this reaction.
- 3. Vapor Products Without Water
- Possible toxic halogens and halogenated compounds would be emitted as vapors.
- 4. Solid or Liquid Products Without. Water
- Some solids are expected to be produced.

#### Caustics + Metals

- 1. Vapor Products in the Presence of Water
- The reaction products are basically the same as those of acids and metals which yield hydrogen and water vapors.
- 2. Solid of Liquid Products in the Presence of Water
- Reaction will form arsenic products in solid form.
- 3. Vapor Products Without Water
- Products again are basically the same as those of acids and metals, except that arsine will probably not be given off.
- 4. Solid or Liquid Products Without Water
- Same as with water except in larger quantities.

## Petroleum Oils + Caustics

- 1. Vapor Products in the Presence of Water
- Many vaporous products will be given off from this violent reaction.
- 2. Solid or Liquid Products in the Presence of Water
- Some solids can be expected to be produced.
- 3. Vapor Products Without Water
- Probably an explosive, flaring reaction with much particulate matter being released.
- 4. Solid or Liquid Products Without Water
- Products would be in the form of a crusty mass of precipitates or a gummy tar.

#### <u>Petroleum Oils + Molten Sulfur</u>

- 1. Vapor Products in the Presence of Water
- Possibly explosive reaction accompanied by fire. Sulfur dioxide and maybe sulfur trioxide would be emitted. Carbon particulates and sulfur comb inations of petroleum products will also be given off.
- 2. Solid or Liquid Products in the Presence of Water
- Solid sulfur and possibly some tars would result.

- 3. Vapor Products Without Water
- The reaction would be violent yielding larger quantities of products and a high probability of fire.
- 4. Solid or Liquid Products Without Water
- Solid sulfur and probably tars would result.

## Hazards Associated with Accidental Releases of Ammonia during Transportation Operations

This discussion presents the results of a detailed site-specific evaluation of potential ammonia spills associated with transportation accidents. These accidents may range from complete loss of the cargo from specialized ammonia transport ships, losses during transfer operations, and losses during trucking of ammonia. Both water and air quality problems associated with these various spill conditions are addressed in this discussion, and consider a typical range of site meteorological conditions, not just worst-case conditions as described earlier using the methods from the *Offsite Consequence Analysis* (EPA 1999) procedure.

## Properties of Ammonia

Ammonia is a colorless gas at atmospheric pressure and normal temperature. It is alkaline and possesses a characteristic penetrating odor. On comp ression and cooling, ammonia gas condenses to a liquid about 60 percent as heavy as water. The liquid has a high vapor pressure at ordinary temperature, and commercial shipment requires pressure containers unless the liquid is refrigerated. Ammonia is readily absorbed in water to make ammonium hydroxide (NH<sub>4</sub>OH). Considerable heat evolves during the solution of ammonia gas in water (1 lb NH<sub>3</sub> gas produces 937 Btu when dissolved in water).

Ammonia does not support ordinary combustion, but it does burn with a yellowish flame in an atmosphere of air or oxygen. The ignition temperature of ammonia-air mixtures is 780°C, and the products of combustion are mainly nitrogen and water. Under certain conditions, mixtures of ammonia and air will explode when ignited. The explosive range for dry ammonia-air mixtures is about 16 to 25 percent ammonia. Admixture with other combustible gases such as hydrogen, admixture of oxygen replacing air, and/or higher than atmospheric temperatures and pressures broaden the explosive range. Because this range is restrictive, the explosion hazard is usually ignored as being highly unlikely, and ammonia is generally treated as a nonflammable compressed gas.

The major hazards associated with ammonia are from the toxic effects on breathing and caustic burns caused by vapor, liquid, or solutions. Also, the cryogenic properties of refrigerated liquid ammonia can present some unique hazards because of the extreme cold. The concentrations of ammonia vapor in the air that will cause various physiological responses in humans are given in Table 4-4. The toxic endpoint of ammonia, as defined in Appendix A to 40 CFR part 68, is 200 ppm (equivalent to 0.14 mg/L), as used by EPA (1999) for offsite consequence analyses.

Table 4-4. Physiological Response to Various Concentrations of Ammonia (Kirk and Othmer)

Physiological Response	Approximate Ammonia Concentration in Air (ppm)
Least detectable odor	50
Maximum concentration allowable for prolonged exposure	100
Maximum concentration allowable for short exposure (1/2-1 hr)	300-500
Least amount causing immediate irritation to throat	400
Least amount causing immediate irritation to eyes	700
Compulsive coughing and possible death	1700
Dangerous for even short exposure (1/2 hr)	2500-4500

## Potential Sources of Accidental Releases

Most leaks and spills of ammonia are caused by failure of equipment or mishandling by personnel. There are many sources for these releases. The most serious and probable of these sources are discussed below. The amounts of releases are estimated for typical design conditions.

#### Vessels

- 1. A catastrophic accident, such as a collision, involving a vessel could release a potential maximum of about 12,000 tons of liquid ammonia.
- 2. The refrigeration system on a vessel could develop a leak from a broken pipe or fitting. During a transfer operation, the loss during a 5-minute shutdown period could amount to about 125 lb, while without a transfer, the loss could be about 42 lb.
- 3. Spills could occur at a terminal during off-loading of a vessel. Because of automatic emergency equipment, the losses would be limited to line drainage between the automatic valves and the break. This loss could be about 7 tons.

#### Trucks and Rail Cars

- 1. Trucks and rail cars could be involved in accidents with subsequent leaks or spills. If there is a tank rupture, the entire ammonia cargo of up to about 20 tons/truck and 80 tons/rail car could be spilled almost instantaneously. A lesser amount could be lost through a tank crack or a broken fitting.
- 2. During the normal loading of a tank truck at a storage terminal, about 1 oz of ammonia vapor may be released to the atmosphere through a vent stack usually about 20 ft high.

#### Venting

Various items of equipment have relief valves that vent ammonia vapor in case of pressure buildup (usually caused by a rise in tempera ture from loss of refrigeration or from a fire.) This venting occurs in a controlled fashion as described below:

- 1. The relief valves on ammonia-carrying vessels can begin to vent after several days without refrigeration. These losses can amount to 200 to 500 lb/hr.
- 2. Large refrigerated storage tanks can vent after about 4 hours without refrigeration. The maximum vent rate can be about 750 lb/hr per tank, which would require an extremely long time to completely vent a tank. Backup electrical generators supply electricity to the refrigeration equipment in case of pro longed power outages, the most probable cause of refrigeration failures.
- 3. The tanks on trucks and rail cars would vent only if they were involved in a fire. In a fire, a full truck tank would empty in about 4.5 hours, and a full rail car would empty in about 18 hours.

## Water Quality Effects

The following discussion pertains to the hazards of spilling anhydrous ammonia during shipping and transfer operations of a facility located on a narrow ship channel. The discussion utilizes the far field prediction models provided in Raj, *et al.* (1974) that are specific for anhydrous ammonia.

Anhydrous ammonia is a cryogenic liquid ( $-28^{\circ}F$ ) at normal atmospheric pressure and floats on the water surface, rapidly dissolving within the water body into ammonium hydroxide (NH<sub>4</sub>OH), while at the same time boiling into the atmosphere as gaseous ammonia (NH<sub>3</sub>). The partition ratio (the quantity of ammonia that dissolves into the receiving water divided by the total quantity spilled) is normally between 0.5 and 0.8 for surface spills, and somewhat higher for underwater spills. For simplicity, the partition ratio for these analyses is assumed to be 0.6 for all spills. Furthermore, all spills are considered to be instantaneous.

If the water body near the site is of a generally one-dimensional nature and lacks advective currents, the spill would be distributed evenly over the cross section of the channel. Furthermore, it is expected that the length of channel affected by the spill would be roughly proportional to the length of time elapsed after the spill. If one further assumes that the concentration is constant longitudinally behind the advancing pollution front, then a single concentration value can be calculated to represent the entire contaminated prism as a function of increasing channel length for a given spill quantity. These functions are plotted on Figure 4-13, which assumes a constant cross-sectional area of 10,000 ft<sup>2</sup> within a ship channel and a speed of advance of the pollution front of

approximately 0.2 ft/sec (if the actual cross-sectional area is larger than 10,000 ft<sup>2</sup>, the resulting concentrations would be correspondingly smaller; if the actual water velocities were greater than 0.2 ft/sec, the times for the indicated concentrations would be correspondingly sooner).

In reality, a well-mixed pollutant diffuses along a one-dimensional channel; is not concentrated evenly along the polluted channel length. The actual concentrations are in inverse proportion to the distance from the spill point. It can be assumed that the single concentration values obtained for a given spill value and channel length (Figure 4-13) best represent those concentration values expected to be measured approximately midway between the spill point and the limit of the channel length affected. The actual values will be greater than those shown near the spill point by a factor of between 1 and 2, and will be less than the plotted concentrations down-channel from the midpoint.

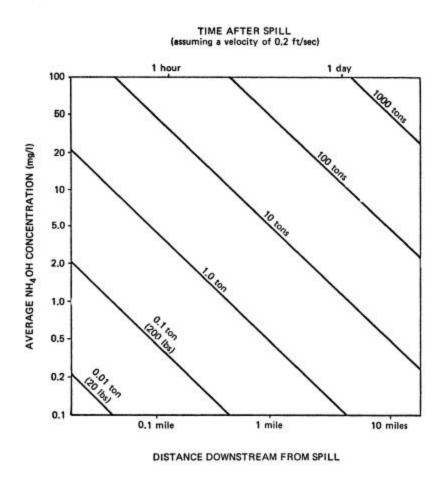


Figure 4-13. Mean ammonium hydroxide concentrations in estuarine prisms for various ammonia spill quantities.

The downstream length before complete mixing across the channel occurs can be estimated using an equation presented by Thomann and Mueller (1987):

$$L_m = \frac{2.6UB^2}{H}$$
 Equation 9

where: U is the stream velocity in ft/second

B is the average stream width in feet, and

H is average stream depth in feet

As an example, consider the following conditions approximating the above example:

U = 0.2 ft/sec B = 285 ftH = 35 ft

In this case, the "complete mixing" length would be about 1200 feet (0.22 mile). About half of this distance would be needed if the discharge location is located at the centerline of the channel. These are relatively short lengths for most of the spills represented in Figure 4-13, and would occur between one and two hours after the ammonia is released.

## Air Quality Effects

The physical processes governing atmospheric dispersion when large quantities (over 1000 tons) of liquid ammonia are spilled instantaneously on or under water are not well understood. However, laboratory, swimming pool, and lake tests provide some insight into the dispersion behavior. These results offer tentative models for estimating potential atmospheric concentrations from spills.

The important parameters needed for analysis of instantaneous ammonia spills are:

- The amount of LNH<sub>3</sub> released
- The actual ratio of LNH<sub>3</sub> that evaporates into the atmosphere when the accident happens on or under the water (one minus the partition ratio)
- The estimated rate of rise of the NH<sub>3</sub> vapor cloud.

The partition ratio of 0.6 (from estimates developed by Raj, *et al.* 1974) was applied in estimating ambient concentrations from spills. Raj and his associates also developed a plume rise model that seemed to agree well with observed cloud center heights and was considered conservative. In these same studies, well-defined Gaussian distributions of concentrations in the horizontal direction were observed. Therefore, Gaussian dispersion models (presented by Turner 1970), using Pasquill-Gifford stability classes, have been applied in estimating the air quality impacts of hypothesized spills on both land and water in the following discussion.

#### **Tank Ruptures on Vessels**

Expected ambient concentrations were calculated for distances of 0.2 to 10 miles downwind from a hypothetical vessel accident in which an entire cargo of liquid ammonia was spilled into the water instantaneously. It was assumed that the entire spill would spread over a circular area with a radius of about 800 ft and that 40 percent of the LNH<sub>3</sub> would evaporate in several minutes (based on projections from Raj, *et al.* 1974).

Since the density of NH<sub>3</sub> is only 0.6 of the density of air at the same temperature and pressure, atmospheric stability will have very little effect on the rate of rise of the NH<sub>3</sub>. Because the rate of rise of the NH<sub>3</sub> is not controlled by atmospheric stability, the only way any part of the plume can reach the ground at a point downwind is through turbulent atmospheric transport. Stability classes A, B, and C are the unstable atmospheric classes, and by definition atmospheric instability fosters turbulent action. Stability class D is called the neutral class, but it embraces both stable and unstable conditions. For such a fast-rising gas (NH<sub>3</sub>), it seems doubtful that the plume can return to the ground, even with unstable conditions. Since stable classes E and F have low levels of turbulence, calculations were made only for classes A, B, C, and D. Even with these unstable conditions, applying the Pasquill-Gifford equation is considered to be a conservative practice, yielding an overestimation of expected ambient concentrations.

Downwind distances to points at which selected concentrations were calculated to occur are summarized in Table 4-5. It should be noted that 0.2 mile is just outside the assumed spill area.. It was assumed that concentrations within the spill area would be at least 5000 ppm (and quickly lethal).

The maximum durations of exposure for the various concentrations will be along the dispersion centerline in the horizontal plane at the ground, in the direct downwind direction. Away from this centerline, durations will be shorter. These duration values are summarized in Table 4-6. The durations are calculated for an instantaneous spill and will increase if the ammonia vapor is released over a longer period; however, concentrations will then be correspondingly lower.

Table 4-5. Estimated Downwind Distances of Four Concentrations of NH<sub>3</sub> - Total Vessel Spill Of 12,000 Tons

Atmospheric	Wind Speed	Downwind Distances (miles) for:					
Stability Class	(mph)	50 ppm	300 ppm	1700 ppm	5000 ppm		
A	5	2.0	0.7	<0.2	<0.2		
В	11	4.4	1.9	0.8	0.4		
С	15	1.2	0.9	0.6	0.4		
	25	9.0	3.5	1.6	1.0		
D	≤15	<0.2	<0.2	<0.2	<0.2		
	25	0.6	0.5	0.4	0.3		
	35	1.1	0.9	0.7	0.5		
	45	2.0	1.5	1.1	0.8		

Table 4-6. Estimated Durations Of Various Concentrations at Several Distances Directly Downwind of an Instantaneous Total Vessel Spill

Atmospheric	Wind	Estimated Duration (minutes) for:							
Stability Class	Speed (mph)	*50 ppm	*300 ppm	*1700 ppm	<b>≈</b> 5000 ppm				
		At a distar	nce of 0.5 mile						
Α	5	19	8	0	0				
В	11	9	7	4	0				
С	15	4	3	1	0				
	25	3	3	2	1				
D	≤15	0	0	0	0				
	25	<1	0	0	0				
	35	1	1	<1	<0.5				
	45	1	1	<1	<0.5				
	At a distance of 1.0 mile								
Α	5	18	0	0	0				
В	11	9	6	0	0				
С	15	3	0	0	0				
	25	3	3	1	0				
D	≤15	0	0	0	0				
	25	0	0	0	0				
	35	<1	0	0	0				
	45	1	<1	<1/2	0				
	At a distance of 5.0 miles								
Α	5	0	0	0	0				
В	11	5	0	0	0				
С	15	0	0	0	0				
	25	4	0	0	0				
D	≤15	0	0	0	0				
	25	0	0	0	0				
	35	0	0	0	0				
	45	0	0	0	0				

The values in Tables 4-5 and 4-6 indicate that:

- For atmospheric stability classes A and B, which involve only low wind speeds, ambient concentrations at a given distance are relatively low, but exposure durations are longer.
- For stability classes C and D, which generally involve higher wind speeds, ambient concentrations are relatively high, but exposure durations are relatively short.

The ammonia cloud is not expected to touch the ground surface within 10 miles for stability classes E and F, because of the small dispersion coefficients and rapid rise of the NH<sub>3</sub> cloud. For all atmospheric stability classes, under certain terrain conditions, ambient concentrations higher than those calculated may occur, depending upon relative altitude and distance from the spill. As an example, a rising plume may strike the ground in an area of extreme topography or if high buildings are nearby.

In fog or low cloud conditions, some spilled  $NH_3$  would react with the water vapor, becoming  $NH_4OH$ . This reaction would cause lower ambient concentrations and longer durations than those shown in Tables 4-5 and 4-6. In fog or a low stratus cloud layer, the lateral spread is expected to be small. In cumulus clouds, there would be greater lateral and vertical spreading. Since an  $NH_4OH$  molecule is about twice as heavy as a water molecule, it is expected that fallout would occur, primarily near the scene of the accident.

#### **Other Malfunctions**

#### Transfer Sills

Transfer spills could occur during the off-loading of a vessel or the loading of a truck or rail car. It was assumed that the LNH<sub>3</sub> from a transfer spill would spread evenly on the land and completely evaporate in one hour or for the duration of the spill (if greater than one hour) and that none of the ammonia would run off into the water. The spill would then act as a continuous source, allowing use of the Gaussian dispersion model for a continuous point ground-level source to predict concentrations downwind. Other malfunctions, such as venting from relief valves on vessels, storage tanks, trucks, and rail cars, can be described by the same model, with the only variation being the rate of venting or evaporation.

The highest concentrations are estimated for stability class D, as discussed previously. The calculations are based on a wind speed of 10 mph because this value represents the most turbulent conditions expected to occur in class D.

#### Venting Leaks

With loss of refrigeration, LNH<sub>3</sub> will begin to boil (vaporize). As heat is absorbed from the surroundings, temperature and pressure inside the tank will rise. Because of heavy insulation of large LNH<sub>3</sub> storage tanks, about 4 hours without refrigeration can elapse before the relief valves begin to vent. Because of higher pressure settings on relief valves on vessels, several days without refrigeration would be required before internal pressure can build to the point where venting begins. Maximum venting rates are expected to be about 200 to 500 lb/hr for vessel tanks.

Trucks and trains are designed to transport ammonia under pressure in liquid form at ambient temperatures. A fire in or near a truck or rail car could cause relief valves to open. The rate capacity of the relief valves is about 4.5 tons/hr of NH<sub>3</sub>. The heat from a fire, in addition to causing the ammonia to boil, would create a strong updraft which would probably cause the ammonia vapors to quickly rise. A fire could also incinerate some of the ammonia vapors. Both of these conditions would combine to reduce ground-level concentrations to below those predicted here.

## Tank Ruptures

Trucks and trains are susceptible to accidents which could create more serious hazard conditions than venting. The worst accident situation would be one in which the tank ruptured and instantaneously spilled 20 tons of LNH<sub>3</sub> for a truck or 80 tons of LNH<sub>3</sub> for a rail car onto the ground without a fire. Without the additional heat from a fire, no special supporting updraft would be created, and the ammonia cloud, though rising, would stay closer to the ground for a greater distance downwind, especially if foggy or rainy. It is assumed that the entire cargo would spread out to a uniform depth of about 3 inches (EPA 1999 assumes a pool depth of 1 cm and the corresponding pool would therefore be about 7.5 times larger, and the total evaporation rate would be similarly larger, but for a shorter duration). Ammonia pools of 3 inches in depth are expected to evaporate in approximately 2 hours. The evaporation

rate would be 40 ton/hr (rail car) and 10 tons/hr (truck). If the LNH<sub>3</sub> is contained in a smaller area, if a smaller total amount spills, or if the atmosphere is in a condition other than class D and/or has higher wind speeds, ammonia concentrations downwind are expected to be less. Similarly, if the pool was 1 cm deep, the ammonia would evaporate in about 15 minutes and the evaporation rate would be about 300 ton/hr (rail car) and 75 tons/hr (truck), and the corresponding downwind concentrations would be about 7.5 times larger than if a 3 inch pool was formed.

## Summary of Effects on the Living Environment

Table 4-7 summarizes expected downwind distances and durations of ammonia concentrations for different spill conditions. The following discussion summarizes the expected impacts associated with these spills.

Table 4-7. Estimated Downwind Distances of Concentrations of NH<sub>3</sub> for Various Transportation Accidents

	Assumed	Maximu				
Malfunction	Evaporation Rate (lb/hr)	50 ppm	300 ppm	1700 ppm	5000 ppm	Assumed Duration
Vessel venting on loss of refrigeration	500	0.05	0.05	<0.01	<0.01	Until refrigeration is re- established and the NH <sub>3</sub> is cooled sufficiently
Truck or rail car transfer line accident	8,000	0.33	0.10	0.03	0.02	1 hr <sup>b</sup>
Truck or rail car venting in a fire	9,000	0.36	0.11	0.04	0.02	1 hr <sup>b</sup>
Vessel transfer line accident	14,000	0.48	0.15	0.05	0.02	1 hr <sup>b</sup>
Truck tank rupture	20,000	0.60	0.19	0.06	0.03	2 hr <sup>b</sup>
Rail car tank rupture	80,000	1.40	0.46	0.15	0.12	2 hr <sup>b</sup>

<sup>&</sup>lt;sup>a</sup> Assumed wind speed, 10 mph; stability class D.

#### **Human Population**

Human physiological responses to various concentrations of ammonia were presented in Table 4-4. Depending on specific atmospheric conditions, it can be expected that people several miles downwind will likely have to be treated for ammonia inhalation effects for a vessel disaster, but no deaths are likely to occur, except possibly very close to a loss. Durations of exposure will increase if the ammonia vapor is released over a longer period of time (not instantaneously), but the concentrations will be correspondingly lower. Other accidents could cause downwind concentrations sufficient to cause odors up to 1.5 miles away. Evacuation might be required for up to 0.5 mile downwind, depending upon the type of accident. Because of ammonia's characteristic odor at relatively low concentrations, people will likely respond by leaving an affected area before official warnings are issued.

#### **Marine and Aquatic Organisms**

In the event of a spill during the off-loading of a vessel, ammonia could be leaked directly into the water. Assuming a line drainage directly into the water, 7 tons of liquid ammonia could be lost. With a partition ratio of 0.6, 4 tons of NH<sub>3</sub> would go into solution as ammonium hydroxide, while the remainder would vaporize into the air. The toxicity of an ammonia solution in water is directly proportional to the concentration of nonionized NH<sub>3</sub> present. The amount of nonionized NH<sub>3</sub> is dependent on pH, temperature, and salinity. With a pH of 8.0, a temperature of 15°C, and zero salinity, the percentage of nonionized NH<sub>3</sub> would be 5.7 percent. At a pH of 9.0, nonionized NH<sub>3</sub> would be 37.7 percent of the total ammonia concentration. A concentration of nonionized NH<sub>3</sub> greater than 1.25 ppm can be toxic to freshwater fish.

With the pH range described above, assuming complete mixing within a channel having a  $10,000 \, \text{ft}^2$  cross-section, a 7-ton spill would produce toxic conditions for fish for a distance of about 1 mile along the channel. There would be a severe fish kill in the immediate vicinity of the spill where the concentrations of NH<sub>3</sub> would be highest. It can also be assumed that mortality of planktonic and benthic organisms would also occur in the vicinity of the spill.

<sup>&</sup>lt;sup>b</sup> If the durations are shorter (pool depths shallower) the concentrations will be greater; similarly, if the durations are longer, the concentrations will be less.

A spill of lesser magnitude could occur if the refrigeration equip ment on a vessel were to develop a leak from a broken pipe or fitting. Such a leak could release from 42 to 125 1b of  $NH_3$  in 5 minutes. The effect of such a release probably would be confined to the local area. However, the possibility of a fish kill within the immediate area is likely.

In the unlikely event that a catastrophic accident were to occur causing the release of an entire vessel's contents, about 12,000 tons of NH<sub>3</sub> could be released into the water. Such a spill could ultimately cause toxic concentrations of NH<sub>3</sub> throughout a large area. The size of the affected area would change as the contaminated water moves downstream. There would be massive mortalities of fish, plankton, shellfish, and other benthic organisms.

A long-term result of any ammonia spill would be increased eutrophication of the receiving waters. The additional nutrient levels could stimulate noxious blooms of algae, which would cause continuous water quality degradation.

#### **Terrestrial Biology**

In high enough concentrations, ammonia is toxic to living organisms (Miner 1969, and Levine 1968). Large amounts of this chemical would be released into the environment in the event of a large leak or spill, such as a total vessel spill. Regardless of where a vessel broke up along an inland route, high concentrations of ammonium hydroxide would likely reach shore. If this chemical floated into any of the wetlands bordering the shipping route, much of the vegetation would be killed, potentially causing destruction of important habitat for waterfowl, shorebirds, and other shore species.

Waterfowl and shorebirds present in the wetlands at the time the ammonium hydroxide came into shore could be directly affected. A large number of birds could be killed by ingestion of the chemical. The ammonium hydroxide could also strip protective oils from the feathers of waterfowl, causing the loss of the birds' natural water repellency. In this case, birds would die either from drown ing or from infections contracted as a result of getting wet.

The ammonia which would escape into the atmosphere would form a plume with a concentration of several thousand ppm at its center. Concentrations of 1700 ppm or more of ammonia would occur for several minutes at sea level for a distance of several miles downwind from the location of a vessel accident or for longer periods but over a smaller area if the ship leaked slowly. It is likely that any bird or animal exposed to these high concentrations of ammonia would be injured or rapidly killed. Re gardless of where the vessel broke up along its route, birds in the vicinity of the accident could possibly become disoriented in their attempts to escape the odor and might fly into the lethal part of the plume. If the vessel broke up near shore, animal and birds could be killed for several miles inland.

Severe damage to vegetation would also occur. The extent of this damage would depend upon the resistance of individual plant species to ammonia and the time of year the spill occurred. Plant species differ in their sensitivity to ammonia (Miner 1969). It is possible that some species may be able to withstand high concentrations of the gas for several minutes. In the spring or summer, a concentrated ammonia plume would probably severely damage most vegetation that it contacts. Perennial species in the natural flora would be most affected by ammonia in the summer and early fall when they are under the greatest physiological stress because of low soil moisture. Since seeds are most resistant to ammonia, annual species in the natural flora would not be greatly affected during summer months. These species would be hardest hit in the spring or fall.

#### Potential Movement and Effects Associated with Oil Spills

This report section is a summary of oil spill analysis and impact reports prepared by Woodward Clyde Consultants, prepared for numerous clients for submission to regulatory agencies. The following discussions are excerpted and summarized from these reports to indicate how impacts associated from oil spills can be evaluated, especially concerning spill movement and dispersion. The fate and effects of oil spills on the environment, based on historical spill incidents, are also described.

#### Parameters Affecting Oil Spill Movement

The movements, and other characteristics, of a spill of petroleum hydrocarbons lost on water are controlled by weather conditions (wind, temperature, and rainfall), ocean conditions (tides and currents), and physical parameters

of the materials which could be spilled. The important physical parameters of the various petroleum hydrocarbons include the following:

Specific gravity (or density)
Evaporation rate
Boiling range
Viscosity
Pour point
Emulsification ability
Water solubility.

Some of these factors are related. For example, the evaporation rate is dependent on weather conditions (especially wind) and the boiling range of the material. Similarly, the spread rate depends on weather, viscosity, and the pour point. Emulsification is a very complex parameter since both oil-in-water and water-in-oil emulsions can be involved and wind and wave conditions are usually controlling. The solubility of most of the materials very limited (below 0.01 g/100g). The significant physical parameters of greatest interest, along with typical values for residual fuel oils (used in the example later in this section) are given in Table 4-8.

Table 4-8. Characteristics of Typical Residual Fuel Oils used in Example

Parameter	Residual Fuel Oils
Specific Gravity (@ 60°F)	0.904 - 1.02
API Gravity (@ 60°F)	7 – 25
Viscosity (Saybolt Universal sec @ 100°F)	45 – 18,000
Flash Point (°F)	150 – 250
Pour Point (°F)Sulfur Content (% by weight)	0.5 or less

#### **Potential Oil Spills**

#### **Submarine Pipelines**

The design and installation of modern submarine pipeline facilities for marine terminals include a number of safety features to prevent oil leakage. In addition, extensive provisions are made to minimize the volume of oil released in the event there is a leak, including:

- Additional steel wall thickness on product transfer lines
- Cathodic protection
- Somastic coatings (or coal tar wrap)
- Concrete weight coating over somastic to provide negative buoyancy for empty lines and increased stability
- Burial of lines in surf zone
- Pressure safety valves
- Submarine hoses of strength several times the operating pressures.

Although these precautions are usually taken, there is still the possibility of damage to the submarine hoses by improper handling, or to the pipeline by man-caused events (dropped material, i.e., anchor or chain, of sufficient weight to cut lines) or natural occurrences. The curtailment of oil released to the sea is dependent upon the rapidity with which the ships or shore pumps are stopped, the vacuum pumps started, and the valves closed. The rate at which petroleum products or crude oil could be released would vary depending upon the extent of the pipeline incident. The magnitude of a spill could range from a few gallons (resulting from a minor leak in the pipeline system) to many barrels (resulting from a major pipeline fracture). The quantity released would also depend upon pipeline operating conditions at the time of the incident, *i.e.*, pumps on line or on standby. The potential spillage magnitude would also vary with the location of the pipeline incident. In submarine installations, the sea water (being of higher specific gravity than fuel oil) would seal off the oil in the sector of pipeline above (upslope) the leak. In the

sector of the line below (downslope) the leak, water would slowly enter the pipe, displacing the crude oil or product. Potential spills volumes are categorized by the National Oil Spill Contingency Plan as follows:

<u>Minor Spill</u> - a discharge of oil less than 10,000 gals (238 bbl\*) in offshore waters <u>Moderate Spill</u> - a discharge of oil of 10,000 to 100,000 gals (238 to 2,380 bbl) in offshore waters <u>Major Spill</u> - a discharge of oil of more than 100,000 gals (2,380 bbl) in offshore waters \*Based on 42 gal/bbl

Pipelines are by far the most common method of transporting crude oil and petroleum products in the United States. The possibility of a crude oil and/or petroleum product spillage could occur at any point along submarine pipelines. An analysis by the National Petroleum Council (1972) of spill incidents from pipeline systems in the United States indicates that approximately 2.8 bbl/mi/yr were lost.

#### **Tanker Operations**

Tankers can contribute to oil pollution of the marine environment through five principal sources:

- Cargo tank cleaning operations
- Discharges from bilge pumping
- Hull leakage
- Spills during cargo handling operations
- Vessel casualties

There are three principal reasons for the unintentional discharge of oil during tanker-terminal operations, namely: mechanical failures, design failures, or human error. Incident reports of spills during tanker-terminal operations show that human error is the pre dominant cause and is the most difficult to remedy. Mechanical failures include cargo transfer hose bursts, and piping, fittings, or flange failures, either on shore or on the tankers. Mechanical failure could also be due to an inherent design fault including the incompatibility of a tanker with a given marine terminal, i.e., improper manifold connections, inadequate mooring facilities, and shoreside loading pumps with excess pumping capacity.

Oil spills that occur during the loading or unloading of crude oil or petroleum products are more often associated with leaky connections, failure to drain cargo hoses, improper mooring, improper valve or manifold alignment, or overfill during loading operations.

## Prediction of the Movement of Oil Spills

The fate of an oil spill in the marine environment depends on the spreading motion of the oil and the translation of the slick by the winds and currents in the surface waters. Both of these mechanisms are well enough understood that oil spill movement predictions can be made, providing adequate input data are available. These required data include oil spreading equations, surface wind speed and direction, tidal currents, and knowledge of the general circulation of the waters of interest.

Fay (1971) developed a prediction equation for the spread of an oil slick considering gravity, inertia, viscous and surface tension forces. This analytical approach, coupled to experimentally determined constants, is considered in some detail by Premack and Brown (1973). From this historic research, it can be shown that simplified estimates of the spread of oil on water can be made using the following equations:

$$A_{\text{max}} = 1.65 \times 10^8 \times V^{3/4}$$
 Equation 10

$$r_{\text{max}} = 72.5 \times V^{3/8}$$
 Equation 11

$$t = \frac{34}{u^{2/3}} \times V^{1/2}$$
 Equation 12

where:

 $A_{\text{max}}$  = maximum area of spread (ft<sup>2</sup>)

 $r_{\text{max}}$  = maximum radius of a circular slick (ft) t = time to reach maximum radius (minutes)

V = spill volume (gallons)

u = spreading coefficient (dynes/cm) (11 dynes/cm for No. 6 fuel oil and 35

dynes/cm for waxy sweet crude)

Ichiye (see James, et al. 1972) and Murray (1972) also considered the impact of oceanic turbulent diffusive processes on the fate of an oil slick. Murray compared Fay's approach and turbulent diffusion theory to observations of slick growth from the Chevron spill of 1970 in the Gulf of Mexico. He concluded that eddy diffusion is a major driving force which cannot be neglected in oil slick growth. Ichiye developed a mathematical model for oil slick expansion and presented theoretical arguments and comparison of data with theory to support the need for applying turbulent forces in the equation for determining oil dispersion at sea. Ichiye also pointed out the significance of wind speed on the spreading rate of a slick. Ichiye's thorough treatment of the subject added a new dimension to oil slick prediction techniques and is considered in the example analysis that follows is this section. However, it should be pointed out that for discontinuous spills under light wind conditions, the two models are in agreement with each other during the time to maximum expansion, as defined by Fay. The consideration of eddy diffusion as a driving force becomes most important at later times and during moderate to high winds.

The transport of oil in an oceanic environment is dependent upon a number of variables. After spreading to its maximum radius, the translation of an oil slick in most near-shore waters will be dominated by wind forces and tidal currents. The direction of the oil slick movement, as influenced by the wind, can be taken as that of the wind as discussed by Murray (1970). The speed of the wind-driven component of the slick movement is generally taken as about 3 percent of the wind speed. Oil slick translation is thus calculated as the vector sum of the tidal currents and the wind stress on the slick. In addition to the translation of the surface slick, one must consider the possibility of the oil aging and mixing vertically with the water column. This requires knowledge of the properties of the oil in question. For examp le, crude oil in a slick can lose its volatile fraction by evaporation in a matter of hours causing a shift in oil density toward that of sea water. Movement of neutrally buoyant oil globules in deeper waters will be influenced by potentially complex and unknown subsurface circulation patterns.

Estimates of initial spill volume and a spreading equation are required to determine the spreading radius of a hypothetical spill as a function of time. Wind speed and direction, local tidal currents, and the general circulation along the coast are required to determine the trajectory of the slick, and estimates of the general circulation of the water body are needed to predict the fate of that fraction of the spill which may mix downward into the water column. The following discussion presents an example analysis of oil spill movement, based on typical offshore oil spill losses, and hypothetical environmental conditions.

#### **Spill Volume and Resulting Spill Dimensions**

In this example, the potential volume of oil that could be released to the environment as a result of a break in a submarine pipeline varies from a minimum of about 500 barrels to a maximum of about 10,000 barrels. A hypothetical oil spill of 500 tons (3750 bbl) is assumed in this example. This volume would be classified as a major spill.

Figures 4-14 and 4-15 describe the oil slick dimensions as a function of time for a 500 ton spill for various wind speeds. It should be noted that the predicted elliptical area defines the envelope in which the oil is found. At later times, and especially under high wind conditions, the slick will have broken up and some fraction will have evaporated and some fraction mixed with subsurface waters.

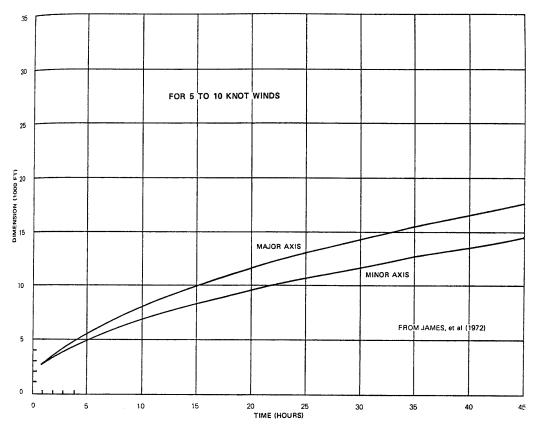


Figure 4-14. Growth of a 500 ton oil spill during five to ten knot winds.

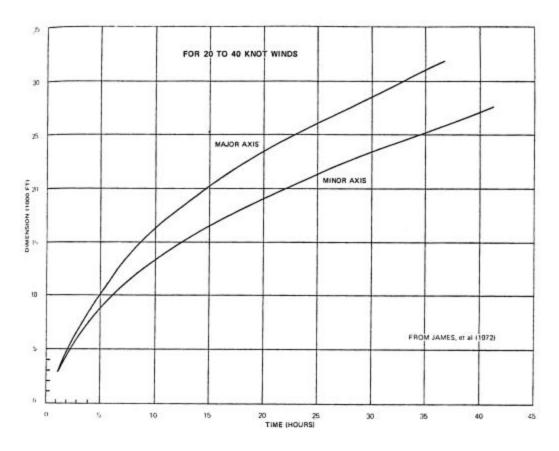


Figure 4-15. Growth of a 500 ton oil spill during twenty to forty knot winds.

#### Calculation of Oil Slick Movement Under Various Selected Wind and Current Conditions

The following example assumes an instantaneous oil spill of 500 tons that will grow radially according to the theory of Ichiye which was plotted in Figures 4-14 and 4-15. The slick movement was determined by the vector sum of tidal or coastal currents and wind-driven currents. Tidal currents had an assumed northerly current paralleling the shore during rising tides and southerly current paralleling the shore during falling tides; an average speed of 0.3 knots over a period of 4 hours for flood and ebb was used. No tidal component is applied during assumed 2 hour periods of slack tides. Wind-driven currents were assumed to have the same direction as the wind and a speed of 3 percent of the wind speed. Figures 4-16 through 4-18 are examples of the predicted fate of this spill occurring at a tanker berth as a result of a ruptured submarine pipeline or a tanker casualty.

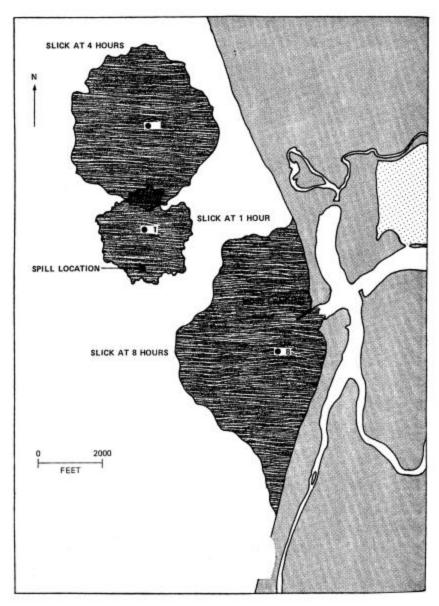


Figure 4-16. Predicted behavior of a 500 ton oil spill under the influence of a 5 knot NW wind and 0.3 knot tidal current (spill initiated at slack water before flooding tide).

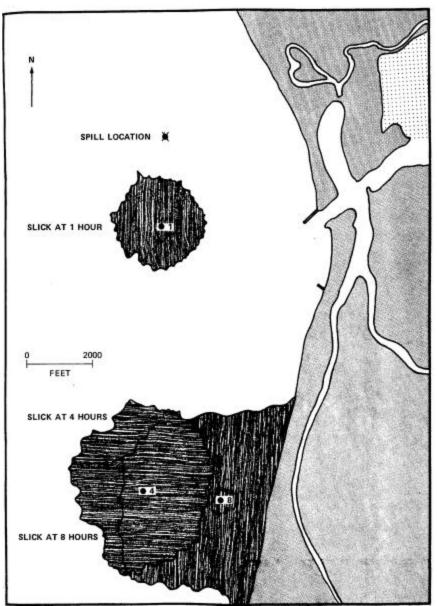


Figure 4-17. Predicted behavior of a 500 ton oil spill under the influence of a 5 knot NW wind and 0.3 knot tidal current (spill initiated at slack water before ebbing tide).

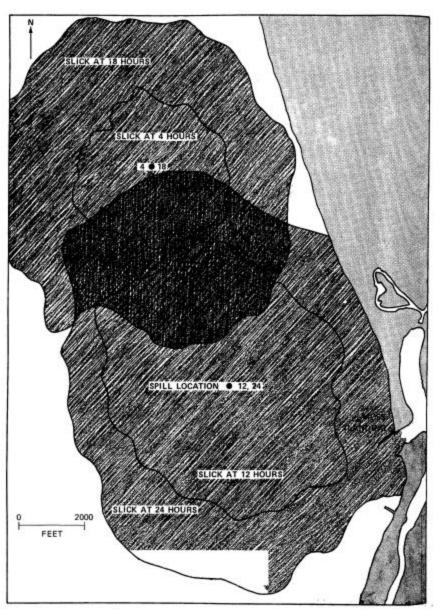


Figure 4-18. Predicted behavior of a 500 ton oil spill under calm winds and a 0.3 knot tidal current (spill initiated at slack water before flood tide).

## Analysis of the Environmental Impact of an Offshore Oil Spill Fate of Oil

The impact of an oil spill will depend upon the volume of spill, duration, type of petroleum product, and physical factors such as wind, wave, and current conditions under which the spill occurs. The fate of oil in an oil spill depends on a complex interaction between the several arbitrarily defined categories, as shown in Figure 4-19, plus a host of other less well-defined variables. Some of the lighter fractions of oil will evaporate very rapidly (evaporation), others are sensitive to sunlight and oxidize to innocuous or inert compounds (photo-oxidation), and still other fractions will either dissolve (dissolution), emulsify (emulsification), or adsorb to sediment particles (sedimentation), depending on their physical properties. The physical fate or dispersion of oil can occur by several methods: littoral deposition, physical removal, dissolution, flushing, elution, sedimentation, microbial oxidation, organic uptake. These are discussed in more detail below.

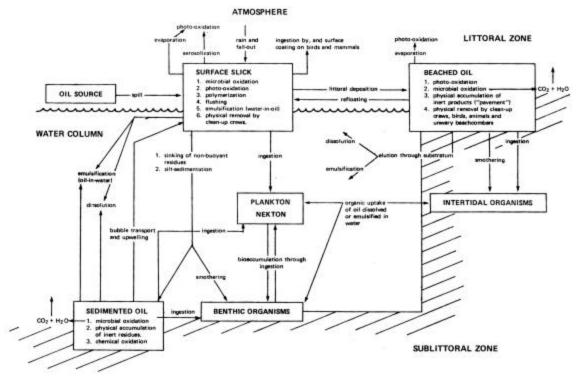


Figure 4-19. Fate of an oil spill in the marine environment.

In an oil spill the relative importance of each of the categories in the fate of an oil spill diagram (Figure 4-19) is influenced by several physical and chemical parameters and other events, including:

- Type of petroleum product (Bunker "C", diesel fuel, naphtha, and others)
- Volume of spill
- Distance from shore
- Sea and weather conditions (air and water temperature, wind direction and speed, wave height, etc.)
- Oceanographic conditions (currents, tide, salinity, etc.)
- Shoreline and bottom topography (sand or rock beaches, relief, degree of exposure to surf, etc.)
- Season of year, especially with reference to biological activities such as breeding, migration patterns, feeding habits, etc.
- Cleanup and restoration procedures.

The type of oil spilled will have a dramatic effect on the resulting effect of the spill. Bunker "C" fuel, for instance, although aesthetically unpleasant, is initially less destructive to marine life than is the more toxic diesel fuel. Oil from a spill occurring when oceanographic and/or meteorological conditions result in rough seas is likely to be more widely dispersed by emulsification, dissolution, wind drift, etc., through the water column and along the shore than one occurring in calm seas. However, the latter can be much more readily contained and/or picked up by mechanical devices such as booms, oil skimmers, and the like.

#### Composition of Petroleum

In order to consider the properties of oil in aqueous environments, it is necessary to know the composition of the oil. Crude oil and several heavy fuel oil fractions are a complex mixture of hydrocarbon and non-hydrocarbon molecules, encompassing a wide range of molecular weights.

Crude oils and most of their distillation products are extremely complex mixtures of organic chemicals with hydrocarbons being the most numerous and abundant and comprising more than 75 percent of most crude and fuel oils. Over 200 hydrocarbons, 90 sulfur-containing organic compounds, and 33 nitrogen-containing organic compounds are present in crude oils. In addition, there are porphyrins, sulfur, trace metals, and residues called asphaltenes in many crude oils. Crude oils and most crude oil products contain a series of n-alkanes with chain lengths of carbon atoms numbering between 1 and 60. The ratio of abundance of odd chain lengths to even chain lengths is approximately 1.0. A series of branched alkanes are also present including isoprenoid alkanes such as pristane, farnesane, and phytane, naphthenes (cyclic alkanes with or without side chains), aromatic hydrocarbons (from alkyl substituted benzenes and naphthalenes to polynuclear aromatic structures), and naphthenoaromatics (naphthenes joined with aromatic ring systems). Alkenes (olefins) are not usually present in crude oils but they are formed in some refining processes and are present in some refined products.

There are three properties of oil in sea water which are important with respect to the impacts of oil on the marine environment. They are: evaporation, emulsification and, to a much lesser degree, dissolution (solubility). Other properties such as density, boiling point, pour point, viscosity, etc., are less important or manifest themselves in the three prime properties listed. The lighter fraction of crude and heavy fuel oil and other volatile fractions (i.e., those of lower molecular weight) will evaporate to the air at a rate primarily dependent on vapor pressure of the oil. However, evaporation will be enhanced by high winds and rough sea conditions which favor formation of aerosols, and increased surface area; the faster and farther the oil spreads, the faster it evaporates. Cobet and Guard (1973) found that as much as much as 13 percent of the Bunker C fuel lost in the San Francisco Bay spill could have evaporated within 3 months and, depending on atmospheric conditions at the time, possibly even more would have evaporated. Fuel oil, lubricating oil, and similar components have few or no volatile components and thus will not evaporate. On the other hand, diesel fuel and other light "cutting" stock is comprised primarily of components which evaporate rapidly. In general, the more toxic fractions are those which evaporate fastest leaving a less toxic, more viscous, and more dense residue in the surface slick.

Oil-in-water and water-in-oil emulsifications do form and that considerable quantities of oil may be bound up in this manner. In general, the lighter fractions will go into an oil-in-water emulsification more easily than heavier fractions but vigorous agitation and/or solvent-emulsifier mixtures are usually required. As the hydrocarbon molecular weight increases, the emulsions become water-in-oil. These water-in-oil emulsions tend to form naturally and easily, especially with some wind and wave agitation, and they are quite stable.

For a given class of hydrocarbons, dissolution of solubility in water decreases with increasing molecular weight (carbon number). For the various classes of hydrocarbons, solubility increases in the following order: alkanes, cycloalkanes, olefins, and aromatics with corresponding solubilities being:

	mg hydrocarbon/liter of water
Alkanes	
ethane $(C_2)$	60
dodecane $(C_{12})$	0.003
Cycloalkanes	
cyclopentane $(C_5)$	156
dimethylhexane $(C_8)$	6
Olefins	
propene (C <sub>3</sub> )	200
1-octene (C <sub>8</sub> )	3
Aromatics	
benzene (C <sub>6</sub> )	1780
isopropylbenzene $(C_9)$	50

Sea water solubilities are approximately 70 percent of those cited for fresh water. Hydrocarbon solution in sea water is only temporary because dis solved hydrocarbons volatilize and evaporate rather rapidly. Because there is no discernible reservoir of hydrocarbons in the atmosphere, with the exception of methane, the equilibrium favors the

transfer of hydrocarbons from the liquid phase (sea water) to the gas phase (air), particularly under turbulent conditions of wind, current, and wave action. Even under the best conditions, relatively little oil is dispersed by dissolution when compared to the amounts dispersed by evaporation, emulsification and physical dispersion.

#### Effects of Oil on Marine Water Quality

The most obvious effect on water quality associated with an oil spill would be the physical presence of floating oil slicks which would deter boaters, bathers, divers, and others from using the affected area. Also, oil coming ashore would be aesthetically objectionable and would interfere with shoreline recreational activities such as picnicking, sunbathing, beachcombing, clam digging, and surf fishing. Depending on the specific oil material, dissolved hydrocarbons could also significantly increase, especially for a material containing large amounts of soluble components (as mentioned previously).

Observations by the U.S. Fish and Wildlife Service during the Santa Barbara oil spill showed small DO reductions even under thin slicks as compared with associated uncontaminated water. The largest decreases in DO were detected in the upper 30 meters under an oil slick. These reductions were insufficient to cause any significant biological damage in that the resultant oxygen levels generally remained above the level considered by the State Water Resources Control Board to be necessary for life (5.0 mg/L) and that the affected area was relatively small. Most observations of DO during oil spills have reported little effect of the spill on dissolved oxygen levels in sea water-petroleum mixtures.

Typical values of  $BOD_5$  of petroleum products in sea water generally range from 2.5 to 5.4 mg  $BOD_5$ /mg hydrocarbon. These  $BOD_5$  values can be high, but the biological activity is generally limited to surface waters where oxygen levels are maintained at high levels due to aeration and photosynthesis. The amount of oxygen required to completely oxidize one gal of crude oil can be calculated as representative of the entire oxygen content of 320,000 gal of typical sea water, assuming no replenishment from the atmosphere or photosynthetic activity. In general, the  $BOD_5$  requirement of oil products would be spread over several days and over a relatively large area, and concentrated in the upper layers of water.

Experimental data has shown that an oily odor is imparted to sea water at relatively low petroleum concentrations (0.05 to 1.0 mg/L). The odor persistence is very much a function of whether or not a slick persists. As the temperature increases, the rapidity with which the odor disappears increases. Odor persistence can range from 1 to 3 days in the absence of a slick, to 1 to 25 days with oil films. Following the *Torrey Canyon* spill, fish and shellfish were tainted by oil.

#### **Dispersion of Oil in the Marine Environment**

#### Physical Dispersion

Crude oil and refined products are physically dispersed to different parts of the marine environment by several mechanisms. The primary forces determining the fate of an oil slick are advective processes such as currents and the wind stress on the slick which determine its trajectory and diffusive processes which are important in determining the growth of the slick after the oil has stopped spreading by inertial and viscous forces (discussed above).

Low viscosity, high API gravity crude oils, and refined products generally break up and dissolve or emulsify in sea water. Individual oil droplets become attached to sediment particles, particularly in the intertidal-shallow sublittoral or surf zones, either by absorption or adherence, and disperse with these suspended particles. By this mechanism, oil becomes diluted and may finally become incorporated in sediments, animals; and plants. On the other hand, high viscosity, low API gravity crude oils and refined products such as Bunker "C" fuel behave like soft asphalt. When lower molecular weight hydrocarbons evaporate or dissolve, the remaining portion of these oils may become more dense than seawater and sink. This will be particularly true if they form water-in-oil emulsions which can also then pick up suspended silt particles and become heavier than water. The sunken oil may reside on the bottom in sediments as relatively inert material or it may undergo further chemical and biological degradation, converting the residues to lighter molecular weight materials which rise to the surface and repeat the original chain of reactions until most of the oil is consumed. Some of these lighter fractions may also dissolve or emulsify on the way back to the surface. These dense oils can form water-in-oil emulsions which may sink or later be cast up on the beach.

With typical on-shore winds and currents, those fractions of oil, especially crude and fuel oil, which are not weathered or lost to evaporation, emulsification, dissolution, sedimentation, or organic uptake while on the water surface or in the water column, are deposited in the littoral or intertidal zone (littoral deposition) by waves and/or receding tides. Diesel fuel and other light fractions evaporate rapidly from rocky beaches, but may penetrate several inches into sand beaches and remain to work their way back to the surface over a long period of time, or work their way through the sand to come out in the shallow sublittoral zone (elution). Crude oil and other heavy fractions are deposited on the beaches in the form of "asphalt" or tar. On rock beaches, this asphalt coats the rocks, weathers, and becomes a semi-permanent substratum. On sand beaches, the asphalt may mix with and become buried under several inches of sand to form a subsurface "pavement" layer. This situation was observed in both the Torrey Canon and Santa Barbara spills. In both cases the "pavement" layer is exposed and covered several times during winter months.

#### **Biological Dispersion**

Hydrocarbons are not foreign to the marine environment; they are synthesized by most, if not all, living organisms. The conditions under which microbial attack occurs and the rate of biodegradation are a function of such diverse factors as the type and number of bacteria in the given marine environment, the quantity and type of oil spilled, the spill concentration, water temperature, salinity, oxygen concentration, nutrients, and pH. Some reported values for marine biodegradation of oils vary from 35 to 55 percent of oxidizable crude oil within 60 hr to between 26 and 98 percent of oil degraded by mixed cultures within 30 days at 77°F.

Early studies have found the presence of oil-oxidizing bacteria in abundance in coastal waters and muds near natural oil seeps. As an example, along the California coast, oil-oxidizing bacteria range from nil to greater than 10 per milliliter of mud, with the largest populations being found in San Pedro Bay and Long Beach Harbor. Microbial degradation appears to be most efficient in removing relatively low concentrations of oil such as thin films. However, oil oxidizing bacteria are sensitive to toxic constituents of oils such as toluene and xylene, as well as phenol and small quantities of nitrogenous, oxygenated, and/or organic sulfur compounds. Therefore, the concentration and composition of oil in a given area affects both the biodegradability and the rate of microbial activity.

Many oleophilic microbes become nutrient limited in that they use up all of the nitrogen or phosphorus or both, which are essential for maintaining life and growth. Both sea water and petroleum have low concentrations of nitrates and phosphates. Once the nitrates and phosphates are depleted, or at least reach very low levels, the microbe populations will be reduced in species diversity and abundance even though a considerable quantity of oil remains. Recent oil spill cleanup activities have therefore included adding substantial amounts of nutrients to affected areas to encourage natural microbial oxidation of residual oils.

#### **Effects of Oil on Marine Ecosystems**

The effect of petroleum products ranging from gasoline to crude oil on one or more components of marine ecosystems has been the topic of numerous symposia, scientific papers, formal and informal lectures, and newspaper articles. Ecological effects are presently receiving close attention by industrial and academic groups under the auspices of the American Petroleum Institute (API), Environmental Protection Agency (EPA), and other industrial, private, state, and Federal agencies. A review of the literature and interviews with these several sources indicate that three kinds of effects and the resultant biotic responses exist. These effects are arbitrarily divided into three categories:

FIRST ORDER EFFECTS include the direct effect of petroleum products on the biota. These effects may be: toxic, physical (such as suffocation), or physiological (such as internal disturbances following ingestion). All of these may result in immediate mortality, torpidity, or poor health. These are generally short-term effects which usually affect all species to some degree and show up within hours or days.

SECOND ORDER EFFECTS include changes in populations of each species with respect to size-frequency and age structure, productivity, standing crop, reproductive abilities, etc. These are generally intermediate-term effects which show up in weeks, months, and for some long-lived species, years.

THIRD ORDER EFFECTS include changes at the community or ecosystem level with respect to relationships within or between trophic levels, species composition and/or abundance, and other aspects of community dynamics. These changes are often the result of subtle, sub-lethal effects which may not show up for months or years.

First order effects have been documented in some detail in several instances. Second and third order effects are generally less well documented, except for a few large spills such as *Torrey Canyon*, *Tampico Maru*, West Falmouth, and Santa Barbara, and even in these cases, the data interpretation may be open to criticism.

Clearly, there are significant impacts on the marine environment from most oil spills. This impact may vary from an aesthetic problem of several days' duration resulting from visible oil slicks and beaches contaminated with oil, to a severe kill of marine organisms and water fowl, and severe disruption of commercial and recreational activities. Long-term effects might range up to several years before ecosystem recovery. The spill may even bring about a permanent change in the ecosystem as evidenced by new and different species of flora and fauna becoming dominant in terms of space or ecological importance.

The severity of both short-term and long-term effects is predicated on certain conditions. The following generally increase the severity of an oil spill:

- 1. A massive oil spill relative to the size of the receiving and affected area
- 2. A spill of primarily refined oil
- 3. The spill being confined naturally or artificially to a limited area of relatively shallow water for a prolonged period
- 4. The presence of sea bird and/or mammal rookeries in the affected area
- 5. The absence of oil-oxidizing bacteria in the marine environment
- 6. The presence of other pollutants, such as industrial and municipal wastes in the affected area
- 7. The application of detergents and/or dispersants as part of the cleaning action.

#### Biological Effects of Recorded Spills

The general aspects of some recent major oil spills are presented in Table 4-9. Of these spills, only four have shown extensive kill of much of the areas' marine life. Three of these, West Falmouth, the *Tampico Maru* incident off Baja California, and the Wake Island spill shared the common factor of a large amount of product being discharged to a small, partially enclosed body of water. The *Torrey Canyon* spill occurred in open waters. In most other spill studies, organism kill was most common in the intertidal zone. A brief description of several historical major older spills follows:

Table 4-9. Summary of Recorded Historical Major Oil Spills

Spill	Date	Quantity Spilled (1000 gal)	Product Tvpe	Detergents Used in Cleanup	Time to Recovery (General Estimate)
Louisiana	1956		Crude	No	several months
Tampico Maru	1957	2,500	Diesel fuel (#2	fuel oil) No	1 - 10 years
Fawley, England	1960	52	Fuel Oil	Yes	> 2 years
Torrey Canyon	1967	29,400	Crude	Yes	> 2 years
Milford Haven	1968	70 - 150	Crude	Yes	Several months
Santa Barbara	1969	4,200	Crude	Yes	Several months
West Falmouth	1969	175	Diesel fuel (#2 f	uel oil) No	< 2 years
Tampa Bay	1970	10	Bunker "C"	Yes	Days to weeks
Nova Scotia	1970	3,800	Bunker "C"	No	Months to years
Platform Charlie, I	LA 1970	42 <sup>a</sup>	Crude	Yes	Days
Wake Island	1970	6,000	Bunker "C"b		<del></del>
San Francisco	1971	840	Bunker "C"	No	10 months +

<sup>&</sup>lt;sup>a</sup>Daily discharge estimated to be 42,000 gal for a three-week period.

<sup>&</sup>lt;sup>b</sup>Also included aviation gasoline and jet fuel, aviation turbine fuel and diesel oil.

Louisiana Spill. On November 17, 1956, an oil well caught fire. and spilled oil for a period of about two weeks into the marshes of Louisiana. Although the original slick covered over 50 square miles, by December the oil had disappeared from the surface except for a light film within Barataria Bay. There was still considerable oil along the shoreline of the Freeport Sulfur Canal. However, as late as February 5, oil could still be stirred from the bottom of areas such as Billet Bay, indicating that considerable oil still covered the bottom. There was no way to determine how much oil escaped from the well. All light fractions must have burned when the well was on fire, and much more evaporated. Thus, most of the lost oil was artificially "weathered" except in the short period of several hours after the fire was extinguished and during which the oil flowed unhindered.

Examination of the impact of the spilled oil on oysters was of prime concern. Data from polluted and nonpolluted areas clearly showed that contact with oil for an extended period had no effect as far as the survival and growth of oysters was concerned. Mortalities of oysters in the area were primarily associated with the incidence of infection of a fungus disease typical of Louisiana and were not related to the distance from the well. Oily taste in the oyster meats could not be identified after two months.

A cursory examination of the organisms associated with oyster reefs showed that control and experimental stations did not differ significantly. Normal reproduction and growth of populations took place during the entire period of study. The oysters themselves spawned normally, and heavy sets of young oysters occurred at some experimental stations. Normal reproduction and growth of populations took place during the entire period of study. The oysters themselves spawned normally, and heavy sets of young oysters occurred at some experimental stations. These young oysters grew rapidly with relatively low mortality, while at the same time large numbers of older oysters died of an epidemic disease probably unrelated to the spill. Growth of the surviving oysters was excellent, as was their condition. Thus, survival, reproduction, growth, and size of oyster meats were not affected by the oil.

<u>Tampico Maru Spill.</u> During the spring of 1957, the oil tanker <u>Tampico Maru</u> went aground off the coast of Baja California. The ship formed a breakwater across a small cove while 60,000 bbl of diesel fuel began leaking from its hull. Damage to the benthic fauna and flora of the cove was extensive, and the shore was littered with dead and dying animals. A month after the accident, a thick viscous sludge of water, oil, and small particles covered most of the bottom of the cove and the tide pools. The sea plants did not seem to be as seriously damaged as the animals. Many plants remained attached and living, although some deterioration was noted. Few animal species survived. Among those that did were the small gastropod, *Littorina planaxis*, and large green anemones, *Anthopleura xanthogrammica*.

By summer, three months after the spill, the cove began to appear fresh and clean; eight months after, no oil was observed, though small quantities may have persisted. Motile animals, such as large fish, sea lions, and lobsters were seen. Smaller organisms, such as bryozoans, began to colonize the barren zones. By far the greatest change was the appearance of a dense and luxurious growth of seaweed.

The No. 2 fuel oil was confined to a small cove by the position of the tanker which in turn reduced the oxygenation of the waters from the breaking waves, resulting in a massive kill among both the fauna and flora. Oil was the primary factor causing the destruction of the organisms. Seaweeds appeared to be more tolerant than the animals. Most of the plant species re-established themselves within a few months, but the animal species reappeared more gradually over a period of 7 years. Seven years afterward, the populations of certain organisms such as grazing sea urchins, abalones, and filter-feeding mussels, were still considerably reduced, and some species present before the shipwreck have not been seen since. Several organisms were observed which are believed to be very tolerant of oil pollution.

<u>Fawley (England) Spill.</u> Effects of this 1960 spill of fuel oil were seen on common intertidal organisms, such as the polychaete worms *Cirriforma tentaculata* and *Cirratulus cirratus*, but it was not certain that fuel oil alone was responsible for mortality. Where oil dispersants were employed, studies indicated a sharp decline in numbers of adults. Two years after the spill, the numbers of adults of *Cirriforma tentaculata* had still not recovered.

Torrey Canyon Spill

The biological effects of the *Torrey Canyon* spill can be divided into two main categories: (1) those caused by, or directly related to, the crude oil itself and (2) those related to the cleanup procedures, especially the application of detergents. It was recognized from the onset of the *Torrey Canon* operations that oil, although it killed several thousand sea birds, was a pollutant mainly destructive to the amenities of shores and beaches, whereas detergents, on the other hand, were known to be destructive to life. Assessment of the biologic damage and recovery in the affected areas was examined in regard to either the presence of crude oil or the presence of crude oil in combination with detergents. Phytoplankton surveys of the channel areas, when compared with past surveys, contained samples having plant populations of the type normally found in a channel in early spring. Both diatoms and dinoflagellates appeared to be healthy at all stations. The overall result of later surveys showed that there were deaths among the smallest flagellates, often after a period of only a few days, in all samples taken from areas of thin or thick oil cover, whereas there were no deaths at stations in uncontaminated water. This indicated that these small flagellates were sensitive to very low concentrations of toxic substances.

Other phytoplankton, such as diatoms and dinoflagellates, appeared to be little affected. Further, most of the colorless dinoflagellates were unaffected, and some of those studied in laboratory cultures grew better in oily sea water than in uncontaminated water. Zooplankton, mainly copepod crustaceans, appeared to be of normal abundance, and all seemed healthy when examined immediately after they were captured. Fish also appeared to be healthy. Some oil was found by divers and fishermen on the sea floor, but there were no external signs of oil contamination on the fish and only a few visible traces of oil within the gut.

Along the rocky shore, heavy oil alone rarely seemed to have any ill effects during the first few days. In some cases, such as Cape Cornwall, moribund limpets were observed under the oil, and it is possible that they had been smothered by thick coatings of oil, or that the oil which enveloped them contained the detergent sprayed at sea. The survival of mussels under heavy oil was seen at Booby's Bay in the first few days of pollution. In the absence of heavy detergent treatments, these mussels had survived. Furthermore, at Portreath mussels were found alive and behaving normally in pools which had a film of oil.

In the Hayle Estuary, oil contamination occurred on March 28 - 29. No detergents were used within the estuary. When examined on April 10, the rich worm fauna of the sandy flats seemed unharmed. Although the black oily rim was still visible on the vertical walls around the estuary and harbor in mid-August, weathering had reduced it considerably. In places, an orange lichen *Xanthoria* was growing through the oil. Perennial salt mash plants and grasses had grown through the oily layer and were spreading over the oil residue. The normal drift-line fauna of small amphipods and wood lice were common under stones. These are good examples of recovery by natural means in the absence of the use of any detergent.

Milford Haven Spill. Crude oil was spilled in Milford haven along the shore at Hazel Beach on November 1, 1968. No evidence of biological damage was observed before cleaning operations commenced, although the rock area was covered with a thick black film of crude oil. Molluscs were attached to rocks and were apparently healthy. Following these observations, the shore was washed twice with an emulsifier applied with a water jet. The most obvious change was the growth of seaweeds in the mid-shore during March, July, and August. By late September, these plants were about 6 in. long, forming a patchy cover on the shore. Following cleaning, the gastropods showed considerable decrease in numbers 3 weeks after the spillage, but when the next survey was made on January 23, the population had largely recovered its previous abundance. In Milford Haven, it is difficult to distinguish between the effects of small, chronic spills and large, rare spills.

Santa Barbara Spill. Oil released from the offshore oil well in the Santa Barbara Channel eventually affected most of the mainland beaches in the channel and some areas of the Channel Islands. Slicks initially covered large areas of the channel and tended to accumulate on the beaches in the upper littoral zone. Phytoplankton studies in the Santa Barbara Channel showed no conclusive evidence of any major effect which could be directly attributed to the spilled oil. These studies were based on 11 stations which were resampled 12 times from 1969 to 1970. The data showed higher productivity occurring inshore, seasonal variations in productivity, and the presence of a phytoplankton bloom in August 1969. No low productivity values resulting from the presence of oil on the surface of the water were found. There was a reduction in the reproduction in *Pollicipes polymerus*, a barnacle. The breeding in *Mytilus californianus*, a mussel, was probably reduced as a result of oil pollution.

The major damage to the marine invertebrates following the Santa Barbara spill resulted principally from the oil-removal operations along the mainland shore. The steam cleaning of rocks to remove the oil killed all sessile invertebrates that were attached to them. Further, cleaning the beaches with skip loaders to remove the oily straw and debris undoubtedly took its toll on some of the invertebrates inhabiting those beaches.

No permanent damage to marine plants was observed by California Department of Fish and Game divers during repeated surveys in 1969. On Santa Cruz Island, the algae *Hespherophycus harveyanus*, originally heavily coated by oil in February, was clean by August. In addition, numerous young plants were found to be present. The surf grass *Phyllospadix torreyi* was heavily coated by oil and suffered high mortalities but the beds had come back. Most of the other plants and algae surveyed on the islands and the mainland appeared relatively unaffected by the oil pollution.

California Department of Fish and Game trawls obtained 14,070 fishes representing 59 species, but they failed to show damage directly related to oil pollution or starvation. U.S. Bureau of Commercial Fisheries personnel found no gross evidence of dead or deformed larvae of fish eggs nor gross changes in the composition of the ichthyoplankton in the channel during February 1969.

West Falmouth Oil Spill. The West Falmouth oil spill of September 16, 1969, involving No. 2 diesel fuel, has been investigated by scientists at the Woods Hole Oceanographic Institute. The controversial studies indicated that a massive kill of benthic invertebrates occurred even before the application of detergents. In addition, wherever fuel oil was detected in the sediments, there was a reported kill; and in areas containing the most oil, the kill was almost complete. The reports state that the kill was caused directly or indirectly by the fuel oil. Affected areas are said to have not been repopulated 9 months after the spill, and marshes are being eroded because of decreased stability following the kill. Up to two years after the spill, fuel oil is still detectable in the sediments.

Nova Scotia Spill. Five months (i.e., July, 1970) after the destruction of the oil tanker S.S. Arrow, carrying Bunker C fuel oil, the marine fauna and flora below the tide levels were healthy, and fishing and lobstering were normal. Background levels of hydrocarbons from the spill had decreased significantly by January 1971. As expected, the intertidal zone was the most severely affected, but only where oiling was exceptionally heavy. An estimated 25 percent of the clams (Mya arenaria) were killed in the early part of the season. Algae, primarily Fucus spiralis, was oiled and became more easily torn loose in storms. Other species appear to have been little affected. Salt marsh cord grass (Spartina alterniflora) suffered high mortality. The lobster season had gotten underway on schedule in early May and the lobsters were in hibernation when the oil was spilled, helping to protect them. Other subtidal organisms appear not to have suffered. Zooplankton in early March were normal, and copepods were observed with oil in their digestive tracts, which generally passed through unaltered and without harm to the animal. Local fisheries were found to be unaffected in the following season.

Gulf Coast Spill. On February 10, 1970, a blowout fire occurred on offshore Platform 2 in Main Pass Block 41 field, 11 miles east of the Mississippi River Delta. The fire burned until March 10 when it was extinguished by explosives. Over the next three-week period, crude oil estimated at 1000 bbl/day escaped before the last well was capped. Oil came onshore only briefly at offshore Breton Island. Investigations revealed no apparent damage to marine organisms. The benthic community consisted of large numbers of species and individuals and showed no measurable effect from the discharged oil. Numerous samples showed large numbers of species of fish and normal size and numbers of shrimp. The shrimp data indicated a normal reproductive cycle, with no effect of oil on reproduction and juvenile stages. The normal attachment of oysters just following the spill further indicated no effect of oil on oyster reproduction or ,juvenile stages.

<u>Wake Island Spill.</u> The Wake Island spill resulted in an estimated kill of 2500 kg of inshore reef fishes plus an unknown number of invertebrates and other fish. There was no evidence of damage to sea birds.

<u>San Francisco Spill.</u> The discharge of 20,000 bbl of Bunker C oil near the Golden Gate Bridge in San Francisco Bay in January 1971 caused extensive coverage of the intertidal zones within portions of the bay and seaward as far north as Bolinas and to a lesser extent south of Half Moon Bay.

An investigation on the effect of the spill on Duxbury Reef, a marine reserve, indicated that heavy oil deposits on the reef area caused kills by smothering certain species such as acorn barnacles and limpets. The same effects were noted at Sausalito. Marine snails suffered less mortality than did the sessile barnacles and other sedentary animals. The normally large population of striped shore crabs (*Pachygrapsus crassipes*) was missing from the rocky crevices. The condition of Duxbury Reef in December 1971 was one of apparent good health; the recruitment of some marine animals appeared to be approaching normal levels and the oil had disappeared from much of the reef surfaces and was barely discernible in the most heavily deluged areas.

#### Summary of Documented Spills

The following is a summary of the effects of the historical oil spills, based on field investigations. The results of the different studies often have quite varied conclusions (likely due to a combination of factors including spill characteristics and material characteristics, and environmental conditions, plus differences in the experimental designs and sampling procedures), but the following is a list of generally accepted conclusions concerning the effects of oil spills:

- 1) The principal damage from oil spills is to birds. The literature is remarkably unanimous on this point. The data are conclusive and can be taken without reservation. While no bird damage has resulted from some spills, it is believed that this resulted from accidental circumstances, and the danger to birds is present wherever a spill occurs.
- 2) The effects in the intertidal zones, beaches, marshes, and rocky shores are sometimes of significant severity. The intertidal zone is subject to heavy concentrations of oil, and damage may be expected if concentrations reach a critical level. Usually the damage to biotic communities from the oil itself is quite small even when heavy concentrations reach the shore. Humans are among the most affected when beaches are made uninhabitable.
- 3) Little documentation has shown any significant damage to marine bottom communities in deep or shallow water. There appears to be an intermediate zone between the intertidal area and "deep" water in which some relatively small damage occurs under adverse circumstances (such as heavy wave action in surf zones).
- 4) Damage to fisheries appears to be confined to those cases where animals (such as the mussel *Mytilus*, oysters, or clams) live in intertidal zones. Any fishery animal can become tainted with oily taste and smell. Considerable losses to the industry may occur when such contamination affects any significant part of the populations.
- 5) Recovery from damage caused by oil spills is usually rapid and complete so far as the marine communities are concerned, and in some cases these communities may be stimulated to higher productivity by the process.
- 6) No significant damage to plankton has been observed in oil spills.

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## Section 5. Community Impacts of Major Transportation Accidents Involving Hazardous Materials

Major transportation accidents involving hazardous materials can produce profound economic, social and psychological impacts in affected communities. These impacts can be both widespread and long lasting. In Section 5, the community impacts of hazardous materials transportation accidents are discussed. As in previous parts of the report, the section begins with a brief illustrative case study. The case study examines a June 1999 pipeline explosion in Bellingham, Washington that killed a man and two children and had a profound effect on the community. Following the discussion of the Bellingham case, the section continues with a more general discussion of the economic, social and psychological effects of hazardous transportation accidents. Here, current scientific research is reviewed, examples are provided, and implications are considered.

## Case Study: Pipeline Explosion, Bellingham, Washington, June 10, 1999

Olympic Pipe Line Company owns and operates a 400-mile system of pipes that carry gasoline, diesel and aviation fuel from several refineries to users in the Puget Sound area of Washington State. This series of pipelines, some sections of which are 35 years old, supplies all the aviation fuel used at the Seattle-Tacoma International Airport. The pipe that ruptured was a 16-inch flexible, high-strength steel pipe. It was designed to withstand loads of soil, rail and car traffic on the outside of the pipe, and the pressure of the fuels flowing within. Normal operating pressures for this pipe were between 1000 and 1400 psi. In the area of the rupture/leak, the pipe was buried eight feet underground.

On June 10, 1999, at 3:18 p.m., Olympic Pipe Line operators at the Renton, WA, control room began switching the operation to supply fuel to a new customer. They had difficulty starting one of the pumps, and the computers that control a series of valves and pumps began malfunctioning. At 3:24 pm, one of the computers crashed. At 3:28 p.m., the backup computer system started up at the same time that a valve in the line closed. The quick closing of the valve caused a pressure surge of up to seven times the normal operating pressure to go back up the pipe. According to initial reports, due to the extreme pressure, a 27-inch gash occurred at a weakened spot in the line. (Later reports in the *Bellingham Herald* on October 2, 1999 stated that a simulation of the line indicated that the pressure in the line at the time of the rupture may not have been above normal operating pressures).

The rupture occurred under Whatcom Creek, near the water treatment plant. The computer malfunction also caused the pumps at the start of the pipeline to shut off, thus preventing fuel from continuing to enter the pipeline. Operators were unaware of the break and so at 3:46 p.m., they restarted the pumps, sending fuel into the broken line. At 4:29 p.m., a leak alarm sounded in the control room. In the meantime, Bellingham residents, starting at 4:24 p.m., called the fire department to report the strong odor of gasoline outside. At 4:31 p.m., the operators started another pump, sending even more fuel into the line. At 4:32 p.m., the pumps shut down automatically, another alarm sounded, and operators began closing off the pipe (*The Seattle Times*, June 11, 1999, June 3, 2000b). At 5:02 p.m., the massive fire is reported (*The Seattle Times*, June 12, 1999, June 24, 1999, June 3, 2000b).

Shortly before the explosion, the Bellingham Fire Department began responding to the calls regarding the strong gas odor. When they approached the park, the firemen saw the fumes rising from the creek. According to firefighter Ryan Provencher, "the creek had turned yellow, a 'river of gasoline'" (*The Seattle Times*, June 13, 1999a). The firefighters immediately began closing off the streets and evacuating the surrounding area. Neighbors also began to alert others. When the gasoline exploded, the fireball reached 30,000 feet into the air and "the fire raced half a mile down the creek until it ran out of fuel." The hottest part of the fire burned itself out in an hour but hotspots remained for another 48 hours. According to Whatcom County's fire chief Gary Crawford, "You can tell how hot it got. It

singed the hills behind it. We had some 2,000-degree heat" (*The Seattle Times*, June 11, 1999). Bellingham's Fire Captain Bill Boyd said the day after the incident, "It was ugly. I've never seen anything like it. It was like Mount St. Helen's" (*Bellingham Herald*, June 11, 1999).

The initial investigation reported that the leak occurred within a mile of where a 1996 test discovered the pipeline wall was thinner than normal but within specification (called by Olympic Pipe Line a 'sub-critical' abnormality). The cause of the pipe weakening was reported to be due to external damage from construction (*The Seattle Times*, June 11, 1999, June 24, 1999, July 1, 1999). According to the National Transportation Safety Board (NTSB) review, the rupture occurred on the pipeline at a location where water lines (as part of an improvement project at the water treatment plant) were installed above and below the pipeline in 1994 and 1995. In 1996, approximately two years since the construction, Olympic Pipe Line had inspected the line using electronic devices ("smart pigs that test the wall thickness) and found anomalies. Based on a review of the data from the "smart pigs," Olympic determined that the anomalies did not warrant additional investigation, which would require excavating the pipe (*The Seattle Times*, October 27, 1999).

Three people were killed as a result of the fire and explosion. Two ten-year old boys, Wade King and Stephen Tsiorvas, were playing along the creek with a plastic fire-starter and ignited the gasoline in the creek. They were burned over 90% of their bodies and died the next morning at the hospital as a result of their injuries. An eighteen-year old fisherman, Liam Wood, suffocated from the gasoline fumes (*The Seattle Times*, June 11, 1999, June 13, 1999b, June 24, 1999).

## Impacts of the Bellingham Pipeline Explosion

The immediate impact was on the families of the boys that were killed. "I held his feet, because those were the only things that were really him any more... I don't know if he heard me tell him how much I loved him." Katherine Dalen was speaking of her son Stephen Tsiorvas. "You worry about cuts and insect stings. You don't worry about the water burning them to death" (*The Seattle Times*, July 28, 1999). Firefighters called Wade King and Stephen Tsiorvas "unwitting heroes," for if the blast had not happened where it did and if the gasoline had traveled further downstream, the loss of life and property would have been "significantly greater." According to one Bellingham firefighter, the fire department found "highly explosive bubbles of gasoline fumes in the sewer system that could have blown up the city's entire sewer system" (*The Seattle Times*, June 13, 1999b).

In the days following the explosion, the community impacts became apparent. City leaders called the accident "the most devastating thing we've ever had happen to this community. This has shaken the community's sense of security to the core" (*The Seattle Times*, June 17, 1999). In an attempt to control public curiosity about the explosion site and fire, the city of Bellingham arranged public tours of the area on the Saturday following the explosion (*The Seattle Times*, July 1, 1999). Reaction among the evacuees to the initial emergency response to the incident was mixed. Evacuation notification was called 'haphazard,' and residents accused officials of taking "an hour to broadcast a warning on the emergency broadcast system. People were left wondering whether their health was threatened by the thick cloud of black smoke" (*The Seattle Times*, June 13, 1999b). Residents have talked among themselves about 'getting back to normal,' but normal was different. Before the disaster, few residents even knew about the pipeline, but now they knew where it was located (a hundred yards from the middle school) and what was in it (*The Seattle Times*, June 13, 1999a, June 17, 1999).

The families of the two ten-year old boys killed in the blast filed lawsuits against Olympic Pipe Line, and against one of its partners, Equilon, for both compensatory and punitive damages for the loss of their children as well as for the pain and suffering. This experience was especially traumatic because the two boys did not die immediately in the blaze, instead they were found and rescued by an older brother of one of the two boys. The family of Liam Woods, the fisherman who drowned when overcome by the fumes, has not filed suit against the companies (*The Seattle Times*, July 28, 1999, September 25, 1999). This accident has also resulted in a federal criminal investigation relating to whether "Olympic met its requirement to closely monitor the construction work [by the City of Bellingham], given that such activity is the leading cause of pipeline ruptures. Also under examination is the company's decision not to inspect the anomaly firsthand after remote sensors discovered it" (*The Seattle Times*, December 9, 1999).

Since the accident, the civil and potential criminal investigations have often conflicted, and these conflicts have delayed a sense of closure for the families. Because of the potential criminal case, several Olympic Pipe Line employees invoked their Fifth Amendment rights during questioning about the accident during the hearings on the civil case. Other delays in the civil case have included the delay of destructive testing of the 20-foot segment of ruptured pipe because of the potential for compromising the criminal defense. In order to not incriminate himself in a criminal case, the president of Olympic Pipe Line requested a one-year delay, to December 2000, in responding to the families' civil lawsuit because of a potential federal criminal investigation. Other Olympic employees have also requested delays in responding to attorneys' questions, and immunity from criminal prosecution has been proposed for some employees who were on duty the day of the explosion (*The Seattle Times*, December 4, 1999).

Olympic accused and later sued a local construction firm who installed the water lines near the pipeline. They accused the firm of fatally damaging the pipeline and failing to notify Olympic of the damage when it occurred. This has led to the local newspaper airing the accusations between the two companies. The construction firm said that they did not damage the pipe and that the faulty valve and resulting pressure wave caused the rupture. Olympic contends that the pipeline would not have ruptured had the pipe been intact/undamaged. When questioned about their availability during the construction in 1994 and 1995, the Olympic spokesperson said that a company representative was on-site during the work, but that they were not present when the damage occurred or when it was covered up. However, according to the president of the construction firm, "They [Olympic] are clearly liable under the law. They are a large corporation, and I can't believe they are blaming their negligence on us and trying to ruin our reputation" (*The Seattle Times*, February 11, 2000).

Residents near the pipeline have also been affected. One resident commented several days after the explosion, "The park was a quiet sanctuary for residents across the region, including her own family. But innocent sounds now jar her emotionally. 'Whenever I hear a jet go over, it's like thunder and feels like the explosions. My nerves are rattled. Some nights I've woken up and it smells like smoke. It's definitely on my mind a lot." Another person, whose home is near the pipeline, but not near the area where the pipe ruptured, said that "now he wonders just how old the pipeline is and whether the earth piled on top of the pipeline from new construction projects ... could become a problem" (*Bellingham Herald*, June 16, 1999). According to Dr. Frank James of Bellingham, he has treated "a Vietnam veteran who believed his home had been napalmed, a young child whose sleep is still disturbed by the vision of a huge black cloud, and a boy who found the body of Liam Wood, the 18-year-old fisherman." As Dr. James said at a public meeting of the state's pipeline safety task force (formed after the accident), "They will not be the same again. It comes as a shock to me how much suffering remains in this community because of this." At the same hearing, Wade King's father said "residents must maintain a 'controlled, reasonable, logical anger' to prevent a recurrence." However, not all residents were as greatly affected as those seen by Dr. James. One resident defended the pipeline with the following statement "When you take the amount of years (the pipeline) has been going through this area, it's been quite well taken care of" (*The Seattle Times*, November 17, 1999).

Several residents along the pipeline found that their houses were now valued at less than they were before the accident. One man seeking a loan for improvements to his home found the value of that loan lowered by half. Another family watched as their house sold for \$8,500 less than expected. Area real estate agents were waiting for the year 2000 tax assessments to determine the extent of the lowered housing values. "Under state and federal law, appraisers must note 'adverse environmental conditions present in the improvements on the site or in the immediate vicinity of the subject property." As a result of this disaster, pipelines may become one of those immediate-vicinity conditions (*The Seattle Times*, September 19, 1999a).

Local utilities were also affected by the explosion. The local water pumping station was destroyed, forcing up to 70,000 system users to heavily restrict their water usage. According to the assistant director of the Bellingham Public Works Department, "For all practical purposes, the pump station was destroyed. The concrete shell was salvageable. All the control systems melted. The fire extinguisher melted" (*Bellingham Herald*, June 11, 1999). For at least a week, 15,000 to 20,000 people had water to cook and drink, but not to bathe or wash clothes. Power lines were also singed, disrupting power to thousands of area residents since the power lines were shut down for protection. The resultant smoke also closed Interstate 5 to traffic for several hours on the evening of the accident (*The Seattle Times*, June 11, 1999).

In addition to the human costs of the disaster, the explosion killed more than 30,000 fish in Whatcom Creek (*The Seattle Times*, June 17, 1999). "As the fire burned and the water temperature rose, the oxygen was sucked out of the water. Some of the fish tried to dive, some hid in the rocks, and those who tried to get to air on the surface were burned to a crisp (*The Seattle Times*, June 13, 1999a). The creek had been the focus of a restoration effort, including attempts to bring back fish that were listed as threatened under the Endangered Species Act (*The Seattle Times*, June 17, 1999). The dead fish, gathered by volunteers and state biologists, included sea-run cutthroat trout, rainbow trout, steelhead, coho and chinook salmon, sculpin, and lamprey. According to Mark Kaufman, an environmental specialist for the Washington Department of Ecology, "This flash destroyed five hard years of stream restoration in a few moments. The stream will recover, but it will be a long recovery" (*The Seattle Times*, June 13, 1999b). The good news for the environment was that two months after the accident, algae had returned, as had tiny mayflies. In addition, green leaves began reappearing on the trees along the creek and ferns covered the ground. As stated in the newspaper, "Olympic Pipe Line pledged millions of dollars toward the reconstruction and recovery of the Whatcom Falls Park, but for now, the community waited and hoped for the annual appearance of the salmon" (*The Seattle Times*, August 10, 1999).

Approximately three months after the accident, Olympic Pipe Line requested permission to reconstruct the pipeline. The City of Bellingham tentatively agreed once federal regulators approved the restart. The new constraints on operation included improved operator training and more detailed standard operating procedures. They also included additional pipeline inspections, testing and replacement (*The Seattle Times*, September 11, 1999). Hydrostatic pressure testing was required on the remaining sections of the line that ruptured. When this test was performed, the pipe burst again, approximately one and one-half miles from where it ruptured in June. This rupture, which occurred before the pressure reached the required test pressure, prompted federal regulators to require testing of all of the older pipeline around the Bellingham area (*The Seattle Times*, September 19, 1999b). Because of additional valve problems on the pipeline and the lack of visual inspections of the defects seen in the 1996 "smart pig" tests, on September 24, 1999, federal regulators required Olympic to reduce the amount of fuel shipped by the still-operating sections of pipeline through a reduction in pipeline pressure of twenty percent (*The Seattle Times*, September 25, 1999b). "The shutdown has been costly to Olympic because it charges field companies for every gallon it transports. The shutdown also contributed to fuel shortages last summer that raised gasoline prices in the West" (*The Seattle Times*, January 19, 2000).

Based upon the newspaper accounts, it appeared that the residents and local officials have mixed feelings about the pipeline. They understood the economic benefits of the pipeline and the fuel it carries. However, they are obviously concerned about the potential safety problems associated with fuel traveling at high pressures below neighborhoods and business areas. In many instances, the question appeared to be one of timely and effective communication. When officials from the areas along the pipeline met in December 1999, "a straw poll found that no one was satisfied with Olympic's responsiveness." According to the Bellevue franchise manager, "We wish we had gotten more information from Olympic. An issue of this nature, if you want to allay people's fears you want to do it on a factual basis" (The Seattle Times, January 21, 2000). Public response to the accident and its impact on regulations was expressed by a resident at a public forum for improving pipeline regulation when he said, "we have to step in and regulate, and regulate - yes - with the cooperation of the industry, but not with the industry calling the shots" (The Seattle Times, September 9, 1999). Olympic held several public forums in 2000, which were set up for pipeline neighbors to ask questions and also to allow Olympic to explain their improved safety and training programs. However, these forums apparently did not necessarily improve the locals' feelings of safety. According to one attendee, "My faith is even more eroded by being here." Referring to the new safety procedures, she said, "You have just started thinking about it. That's what worries me" (The Seattle Times, March 17, 2000). U.S. Representative Jack Metcalf, from Langley, WA, stated, "Testing along full length of the pipeline will help ease the fears of state residents, and serve as an excellent indicator of the overall safety of the pipeline." The Olympic Pipe Line spokesperson responded, "We don't think that's necessary," and added that "pressure tests stress the pipes." Olympic proposed the use of electronic devices to inspect the pipeline from inside (*The Seattle Times*, October 8, 1999). When Olympic requested re-opening the line in January 2000, without subjecting the complete line to the more rigorous tests, Congressman Jay Inslee of Bainbridge Island commented, "I think the folks in Snohomish and East King County are deserving of the same level of confidence that was obtained in Whatcom County before it is reopened" (The Seattle Times, January 19, 2000).

According to Wade King's father, "This company is an outrage. They basically have no requirements on them whatsoever. They put profits before people." However, he recognizes that the Office of Pipeline Safety allowed Olympic to operate in that manner. Therefore, he does not completely blame Olympic Pipe Line. "I blame the Office of Pipeline Safety for not doing their job. I loved my son so much that I can't allow that he be buried along with the pipeline. His death has to stand for something" (*The Seattle Times*, March 12, 2000). When discussing the Congressional hearings on the Bellingham disaster and pipeline safety, NTSB chairman Hall stated, "It is a sad state of affairs that regulatory oversight is basically coming out of the Department of Justice and not the Department of Transportation" (*The Seattle Times*, October 28, 1999). Regulatory response to the accident has included a proposal to require federal certification of pipeline operators, increase pipeline inspections and allow states to impose stricter regulations than the federal ones. The proposal also would require internal inspections and pressure testing every five years, the reporting of small spills (40 gallons or more), and the creation of an Internet site that shows where the pipelines are located. It would also require research into whether pipelines should be buried deeper and what leak detection and prevention equipment (double-walls, leak detection systems) should be installed. Additional legislation would increase the public's right to know about safety problems and increase the funding for pipeline inspectors (*The Seattle Times*, February 1, 2000).

The first penalty, \$3.05 million, imposed upon Olympic Pipe Line Company, resulted from the findings of the Department of Transportation investigation which concluded Olympic "failed to properly inspect and operate its pipeline and train its workers." According to Stephen Tsiorvas' grandmother, "I certainly think it's appropriate. I don't know what would ever be adequate" (*The Seattle Times*, June 3, 2000c).

The local and regional newspapers, including *The Bellingham Herald, The Seattle Times*, and *The Seattle Post-Intelligencer*, has helped keep the issue alive both through their reporting of the investigations and through their use of human interest stories regarding how people are coping with the aftermath of the explosion. On June 3, 2000 (a), *The Times* ran a feature story on the three people killed in the explosion. This was a very effective technique for reminding people about the human cost, especially since most of the recent discussion had been about the legal matters. The Internet is also being used to assist people in locating additional information about the accident and the follow-up investigations. *The Seattle Times* has listed four websites where the public can find this additional information. The federal Office of Pipeline Safety can be located at <a href="http://ops.dot.gov">http://ops.dot.gov</a>. The website for the NTSB is <a href="http://www.ntsb.gov">http://www.ntsb.gov</a>. Olympic Pipe Line has a website at <a href="http://www.olypipeline.com">http://www.olypipeline.com</a> Finally, the community group lobbying for improved pipeline regulations, SAFE Bellingham, has a website at <a href="http://www.safebellingham.org">http://www.safebellingham.org</a> (*The Seattle Times*, June 4, 2000a). Also, a memorial gathering and march was planned. The gathering would mark the disaster but also "celebrate the beginning of the restoration of Whatcom Park (*The Seattle Times*, June 4, 2000b).

# Community Impacts of Transportation Accidents Involving Hazardous Materials: Research, Examples and Implications

As the Bellingham case study dramatically demonstrates, transportation accidents involving hazardous materials can produce profound economic, social and psychological impacts in affected communities. These impacts can range from short-term financial losses to long-term emotional distress, community division, loss of trust, and social stigma.

#### Evacuation

Some of the most immediate effects of toxic transportation emergencies can result when an accident forces people to evacuate. Evacuations are highly disruptive, affecting businesses, schools, and every aspect of community life. For example, during the first 6 days after the Dunsmuir, California train derailment and pesticide spill, 483 residents left homes and went to evacuation centers. While some people's stays in the centers were short-term, others were there for several weeks. Many other residents also left the area and went to the homes of relatives or friends in unaffected communities (Bowler, Mergler, Huel, Cone 1994a).

The 1979 train accident in Missasauga, Ontario provides a vivid illustration of widespread evacuation-related disruption after a major incident. A train consisting of 3 engines, a caboose and 106 cars derailed at a level crossing.

In the wreckage were 11 cars of propane, 4 cars of caustic soda, 3 cars of styrene, and — most worryingly — a car of chlorine. Not long after the derailment, a massive propane explosion occurred, followed by two other propane explosions within 25 minutes. As a result of serious concerns about the threat posed by the chlorine, a large-scale evacuation was ordered. This was no small undertaking: Missasauga is one of Canada's biggest suburban cities. In all, 217,000 people were evacuated. This included not only residences and businesses, but also a range of institutions and facilities such as major hospitals (Scanlon 1989).

### Economic Effects

The economic effects of toxic emergencies can be considerable. Contamination, or even the *perception* of contamination, can seriously damage industries such as farming, fishing and tourism, resulting in unemployment and loss of financial security. As was evident from the Bellingham case study, property values can decrease in the aftermath of an incident. In addition, response operations after hazardous materials emergencies can also be costly. The Dunsmuir train derailment again provides a useful illustration. The accident spilled approximately 18,000 gallons of metam sodium into the Upper Sacramento River. The pesticide was carried downstream for 40 miles, killing fish and aquatic life and contaminating vegetation. State and local expenses related to the July 1991 train derailment and pesticide spill exceeded \$1.4 million. Meanwhile, other expenses (i.e., clean-up, medical, economic, etc.) came to over \$2 million (Committee on Government Operations 1992).

### Psychological Impacts

Less apparent than immediate disruption and economic effects -- but potentially more problematic and complex to address -- are the psychological effects of accidents involving hazardous materials. Disaster specialist James Thompson (1990) suggests that, in terms of chronic effects, the number of people psychologically affected by a chemical accident can far exceed the immediate casualty list. "From some of the data we have on chemical and 'contamination' incidents, it might well be that the psychological impact rate is about one order of magnitude higher."

Baum and other researchers have argued that technological disasters are more likely to produce chronic, widespread psychosocial sequelae than natural disasters (Baum, Fleming & Davidson 1983; Baum 1987; Baum, Fleming & Singer 1983; Weisaeth 1994). Just why this should be the case relates to the particular nature of technological accidents, particularly those involving hazardous materials. Natural disasters like a tornado have a low point, after which things can be expected to get better. Damage is visible and can be assessed, after which people may begin a process of recovery. In disasters involving possible exposure to toxic agents, however, there is no clear low point for those who may have been affected. There is usually considerable uncertainty about the consequences of exposure. Medical knowledge is frequently limited, and both contaminants and their resulting damage may be invisible. Further, potential long-term health consequences (e.g., cancer) may take years or even decades to develop. Thus it is not clear to people whether the worst is over or whether the worst is yet to come (Baum, Fleming and Davidson 1983).

"In a sense," Baum (1987) explains, "this pattern of influence extends the duration of victimization." Rather than being struck and then having a chance to recover, as in the case of a flood, the threat here is seen as a chronic and continuing one. "One does not know when the impact of what happened is really going to hit." (Reko 1984a) People wonder whether contaminants have entered their bodies, and they worry about their health and the health of loved ones (especially children). Even when an accident is officially declared "over," it is, in an important sense, not really over for those who may have been exposed (Erikson 1995) The "point of worst impact may not pass with the event. Perceived threats may continue indefinitely." (Baum, Fleming & Singer 1983)

As Ursano, McCaughey and Fullerton (1994) have written, contamination incidents "produce long-term anticipatory stress of the possible, the probable and the imagined risks to health and family." At the same time, in the face of medical uncertainty, a need to rely on expert assessments, and the invisibility of contaminants, people can feel a continuing sense of vulnerability and powerlessness: they cannot be certain what is going on, nor can they do anything to protect themselves (Brown & Mikkelsen 1990; Aaronen & Mikkelsen 1993). Victims of chemical or radiological accidents, then, often live in what Erikson characterizes as a "permanent state of alarm and anxiety." Beyond whatever possible toxicological or other health effects people may experience in the aftermath of a chemical

accident, such unremitting tension and profound apprehension about the future can take its own considerable toll on health and well-being (Erikson 1993).

Another characteristic of technological accidents that has psychosocial implications concerns the matter of responsibility and blame. Erikson (1995), employing the analytic comparison with natural disasters, has said the following:

Natural disasters are almost always experienced as acts of God or caprices of nature. They happen to us. The visit us, as if from afar. Technological disasters, however, being of human manufacture, are at least in principle preventable, so there is always a story to be told about them, always a moral to be drawn from them, always a share of blame to be assigned.

In the aftermath of technological disasters, people want to know why technology under human control has failed, why suffering that could have been avoided has not been. Thus, rather than ultimately producing resignation or acceptance, human-caused disasters generate mistrust, anger, fear and outrage. Erikson's (1995) words are again germane here:

[P]eople who are victimized by such events feel a special measure of distress when they come to think that their affliction was caused by other human beings. And that sense of injury becomes all the sharper and more damaging when those other human beings respond to the crisis with what is seen as indifference or denial.

Human-made disasters, argues Weisaeth (1994), "frequently cause withdrawal and social isolation." Indeed, the more clearly people perceive a human cause behind a disaster, the more distressing and potentially pathogenic the situation seems to be (Weisaeth 1994; Brown & Mikkelsen 1990) As Vyner (1988) has written, accidents involving hazardous materials can be highly traumatic: "All evidence indicates that adapting to an invisible exposure is a toxic process. It is a process that can severely traumatize the exposed persons and change their lives for the worse.

Various examples of the psychological impacts of transportation accidents involving hazardous materials may be found in the scientific literature. One example is provided by the March 1989 Exxon Valdez oil spill. The accident, in which a tanker ran aground on a reef, spilled 258,000 barrels of crude oil into Alaska's Prince William Sound (Davis 1996). A follow-up study conducted a year after the accident (Palinkas, Downs, Petterson, Russell 1993) found a significant relationship between exposure to the spill and the prevalence of psychiatric disorders. Problems included increased post-spill rates of generalized anxiety disorder, post-traumatic stress disorder, and depression. Indeed, 43 percent of people in the "high exposed" groups were reported to have experienced one or more such problems.

Studies of other transportation-related accident have also identified various psychological sequelae. Bowler, *et al.* (1994a) conducted follow-up research after the July 1991 freight train derailment at Dunsmuir, California. Researchers found a wide range of psychological, psychosocial, and psychophysiological effects in people from the affected area. In comparison with controls, the exposed group experienced higher blood pressure and more sleep disorders, headaches, visual problems, skin rashes, gastrointestinal symptoms, cardiac/respiratory symptoms, anxiety symptoms and depression symptoms.

An analysis by Gill and Picou (1998) of a 1982 train derailment in Livingston, Louisiana provides further evidence of psychological effects after a hazardous materials transportation incident. The accident caused 43 cars to derail, including 36 cars containing hazardous materials. Most of these leaked, burned or exploded, forcing the evacuation of approximately 2,500 people for up to 17 days. Despite the fact that there were no deaths or serious injuries, and although property destruction was limited, the level of event-related psychological stress was significant. According to the researchers, this was clearly evident on the Impact of Events (IES) Scale, which is used to measure "stress arising from traumatic events that are generally outside the range of human experience." (Gill and Picou 1998) On the "Intrusive Stress" subscale, which measures "recurring, unbidden, and distressing thoughts and feelings," the mean among Livingston residents was 13.7. In the words of Gill and Picou (1998), "the mean levels of intrusive

stress observed for... Livingston (13.7)... were comparable with that experienced by clinical patients 6 months after therapy for bereavement resulting from the death of a parent (13.8)...."

Studies also suggest that some groups may be especially at risk for psychological effects after contamination incidents. For example, work carried after the 1989 Exxon Valdez oil spill (Palinkas, *et al.* 1993; Picou, *et al.* 1992) identified several groups as being among those who were particularly hard hit. In the words of Palinkas, *et al.* (1993):

Younger age groups, women, and Alaskan Native residents of these communities appear to have been especially vulnerable to these negative impacts as evidenced by higher rates of psychiatric disorders.

In addition, other research has called attention to the mental health impacts of chemical contamination episodes on children (Breton, Valla and Lambert 1993).

### Social Impacts

Just as hazardous materials accidents can have substantial and long-lasting mental health effects, so too can they leave profound social impacts in their wake. One such impact that is frequently experienced is social division (Edelstein and Wandersman 1987; Kroll-Smith and Couch 1993; Couch and Kroll-Smith 1985). Here, the contrast with natural disasters is again useful. In the post-impact phase of natural disasters, people typically pull together to overcome a common problem and to get things back to normal. In the context of a sense of "common suffering and altruistic concern," a kind of therapeutic community emerges, providing an ambience of camaraderie, solidarity, unity of purpose, and mutual support (Cuthbertson and Nigg 1987).

In the case of chemical and radiological accidents, however, this is often not the case. More than anything else, contamination situations are characterized by haziness and ambiguity. Hazardous agents are often invisible, so there is great uncertainty as to which areas have been exposed and who has been affected. The uneven spread of contaminants frequently means that people who live near each other -- even on the same street -- can have vastly different experiences of the problem. People's assessments of the degree of risk posed by the contamination may differ enormously, and their views as to what should be done may clash as well (Cuthbertson & Nigg 1987; Kroll-Smith & Couch 1993). Then, too, the matter of assigning blame for the accident can be a source of disagreement as well.

With high-stakes issues involved (e.g. health, children's well-being, property values), such differing definitions of the situation can produce hostility, factionalism and fragmentation. Environmental accident situations "produce increased conflict and deleterious long-term strain on community structures...." (Couch & Kroll-Smith 1985). They have the capacity to damage the very fiber of a community, to be, in a sense, what Taylor (1986 and 1989) calls "sociotic." Rather than producing consensus and a therapeutic community, they have a tendency to create the very opposite: social division and a dissensus community (Edelstein and Wandersman 1987). Such social division can impair the social support network that people normally rely upon in time of crisis.

Evidence of social conflict has been found in various studies of communities affected by transportation accidents. In the aftermath of the Exxon Valdez oil spill, for example, researchers noted conflicts among friends and family members, arguments between community members and outsiders, divisiveness over whether or not to work for Exxon as part of the cleanup, friction over compensation issues, and other social impacts (Palin kas, *et al.* 1993).

Studies have also identified various social impacts after hazardous materials train derailments. In the aftermath of the Dunsmuir accident, Bowler, *et al.* (1994a) noted the presence of a split in the community. In addition, the researchers found that on the Perceived Social Support Scale, there was a significant difference between people in the exposed group and matched controls. The Perceived Social Support Scale measures an individual's perception of the extent to which he or she has access to emotional support systems. According to Bowler, *et al* (1994b), in the aftermath of the accident, spill residents "had significantly ... lower perceived social support than their matched controls."

Another important social impact is stigma, which is also common after environmental accident situations. Residents of affected communities may be seen by others as "tainted" and as "people to be avoided." (Edelstein 1988; Kroll-Smith and Couch 1993) The point is well illustrated by the words of a local councilwoman from Triana, a small North Alabama town that was contaminated with DDT: "Once you are branded a contaminated person, you are a contaminated person. You are branded everywhere you go. That's our schoolchildren. That's everybody." (Birmingham Post-Herald, Nov. 1, 1997)

Social stigma can be powerful and pervasive. Following a radiological contamination incident in Goiania, Brazil, people from the city found themselves the focus of fears and the target of discrimination. As Kasperson and Kasperson (1996) have noted: "Hotels in other parts of Brazil refused to allow Goiania residents to register. Some airline pilots refused to fly airplanes that had Goiania residents aboard. Cars with Goias license plates were stoned in other parts of Brazil."

Community division and stigma are by no means the only important social impacts of chemical and nuclear accidents. Other effects include chronic loss of trust (Levine 1982 and 1983) and impairment of the pattern of community life due to destruction of natural resources (Dyer, Gill and Picou 1992). In addition, the experience of a contamination episode can powerfully alter people's view of their place of residence. As Gill and Picou (1998) have commented:

When communities experience a technological disaster, one response is to contemplate leaving one's place of residence. Contamination and subsequent uncertainty regarding exposure, long-term environmental damage, and the alteration of a lifescape reduce the quality of life in contaminated communities.

This point was apparent in research carried out after the Livingston, Louisiana train derailment. Whereas only 28 percent of people in a control community expressed a desire to move, the figure for Livingston was 48 percent. Even more strikingly, whereas only 1 percent of those in the control community indicated that they *expected* to move, the figure for Livingston was 14 percent (Gill and Picou 1998).

Finally, sometimes, the effects of a hazardous materials accident can be so widespread that they tear apart the very fiber that holds a community together. The contamination and resulting evacuation of a small Missouri town in 1983 is probably one of the best-know examples of an environmental accident producing what Erikson (1976) terms "loss of communality." When Times Beach was found to be heavily contaminated with dioxin -tainted waste oil that had been applied to area roads, officials evacuated the town's 2,240 residents, erected a security fence to keep anyone from entering the area, and officially closed the town. The evacuation tore apart the tight-knit community bonds upon which people had relied in the past. Further, once former residents had been scattered through relocation, they were unable to find each other, since privacy laws prevented government officials from sharing their lists of new addresses with victims. Thus, even as the frightening reality of dioxin contamination was still settling in, victims "lost their sense of place and identity as the social fabric of the community disintegrated." (Re ko 1984a)

To sum up, then, hazardous materials accidents can produce a wide range of damaging community impacts. This complex constellation of economic, psychological and social effects can harm individuals, families and entire neighborhoods. Given the severe psychosocial damage that such accidents can cause, Baum (1987) has argued that these events can be thought of as disasters regardless of how controversies about biological impacts are resolved. Such "human-made accidents involving toxic substances are disasters, whether or not the amount of toxic exposure involved can be proven to be dangerous to health."

### Strengthening Preparedness and Response Capabilities

It is clear from the previous discussion that social, psychological and other community impacts are among the most significant consequences of major transportation-related hazardous materials accidents. At the present time, however, states and localities across the U.S. are only beginning to recognize such issues and fully integrate them into preparedness and response mechanisms. For example, response plans and protocols rarely devote adequate attention to the psychosocial effects of contamination incidents. When psychosocial content is included, it is usually limited to *generic* information about dis asters, debriefing, and mental health. Plans rarely include *specific* 

information about contamination incidents and the complex psychosocial challenges – immediate and longer term – that they pose. Thus, guidance related to the specific challenges posed by hazardous materials accidents – fears associated with invisible agents, the stress of being in a potentially-contaminated environment, the problem of social stigma – is generally absent. This is particularly true with regard to social impacts and longer-term psychological effects. So, even though a great deal is now known about the psychosocial challenges posed by environmental contamination situations, current plans for managing such disasters do not usually reflect this knowledge.

The same is true with regard to training. The emergency management community is now quite good at practicing various technical aspects of hazardous materials accident management. Likewise, health care professionals are becoming quite adept at creating exercises to improve the medical response to a contamination incident. These efforts are vital. Unfortunately, however, social and psychological issues are not generally incorporated in a way that fully reflects their importance in actual large-scale hazardous materials accidents. Again, this is particularly true with respect to social impacts and longer-term psychological effects.

Thus, it will be important in the coming years to better incorporate social and psychological considerations into preparedness and response mechanisms for dealing with hazardous materials transportation accidents. Given what is now known about such accidents, it would be useful for such mechanisms to include not only immediate response issues but longer-term effects as well. In addition, it would be valuable for training exercises to include more attention to psychosocial issues and more realistic social-behavioral assumptions.

Based on experience from past accidents, it is evident that social stigma is a serious problem after nuclear and radiological accidents. It is a problem in and of itself, and it also complicates efforts to deliver services and rehabilitate communities. It would be beneficial, therefore, for strategies to prevent and mitigate stigma to be developed and integrated into large-scale contamination incident plans. Likewise, strategies to mitigate other social impacts (e.g., social division) would be useful.

In addition, there is a need for special materials and interventions for high-risk populations. In natural disaster situations, there are coloring books for children that help them to understand what has happened. Few such materials are available for chemical and radiological accidents. Clearly, the development of appropriate materials, as well as tailored interventions for high-risk populations, needs to be a priority, too.

Finally, there is the issue of information. In considering ways to reduce the community impacts of major hazardous materials transport accidents, information stands out as a crucial factor. Research suggests that an early lack of accurate information can contribute to both anger and fear (Bowler, *et al.* 1994a). Such a situation may increase long-term psychological morbidity, undermine trust and damage public confidence, greatly hindering individual and community recovery after a major accident.

In an analysis of the Dunsmuir train derailment, for example, Bowler, *et al.* (1994b) concluded that the inability of authorities to provide residents with accurate and early information on the possible adverse health effects of the spilled chemical (metam sodium) "was reported overwhelmingly as a contributing cause of fears and worries." According to the researchers, "this early lack of information contributed to a lingering anger at the authorities and heightened fear of future illness."

If information is a vital factor in reducing community impacts after a chemical or radiological accident, it is also crucial beforehand as well. Long before an accident occurs, members of the public need to be aware of the particular hazards in their community and of how to respond in an emergency situation. Furthermore, prior familiarity with, and understanding of, hazards may also help to reduce psychological morbidity should a major accident actually occur.

At the present time, mechanisms for *post-accident* communication are relatively well established. Public Safety, emergency management, environmental, public health and other officials have amassed considerable experience with television, radio and other means of information transmission that would be utilized after a major transportation-related accident. In terms of *pre-accident* communication, however, the picture is more mixed. Unfortunately, at the present time, only a small number of local emergency planning committees in Alabama have

the resources they need to communicate with the public on a regular basis. For example, Title III (Emergency Planning and Community Right-to-Know) newsletters are rare. Likewise, only a few LEPCs in the state have websites.

While a number of Alabama LEPCs are making valiant efforts, LEPC communication activities are clearly hampered by a lack of funding. A comprehensive analysis prepared by the National Governors' Association found that in contrast to many other states, the State of Alabama provides no funding for LEPC activities (Finegold 1997). The lack of resources for newsletters and especially, websites, means that pre-accident communication with the public remains limited. Thus, as part of overall efforts to improve preparedness for major transportation accidents involving hazardous materials, it would be advantageous for funds to be allocated to Alabama's local emergency planning committees.

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## **Photographs of The Bellingham Pipeline Explosion**



Figure 5-1. Seen for miles around; Thursday's explosion in Bellingham is photographed from Lummi Shore Drive, miles away. The plume of smoke eventually rose to an estimated 30,000 ft (photo by Kim Walker, copyright Bellingham Herald, permission requested).



Figure 5-2. Smoke towers over city (photo by Angela Lee Holstrom, copyright June 10, 1999 Bellingham Herald, permission requested).



Figure 5-3. Aerial photo of explosion scene (copyright Bellingham Herald June 11, 1999, permission requested).



Figure 5-4. Burned Whatcom Creek from the air on Sunday June 20, ten days after the explosion that took the lives of three boys in Bellingham (photo by David Willoughby copyright Bellingham Herald, permission requested).



Figure 5-5. Fire fighters from Tosco Refinery spray foam on hot spots along Woburn St. (copyright June 10, 1999 Bellingham Herald, permission requested).



Figure 5-6. An unidentified person walks the point where Park Creek enters Whatcom Creek in Whatcom Falls Park in Bellingham, WA (copyright June 10, 1999 Bellingham Herald, permission requested).



Figure 5-7. Larry Bateman, operations supervisor for the Bellingham Public Works Dept. walks past a crater near the water treatment plant Friday afternoon, June 11, 1999 (copyright June 11, 1999 Belligham Herald, permission requested).



Photo 5-8. Photo of where the 277,200 gallon gasoline leak occurred (copyright 1999 nwcitizen.com. Reprinted with permission).

### **Section 6. Conclusions and Recommendations**

Accidents involving chemicals or radioactive materials represent a significant threat to the environment, public health and safety, and community well-being. In an increasingly complex and interconnected world, no community is immune from the threat posed by environmental accidents and contamination. Even communities far removed from industrial production or storage facilities can still be at ris k from accidents associated with the transport of hazardous materials. While a variety of studies have been conducted on aspects of major transportation accidents, few have attempted to examine both environmental and community aspects of the problem. In contrast, this report takes an integrated approach to hazardous transportation accidents by considering environmental, safety, economic, and psychosocial issues. The purpose of the project is to (1) quantify transportation-related accidents involving hazardous materials in the state, and (2) identify key longer-term environmental health, public safety, and social impacts that are often overlooked after major transportation-related hazardous materials accidents.

The project was comprised of four main tasks: consultation with key stakeholders; summarizing and analyzing representative transportation-related accidents involving hazardous materials that have occurred in Alabama since 1990; presentation of simplified chemical transport and fate models; and presentation of information to help anticipate important social, psychological and related community impacts that can occur after major transportation-related hazardous materials accidents.

Three case studies of transportation accidents involving hazardous materials are presented. The first, which took place near Dunsmuir, CA, in 1991, involved a train derailment that spilled a large quantity of the pesticide metam sodium. The second case study, a truck accident on Interstate-65 in Alabama, was far smaller and far less serious than the Dunsmuir case. It is noteworthy, however, because it illustrates how an accident involving even a very small quantity of hazardous material can produce significant problems. The third case study presented is of a massive gasoline pipeline break and resulting explosion that occurred in 1999 in Bellingham, WA. All three of these case studies present extensive discussions of community impacts, along with descriptions of the physical problems that occurred during the accidents.

Alabama hazardous material transportation related accident information was collected and analyzed using data from the National Response Center. The purpose of this task was to identify the most common hazardous materials lost, where the accidents occurred, and which media was affected. This information was used to present procedures that can be used to predict the movement and dispersion of the lost material.

More than 1700 transportation related accidents involving hazardous materials occurred in Alabama during the past ten years, involving a large number of different materials, although many petroleum hydrocarbons were the most common hazardous material lost. Of the 226 reported accidents in 1998, there were 20 deaths and 27 injuries. In addition, four accidents caused property damage, two accidents resulted in evacuations, and nine accidents resulted in road closures. The locations with the most frequent spills are the USS Alabama Battleship and the hazardous waste landfill at Emelle, probably due to diligent reporting by the site operators. Additional locations of frequent spills include several sites where chemicals are transferred from marine craft to land vehicles, such as trains and trucks.

The report presents several procedures to predict the fate and transport of spilled hazardous materials. The initial discussion is a general procedure that stresses downwind toxic and explosive hazards, summarized from a recent EPA manual and is applicable for a wide range of hazardous materials. Two detailed examples are also presented describing problems associated with spills of petroleum hydrocarbons, by far the most common material lost in Alabama transportation accidents, and losses of ammonia, an example of a toxic gas.

Major transportation accidents involving hazardous materials can produce profound economic, social and psychological impacts in affected communities. These impacts can be both widespread and long lasting. Details

from the Bellingham pipeline explosion are presented, along with a more general discussion of the economic, social and psychological effects of hazardous transportation accidents. Current scientific research is reviewed, examples are provided, and implications are considered.

Stakeholders raised a number of issues that need to be addressed in future state planning for transportation accidents involving hazardous materials:

- From a planning standpoint, concerns were raised about the routing of hazardous materials in the state, particularly in relation to the tunnel in Mobile.
- Shipments of transuranic waste from both Oak Ridge and Savannah River are scheduled to travel through Birmingham on I-59/I-20. Concern was expressed about whether public safety personnel would be notified when shipments are scheduled to pass through the state. These shipments will pass through the most populated city in the state and are likely to be contentious.
- Several of the larger fire departments (Birmingham, Tuscaloosa, Montgomery, Mobile and Huntsville) have hazardous materials responders who have had the required training. Fort Rucker also has its own hazmat responder unit. However, much of the state is served by volunteer/semi-volunteer fire departments. Most of the departments are not prepared to assist in a hazardous materials incident. In order to combat this lack of preparedness, several volunteer fire departments have begun cooperating with each other to create a hazmat unit for a county/region. This cooperative effort would require each department in the area to contribute equipment and/or personnel for the endeavor, but it would mean that each department would not have to have its own functioning hazmat unit. Greater support for such efforts is needed so that small fire departments can obtain needed training and equipment.
- Concern was expressed over the limited resources available both to responder agencies and local emergency planning committees (LEPCs) in Alabama. Mandated under the Emergency Planning and Community Right to Know Act of 1986, LEPCs are a key component in preparedness and response for contamination incidents. Concern was expressed that current responder agency and LEPC resources are not adequate.

Other concerns raised during stakeholder meetings included (1) recovery of resources spent on a hazmat incident, (2) communications difficulties during an incident, and (3) appropriateness of response to "unusual" chemicals. First, the State has no mechanism for recovering its expenses relating to a hazardous materials incident response. Not only is there no money in the state budget for expenses relating to this type of emergency, but there are no requirements for the responsible party to reimburse the state for the money expended on a response. Second, there is no uniform standard for communications equipment between Department of Public Safety (DPS) and local police, fire and emergency responder departments. Even inside the DPS there are three communications systems, which can cause major problems with internal coordination, much less trying to coordinate with outside departments. Third, there is a concern about responders, especially local departments, having the knowledge or ability to get the knowledge quickly to respond to incidents involving 'unusual' chemicals, i.e., those chemicals that are not encountered frequently during a traffic accident.

The chemical groups that responders generally were not prepared and equipped to deal with were water-reactive chemicals, corrosives, elevated temperature materials, regulated medical waste, and precursor chemicals for clandestine laboratories. The typical response of a local fire department would be to put water on the chemical and wash it off the roadway. However, in the case of water-reactive chemicals, this may make a small problem a significantly larger one. When dealing with elevated temperature materials, the departments do not have the appropriate gear, i.e., do not wear rubber suits near a 250°C fire, and one example of a commonly transported elevated temperature material was liquid asphalt. Regulated medical waste is a concern because of the variety of vehicles in which it can be transported and because of the lack of information that may be available about the exact nature of the waste. The last chemical group is the precursor chemicals for clandestine laboratories. These shipments are not placarded and there is no paperwork on what the truck contains. In many cases, these are rental trucks. Therefore, personnel responding to an accident likely does not know that he/she is entering a chemical hazard area, and they are not appropriately protected.

In addition to their environmental impacts, major transportation accidents involving hazardous materials can produce significant economic, social and psychological impacts in affected communities. People in Bellingham, for example, viewed the pipeline explosion as "the most devastating thing we've ever had happen to this community. This has shaken the community's sense of security to the core." Furthermore, as both the scientific literature and the case studies presented in the report illustrate, these impacts can be traumatic, widespread and long lasting. "It comes as a shock to me how much suffering remains in this community because of this," a Bellingham doctor noted. And as a Dunsmuir resident made clear, the lingering effects of a contamination accident make getting "back to normal" difficult. "We all want to forget the spill, but we, as people who have been forced to live in the midst of the disaster, have changed. The spill affects our lives daily and will for a very long time."

Some of the most immediate effects of toxic transportation emergencies can result when an accident forces people to evacuate. Evacuations are highly disruptive, affecting businesses, schools, and every aspect of community life. The economic effects of toxic emergencies can also be considerable. Response and clean-up operations are expensive, and contamination, or even the *perception* of contamination, can lower property values and seriously damage industries such as farming, fishing and tourism.

Less apparent than immediate disruption and economic effects — but potentially more problematic and complex to address — are the psychological effects of accidents involving hazardous materials. Concerned about their health and the health of loved ones, victims of chemical or radiological accidents live in what Erikson (1995) characterizes as a "permanent state of alarm and anxiety." Studies suggest that people who have suffered through transportation accidents involving hazardous materials are at increased risk of a range of psychological problems. "All evidence indicates that adapting to an invisible exposure is a toxic process. It is a process that can severely traumatize the exposed persons and change their lives for the worse." (Vyner 1988) Furthermore, just as hazardous materials accidents can have substantial and long-lasting mental health effects, so too can they leave profound social impacts in their wake. Loss of trust, social conflict and division are common, as are social stigma and a sense of a reduced quality of life in affected communities.

Clearly, then, social, psychological and other community impacts are among the most significant consequences of major transportation-related hazardous materials accidents. At the present time, however, states and localities across the U.S. are only beginning to recognize such issues and fully integrate them into preparedness and response mechanisms. To enhance our ability to prevent and mitigate community impacts, it will be crucial to better incorporate social and psychological considerations into preparedness and response mechanisms for dealing with hazardous materials transportation accidents. Given what is now known about such accidents, it would be useful for such mechanisms to include not only immediate response issues but longer-term effects. In addition, it would be valuable for training exercises to include more attention to psychosocial issues and more realistic social-behavioral assumptions. It would also be beneficial, for strategies to prevent and mitigate stigma to be developed and integrated into large-scale contamination incident plans. Likewise, strategies to mitigate other social impacts (e.g., social division) would be very useful. The development of appropriate materials, as well as tailored interventions for high-risk populations, needs to be a priority, too.

Finally, there is the issue of information. In considering ways to reduce the community impacts of major hazardous materials transport accidents, information stands out as a crucial factor. It is vital in reducing community impacts *after* a chemical or radiological accident, and it is also crucial *beforehand*. Long before an accident occurs, members of the public need to be aware of the particular hazards in their community and of how to respond in an emergency situation. Furthermore, prior familiarity with, and understanding of, hazards may also help to reduce psychological morbidity should a major accident actually occur.

While mechanisms for *post-accident* communication are relatively well established, the situation with respect to *preaccident* communication remains mixed. Unfortunately, at the present time, only a small number of local emergency planning committees in Alabama have the resources they need to communicate with the public on a regular basis. For example, only a few LEPCs in the state have websites. While a number of Alabama LEPCs are making valiant efforts, LEPC communication activities are clearly hampered by the fact that, in contrast to many other states, the State of Alabama provides no funding for LEPCs. As part of overall efforts to improve preparedness for major transportation accidents involving hazardous materials, it would be advantageous for funds to be allocated to Alabama's local emergency planning committees.

# Appendix A. Alabama Transportation Accidents Involving Hazardous Materials (1990 – 1999)

NRC Report No	Date Call Reported	Call Tyne	Incident Date and Time Incident Tyme Incident I ocation	Incident Tyne	Incident I ocation	Country	). J	Zin Code	Suspected Responsible	SDD Address
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870	1/12/90	Std Report	1/12/90 13:30	Fixed	Dock 12	Mobile			International Paper Company	Paper Mill Rd, Mobile, AL 36653
0	000				3	;			International Paper	Paper Mill Rd, Mobile,
606	1/14/90	эта кероп	00.01 06/61/1	Fixed	Building 3267	Mobile			Company	AL 36653
956	1/14/90	Std Report	1/13/90 16:00	Fixed	Building 3267	Mobile			International Paper Company	Paper Mill Rd, Mobile, AI. 36653
					D					
1073	1/16/90	Std Report	1/13/90 15:30	Pipeline	2101 E Pacific Coast Hwy	Mobile	Citronelle		Douglas Oil Co.	POB 305, Citronelle, AL 36522
									International Paper	Paper Mill Rd, Mobile,
1161	1/11/90	Std Report	1/13/90 15:00	Fixed	Galveston Terminal Docks	Mobile			Company	AL 36653
1161	1/17/90	Std Report	00-51 06/21/1	Fixed	Galveston Terminal Docks	Mohile			International Paper Company	Paper Mill Rd, Mobile,
						all court			Company	
1911	06/11/1	Std Report	1/13/90 15:00	Fixed	Galveston Terminal Docks	Mobile			International Paper Company	Paper Mill Rd, Mobile, AL 36653
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1299	1/18/90	Std Report	1/18/90 14:30	Railroad	19-31W	Colbert	Sheffield		Railroad	GA 30303
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1351	1/19/90	Std Report	1/19/90 12:15	Fixed	Section 20, 4 S, 3 W	Cullman			292 Truckstop	AL
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1439	1/20/90	Std Report	1/20/90 15:00	Marine	19-31W	Baldwin	Magnolia Springs		Unknown	
1630	1/23/90	Std Report	1/23/90 15-30	Pineline	Great Salt Lake 8 mi W O	Colher	Shefffeld		Reynolds Metals	POB 120, Sheffield, AL
									Chemical Waste	POR S
1791	1/25/90	Std Report	1/25/90 7:30	Highway		Sumter			Management	35459
1812	1/25/90	Std Report	1/25/90 11:00	Offshore	Hwy 146 & Texas Av	Marion	Offshore		Mobil Oil Co	1250 Poydras St, New Orleans, LA 70013
		-								
6576	2/1/90	Std Report	2/1/90 10:00	Highway	2725 N Wood Rd	Jefferson	Fultondale	35207	Bunt Construction Co	PO Box 321035, Birmingham, AL 35232
									Unnamed Used Parts	Hwy 75 N, Albertville.
0569	2/5/90	Std Report	2/1/90 12:00	Fixed	Hwy 75 N	Marshall	Albertville	35950	Shop	AL
6977	2/5/90	Std Report	2/5/90 4:00	Highway	1-20/59 & 1-65	Jefferson	Birmingham		Builder's Transport, Inc.	POB 7005, Camden, SC 29020
7216	2/6/90	Std Report	2/6/90 9:00	Fixed	Paper Mill Rd	Mobile	Mobile	36692	International Paper Company	PO Box 2448, Mobile, AL 36692
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Transportation Accidents with Hazardous Materials, 1990-1999

NRC Report No.	NRC Report No. Reported Cause	Incident Description	Medium Description	Medium Affected D	Injuries Deaths Reported Reported		Damages	Evacuations	Number Evacuated	Airway Closure
7560	Dumping	Caller says Mr. Transmission suspected of dumping oil	Unknown Creek	Water			No O	N <sub>o</sub>		
7950	Operator Error	Transformer hit by forklift	Concrete	Land			No	No		
8419	Equipment Failure	Open valve on tank/unknown why valve open/Trash on an embankment near tracks/036550 MD License Plate.	Cement sump	Land		,-	No.	No.		
8593	Equipment Failure	Transfer pump flange came loose	Land	Land			No	No		
8671	Орегатог Еггог	Fuel tank overflow	Theodore Ship Channel	Water			92.	%		
9916	Unknown	Oil discovered on water	Mobile Bay	Water			No No	°Z		
0856	Equipment Failure	M/V Glomar Pacific hose busted while pumping waste oil into container	Mobile River	Water			No	No	The state of the s	
11803	Unknown	Tractor trailer struck by train/Truck cargo (125 automobile batteries) spilled	Tennessee River	Water			No	No		
11804	Transport Accident	Transport Accident Tank truck driven off road and rolled over	Ground	Water		-	1 No	o <sub>Z</sub>		
12023	Орегаtот Еггог	Material spilled during offloading of material from vessel	Concrete pad & pit	Land			No	°Z		
12050	Equipment Failure	Barrel on flatbed truck leaking	Truck bed	Unknown/ Other			No	No		
12185	Орегатог Епог	Dump truck tailgate open	Ground	Land			Ŷ.	Š		
12341	Орегатог Етгог	Caller reported RP dumping oil on field behind his garage	Mobile Bay	Water			No	%		
12824	Other	Someone tied back fuel triggers at two terminal pumps	Concrete sumps	Land			No	N <sub>O</sub>		

Transportation Accidents with Hazardous Materials, 1990-1999

NRC Report No. Road Closure Damages (\$)		CHRIS Code	Name of Material	Unit of Quantity Spilled Measure	Quantity in Units of Water Measure	of Name of Railroad	Train Number	Derailed?
7560		<u> </u>	Oil, unknown	0 unk	0 nuk			
7950		PCB	Polychlorinated biphenyls	10 gal	о поп			
8419	1	DCM	Dichloromethane	15393 lbs	uou 0			
8593		SFA	Sulfuric acid	sq1 0066	0 non			
8671		SGO	Oil: diesel	70 gal	60 gal			
9916	9	OUN	Oil, unknown	0 unk	0 unk			
9580	J	OWA	Waste oil	2 gal	2 gal			
11803		NCC	Black liquor	0 unk	0 unk			
11804	7	NCC	Black liquor	0 unk	0 unk			
12023		NCC	Copper chromium arsenic compound	3000 gal	0, non			
12050		NCC	Flammable liquid, waste	20 gal	uou 0			
12185		NCC	Battery plant trash	40 lbs	uou 0			
12341		TIO	Oil: crude	3000 gal	3000 gal			
12824		SGO	Oil: diesel	293 gal	0 unk			

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NRC Report No.	NRC Report No. Reported Cause	Incident Description	Medium Description	Medium Affected D	Injuries Deaths Reported Reported	Injuries Reported	Damages	Evacuations	Number Evacuated	Airway Closure
14254	Operator Error	Vehicle transmission busted open when driver ran over curb	Ground	Land			No	°Z		
14525	Unknown	Caller states major road accident spilled gravel contaminated with old RR ballast	Topsoil & Asphalt	Land			No No	No		
14525	,									
15064	Equipment Failure	T/B S-2511 crack above waterline leaking	Mobile River	Water			No	No		
15410	Equipment Failure	Equipment Failure Tent enclosure/Equipment failure	Sump > Ash pond Water	Water			No	No		
15437	Equipment Failure	Dump truck tailgate leaked due to rainwater infiltration	Company scales	Land			No	ĝ		
15459	Equipment Failure	Equipment Failure Dump trailer leaked	Asphalt	Land			No	Š		
15536	Equipment Failure	Dump truck tailgate leaking	Pavement	Land			No	ŝ		
15653	Transport Accident	Transport Accident Tractor trailer overturned on highway	Highway	Land			No	°Z		
15949	Орегатог Етгог	Case of material dropped during loading	Concrete	Land			No	No		
16002	Operator Error	Multiple oil spills at fuel tank storage area, saturating area's soil	Soil	Land		0	0 No	o <u>N</u>		
16206	Natural Phenomenon	Storage tank overturned during March 17 flooding	Murder Creek	Water			No	No		And the second of the second o
16461	Equipment Failure	M/V Nicor Texas hose leaked while transferring material to platform	Gulf of Mexico	Water	:		No	No		
16552	Equipment Failure	Tank car RC RX 1296 dome leaking	Tank car side	Unknown/ Other			No	No		
16797	Unknown	Tank overflowed due to unknown cause	Mobile River	Water			No	No		

NRC Report No Road Closure Damages (\$)	Reported Damages (\$)	CHRIS	Name of Material	Unit of Quantity Spilled Measure		Quantity in Water	Units of Measure	Name of Railroad	Train Number	Derailed?
14254		ОТН	Transmission fluid		5 gal		0 gal			
14525		EGL	Ethylene glycol	0	0 unk	0	0 unk			
14525		SHD	Sodium hydroxide	-0	0 unk	0	0 unk			
15064		OIL	Oil: crude		l gal	_	gal		AND THE PROPERTY OF THE PROPER	
					,					
15410		NCC	Asbestos	0	0 nnk	9	0 umk			
15437		NCC	RCRA waste runoff, D008 (lead)	1	gal	-	gal			
15459		NCC	Waste lead D008	2	2 gal	0	0 non			
15536		NCC	Blast furnace slag	- 7	2 gal	0	0 non			
15653		PAC	Phosphoric acid (80 %)		l gal	0	0 non			:
15949		CRF	Chloroform	17 lbs	lbs	5	0 non			
16002		OMT	Oil, misc: motor	1000 gal	gal		0 unk			
16206		ODS	Oil: diesel	0	0 unk	0	0 unk			
16461		ODS	Oil: diesel	15.	15 gal	15	15 gai			
16552		HCL	Hydrochloric acid	2	2 gal	0	0 non			
16797		gcs	Gasoline: casinghead	2.5 gal	gal	2.5	2.5gal			

	index nin									
POB 482, Mobile, AL 36601	Bender Ship Building and Repair		Mobile	Mobile	Alabama State Docks Berth North B2	Fixed	5/9/90 15:45	Std Report	2/9/90	21212
185 Spring St, Auanta, GA 30303	Nortolk Southern Railroad		Selma	Dallas	Selma railyard MP 193N	Railroad	5/8/90 23:30	Std Report	2/9/90	21062
35602	Decatur Transport Inc.		Decatur	Morgan	LDB	Marine	5/6/90 8:30	Std Report	2/1/90	20792
POR 1784 Decatur Al		08.10			Tennessee River MP 304 1				:	
			Mobile	Mobile	Blakeley Island at Coastal Fuel Dock	Marine	5/6/90 12:30	Std Report	9/9/9	20694
			Mobile	Mobile	Mobile River MP 1	Marine	5/4/90 12:00	Std Report	5/4/90	20566
Ross Clark Cir NW (next to 3771), Dothan,	Swifty Oil Lube	36304	Dothan	Houston	Ross Clark Cir NW (next to 3771)	Fixed	5/2/90 17:00	Std Report	5/3/90	20332
PO Box 55 Alabama Hwy 117 at MM 163, Emelle, AL 35459	Chemical Waste Management	35459	Emelle	Sumter	AL Hwy 17 at MM 163	Fixed	4/30/90 22:30	Std Report	2/1/90	19964
PO Box 29, Powder Springs, GA 30073	Marathon Oil Co.		Montgomery	Montgomery	1104 Hunter Loop Rd	Fixed	4/29/90 20:00	Std Report	2/1/90	86861
ATZN-FEE, Fort McClennan, AL 36205	US Army Fort McClellan	36205	Fort McClellan	Calhoun	Fuel point at base	Fixed	4/30/90 8:30	Std Report	4/30/90	19753
PO Box 2386, Mobile, AL 36652	National Marine, Inc.		Mobile	Mobile	Weeks Bay	Marine	4/28/90 16:00	Std Report	4/28/90	19620
	Oil Fields Services		Buhl	Tuscaloosa	1 mi S of Rte 1 Box 171	Highway	4/27/90 20:30	Std Report	4/27/90	19483
185 Spring St, Atlanta, GA 30303	Norfolk Southern Railroad		Birmingham	Jefferson	Norris Yard MP 791	Railroad	4/24/90 12:20	Std Report	4/24/90	18808
1101 Market St 3F Blue Ridge Place, Chattanooga, TN 37402	Tennessee Valley Authority		Stevenson	Jackson	Widow's Creek Fossil Plant Hwy 72 MP 407	Fixed	4/24/90 9:00	Std Report	4/24/90	18798
POB 11526, Chickasaw, AL 36611	Coastal Tug & Barge, Inc.		Mobile	Mobile	Mobile Harbor Coastal Fuels & Marketing S dock	Marine	4/24/90 4:00	Std Report	4/24/90	18732
			Mobile	Mobile	Vessel moored at LL&E terminal	Marine	4/16/90 10:30	Std Report	4/16/90	17387
103 Dauphin St. Suite 501, Mobile, AL 36602	Total Tankering		Mobile	Mobile	Alabama State Docks Pier	Marine	4/13/90 22:00	Std Report	4/14/90	17184
4900 Osborn Rd, Richmond, VA 23231	CSX Transportation		Mobile	Mobile	Seibert rail yard	Railroad	4/13/90 11:30	Std Report	4/13/90	17115
SRP Address	Suspected Responsible Party	Zip Code	City	County	Incident Location	Incident Type	Incident Date and Time Incident Type Incident Location	Call Type	Date Call Reported	NRC Report No.

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Transportation Accidents with Hazardous Materials, 1990-1999

NDC Deport No	Date Call	Cell Type	Incident Date and Time	Incident Type	Incident Location	Country	, Ali	7 in Code	Suspected Responsible	SDD Address
The metall the	Palloday	odi ino	Alabama Carta Day	add i momani	Alebama Ctota Deal: Court	Commo			6m.	DOD 1800 Mobile A1
21386	2/10/90	Std Report	5/9/90 12:00	Marine	Alabama State Dock South	Mobile	Mobile		Lott Ship Agency	PUB 1802, Mobile, AL 36602
21757	5/14/90	Std Report	5/14/90 6:40	Marine	In front of Wheeler Hydro Plant, Rte 2	Lawrence	Town Creek	35672	Tennessee Valley Authority	Lookout Place 3 S 155 FF-C, 1102 Market St., Chattanooga, TN 37402
21757	5/14/90	Std Report	5/14/90 6:40	Marine	In front of Wheeler Hydro Plant, Rte 2	Lawrence	Town Creek	35672	Tennessee Valley Authority	Lookout Place 3 S 155 FF-C, 1102 Market St, Chattanooga, TN 37402
21757	5/14/90	Std Report	5/14/90 6:40	Marine	In front of Wheeler Hydro Plant, Rte 2	Lawrence	Town Creek	35672	Tennessee Valley Authority	Lookout Place 3 S 155 FF-C, 1102 Market St, Chattanooga, TN 37402
21757	5/14/90	Std Report	5/14/90 6:40	Marine	In front of Wheeler Hydro Plant, Rte 2	Lawrence	Town Creek	35672	Tennessee Valley Authority	Lookout Place 3 S 155 FF-C, 1102 Market St, Chattanooga, TN 37402
21757	5/14/90	Std Report	5/14/90 6:40	Marine	In front of Wheeler Hydro Plant, Rte 2	Lawrence	Town Creek	35672	Tennessee Valley Authority	Lookout Place 3 S 155 FF-C, 1102 Market St., Chattanooga, TN 37402
22183	5/16/90	Std Report	00:6 06/91/5	Highway	Hwy 216, 2 mi W of 1-59	Tuscaloosa	Million Dollar Lake			
22553	5/18/90	Std Report	5/17/90 20:40	Marine	Mobile River MM 30	Mobile	Mobile		Coastal Refinery	POB 9037, Houma, LA 70360
22828	5/20/90	Std Report	5/19/90 20:30	Fixed	Wolf Creek Rd County Hwy 27	Franklin	Vandiver			Wolf Creek Rd, County Hwy 27, Vandiver, AL
23079	5/21/90	Std Report	5/21/90 15:30	Railroad	Norfolk Southern rail yard	Mobile	Mobile		Norfolk Southern Railroad	185 Spring St, Atlanta, GA 30303
23179	5/22/90	Std Report	5/22/90 8:35	Fixed	Alabama Power Co 1-65	Lamar	Vernon		Heavi-Duty Electric	POB 268, Goldsboro, NC 27530
23928	2/26/90	Std Report	5/26/90 15:10	Marine	Mobile River MM 6	Mobile	Mobile		M/V Eagle One	
24143	5/29/90	Std Report	5/29/90 8:45	Unknown	Okatuppa Creek MP 123	Choctaw	Gilberttown			

Transportation Accidents with Hazardous Materials, 1990-1999

NRC Report No. Reported Cause		Incident Description	Medium Description	Medium Affected De	Injuries Deaths Reported Reported	Damages	Evacuations	Number Evacuated	Airway Closure
21386	Equipment Failure	M/V Giant pumped bilge	Mobile River	Water		No No	o No		
21757	Unknown	Vesseloat sank at dam face	Wheeler Lake	Water	,	ž	°,		* A 700 *
21757	Unknown	Vesseloat sank at dam face	Wheeler Lake	Water		°Z	°Z.		
21757	Unknown	Vesseloat sank at dam face	Wheeler Lake	Water		N <sub>o</sub>	N <sub>o</sub>		
21757	Unknown	Vesseloat sank at dam face	Wheeler Lake	Water		°Z	Š		
21757	Unknown	Vesseloat sank at dam face	Wheeler Lake	Water		°N.	°Z.		
22183	Transport Accident	g accident	Million Dollar Lake	Water		No	No 0X		
22553	Equipment Failure	Tugboat hull crack leaking	Mobile River	Water		No	No		
22828	Dumping	Car battery residue dumped	Wolf Creek	Water		°Z	N <sub>O</sub>		
23079	Equipment Failure	Rail car (DUPX 14633) top sloshed material while in motion	Railcar & ballast Land	Land		SZ.	No		
23179	Equipment Failure	Transformer turned over in back of truck during offloading	Truck bed & concrete	Land		No	No V		
23928	Equipment Failure	M/V Eagle One	Mobile River	Water		No Vo	No		-
24143	Unknown	Oil found in river	Okatuppa Creek	Water		No	No		

Transportation Accidents with Hazardous Materials, 1990-1999

NRC Report No. Road Closure Damages (\$)	Reported  Damages (\$)	CHRIS Code	Name of Material	Unit of Quantity Spilled Measure	Quantity in Units of Water Measure	Name of Railroad	Train Number	Derailed?
21386		NUO	Bilge oil	20 gal	20 gal			
17516		M	Mathul othul Patrina	viii o	, e			
21757		OMT	Oil, misc. motor	o o	o nuk			
21757		ODS	Oil: diesel	0 unk	0 unk			
21757		UNK	Vinyl paint	O nuk	0 nuk			
21757		UNK	Waste paint	0 nuk	0 unk			
22183		NCC	Coal	0 unk	0 unk			
22553		OSX	Oil, fuel: no. 6	l gal	i gal			
22828		UNK	Unknown Material	0 unk	0 unk			
23079		SFA	Sulfuric acid	0.5 gal	0 non			
23179		PCB	Polychlorinated biphenyls	8 gal	non 0			
23928		NOO	Oil, unknown	0 unk	0 unk	and the second s		
24143		OUN	Oil, unknown	0 unk	0 unk			

Transportation Accidents with Hazardous Materials, 1990-1999

Transportation Accidents with Hazardous Materials, 1990-1999

NRC Report No.	NRC Report No. Reported Cause	Incident Description	Medium Description	Medium Affected	Injuries Deaths Reported Reported	Injuries Reported	Damages	Evacuations	Number Evacuated	Airway Closure
24341	Equipment Failure	Tank car (UTLX 84328) faulty gasket leak	Atmosphere	Air			No	No		
24358	Transport Accident	Vehicle hit pole knocking down 2 transformers	Ground	Land			No	No		
24434	Other	Rail tank car leaked while being washed	Ground	Land			No.	Z		
	Unknown	ext to	ee River	Water			No	92		
24896	Unknown	Oil sheen sighting	Mobile Bay	Water			oN S	No		
24958	Operator Error	Tractor trailer ran off road	Unknown Creek	Water			No	No O		
25107	Transport Accident	Transport Accident Transformer hit by truck	Ground	Land			No	No		
25187	Unknown	Greyhound facility	Roadway & drains	Water			No No	No		
25638	Transport Accident	Train (28 cars) derailed when train truck tractor trailer truck	Land	Land		4	4 Yes	No		
25843	Equipment Failure	Locomotive derailment/Fuel tank leaked	Gravel & dirt	Land			°Z	Š		
26027	Unknown	AV 257 tank battery leaking	Unnamed stream	Water			No No	o <sub>Z</sub>		
26618	Equipment Failure	Rolloff box fell when truck hydraulic system failed	Ground	Land			No	N <sub>o</sub>		
26692	Transport Accident Truck accident	Truck accident	Roadway	Land			No	Yes		0
26783	Transport Accident	Transport Accident Truck with used motor oil in accident	Storm sewer	Water			No	No		
26901	Other	Oil in barge when barge full of water/Oil pumped out with water	Tombigbee River Water	Water		;	S,	Š		
26946	Equipment Failure	Equipment Failure   Tank truck leaking	Roadway & ditch Land	Land			No	No		

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NRC Report No. Road Closure Damages (\$)	Closure L	Reported Damages (\$)	CHRIS Code	Name of Material	Unit of Quantity Spilled Measure		Quantity in Water	Units of Measure	Name of Railroad	Train Number	Derailed?
24341			EDA	Ethylenediamine	0	0 unk	J	0 non			
24358			PCB	Polychlorinated biphenyls (>50 ppm)	9	6 gal	0	0 non			
24434			CCT	Creosote	500 gal	gal	3	0 non		THE CASE OF THE CA	
24738			NY S	Unknown Blue Material	ō	0 nnk		0 unk			
24896			OIL	Oil: crude	100	0 unk		0 unk			
24958			SOO	Oil: diesel	0	0 unk		0 nnk			
25107			PCB	Polychlorinated biphenyls	500 gal	gal		0 non	And the second s		
25187			OTW	Oil, fuel: no. 2	0	0 unk	9	0 unk			
25638		20000	500000 OTH	Petroleum oil	23000 gal	gal		0 non			
25843			OTW	Oil, fuel: no. 2	50 gal	gal	J	0 non			
26027			OIL	Oil: crude	3000 gal	gal		0 unk			
26618			NCC	Sweeper trash, D007	10	10 gal		0 non			
2693			UNK	Unknown Material	0	0 unk		0 unk			
26783			OMT	Oil, misc: motor	0	0 unk	)	0 unk			
26901			NOO	Oil, unknown	55	55 gal	55	55 gal			
26946			UNK	Unknown red material	0	0 unk		0 non			

# Transportation Accidents with Hazardous Materials, 1990-1999

		No	No		Land	Soil	Equipment Failure Rail car hose disconnected	Equipment Failure	29737
		No	No V		Subsurface	Subsurface soil Su	Underground storage tank/Gas removed from tank and tank filled with water to do tank test	Other	29308
		N <sub>o</sub>	No		Water	Terry Cove W	M/V Porkies II	Unknown	28997
		No	3 No		Water	Mobile River W	Floating dry dock tank ruptured in explosion and fire	Other	28910
		No	3 No		Water	Mobile River W	Floating dry dock tank ruptured in explosion and fire	Other	28753
		No	No		ater	Tombigbee River Water	Barge tank overfilled	Operator Error	28460
		No	No		Water	Mobile River W	Oil bubbles coming from stern of barge assumed to have pinhole leaks/Barge capacity 12,000 BBLS	Unknown	28400
		N <sub>o</sub>	0 No	0	Water	Gulf of Mexico W	P/C Soon Came pumping bilges	Орегатог Еггог	28118
		°Z	, v		Land	Garbage Pit	Dunlop putting hazardous materials into trash/Caller had to separate hazmat so trash could be burned/Caller says he is contaminated	Dumping	27952
		No No	1 No		Unknown/ Other	'n	Van rear ended	Operator Error	27623
		S.	9Z		Water	Tennessee River W	M/V Frankie Lee sank	Other	27556
		No Vo	N <sub>o</sub>		Water	Tennessee River W	M/V Frankie Lee sank	Other	27556
		ON.	No		Land	Unknown	Tank truck overfilled	Operator Error	27367
		°Z	No V		Water	Pinhook Creek W	Gasoline found on water	Unknown	27029
Airway Closure	Number Evacuated	Evacuations	Damages	Injuries Deaths Reported Reported	Medium Affected Deaths R	Medium Medium Description Af	Incident Description	NRC Report No. Reported Cause	NRC Report No

Derailed?														
Train Number		:												
Name of Railroad													:	
Units of Measure	0 unk	0 non	20 gal	0 unk		0 110011	0 unk	0 gal	1000 lbs	500 gal	0 non	0 unk	0 unk	0 non
Quantity in Water			7						100	90				
Unit of Measure	0 unk	50 gal	20 gal	0 unk		0 nuk	0 unk	0.01 gal	1000 lbs	500 gal	0 unk	0 unk	0 unk	5000 gal
Unit of Quantity Spilled Measure								0	2					
Name of Material	Gasoline: automotive (4.23 g Pb/gal)	Naphtha: solvent	Paint	Oil: diesel		Cobalt bromide (OUS)	Waste oil/lubricants	Oil, fuel: no. 6	Sodium hydroxide	Waste oil and water mixture	Gasoline: automotive (4.23 g Pb/gal)	Oil: diesel	Gasoline: automotive (4.23 g Pb/gal)	Sulfuric acid
CHRIS	GAT	NSV	NCC	. SQO		COB	WTO	XSO	SHD	OWA	GAT	SOO	GAT	SFA
Reported Damages (\$)	:			;					THE COLUMN TO					
Road Closure									B					
REPORT No. Road Closure Damages (\$)	27029	27367	27556	27556	27623	27952	28118	28400	28460	28753	28910	28997	29308	29737

	Date Call								Suspected Responsible	
NRC Report No.	Reported	Call Type	Incident Date and Time Incident Type Incident Location	Incident Type	Incident Location	County	City	Zip Code	Party	SRP Address
									Dancka Truck Lassing	312 28th St N
29766	06/9/L	Std Report	7/6/90 9:30	Fixed	312 28th St N	Jefferson	Birmingham	35203	Co.	Birmingham, AL 35203
30078	06/6/L	Std Report	7/9/90 10:00	Railroad	MM 149 MB N of rail yard	Mobile	Mobile		Norfolk Southern Railroad	185 Spring St, Atlanta, GA 30303
		•			Mobile Bay near N Sand				Woodson Construction	PO Box 80337,
30138	06/6/L	Std Report	7/6/90 11:23	Marine	Island	Mobile	Mobile		Co.	Lafayette, LA
30140	06/6/L	Std Report	7/7/90 16:30	Marine	Mobile Bay near N Sand Island	Mobile	Mobile		Woodson Construction Co.	PO Box 80337, Lafayette, LA
30141	06/6/L	Std Renort	7/8/90 11:38	Fixed	Mobile Bay near N Sand	Mohile	Mohile		Woodson Construction	PO Box 80337,
										The same same same same same same same sam
					; ; ;				;	101 Market St 3 S 153 F
30197	7/10/90	Std Report	7/9/90 19:00	Unknown	Widow's Creek Fossil Plant Hwy 72 MP 407	Jackson	Bridgeport	37402	Tennessee Valley Authority	Lookout Place, Chattanooga, TN 37402
30312	06/01/2	Std Report	00.00/8/2	Fixed	848 Seacliff Dr	Baldwin	Fairhone		Fastern Shore Marine	848 Seacliff Dr,
		•	The state of the s		Hwy 17 & Hwy 28, 6 mi S		<b>L</b>		Jack Grav Transport	4600 E 15th Av. Garv.
30334	06/11//2	Std Report	7/10/90 21:45	Highway	of Emelle	Greene	Emelle		Inc.	IN 46403
					Over reservoir and nearby land (possibly Lake					834 ABW/DEEV, Hurlburt Field, FL
30546	7/12/90	Std Report	7/11/90 0:00	Air	Martin)	Elmore	Opelika		USAF - Hurlburt Field	32544
		- -		,		;	:			7275 Congo Rd,
30702	7/13/90	Std Report	7/12/90 14:00	Marine	Fowl River Marina	Mobile	Mobile			Theodore, AL 36582
30901	7/14/90	Std Report	7/13/90 22:30	Marine	Hwy 43 N PO Box 1028, Tombigbee River MM 216.7	Marengo	Demopolis	36732	Bertucci Barge Line	POB 10563, Jefferson, LA 70181
31526	7/18/90	Std Report	7/18/90 10:45	Marine		Mobile			Mobil Oil Co	1250 Poydras St, New Orleans, LA 70013
31950	7/21/90	Std Report	7/20/90 5:30	Fixed	Pinto Island, S end	Mobile	Mobile	36633	S.A.F.E. Inc.	PO Box 1691, Mobile, AL 36633
33038	7/28/90	Std Report	7/28/90 9:30	Offshore	Orange Beach, S half of Wolf Bay, Hwys 180 & 161	Baldwin	Orange Beach			
33188	7/30/90	Std Report	7/29/90 23:10	Railroad	Boyles Railyard	Jefferson	Birmingham		CSX Transportation	500 Water St, Jacksonville, FL 32202
33342	7/30/90	Std Report	7/28/90 17:00	Marine	Mobile Bay	Mobile	Mobile		Woodson Construction Co.	PO Box 80337, Lafayette, LA
33567	8/1/90	Std Report	8/1/90 6:50	Fixed	MM 31	Mobile	Bucks		Alabama Power Co.	POB 70, Bucks, AL 36512

Transportation Accidents with Hazardous Materials, 1990-1999

NRC Report No.	NRC Report No. Reported Cause	Incident Description	Medium Description	Medium Affected Death	Injuries Deaths Reported Reported	Damages	Evacuations	Number Evacuated	Airway Closure
29766	Other	Underground pipe leading from underground tank ruptured when truck struck fuel pump	Subsurface soil	Subsurface		 No	No		
30078	Equipment Failure	Rail car (VSCX143) bottom outlet valve leaking		Unknown/ Other		No	No		
30138	Unknown	Material washed off Woodson Lay barge no. 1	Mobile Bay	Water		 No	No		
30140	Unknown	Woodson Lay barge no. I gear box leaked	Mobile Bay	Water		 No No	ŏZ		
30141	Operator Error	Fuel system valve accidently left open after fueling welding machine	Mobile Bay	Water		No	No		
30197	Unknown	Oil discharged with source unknown	Tennessee River	Water		No	No		
30312	Equipment Failure	Equipment Failure Leaking diesel pump	Mobile Bay	Water		No	No		
30334	Transport Accident	Transport Accident Tractor trailer in accident	Roadway	Land		No	No		
30546	Equipment Failure	H-53 Helicopter/Due to engine problems, had to jettison fuel in order to perform safe landing	Reservoir	Water		N <sub>o</sub>	No.		
30702	Unknown	Vessel sunk	Fowl River	Water		Š	% %		
30901	Equipment Failure	Equipment Failure M/V Captain Anthony hose ruptured	Tennessee River Water	Water		No	N <sub>o</sub>		
31526	Equipment Failure	Barge swivel leaking	Mobile Bay	Water		No.	% %	,	
31950	Operator Error	Bucket spilled while being moved by crane	Mobile Bay	Water		N <sub>o</sub>	No		
33038	Unknown	Oil slick covering south half of Wolf Bay	Wolf Bay	Water		Š	No		
33188	Transport Accident	Tank car developed hole when bumped into another car	Soil	Land		No	°Z		
33342	Equipment Failure	Engine problems released fuel from exhaust pipe of crewboat Miss Margie	Mobile Bay	Water		No	No		
33567	Unknown	Facility intake pipes bringing in oily sheen with cooling water from Mobile River	Mobile River	Water		No	No		

Transportation Accidents with Hazardous Materials, 1990-1999

Reported No. Road Closure Damages (\$)	CHRIS Code	Name of Material	Unit of Quantity Spilled Measure	Quantity in Units of Water Measure	s of Name of Railroad	Train Number	Derailed?
29766	WLO	Oil, fuel: no. 2	700 gal	0 non			
30078	NCC	Vitropylamine	3 gal	non 0			
30138	WTO	Oil, fuel: no. 2	0 unk	0 nuk			
30140	OLB	Gear oil	0.25 gal	0.25 gal			-
30141	OTW	Oil, fuel: no. 2	2 gal	2 gal			
30197	ODS	Oil: diesel	0 unk	0 unk			
30312	ods	Oil: diesel	0 unk	0 unk			
30334	сом	Cumene	0 unk	0 unk			
30546	JPF	Jet fuel: JP-4	5000 lbs	\$000 lbs			
30702	GAT	Gasoline: automotive (4.23 g Pb/gal)	. c				
30901	ODS	Oil: diesel	30 gal	30 gal			
31526	OLB	Oil, misc: lubricating	0.99 gal	0.99 gal			
31950	ОТН	Drilling fluid	20 gal	0 unk	THE PROPERTY OF THE PROPERTY O		
33038	ODS	Oil: diesel	0 unk	0 unk			
33188	css	Caustic soda solution	0 unk	non 0			
33342	OTW	Oil, fuel: no. 2	0 unk	0 unk			
33567	Sdo	Oil: diesel	0 unk	0 unk			

Transportation Accidents with Hazardous Materials, 1990-1999

NRC Report No.	Date Call Reported	Call Type	Incident Date and Time Incident Type Incident Location	Incident Type	Incident Location	County	City	Zip Code	Suspected Responsible Party	SRP Address
33726	8/2/90	Std Report	8/2/90 9:00	Unknown	2407 Fraim Dr	St. Clair	Pell City	35125		
33899	8/3/90	Std Report	8/3/90 8:30	Railroad	MP 149MB	Mobile	Mobile		Norfolk Southern Railroad	185 Spring St, Atlanta, GA 30303
33909	8/3/90	Std Report	8/3/90 10:45	Fixed	Fowl River at Dauphin Island Pkwy at Quartering Barge	Mobile	Mobile		Offshore Pipelines, Inc.	PO Box 758, Dauphin Island, AL 36528
34032	8/4/90	Std Report	8/4/90 12:30	Fixed	ON HWY 21 TALLADEGA HWY, ROUTE 2	Talladega	Winterboro		Floyd and Beasley Transfer	Hwy 21 and Talladega Hwy, Rte 2, Winterboro, AL
34037	8/4/90	Std Report	8/4/90 12:20	Railroad	River Rd	Dallas	Selma	36702	Hammermill Paper	River Rd, Selma, AL 36702
34132	8/5/90	Std Report	8/5/90 15:00	Highway	Rte 4 Box 316	Tuscaloosa	Cottondale	35453		
34324	8/7/90	Std Report	8/7/90 4:30	Railroad	Brewton City Hall	Escambia	Brewton		CSX Transportation	PO Box 1030, Mobile, AL 36633
34966	8/11/90	Std Report	8/11/90 5:11	Marine	Blakeley Island	Mobile	Mobile		Coastal Fuel Marketing, Inc.	, Blakeley Island, Mobile, AL
34967	8/11/90	Std Report	8/11/90 5:00	Fixed	Blakeley Island Terminal Hwy 98	Mobile	Mobile	36652	Belcher Oil/Coastal Oil	Blakeley Island Terminal, Hwy 98, Mobile, AL 36652
34979	8/11/90	Std Report	8/11/90 6:52	Railroad	MP 88.3	Walker	Jasper		Norfolk Southern Railroad	185 Spring St, Atlanta, GA 30303
35245	8/13/90	Std Report	08:11.06/81/8	Highway	1-65	Conecuh	Evergreen		W.P. Ballard	1200 2nd Av N, Birmingham, AL 35203
35692	06/91/8	Std Report	8/16/90 14:05	Marine	Bender Shipyard, I-10	Mobile	Mobile		USACE	Custom House, 2nd and Chestnut, Philadelphia, PA 19106
35956	8/18/90	Std Report	8/18/90 11:30	Fixed	Rte 1	Mobile	Creola	36525	GFS Seismograph	LA
35967	8/18/90	Std Report	8/18/90 10:55	Railroad	Norris Yard MP 791	Jefferson	Birmingham		Norfolk Southern Railroad	185 Spring St, Atlanta, GA 30303
35968	8/18/90	Std Report	8/17/90 16:00	Fixed	Seacliff Dr	Baldwin	Fairhope	36532	Eastern Shore Marine	Sea Cliff Dr, Fairhope, AL 36532
36149	8/20/90	Std Report	8/20/90 10:45	Railroad	Mobile Railyard MP 149	Mobile	Mobile		Norfolk Southern Railroad	185 Spring St, Atlanta, GA 30303
36185	8/20/90	Std Report	8/20/90 15:00	Fixed	1200 2nd Av N	Jefferson	Birmingham	35203	W.P. Ballard	1200 2nd Av N, Birmingham, AL 35203
36229	8/21/90	Std Report	8/21/90 6:00	Marine	Mobile Bay Channel at Mobil Oil Slip	Mobile	Theodore		M/V Nicor Sailor	

Transportation Accidents with Hazardous Materials, 1990-1999

			Medium			ſ	٠.	Number	Airway
NKC Keport No. Keported Cause	Reported Cause		Description	Affected	Deaths Reported Reported	 Damages	Evacuations	Evacuated	Closure
32728	Tinknown	Construction site on water 200 yds from	Logan Martin	Water		Š	Ž		
	Cilwing		Lanc	11 4101		ONT	011		
33899	Unknown	Railroad tank car vent valve leaking vapor	Atmosphere	Air		No	No		
		Fuel transfer hose from tug Miss Rachel to							
33909	Equipment Failure	quartering barge Carolyn slipped	Fowl River	Water		No	No		
34032	Unknown	Tire fire		Air		No	No No		
34037	Equipment Failure	RR car dome flange	Air	Air		No	No No		
34132	Dumping	2 tanker trucks dumping loads on road in front of caller's property	Soil	Land		No	No		NOT 7 YEAR OLD THE STREET
34324	Equipment Failure	Pump stuck open, overfilling locomotive fuel tank	Roadway & ballast	Land		N <sub>o</sub>	N N		
34966	Equipment Failure	Leaking flange on barge IV1109	Mobile Bay	Water		No	N <sub>o</sub>		
		:							
34967	Unknown	Fueling line	Mobile River	Water		 No	So.		
34979	Unknown	Tank car top dome leaking	Soil	Land		No	No		
35245	Equipment Failure	After blowout, truck tire severed pipe	Soil	Land		Š	Ž		
	-								
35603	- International	TISS MAEndand Indicine honouth money	Mobile Bires	Motor		2	, , , , , , , , , , , , , , , , , , ,		
	Dumping		Dead Lake	Water		No	No		
35967	Equipment Failure	Rail tank car pressure valve leaking	Railcar	Land		No No	No No		
35968	Dumping	Foam sheeting used to protect paint dumped	Fly Creek	Water		No No	No		
36149	Unknown	Rail tank car (UTLX66437) dome leaking only when moving	Ground	Land		No	No		
36185	Equipment Failure	Tank truck overflowed while pumping from storage tank due to meter malfunction on truck	Asphalt	Unknown/ Other	managama na para-re percena	No	°Z		
36229	Operator Error	M/V Nicor Sailor receiving tank overfilled during internal fuel transfer	Mobile Bay	Water		oN S	No		

Transportation Accidents with Hazardous Materials, 1990-1999

Transportation Accidents with Hazardous Materials, 1990-1999

Date Call Reported	Call Type	Incident Date and Time	Incident Type	Incident Location	County	City	Zip Code	Suspected Responsible Party	SRP Address
8/21/90	Std Report	8/21/90 11:00	Marine	Coastal Fuel Docks, Blakeley Island	Mobile	Mobile		USNS Cape Flattery	Norfolk, VA
8/22/90	Std Report	8/22/90 16:00	Highway	1002 Hoke Av	Jefferson	Dolomite	35061	Safety Kleen Corp.	1002 Hoke Av, Dolomite, AL 35061
8/23/90	Std Report	8/22/90 16:00	Highway	Safety Kleen Corp., 1002 Hoke Av	Jefferson	Dolomite		Montgomery Tank Lines	2250 E 15th St, Gary, IN 46402
8/23/90	Std Report	8/23/90 18:15	Fixed	AL Hwy 17 at MM 163	Sumter	Emelle	35459	Chemical Waste Management	PO Box 55 Alabama Hwy 117 at MM 163, Emelle, AL 35459
8/26/90	Std Report	8/26/90 3:00	Marine	LL&E on Mobile River, E bank	Mobile	Mobile		Page and Jones, Inc.	52 N Jackson St, Mobile, AL 36602
8/26/90	Std Report	8/26/90 15:30	Railroad	Norris Yard MP 791	Jefferson	Irondale		Norfolk Southern Railroad	185 Spring St, Atlanta, GA 30303
8/26/90	Std Report	8/24/90 18:00	Fixed	7 mi E of Athens between MP 84 and MP 85 of Hwy 72	Limestone	Athens		Carriage Motor Company	7 Miles E of Athens btwn Mile 84 and 85 on Hwy 72, Athens, AL
8/27/90	Std Report	8/27/90 14:40	Fixed	McDuffic Coal Terminal Port of Mobile	Mobile	Mobile		Lott Ship Agency	259 N Conception St, Mobile, AL 36603
8/27/90	Std Report	8/27/90 14:50	Marine	McDuffie Coal Terminal, off Virginia St, Berth 2	Mobile	Mobile	1. (0) (0)	Mid Stream Fuel Service	11 Government St, Mobile, AL 36652
06/05/8	Std Report	8/29/90 16:00	Highway	Hwy 5 E of Hwy 82 on Don MacMillan Bridge	Bibb	Brant		Belcher Oil Company	
9/2/90	Std Report	9/2/90 1:30	Highway	I-59 N MM 163	St. Clair	Gadsden		CTX Trucking	Geismor, GA
9/2/90	Std Report	9/2/90 12:00	Marine	LL&E Terminal	Mobile	Mobile		Sabine Towing	PO Box 1528, Groves, TX 77619
9/2/90	Std Report	9/2/90 12:15	Marine	Industrial Pkwy	Mobile	Saraland		LL and E	Industrial Pkwy, Saraland, AL
9/3/90	Std Report	9/2/90 1:30	Highway	I-59 N MM 163	St. Clair	Ashville		Continental Transport Express	PO Box 228, Geismar, LA 70734
9/4/90	Std Report	9/4/90 8:30	Fixed	McHugh Oil Field Services 1112 DeSoto Dr	Mobile	Dauphin Island		Offshore Pipelines, Inc.	PO Box 758, Dauphin Island, AL 36528
9/4/90	Std Report	9/4/90 8:30	Fixed	McHugh Oil Field Services 1112 DeSoto Dr	Mobile	Dauphin Island		Offshore Pipelines, Inc.	PO Box 758, Dauphin Island, AL 36528
	8/21/90 8/21/90 8/22/90 8/23/90 8/26/90 8/26/90 8/26/90 8/26/90 9/2/90 9/2/90 9/2/90		Std Report	Std Report	Call Type         Incident Date and Time         Incident Type           Std Report         8/21/90 11:00         Marine           Std Report         8/22/90 16:00         Highway           Std Report         8/22/90 16:00         Highway           Std Report         8/26/90 3:00         Marine           Std Report         8/26/90 15:30         Railroad           Std Report         8/24/90 18:00         Fixed           Std Report         8/24/90 14:40         Fixed           Std Report         8/24/90 14:50         Highway           Std Report         9/2/90 15:00         Highway           Std Report         9/2/90 12:05         Marine           Std Report         9/2/90 12:05         Highway           Std Report         9/2/90 13:0         Fixed           Std Report         9/2/90 13:0         Fixed	Call Type         Incident Date and Time         Incident Type         Incident Date and Time         Incident Location           Sid Report         8/21/90 11:00         Marrine         Blakeley Island           Sid Report         8/22/90 16:00         Highway         1002 Hoke Av           Sid Report         8/22/90 16:00         Highway         Hoke Av           Sid Report         8/26/90 16:00         Marine         LL&E on Mobile River, E           Sid Report         8/26/90 16:00         Railroad         Norris Yard MP 791           Sid Report         8/26/90 16:00         Railroad         Norris Yard MP 791           Sid Report         8/24/90 18:00         Fixed         7           Sid Report         8/27/90 14:40         Fixed         Dro Abuffie Coal Terminal           Sid Report         8/29/90 16:00         Highway         LA&E Terminal           Sid Report         9/2/90 12:00         Marine         LL&E Terminal           Sid Report         9/2/90 12:00         Marine         LL&E Terminal           Sid Report         9/2/90 12:00         Highway         L-59 N MM 163           Sid Report         9/2/90 12:00         Highway         L-59 N MM 163           Sid Report         9/2/90 13:00         Highway	Call Type         Incident Date and Time         Incident Location         County           Std Report         8/21/90 11:00         Marine         Blakeley Island         Mobile           Std Report         8/22/90 16:00         Highway         1002 Hoke Av         Jefferson           Std Report         8/22/90 16:00         Highway         1002 Hoke Av         Jefferson           Std Report         8/22/90 16:00         Highway         Hoke Av         Jefferson           Std Report         8/26/90 3:00         Marine         Jefferson         Jefferson           Std Report         8/26/90 15:30         Raliroad         Norris Yard MP 791         Jefferson           Std Report         8/26/90 16:00         Highway         Prixed         AL Hwy 17 at MM 163         Sumter           Std Report         8/26/90 16:00         Marine         Bank         All Hwy 1791         Jefferson           Std Report         8/27/90 14:40         Fixed         McDuffic Coal Terminal         Mobile           Std Report         8/29/90 16:00         Highway         Like Terminal         Mobile           Std Report         9/29/0 12:05         Marine         Like Terminal         Mobile           Std Report         9/29/0 12:05         Marine	Std Report         872190 11:00         Marine         Genetal Fuel Docks, Anobile         Mobile         Anobile         Zip Code           Std Report         872190 11:00         Marine         Guestal Fuel Docks, Anobile         Mobile         Mobile         35061           Std Report         872290 16:00         Highway         1002 Hote Av         Jefferson         Dolomite         35061           Std Report         872290 16:00         Highway         1002 Hote Av         Jefferson         Dolomite         35061           Std Report         872590 16:00         Marine         Lake on Mobile River, E         Mobile         Mobile         35459           Std Report         872690 18:00         Fixed         At Hwy 17 at MM 163         Sunfer         Emelle         35459           Std Report         872690 18:00         Marine         In E of Adhers between         Anobile         Mobile         Adhers           Std Report         872690 14:40         Fixed         Andres of Hwg 85 of Hwy         Limestone         Adhers           Std Report         872790 14:40         Fixed         McDuffie Coal Terminal         Mobile         Mobile         Mobile           Std Report         97290 12:0         Marine         Highway         159 N MM 163	Std Report   St2190 11-500   Marine   Guestal Fuel Docks,   Mobile   Mobile   Mobile   Std Report   St2190 11-500   Highway   1002 Hoke Av   Lefferson   Dolomite   33661

Transportation Accidents with Hazardous Materials, 1990-1999

Closure																
Evacuated CI											300			300		
Evacuations E	No	%	No		0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0	No.	N <sub>o</sub>	ON O	o N	No	Yes	No	No	Yes	°N,	No
Damages	No	No	No		0		N <sub>O</sub>	No	No	No	Unknown	No	No	No	Š	No
Reported																
Deaths Reported Reported											0					
Affected	Water	Land	Land	-	Land	Land	Air	Water	Water	Land	Land	Water	Water	Land	Water	Water
Description	Mobile River	Ground	Soil		Aspnant Mobile River	Land	Atmosphere	Mobile Bay	Mobile River	Soil	Pavement & soil	Intracoastal Waterway	Chickasaw Bogue Water	Roadway	Bay Aloe, Mississippi Sound	Bay Aloe, Mississippi Sound
	USNS Cape Flattery starboard stem struct bearing possibly leaking	Tanker truck release during transfer	Tank truck fitting leaking	-	MA FOILITICALLITE III DOX OTORE GUITING SINDINEII A	leaking during motion	Burning used tires on open lot	Transfer hose from pier improperly connected	Fuel released during transfer from T/B MF727 to M/V Lady Bird	Caller discovered gas truck dumping oil	Tanker truck overturned	Pinhole leak in no. 2 tank of barge STCO 211	Pinhole leak in hull of barge STCO-211	MC307 Cargo tank dome leak	Crane fell in water while loading out on barge	Crane fell in water while loading out on barge
	L Equipment Failure p	Operator Error	Equipment Failure 1		Operator Error	ure	Other	Operator Error	F Operator Error	Dumping	Transport Accident	Unknown	Equipment Failure	Transport Accident	Unknown	Unknown
NRC Report No. Reported Cause	36272 E	36485	36628 E		36974	37029 E	37037	37176	37181 C	37671	37891	37898	37899 I	37932	38083	38083

Transportation Accidents with Hazardous Materials, 1990-1999

Transportation Accidents with Hazardous Materials, 1990-1999

										T						
SRP Address	PO Box 1220, Orange Beach, AL 36561	185 Spring St, Atlanta, GA 30303	PO Box 1309, Kenner, LA 70063	Chattanooga TVA Office Complex, Chattanooga, TN		Green Wood, FL	4800 S Old Peachtree Rd, Norcross, GA 30071	1002 Hoke Av, Dolomite, AL 35061	Birmingham Municipal Airport, Birmingham, AL 35212		POB 2248, Decatur, AL 35602	PO Box 55 Alabama Hwy 117 at MM 163, Emelle, AL 35459	185 Spring St, Atlanta, GA 30303	USNS Range Sentinel	1250 Poydras St, New Orleans, LA 70013	3300 Ball St, Birmingham, AL 35234
Suspected Responsible Party	Zeke's Landing Marina	Norfolk Southern Railroad	Shell Oil Company	Tennessee Valley Authority	Shrimp Boat Miss Dana	Shane Gardner Co.	Safety Kleen Corp.	Safety Kleen Corp.	Northwest Airlines	Tug Sibley and Barge	Bungee Corporation	Chemical Waste Management	Norfolk Southern Railroad	US Navy	Mobil Oil Co	Ashland Chemical Co.
Zip Code									35232			35459				35234
City	Orange Beach	Birmingham		Bridgeport	Mobile	Spring Hill	Brewton	Anniston	Birmingham	Akron	Decatur	Emelle	Decatur	Mobile		Birmingham
County	Baldwin	Jefferson	Mobile	Jackson	Mobile	Russell	Escambia	Calhoun	Jefferson	Greene	Morgan	Sumter	Morgan	Mobile	Mobile	Jefferson
ncident Location	26619 Perdido Beach Blvd	Norris Yard MP 791		Widow's Creek Facility MP 415	Marina on Dauphin Island Pkwy	SW corner of county	Kenny's Yamaha, 338 Douglas Av	Calhoun Disposal 1106 Old Gadsden Hwy	Birmingham Municipal Airport Concourse B-2	Black Warrior River MM 270	1400 Market St NE	AL Hwy 17 at MM 163 PO Box 55	Decatur Railyard MP 363A	Bendor Shipyard no. 9, 265 S Water St		3300 Ball St
Incident Type	Fixed	Railroad	Offshore	Fixed	Marine	Air	Highway	Fixed	Air	Marine	Marine	Highway	Railroad	Marine	Offshore	Fixed
Incident Date and Time Incident Type Incident Location	9/14/90 10:40	9/16/90 20:50	9/17/90 8:45	9/18/90 17:15	9/17/90 14:00	9/20/90 13:20	9/18/90 10:30	10/2/90 11:30	06/9/01	00:81 06/L/01	10/9/90 8:30	10/9/90 10:30	10/9/90 14:10	10/9/90 20:44	10/10/90 7:30	10/11/90 8:00
Call Type	Std Report	Std Report	Std Report	Std Report	Std Report	Std Report	Std Report	Std Report	Std Report	Std Report	Std Report	Std Report	Std Report	Std Report	Std Report	Std Report
Date Call Reported	9/14/90	9/11/90	9/11/60	0/18/60	06/61/6	9/20/90	9/21/90	10/2/90	06/9/01	10/8/90	06/6/01	10/9/90	10/9/90	10/9/90	10/10/90	06/11/01
NRC Report No.	39332	39552	39572	39831	39930	40129	40400	41972	42596	42679	42832	42882	42918	42953	42993	43207

Transportation Accidents with Hazardous Materials, 1990-1999

										Т	1	1	T	1	1	
Airway Closure																
Number Evacuated			1911									A. A				
Evacuations	N <sub>O</sub>	No	No	°Z	oN ON	No	No	°Z	No No	No No	No	No No	Š.	No	No	No
Damages	No	No	S <sub>o</sub>	No	No	No	No	N <sub>o</sub>	N <sub>o</sub>	No No	o <u>X</u>	No.	No No	S S	N <sub>O</sub>	N <sub>o</sub>
Injuries Reported											5		AND THE PARTY OF T			
Injuries Deaths Reported Reported																
Medium Affected	Water	Air	Water	Water	Water	Land	Land	Land	Land	Water	Water	Land	Land	Water	Water	Land
Medium Description	Cotton Bayou	Atmosphere	Gulf of Mexico	Tennessee River	Fowl River	Forested area	Ground	Concrete & gravel	Concrete surface	Black Warrior River	Tennessee River	Soil	Soil	Mobile Bay	Mobile Bay	Soil
Incident Description D	Fuel pump hose nozzle leaking due to pressure buildup	Tank car GATX-26769 dome leaking	M/V New Venture cable hole	Stored oil filters seeped from storage area	M/V Miss Dana sunk at dock	Crop duster crashed	16-gal drum damaged during unloading	C Transport Accident 16-gal drum tipped over	Left wing fuel tank of Boeing 727/Accidentally overfilled fuel tank.	Oil sheen observed around tug Sibley and barge R	Barge Uluage cargo compartment overfilled	Tank truck valve left open	Tank car CELX1055 bottom valve faulty	Transfer hose clamp came loose while taking on fuel from fuel barge	Transfer hose release	Operator hooked up wrong line while transferring material to storage tank
	Equipment Failure	Equipment Failure	Equipment Failure	Other	Other	Transport Accident	Operator Error	Fransport Accident	Operator Error	Unknown	Operator Error	Operator Error	Equipment Failure	Equipment Failure	Operator Error	Equipment Failure
NRC Report No Reported Cause	39332	39552	39572 I	39831	39930	40129	40400	41972	42596	42679	42832	42882	42918	42953	42993	43207

Transportation Accidents with Hazardous Materials, 1990-1999

Transportation Accidents with Hazardous Materials, 1990-1999

SRP Address	Houston, TX	PO Box 3064, Mobile, AL 36652	AB 44, Marshall Space Flight Center, Huntsville, AL 35812	PO Box 646, Venice, LA 70038	POB 80337, Lafayette, LA 70598	No. 3 Mobile Infirmary Cir, Mobile, AL 36607	185 Spring St, Atlanta, GA 30303		Conception St Rd, Mobile, AL 36602	Box 3064, Mobile, AL 36652	185 Spring St, Atlanta, GA 30303	Houma, LA	PO Box 340, Dauphin Island, AL 36528	PO Box 24999, Greenville, SC 29616		PO Box 55 Alabama Hwy 117 at MM 163, Emelle, AL 35459
Suspected Responsible Party	Woodson Engineering	Radcliff Marine	NASA	Chevron USA	Woodson Construction Co.	Mobile Infirmary Medical Center	Norfolk Southern Railroad	At the state of th	Gulf Lumber Company	Radcliff Marine	Norfolk Southern Railroad	Lebeouf Brothers Towing Company	Woodson Construction Co.	Grace Transportation Svc.		Chemical Waste Management
Zip Code		E de la constante de la consta	35812			36607										35459
City	Mobile	Dauphin Island	Marshall Space Flight Center		Mobile	Mobile	Irondale	Dauphin Island	Mobile	Mobile	Decatur	Mobile	Dauphin Island	Mobile	Mobile	Emelle
County	Mobile	Mobile	Madison	Mobile	Mobile	Mobile	Jefferson	Mobile	Mobile	Mobile	Morgan	Mobile	Mobile	Mobile	Baldwin	Sumter
Incident Type Incident Location	1 mi S of Dauphin Island at mouth of Mobile Bay	E end of Dauphin Island	Saturn 5 Blvd		N of Sand Island	No. 3 Mobile Infirmary Cir	Norris Yard MP 791	Across from McHugh's Dock on DeSoto St	Gulf Lumber Co., Conception St	Gulf Lumber Co on Conception Rd	Norfolk Southern yard MP 363A	Amarada Hess Corp., Magazine Point	Mobile Bay 3000 ft S of Mobil Platform 76 Aux	1-65	I-10 near exit 98, W of City, Mobile Bay	AL Hwy 17 at MM 163
Incident Type	Marine	Marine	Fixed	Offshore	Offshore	Fixed	Railroad	Marine	Highway		Railroad	Marine	Marine	Highway	Highway	Highway
Incident Date and Time	10/13/90 7:30	10/13/90 20:50	10/19/90 10:00	10/22/90 16:30	10/8/90 23:00	10/25/90 7:45	10/28/90 3:00	10/30/90 13:30	11/1/90 1:00	11/1/90 0:33	11/5/90 13:30	11/6/90 8:45	11/7/90 1:45	11/9/90 2:30	08:30	11/12/90 9:15
Call Type	Std Report	Std Report	Std Report	Std Report	Std Report	Std Report	Std Report	Std Report	Std Report	Std Report	Std Report	Std Report	Std Report	Std Report	Std Report	Std Report
Date Call Reported	10/13/90	10/13/90	10/19/90	10/22/90	10/23/90	10/25/90	10/28/90	10/30/90	11/1/90	11/1/90	11/5/90	06/9/11	11/7/90	11/9/90	11/9/90	11/12/90
NRC Report No.	43469	43477	44298	44626	44702	44990	45341	45660	45863	45891	46491	46559	46718	46996	47025	47342

Transportation Accidents with Hazardous Materials, 1990-1999

NRC Report No. Road Closure Damages (\$)	Reported re Damages (\$)	CHRIS Code	Name of Material	Unit of Quantity Spilled Measure	Quantity in Units of Water Measure	Name of Railroad	Train Number	Derailed?
43469		OTW	Oil fuel no 2	803	\$ 603			
43477		WTO	Oil, fuel: no. 2	S gal	S gal			
44298		GAT	Gasoline: automotive (4.23 g Pb(gal)	15 gal	0 non			
44626		ОТН	Oil-based mud	20 gal	20 gai			1
44702		ОНУ	Oil: hydraulic oil	20 gal	20 gal			
44990		PCB	Polychlorinated biphenyls	20 gal	0 non			
45341		HCL	Hydrochloric acid	0.99 gal	uou 0			
45660		OUN	Oil, unknown	0 unk	0 unk			
45863		GAT	Gasoline: automotive (4.23 g Pb/gal)	100 gal	100 gal		and the second property of the second	
45891		GAT	Gasoline: automotive (4.23 g Pb/gal)	100 gal	uou 0			
46491		NCC	Octyl mercaptas	1.5 gal	non 0			
46559		SGO	Oil: diesel	2 gal	2 gal			
46718		ОНУ	Oil: hydraulic oil	lag.	l ga			
46996		HCL	Hydrochloric acid	50 gal	0 non			
47025		GAT	Gasoline: automotive (4.23 g Pb/gal)	0 unk	0 unk			
47342		NCC	Lead, unknown type	2 gal	non 0			

NRC Report No.	Date Call Reported	Call Type	Incident Date and Time		Incident Type Incident Location	County	City	Zip Code	Suspected Responsible Party	SRP Address
47983	11/16/90	Std Report	11/16/90 18:15	Marine	Mobile Bay near Point Monluis Island, inlet of Fowl River	Mobile	Belle Fountain		Steiner Ship Yard	Bayou La Batre, AL
48035	06/21/11	Std Report	11/17/90 9:30	Marine	Mobile Bay 1/8 mi W of Mobile Platform auxiliary 76	Mobile	Mobile		Woodson Construction Co.	PO Box 340, Dauphin Island, AL 36528
48235	06/61/11	Std Report	11/1/90 15:00	Highway	AL Hwy 17 at MM 163	Sumter	Emelle		Matlack, Inc.	1413 Folk Rd, Wilmington, DE 19989
48597	11/23/90	Std Report	11/22/90 21:17	Магіпе	Pascagoula River	Jackson	Pascagoula			
48601	11/23/90	Std Report	11/23/90 7:55	Marine	Coastal Fuel Docks, Blakeley Island	Mobile	Mobile	36652		POB 190352, Mobile, AL 36619
48602	11/23/90	Std Report	11/23/90 7:55	Marine	Coastal Fuel Docks, Blakeley Island	Mobile	Mobile		Coastal Tug & Barge, Inc.	870 W Flagler, Miami, FL
48664	11/23/90	Std Report	11/23/90 14:23	Marine	W side of Dutch Harbor	Unknown				
48924	11/26/90	Std Report	11/26/90 14:30	Highway	Redstone Arsenal US Army Missile Command	Madison	Redstone Arsenal	35898	US Army Missile Command	US Army Missile Command ATTN: AMSMI-EQ, Redstone Arsenal, AL 35989
48999	11/27/90	Std Report	11/26/90 15:35	Offshore	Mobile Bay	Mobile			Mobil Oil Co	1250 Poydras St, New Orleans, LA 70013
49055	11/27/90	Std Report	11/27/90 10:35	Marine	Bayou La Batre Gulf City Fish Docks	Mobile				
49056	11/27/90	Std Report	11/27/90 13:17	Marine	Bayou La Batre International Oceanic Enterprises, Inc. Dock	Mobile				
49173	11/28/90	Std Report	11/27/90 16:00	Fixed	Metfuel Co. Hwy 69 & Maxwell Crossing	Tuscaloosa	Tuscaloosa		McKenzie Tank Lines	POB 1200 122 Appleyard Dr, Tallahassee, FL 32304
49240	11/28/90	Std Report	11/28/90 16:50	Marine	Fairway Field	Mobile			Shell Offshore, Inc.	POB 4464, Houston, TX 77210
49250	11/28/90	Std Report	11/28/90 20:00	Highway	E Bank of Mobile River above Cochran Bridge site	Mobile	Mobile		Smith Trucking Co.	POB 226, Cleveland, SC 29635
49487	11/30/90	Std Report	11/30/90 12:30	Railroad	MP 804.2	Monroe	Fountain		Burlington Northern Railroad	3253 E Chestnut Expressway, Springfield, MO 65802

Transportation Accidents with Hazardous Materials, 1990-1999

				T									İ		
Airway Closure															
Number Evacuated															
Evacuations	No	Š.	N <sub>o</sub>	No	N <sub>o</sub>	N <sub>o</sub>	S <sub>S</sub>	ν̈́	No	No	No No	S N	No	No No	No
Damages	No	No	No	No	N <sub>o</sub>	N <sub>O</sub>	No	2 No	No	SZ.	No	No.	No No	No	No
Injuries Reported								2				_			
Injuries Deaths Reported Reported															
Medium Affected	Water	Water	Unknown/ Other	Water	Water	Water	Water	Land	Water	Water	Water	Land	Water	Water	Unknown/ Other
Medium 1 Description	Mobile Bay	Mobile Bay	Trailer Insides	Pascagoula River Water	Mobile River	Mobile River	Iliuliuk Bay	Asphalt parking lot	Mobile Bay	Bayou La Batre	Bayou La Batre		Gulf of Mexico	Mobile River	
N Incident Description	2 barges/Valve may have been opened	Work barge (8 ft by 42 ft) sank/Hydraulic unit leaked	Transport Accident 55-gal drum punctured during transit	Equipment Failure F/S Two Fools bilge pump discharged overboard P	T/B Coastal barge 35 developed hole in hull	T/B Coastal barge 35 developed hole in bottom	55-gal drum smashed between T/B Wayueh Jireh and F/V Alaska Ocean	40-gal drum tipped over in back of parked delivery A vehicle	Oil hose cap came off	M/V Amy Leshay sunk	M/V Capt. Dawn pumping bilge	Tank fell over during loading	Containment dike around vent line on M/V Sage/Rough seas	Transport Accident Truck fuel tank caught on ferry ramp	Transport Accident 5 tankcars and 1 flatcar (6 units total) derailed
	Other	Other	Fransport Accident	Equipment Failure	Equipment Failure	Equipment Failure	Unknown	Other	Other	Other	Dumping	Equipment Failure	Other	Transport Accident	Transport Accident
NRC Report No. Reported Cause	47983	48035	48235	48597	48601	48602	48664	48924	48999	49055	49056	49173	49240	49250	. 49487

Transportation Accidents with Hazardous Materials, 1990-1999

Reported No. Road Closure Damages (\$)	Reported losure Damages	es (\$) Code	Name of Material	Unit of Quantity Spilled Measure		Quantity in Water	Units of Measure	Name of Railroad	Train Number	Derailed?
47983		NJO	Oil, unknown	0 unk		0	0 unk			
48035		YHO	Oil: hydraulic oil	0 unk		0	0 unk			
48235		NCC	Alkaline corrosive material	25 gal		0	0 non			
48597		WTO	Waste oil/lubricants	0 unk	-24	0	0 unk			
48601		XSO	Oil, fuel: no. 6	l gal		1	l gal			
48602		XSO	Oil, fuel: no. 6	2 gal		2	2 gal			
48664		OUN	Oil, unknown	55 gal		55 gal	gal			
48924		NCC	Polyoxypropylenediamine	4.5 gal		0	0 non			
48999		OMT	Oil, misc: motor	0.99 gal	-	0.99 gal	gal			
49055		NUO	Oil, unknown	0 unk		0	0 unk			
49056		NNO	Oil, unknown	0 unk	*	0	0 mk			
49173		HTA	Potassium hydroxide	1200 gal	1	0	ou ou			
49240		SOOS	Oil: diesel	0.99 gal	1	0.99 gal	gal			
49250		SOO	Oil: diesel	50 gal		8	3 gal			
49487		CSS	Caustic soda solution	0 non	ŭ	0	0 non			

Transportation Accidents with Hazardous Materials, 1990-1999

23002	Сотрапу		INIODING	INIOUIIC	MULIUIN SOUMOIN IAIL JANA	TABILLORG	11.01 10.111	andau mo	171111	
POB 2204, Decatur, AL	Monsanto Chemical		Mobile	Mobile	Norfoll Southern rail ward	Pailroad	1/7/91 10:41	Std Report	10/7/1	53914
185 Spring St, Atlanta, GA 30303	Norfolk Southern Railroad		Mobile	Mobile	Rail yard	Railroad	1/7/91 10:45	Std Report	1/2/91	53901
Orleans, LA 70013	Mobil Oil Co			Moone		Maline	17/71 14:43	noday me	111121	12024
1250 Poydras St, New	O ii O			Mobile		Marine	1/5/91 14:45	Std Report	16/2/1	53894
	T/V R. Hal Dean			Unknown	Pascagoula Ship Channel	Marine	1/2/91 8:40	Std Report	1/2/91	53250
Hwy 98, Mobile, AL 36603	Coastal Fuel Marketing, Inc.	36603	Mobile	Mobile	Coastal Marketing, Inc., Blakely Island	Marine	1/2/91 5:40	Std Report	1/2/91	53224
	Battleship		Mobile	Mobile	Battleship Park	Marine	12/21/90 13:00	Std Report	12/21/90	51969
POB 65, Mobile, AL	USS Alabama								:	1
New Orleans, LA	Towing Company		Saraland	Mobile	Hwy 158	Marine	12/20/90 22:30	Std Report	12/21/90	51936
	Lebeouf Brothers				Chickasaw Creek at La Land Exploration Dock					
POB 391, Ashland, KY 41114	Ashland Oil Company		Stevenson	Jackson	Tennessee River MP 405.2	Marine	12/17/90 18:00	Std Report	12/17/90	51437
515 S Post Oak St, Houston, TX 77027	Western Co. NA		Tuscaloosa	Tuscaloosa	Hwy 69 N	Highway	12/16/90 9:30	Std Report	12/17/90	51345
PO Box 55 Alabama Hwy 117 at MM 163, Emelle, AL 35459	Chemical Waste Management		Emelle	Sumter	AL Hwy 17 at MM 163	Highway	12/12/90 20:15	Std Report	12/13/90	50948
POB 61122, New Orleans, LA 70161	Shell Offshore, Inc.			Mobile	2 mi S of Dauphin Island	Offshore	12/13/90 9:20	Std Report	12/13/90	50887
Hwy 98, Mobile, AL 36603	Coastal Fuel Marketing, Inc.	36603	Mobile	Mobile	Hwy 98	Marine	12/11/90 21:00	Std Report	12/11/90	50730
	KW Plastics	35459	Emelle	Sumter	AL Hwy 17 at MM 163	Highway	12/11/90 9:00	Std Report	12/11/90	15905
4600 River Rd, Columbus, GA 21991	Columbus Mills		Phenix City	Russell	N Bypass GA/AL state line	Highway	12/3/90 12:30	Std Report	12/3/90	49737
3253 E Chestnut Expressway, Springfield, MO 65802	Burlington Northern Railroad		Fountain	Monroe	MP 804.2	Railroad	11/30/90 12:30	Std Report	11/30/90	49487
3253 E Chestnut Expressway, Springfield, MO 65802	Burlington Northern Railroad		Fountain	Monroe	MP 804.2	Railroad	11/30/90 12:30	Std Report	11/30/90	49487
SRP Address	Suspected Responsible Party	Suspe Zip Code Party	City	County	Incident Location	Incident Type	Incident Date and Time Incident Type Incident Location	Call Type	Date Call Reported	NRC Report No.

Transportation Accidents with Hazardous Materials, 1990-1999

Transportation Accidents with Hazardous Materials, 1990-1999

Transportation Accidents with Hazardous Materials, 1990-1999

NRC Report No.	Date Call Reported	Call Type	Incident Date and Time Incident Type Incident Location	Incident Type	Incident Location	County	City	Zip Code	Suspected Responsible Party	SRP Address
54105	16/8/1	Std Report	00:61 16/8/1	Marine	Chickasaw Port Authority	Mobile	Chickasaw		T/S Star	Chickasaw, AL
54151	1/9/91	Std Report	1/8/91 11:30	Fixed	Loran 3017.4 8744.4	Baldwin	Gulf Shores		Boat and Propeller Shop	
54188	16/6/1	Std Report	1/8/91 11:55	Marine	Midstream fuel dock MP	Mobile	Mobile		Mid Stream Fuel Service	11 Government St, Mobile, AL 36652
54780	1/13/91	Std Report	1/13/91 16:30	Marine	Theodore Industrial Canal, Ideal Cement Co.	Mobile	Theodore	36609		
54867	1/14/91	Std Report	1/14/91 8:40	Fixed	Paper Mill Rd	Mobile	Mobile	36692	International Paper Company	PO Box 2448 Paper Mill Rd, Mobile, AL 36692
55244	1/16/91	Std Report	1/15/91 14:30	Marine	Alabama State Dock Pier 5	Mobile	Mobile	36651	Bulk Shipping, Inc.	Sneeuwbeslaan 14, Antwerp, Netherlands
55443	16/21/1	Std Report	1/16/91 17:00	Fixed	AL Hwy 231 MP 48	Dale	Ozark	36361	Ozark Truck Stop, Inc.	PO Box 1669, Ozark, AL 36361
56112	1/22/91	Std Report	1/22/91 15:05	Railroad	Rail yard	Jefferson	Birmingham		Norfolk Southern Railroad	185 Spring St, Atlanta, GA 30303
56113	1/22/91	Std Report	1/22/91 17:00	Offshore	Alabama State Docks at foot of Water St.	Mobile	Mobile		US Navy	
56586	1/25/91	Std Report	1/25/91 15:00	Fixed	2429 N 19th St	Jefferson	Bessemer		Crown Central	11 W Oxmoor Rd, Homewood, AL 35209
56745	1/27/91	Std Report	1/27/91 11:30	Marine	1101 Ezre Trice Blvd Dock Facility	Mobile	Mobile	36603	Pacific Molasses Co.	1101 Ezra Trice Blvd, Mobile, AL 36603
56765	1/28/91	Std Report	1/28/91 0:01	Marine	Douglas Oil Co. at Mobile Harbor Pier	Mobile	Mobile		Barge Transport Co.	12941 I-45 N, Houston, TX 77060
56766	1/28/91	Std Report	1/28/91 0:40	Marine	Mobile Harbor	Mobile	Mobile		Barge Transport Co.	12941 I-45 N, Houston, TX 77060
59064	2/12/91	Std Report	2/11/91 16:00	Fixed	Alba Rd	Mobile	Coden	36523	Rodriguez Ship Builders	Alba Rd, Coden, AL 36523
59440	2/14/91	Std Report	2/14/91 0:15	Marine	Hwy 98, Blakeley Island Coastal Fuels pier	Mobile	Mobile			
60159	2/19/91	Std Report	2/19/91 7:00	Fixed	Hwy 20 W	Morgan	Decatur	35602	McPherson Oil Co.	Hwy 20 W, Decatur, Al 35602
60709	2/22/91	Std Report	2/22/91 8:45	Fixed	Hwy 98, Blakeley Island	Mobile	Mobile	36633	Coastal Fuel Marketing, Inc.	Hwy 98 Blakeley Island, Mobile, AL 36633

Transportation Accidents with Hazardous Materials, 1990-1999

NRC Report No.	NRC Report No. Reported Cause	Incident Description	Medium Description	Medium Affected D	Injuries Deaths Reported	Damages	Evacuations	Number Evacuated	Airway Closure
54105	Dumping	T/S Star pumping bilges 1 - 2 times weekly	Mobile River	Water		No	No		
54151	Unknown	Caller saw oil sheen behind facility		Water		No	No		
54188	Operator Error	1-gal sample container dropped overboard	Mobile River	Water		N <sub>o</sub>	No		1
54780	Unknown	Oil on deck of tank barge flowed into water	Theodore Industrial Canal	Water		No	No	,,	
54867	Equipment Failure	Bearing in primary air fan failed and caused shutdown of system	Air	Air		No	No		
55244	Operator Error	Painting vessel hull spilling paint into water	Mobile River	Water		No	No.		
55443	Operator Error	Flushing fuel tanks	Clay	Land		No	No		
56112	Equipment Failure	Tank car plug loose	Atmosphere	Air		No	No		
56113	Unknown	Caller reports sheen sighting near Sealift Command Naval Vessel	Mobile River	Water		No	No		
56586	Unknown	Service station gas pump	Concrete	Land		Š	o <u>N</u>		
56745	Equipment Failure	Equipment Failure Leaking cargo hose on barge no. PAVERSYT 302	Mobile Bay	Water		%	No		
56765	Equipment Failure	T/B Danielle pinhole leak below waterline	Mobile Harbor	Water		No	No		A CAMPAN CONTRACTOR CO
56766	Unknown	Tank barge Danielle	Mobile River	Water		No	No		
59064	Other	Paint boat on water releasing paint into water	Coden Bayou	Water		No	No		
59440	Dumping	Barge Apex 3603 pumping void tank	Mobile River	Water		S.	o N		
60159	Natural Phenomenon	Heavy rain caused oil water separator to overflow	Bettye Rye Branch > Tennessee River	Water		N <sub>o</sub>	°Z		
60709	Equipment Failure	Oil water separator storage tank pump failure	Storm drain	Water		Š.	No		

Transportation Accidents with Hazardous Materials, 1990-1999

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NRC Report No.	Date Call Reported	Call Type	Incident Date and Time Incident Type Incident Location	Incident Type	Incident Location	County	City	Suspe Zip Code Party	Suspected Responsible Party	SRP Address
61107	2/25/91	Std Report	2/25/91 13:40	Marine		Baldwin				
61849	3/2/91	Std Report	3/2/91 15:00	Fixed	State Docks Rd	Russell	Phenix City	36868		
62208	3/5/91	Std Report	3/5/91 10:50	Marine	Кет McGee Dock, Blakeley Island	Mobile	Mobile		Brown Marine Service, Inc.	POB 1415, Pensacola, FL 32956
62393	3/6/91	Std Report	3/6/91 5:05	Fixed	Hwy 10 E	Wilcox	Pine Hill		MacMillan Bloedel	POB 336, Pine Hill, AL 36769
62978	3/11/91	Std Report	3/11/91 10:43	Marine	Abba Av	Mobile	Coden	36523	Rodriguez Ship Builders	Abba Av, Coden, AL 36523
63372	3/13/91	Std Report	3/6/91 4:00	Highway	Hwy 134 & West Bypass	Coffee	Enterprise		Sikorsky Aircraft Services	6900 Main St, Stratford, CT 06601
63791	3/16/91	Std Report	3/16/91 12:45	Air	Airport Terminal Concourse C-2	Jefferson	Birmingham		Hanger One	4725 65th Pl N, Birmingham, AL 35206

			Medium	Medium	I	Injuries			Number	Airway
NRC Report No	NRC Report No. Reported Cause	Incident Description	Description	Affected	Deaths Reported Reported	Reported	Damages	Evacuations	Evacuated	Closure
61107	Equipment Failure	Equipment Failure M/V Ship Island transfer line leaked residue	Gulf of Mexico	Water			S <sub>o</sub>	°Z		
		Sheen sighted during transfer operation from T/B	Chattahoochee							
61849	Unknown	TB 2010	River MM 153	Water			No	No		
62208	Unknown	Barge Brown 290	Mobile River	Water			o <sub>N</sub>	No		
62393	Equipment Failure	Equipment Failure Pipe leak due to corrosion	Air	Air			No	No		
62978	Dumping	M/V Valiant Lady pumping out bilge	Coden Bayon	Water			No	No		
63372	Transport Accident	Transport Accident Tank truck in accident	Soil & asphalt	Land			Yes	N OX		
63791	Operator Error	Aircraft Overfill	Tarmac	Land			N <sub>o</sub>	N <sub>o</sub>		

Reported No. Road Closure Damages (\$)	Road Closure	Reported Damages (\$)	CHRIS Code	Name of Material	Unit of Quantity Spilled Measure		Quantity in Units of Water Measure	Units of Measure	Name of Railroad	Train Number	Derailed?
61107			SGO	Oil: diesel	0.99 gal	gal	0.99 gal	gal			
61849			ОТН	Oil: Feed stock oil	1	gal	-	gal			
62208			OUN	Oil, unknown	0	0 unk	0	0 unk			
62393			CLX	Chlorine	09	60 lbs	0	0 non			
62978			WTO	Waste oil/lubricants	0	0 nnk	0	0 unk			
63372		100000 JPF	JPF	Jet fuel: JP-4	280 gal	gal	0	0 non			
63791			JPV	Jet fuel: JP-5 (kerosene, heavy)	75	75 gal	0	0 unk			

## Appendix B. Multiple Chemical Spills Sorted by Location (locations having greater than two incidents shown)

No. of						
Incidents	Incident Location	Incident 1 Chemicals	Incident 2 Chemicals	Incident 3 Chemicals	Incident 4 Chemicals	Incident 5 Chemicals
75	Battleship Pkwy (No. 2703), Mobile, Mobile	Oil, fuel: no. 5 (0.07 gal)	Oil, fuel: no. 5 (0.06 gal)	Oil, fuel: no. 6 (0.13 gal)	Other oil (unknown)	Oil, fuel: no. 5 (0.03 gal)
49	Hwy 17 at MM 163, Emelle, Sumter	Boiler fly ash (40 lbs)	D004, D008, D009, D0011, D0019 (0.25 lbs)	Waste alkaline (5 gal)	D004 (5 gal)	Paint filters, D007 (1 gal)
88	MP 791 (Norris Yard), Birmingham, Jefferson	Ammonia, anhydrous (unknown)	Butadiene (unknown)	Toluene 2,4-diisocyanate (0.99 gal)	Hazardous waste solid (3 lbs)	Liquefied petroleum gas (1 gal)
50	Tenn-Tom Waterway MM 216 (Demopolis Yacht Basin), Hwy 43 N, Demopolis, Marengo	Oil: diesel (0.25 gal)	Oil, fuel: no. 2-D (0.02 gal)	Gasoline: automotive (4.23 g Pb/gal) (3 gal)	Oil: diesel (0.5 gal)	Oil, fuel: no. 2-D (0 03 gal)
44	Alabama State Dock, Mobile, Mobile	Oil, misc: lubricating (2.6 gal)	Oil, misc: lubricating (3 gal)	Oil, fuel: no. 6 (5 gal)	Other oil (unknown)	Oil, unknown (unknown)
32	MP 401-A (Sheffield Railyard), Sheffield, Colbert	Nitrogen, liquified fertilizer (40 gal)	Waste oil/lubricants (15 gal)	Oil, fuel: no. 2-D (400 gal)	Phosphoric acid (0.13 gal)	Oil: hydraulic oil (30 gal)
25	Water St S (No. 265) (Bender Shipyard), Mobile, Mobile	Oil: crude (7 gal)	Waste oil (30 gal)	Oil: diesel (15 gal)	Oil: hydraulic oil (0.99 gal)	Oil: diesel (50 gal)
24	Industrial Pkwy (LL&E Dock), Saraland, Mobile	Gasoline: automotive (4.23 g Pb/gal) (1 gal)	Waste oil (40 gal)	Jet fuel: JP-12 (unknown)	Gasoline: automotive (unleaded) (120 gal) Oil: diesel (6 gal)	Oil: diesel (6 gal)
23	Blakeley Island (Atlantic Martne), Mobile, Mobile	Other oil (IFO) (125 gal)	Oil, unknown (unknown)	Waste oil (75 gal)	Oil: hydraulic oil (mixed with water) (3	Oil, fuel: no. 2-D (0.25 gal)
21	Blakeley Island (Coastal Fuel Docks), Mobile, Mobile	Oil, unknown (unknown)		Waste oil (2 gal)	fuel: no. 6 (2 gal)	Oil, fuel: no. 6 (1 gal)
20	MP 363.4A (Railyard), Decatur, Morgan	Sodium hydroxide (10 gal)	Octyl mercaptas (1.5 gal)	Ethylene glycol (unknown)	Sewage (unknown)	Oil, fuel: no. 2-D (20 gal)
17	MP 149 MB (Mobile railyard), Mobile, Mobile	Sulfate turpentine: crude (1 gal)	Chlorine (unknown)	Methyl acetoacetate (unknown)	Oil: diesel (1500 gal)	Sulfuric acid (0.5 gal)
91	Mobile (unknown location), Mobile	Oil, unknown (unknown)	Oil: diesel (40 gal)	Unknown oil (drilling mud) (100 gal)	Gasoline: automotive (unleaded) (0.06 gal) Oil, fuel: no. 2-D (10 gal)	Oil, fuel: no. 2-D (10 gal)
15	Blakeley Island (Midstream Fuel), Mobile, Mobile	Oil, fuel: no. 6 (0.99 lbs)	Oil, fuel: no. 6 (0.8 gal)	Oil, fuel: no. 6 (5 gal)	Oil, fuel: no. 6 (5 gal)	Oil, unknown (unknown)
13	Tennessee River MM 407.5 (Widow's Creek Fossil Plant), Bridgeport, Jackson	Oil: hydraulic oil (0.25 gal)	Asbestos (unknown)	Oil, misc: lubricating (0.99 gal)	Oil: diesel (unknown)	Oil, mise: lubricating (unknown)
12	ATC USCG Aviation Training Center, Mobile, Mobile	Jet fuel: JP-4 (25 gal)	Jet fuel: JP-4 (5 gal)	Jet fuel: JP-4 (10 gal)	Jet fuel: JP-4 (5 gal)	Oil, misc: motor (0.03 gal)
=	Mobile Bay (unknown location), Mobile	Other oil (0.99 gal)	Oil: diesel (10 gal)	Oil, fuel: no. 2 (unknown)	Oil: hydraulic oil (0.2 gal)	Oil: hydraulic oil (5 gal)
01	Fairlawn Rd (No. 1855) Barge Dock, Tuscaloosa, Tuscaloosa	Asphalt (1 gal)	Gasoline: automotive (unleaded) (1000 gal)	Oil: crude (1 gal)	Sludge (100 gal)	Asphalt (5 gal)
10	Perdido Beach Blvd (No. 26619), Orange Beach, Baldwin	Oil: diesel (0.99 gal)	Oil, unknown (unknown)	Oil, unknown (unknown)	Oil, unknown (unknown)	Oii, unknown (unknown)
10	Pinto Island, Mobile, Mobile	Waste oil (2 gal)	Oil, fuel: no. 2 (10 gal)	Oil: diesel (unknown)	Oil, fuel: no. 6 (unknown)	Oil: hydraulic oil (20 gal)
01 01	Viaduct Kd (No. 50), Chickasaw, Mobile Water St S (No. 5) (Radcliff Economy Marine), Mobile, Mobile	Waste oil (30 gal) Oil: diesel (10 gal)	Oil, misc: lubricating (1 gal) Oil, fuel: no. 2-D (unknown)	Waste oil (unknown) Oil fuel no. 2-D (1 pal)	Oil, unknown (unknown) Oil diesel (5 gal)	Waste oil/lubricants (5 gal) Oil mise: motor (1 gal)
8	Blakeley Island (unknown location), Mobile, Mobile	1	Gasoline: automotive (4.23 g Pb/gal) (10 gal)	Jet fuel: JP-4 (0.5 gal)	Jet fuel: JP-9 (40 gal)	Jet fuel (0.13 gal)
8	Dauphin Island Pkwy (No. 7778), Theodore, Mobile	Oil: diesel (70 gal)	Oil, fuel: no. 2-D (2 gal)	Oil: diesel (0.03 gal)	Oil, fuel: no. 2-D (10 gal)	Oil, fuel: no. 2 (0.99 gal)
×	Dunlap Dr Gate B Pinto Island, Mobile, Mobile	Waste oil/lubricants (15 gal)	Oil, misc: lubricating (25 gal)	Oil, fuel: no. 6 (20 gal)	Oil: crude (27 gal)	Oil: diesel (unknown)
7	Finley Island Rd (BP Amoco Chemical Company), Decatur, Morgan	p-Xylene (1000 gal)	o-Xylene (360 gal)	Sulfur dioxide (unknown)	Oil: diesel (unknown)	Oil: diesel (25 gal)
7	OCSG 5753 Platform A, Mobile	Oil, unknown (unknown)	Oil, unknown (unknown)	Oil, misc. motor (0.5 gal)	9 gal)	Waste oil (0.1 gal)
7	Viaduct Rd (No. 200), Chickasaw, Mobile	Asphalt (20000 gal)	Oil, misc: lubricating (2 gal)	Oil, unknown (unknown)	Oil, unknown (unknown)	Oil: diesel (10 gal)
9	Erie St (No. 1327), Birmingham, Jefferson	Coal tar pitch (8000 gal)	Benzo(a)pyrene (330 lbs)	Creosote (1000 lbs)	Oil, mise: coal tar and water (200 gal)	Creosote (500 gal)

Incidents Sorted by Location

No. of						
Incidents	Incident Location	Incident 6 Chemicals	Incident 7 Chemicals	Incident 8 Chemicals	Incident 9 Chemicals	Incident 10 Chemicals
75	Battleship Pkwy (No. 2703), Mobile, Mobile	Oil, fuel: no. 5 (0.5 gal)	Oil: diesel (0.06 gal)	Oit, fuel: no. 5 (0.06 gal)	Oil, fuel: no. 5 (0.13 gal)	Oil, fuel: no. 5 (unknown)
2	Hwy 17 at MM 163, Emelle, Sumter	Hazardous waste (400 gal)	Lead battery, liquid (10 gal)	Wastewater treatment sludge (20 gal)	Hazardous waste, D008 (0.5 gal)	Bag house dust, D006, D008 (1 gal)
88	MP 791 (Norris Yard), Birmingham, Jefferson	Oil: diesel (10 gal)	Oil: hydraulic oil (unknown)	Pulpmill liquids (1 gal)	Hydrochloric acid (0.99 gal)	Styrene (1 gal)
90	Tenn-Tom Waterway MM 216 (Demopolis Yacht Basin), Hwy 43 N, Demopolis, Marengo	Oil, unknown (unknown)	Oil: diesel (1 gal)	Oil, unknown (unknown)	Oil: diesel (unknown)	Oil, unknown (unknown)
44	Alabama State Dock, Mobile, Mobile	Oil. diesel (3 gal)	Oil, unknown (unknown)	Gasoline: automotive (4.23 g Pb/gal) (40 gal)	Oil: hydraulic oil (0.5 gal)	Bilge material (5 gal)
32	MP 401-A (Sheffield Railyard), Shefffeld, Colbert	Oil, mise: lubricating (0.25 gal)	Oil: hydraulic oil (10 gal)	Alcohol, flammable (0.13 gal)	Oil, fuel: no. 2-D (2 gal)	Oil, edible: soya bean (3 gal)
25	Water St S (No. 265) (Bender Shipyard), Mobile, Mobile	Oil: diesel (3 gal)	Waste oil/Iubricants (15 gal)	Oil, fuel: no. 6 (80 gal)	Kerosene (unknown)	Oil: hydraulic oil (0.13 gal)
24	Industrial Pkwy (LL&E Dock), Saraland, Mobile	Oil: diesel (5 gal)	Naphtha: solvent (25 gal)	Jet fuel (15 gal)	Gasoline: automotive (unleaded) (40 gal)	Oil: diesel (5 gal)
23	Blakelev Island (Atlantic Marine), Mobile. Mobile	Oil, unknown (5 gal)	Oil, unknown (20 gal)	Oil unknown (2 eal)	Oi; unknown (unknown)	Oil_unknown (unknown)
21	Blakeley Island (Coastal Fuel Docks), Mobile, Mobile	Oil, fuel: no. 6 (8 gal)	Oil, fuel: no. 6 (5 gal)	Oil, misc: lubricating (3 gal)	Oil, fuel: no. 6 (0.01 gal)	Oil, unknown (unknown)
20	MP 363.4A (Railyard), Decatur, Morgan	Oil: hydraulic oil (30 gal)	Sodium hydroxide solution (10 gal)	Terephthalic acid (5 gal)	Oil, unknown (100 gal)	Oil: diesel (unknown)
17	MP 149 MB (Mobile railyard), Mobile, Mobile	Suffuric acid (1 gal)	Sodium hydroxide (0.5 gal)	Sulfuric acid (unknown)	Sulfur dioxide (1 lbs)	Acrylonitrile propionitric Rmenral (6 gal)
16	Mobile (unknown location), Mobile	Oil: diesel (0.1 gal)	Oil, unknown (unknown)	Gasoline (unknown)	Oil: diesel (50 gal)	Unknown Material (unknown)
15	Blakeley Island (Midstream Fuel), Mobile, Mobile	Oil: diesel (10 gal)	Oil, misc: lubricating (10 gal)	Oil, misc: lubricating (1.5 gal)	Oil, fuel: no. 6 (30 gal)	Oil, misc: motor (1 gal)
13	Tennessee River MM 407.5 (Widow's Creek Fossil Plant), Bridgeport, Jackson	Oil, unknown (unknown)	Oil: diesel (2500 gal)	Waste oil (2 gal)	Oil, unknown (unknown)	Oil, unknown (unknown)
12	ATC USCG Aviation Training Center, Mobile, Mobile	Jet fuel: JP-4 (15 gal)	Jet fuel: JP-4 (3 gal)	Jet fuel: JP-5 (kerosene, heavy) (10 gal)	Jet fuel: JP-4 (20 gal)	Oil, misc: lubricating (5 gal)
=	Mobile Bay (unknown location), Mobile	Gasoline: automotive (unleaded) (unknown)	Oil, misc: motor (0.99 gal)	Oil: diesel (1 gal)	Other oil (15 gal)	Oil: diesel (unknown)
10	Fairlawn Rd (No. 1855) Barge Dock, Tuscaloosa, Tuscaloosa	Oil diesel (8 gal)	Other oil (fuel oil no. 5) (5 gal)	Asphalt (6 gal)	Oil: crude (2 gal)	Other oil (fuel oil no. 5) (1 gal)
10	Perdido Beach Blvd (No. 26619), Orange Beach, Baldwin	Oil: diesel (unknown)	Oil, unknown (unknown)	Oil: diesel (unknown)	Oil, unknown (unknown)	Oil, unknown (unknown)
0 9	Pinto Island, Mobile, Mobile	Oil: diesel (15 gal)	Oil: hydraulic oil (6 gal)	Drilling fluid (20 gal)	Oil, unknown (unknown)	Oil, unknown (unknown)
01	Water St S (No. 5) (Radcliff Economy Marine), Mobile, Mobile	Oil, fuel: no. 2-D (400 gal)	waste on (1 gar) Oil: diesel (20 gal)	Oil, misc: tuoricating (0.13 gat) Oil, fuel: no. 2-D (3 gal)	Oil, unknown (unknown) Oil: diesel (30 gal)	Oil, misc: lubricating (1 gal) Oil, fuel: no. 2-D (5 gal)
æ	Blakeley Island (unknown location), Mobile, Mobile		Oil, unknown (unknown)	Oil: hydraulic oil (4.5 gal)		
∞	Dauphin Island Pkwy (No. 7778), Theodore, Mobile		Sodium hypochlorite (15% or less) (3500 gal)	Oil, fuel: no. 2-D (1 gal)		
	Dunlap Dr Gate B Pinto Island, Mobile, Mobile	Epoxy cure (42 gal)	Oil, unknown (25 gal)	Other oil (2 gal)		
7	Finley Island Rd (BP Amoco Chemical Company), Decatur, Morgan	Oil, unknown (unknown)	m-Xylene (1 gal)			
7	OCSG 5753 Platform A, Mobile	Oil, unknown (unknown)	Jet fuel (30 gal)			
7	Viaduct Rd (No. 200), Chickasaw, Mobile	Unknown oil (possibly bilge slops) (unknown)	Oil, fuel: no. 6 (unknown)			
9	Erie St (No. 1327), Birmingham, Jefferson	Creosote (100 gal)			2	

Incidents Sorted by Location

No. of Incidents	Incident Location	Incident 11 Chemicals	Incident 12 Chemicals	Incident 13 Chemicals	Incident 14 Chemicals	Incident 15 Chemicals
75	Battleship Pkwy (No. 2703), Mobile, Mobile	Oil, fuel: no. 5 (1 gal)	Oil, fuel: no. 5 (unknown)	Oil, fuel: no. 5 (0.13 gal)	Oil, fuel: no. 6 (0.06 gal)	Oil: diesel, bunker C (0.13 gal)
2	Hwy 17 at MM 163, Emelle, Sumter	D006, D007, D009, D018 (5 gal)	Waste Fuel, D008 (1 gal)	Hazardous waste, D001, F003, F005, U056 (1 gal); Hazardous waste, U120, U156, U188 (1 gal)	Benzene (3 gal)	Lead, liquid (2 gal)
28	MP 791 (Norris Yard), Birmingham, Jefferson	Oil: diesel (5 gal)	Oil, misc: lubricating (100 gal)	Oil, misc: lubricating (50 gal)	Liquefied petroleum gas (unknown)	Turpentine (2 gal)
90	Tenn-Tom Waterway MM 216 (Demopolis Yacht Basin), Hwy 43 N, Demopolis, Marengo	Oil, unknown (unknown)	Oil, fuel: no. 2-D (0.13 gal)	Oil: diesel (unknown)	Oil, fuel: no. 2-D (10 gal)	Oil, fuel: no. 2-D (1 gal)
44	Alabama State Dock, Mobile, Mobile	Oil, fuel: no. 6 (unknown)	Oil, fuel: no. 2-D (100 gal)	Oil, misc: motor (5 gal)	Oil: diesel (10 gal)	Oil, fuel: no. 2-D (10 gal)
32	MP 401-A (Sheffield Railyard), Sheffield, Colbert	Phosphoric acid (0.25 gal)	Hominy feed (200 lbs)	Gasoline: automotive (unleaded) (0.06 gal) Gasoline: automotive (unleaded) (1 gal)	Gasoline: automotive (unleaded) (1 gal)	Ammonia, anhydrous (unknown)
25	Water St S (No. 265) (Bender Shipyard), Mobile, Mobile	Oil: diesel (unknown)	Oil: hydraulic oil (15 gal)	Oil: hydraulic oil (0.25 gal)	Oily waste (20 gal)	Oil: diesel (unknown)
24	Industrial Pkwy (LL&E Dock), Saraland, Mobile	Jet fuel: JP-11 (1 gal)	Oil: diesel (unknown)	Heavy olefin seed (35 gal)	Gasoline: automotive (4.23 g Pb/gal) (800 gal)	Oil, unknown (unknown)
23	Blakeley Island (Atlantic Marine), Mobile, Mobile	Other oil (5 gal)	Oii: hydraulic oil (15 gal)	Oil: hydraulic oil (unknown)	Bilge slops (300 gal)	Oil, mise: tranmission (1.1 gal); Ethylene glycol (2.3 gal); Oil, mise: lubricating (2.7 gal); Oil: diesel (6 gal)
21	Blakeley Island (Coastal Fuel Docks), Mobile, Mobile	Oil, fuel: no. 6 (5 gal)	Oil, fuel: no. 6 (unknown)	Vacuum gas oil (2 gal)	Oil: diesel (1 gal)	Oil, fuel: no. 6 (0.99 gal)
20	MP 363.4A (Railyard), Decatur, Morgan	Oil, misc: lubricating (50 gal)	Chlorine (unknown)	Terephthalic acid (2 lbs)	Sodium hydroxide (2 gal)	Other oil (condensate) (0.25 gal)
17	MP 149 MB (Mobile railyard), Mobile, Mobile	Hydrochloric acid (2 gal)	Propionitrile (0.99 lbs)	Carbon dioxide (unknown)	Vitropylamine (3 gal)	Sulfuric acid (1 gal)
91	Mobile (unknown location), Mobile	Oil, unknown (unknown)	Oil, unknown (unknown)	Oil, unknown (unknown)	Oil: diesel (0.1 gal)	Waste oil/lubricants (unknown)
15	Blakeley Island (Midstream Fuel), Mobile, Mobile	Oil, fuel: no. 2 (0.5 gal)	Oil, fuel: no. 2-D (10 gal)	Oil, unknown (unknown)	Oil, fuel: no. 2-D (5 gal)	Ammonia, anhydrous (unknown)
13	Tennessee River MM 407.5 (Widow's Creek Fossil Plant), Bridgeport, Jackson	Oil, misc. lubricating (5 gal)	Unknown Material (unknown)	Oil, unknown (unknown)		
71	ATC USCG Aviation Training Center, Mobile, Mobile	Oil: hydraulic oil (0.1 gal)	Oil, unknown (unknown)			
11	Mobile Bay (unknown location), Mobile	Oil, unknown (unknown)				
01	Fairlawn Rd (No. 1855) Barge Dock, Tuscaloosa, Tuscaloosa					
01	Perdido Beach Blvd (No. 26619), Orange Beach, Baldwin					
01 01	Pinto Island, Mobile, Mobile Viaduct Rd (No. 50), Chickasaw, Mobile					
10	Water St S (No. 5) (Radcliff Economy Marine), Mobile, Mobile					
8	Blakeley Island (unknown location), Mobile, Mobile					
8	Dauphin Island Pkwy (No. 7778), Theodore, Mobile					
*	Dunlap Dr Gate B Pinto Island, Mobile, Mobile					
7	Finley Island Rd (BP Amoco Chemical Company), Decatur, Morgan					
7	OCSG 5753 Platform A, Mobile					
7	Viaduct Rd (No. 200), Chickasaw, Mobile					
9	Erie St (No. 1327), Birmingham, Jefferson					

Incidents Sorted by Location

No. of Incidents	Incident Location	Incident 16 Chemicals	Incident 17 Chemicals	Incident 18 Chemicals	Incident 19 Chemicals	Incident 20 Chemicals
75		Oil, fuel: no. 5 (0.06 gal)	Oil, fuel: no. 5 (1 gal)	Oil, fuel: no. 5 (0.13 gal)	Oil, fuel: no. 5 (0.03 gal)	Oil, fuel: no. 5 (0.06 gal)
64	Hwy 17 at MM 163, Emelle, Sumter	Leachate, F039 (5 gal)	Chemical waste products (2 gal)	Hexachlorobutadiene (0.5 gal)	Flammable waste liquids (NOS) (30 gal)	Dichloromethane (25 gal)
88	MP 791 (Norris Yard), Birmingham, Jefferson	Sulfuric acid (20 gal)	Oil, fuel: no. 2-D (1 gal)	Oil, fuel: no. 2-D (50 gal)	Carbon dioxide (refrigerant) (unknown)	Oil, misc: lubricating (50 gal)
20	Tenn-Tom Waterway MM 216 (Demopolis Yacht Basin), Hwy 43 N, Demopolis, Marengo	Oil, fuel: no. 2-D (3 gal)	Finish (170 gal)	Oil, fuel: no. 2-D (10 gal)	Unknown Material (unknown)	Oil: diesel mixed with water (2.5 gal)
44	Alabama State Dock, Mobile, Mobile	Oil: diesel (3 gal)	Oil, misc: lubricating (0.25 gal)	Oil: diesel (10 gal)	Oil, misc: lubricating (150 gal)	Waste oil (500 gal)
32	MP 401-A (Sheffield Railyard), Sheffield, Colbert	Oil: diesel (150 gal)	Hydrochloric acid (unknown)	Oil: diesel (unknown)	Oil: hydraulic oil (25 gal)	Sulfur dioxide (unknown)
25	Water St S (No. 265) (Bender Shipyard), Mobile, Mobile	Waste oil and water mixture (500 gal)	Gasoline: automotive (4.23 g Pb/gal) (unknown)	ODS and OSX mixture (1 gal)	Oil: diesel (10 gal)	Oil, fuel: no. 6 (unknown)
24	Industrial Pkwy (LL&E Dock), Saraland, Mobile	Oil, unknown (unknown)	Oil, unknown (unknown)	Jet fuel: JP-8 (5 gal)	Jet fuel: JP-10 (unknown)	Oil: crude (unknown)
23	Blakelev Island (Atlantic Marine) Mobile Mobile	Oil no 6 oil mixed with diesel (0.28 sal)	Оі] пакаома (пакаома)	Oil: diesel (5 val)	Oil: bydraulic oil (2 6 gal)	Oil: hydraulic oil (30 oal)
21	Blakeley Island (Coastal Fuel Docks), Mobile, Mobile	Oil, fuel: no. 6 (unknown)	Oil: Marine diesel (unknown)	Waste oil/lubricants (20 gal)	Oil: diesel (unknown)	Oil, fuel: no. 6 (12000 gal)
20	MP 363 4A (Railyard), Decatur, Morgan	Acetic acid (1 gal)	Nitrogen fertilizer solution (1 gal)	Sodium hydroxide solution (0.13 gal)	Nitrogen fertilizer solution (0.25 gal)	Sodium hydroxide (1 gal)
17	MP 149 MB (Mobile railyard), Mobile, Mobile	Iso-butyraldehyde (0.1 gal)	p-Xylene (5 gal)			
16	Mobile (unknown location), Mobile	Oil, unknown (unknown)				
15	Blakeley Island (Midstream Fuel), Mobile, Mobile			-		
13	Tennessee River MM 407.5 (Widow's Creek Fossil Plant), Bridgeport, Jackson					
12	ATC USCG Aviation Training Center, Mobile, Mobile					
	Mobile Bay (unknown location), Mobile					
10	Fairlawn Rd (No. 1855) Barge Dock, Tuscaloosa, Tuscaloosa					
10	Perdido Beach Blvd (No. 26619), Orange Beach, Baldwin					
0.	Pinto Island, Mobile, Mobile					
2 2	Water St S (No. 5) (Radcliff Economy Marine), Mobile, Mobile					
∞	Blakeley Island (unknown location), Mobile, Mobile					:
∞	Dauphin Island Pkwy (No. 7778), Theodore, Mobile					
∞	Dunlap Dr Gate B Pinto Island, Mobile, Mobile					
7	Finley Island Rd (BP Amoco Chemical Company), Decatur, Morgan					:
7	OCSG 5753 Platform A, Mobile					
7	Viaduct Rd (No. 200), Chickasaw, Mobile					
9	Erie St (No. 1327), Birmingham, Jefferson					

Incidents Sorted by Location

No. of	_					
Incidents	Incident Location	Incident 21 Chemicals	Incident 22 Chemicals	Incident 23 Chemicals	Incident 24 Chemicals	Incident 25 Chemicals
75	Battleship Pkwy (No. 2703), Mobile, Mobile	Oil, fuel: no. 5 (unknown)	Oil, fuel: no. 5 (unknown)	Oil, fuel: no. 6 (unknown)	Oil, fuel: no. 5 (unknown)	Oil, fuel: no. 6 (0.03 gal)
64	Hwy 17 at MM 163, Emelle, Sumter	Hazardous liquid waste, F034 (10 lbs)	Soil, creosote contaminated and debris (10 gal)	Fuel waste (5 gal)	RCRA incinerator ash (5 gal)	Incinerator ash (3 lbs)
88	MP 791 (Norris Yard), Birmingham, Jefferson	Carbon dioxide (refrigerant) (0.13 gal)	Oil, mise: lubricating (0.5 gal)	Oil: hydraulic oil (50 gal)	Sodium chlorate (0.25 gal)	Ammonia, anhydrous (unknown)
50	Tenn-Tom Waterway MM 216 (Demopolis Yacht Basin), Hwy 43 N, Demopolis, Marengo	Oil, unknown (unknown)		Oil, unknown (unknown)	Gasoline: automotive (unleaded) (0.5 gal)   Oil, unknown (unknown)	Oil, unknown (unknown)
44	Alabama State Dock, Mobile, Mobile	Oil: fuel oil (26 gal)	Oil: fuel oil (26 gal)	Oil, misc: motor (1.5 gal)	Oil, fuel: no. 2-D (40 gal)	Oil: diesel (unknown)
32	MP 401-A (Sheffield Railyard), Sheffield, Colbert	Oil: hydraulic oil (30 gal)	(unknown)	Hydrochloric acid (0.06 gal)		Carbutadienes, inhibited (unknown)
25	Water St S (No. 265) (Bender Shipyard), Mobile, Mobile	Benzyl chloride (15 gal)	Oil: hydraulic oil (15 gal)	Oil, fuel: no. 2-D (1 gal)	)	Oil, fuel: no. 2-D (200 gal)
24	Industrial Pkwy (LL&E Dock), Saraland, Mobile	Jet fuel (1 gal)	Oil, fuel: no. 2-D (120 gal)	Gasoline: automotive (4.23 g Pb/gal) (2000 gal)	Oil, fuel: no. 2 (0.25 gal)	
23	Blakeley Island (Atlantic Marine), Mobile, Mobile	Other oil (10 gal)	Oii, unknown (unknown)	Other oil (gas oil) (unknown)		
21	Blakeley Island (Coastal Fuel Docks), Mobile, Mobile	Oil, 90% fuel: no. 6, 10% diesel fuel (unknown)				
20	MP 363.4A (Railyard), Decatur, Morgan					
17	MP 149 MB (Mobile railyard), Mobile, Mobile					
91	Mobile (unknown location), Mobile					
15	Blakeley Island (Midstream Fuel), Mobile, Mobile					
13	Tennessee River MM 407.5 (Widow's Creek Fossil Plant), Bridgeport, Jackson					
13	ATC USCG Aviation Training Center, Mobile, Mobile					
=	Mobile Bay (unknown location), Mobile					
10	Fairlawn Rd (No. 1855) Barge Dock, Tuscaloosa, Tuscaloosa					
10	Perdido Beach Blvd (No. 26619), Orange Beach, Baldwin					
2 2	Pinto Island, Mobile, Mobile Viaduct Rd (No. 50), Chickasaw, Mobile					
01	Water St S (No. 5) (Radcliff Economy Marine), Mobile, Mobile					-
8	Blakeley Island (unknown location), Mobile, Mobile					
œ	Dauphin Island Pkwy (No. 7778), Theodore, Mobile					
<b>%</b>	Dunlap Dr Gate B Pinto Island, Mobile, Mobile					
7	Finley Island Rd (BP Amoco Chemical Company), Decatur, Morgan					
7	OCSG 5753 Platform A, Mobile					
7	Viaduct Rd (No. 200), Chickasaw, Mobile					
9	Erie St (No. 1327), Birmingham, Jefferson					

Incidents Sorted by Location

No. of						
HIGHGENS	_	meturit 20 Circinicais	motion 27 Chemicals	incident 28 Chemicals	incident 29 Chemicals	Incident 30 Chemicals
75	Battleship Pkwy (No. 2703), Mobile, Mobile	Oil, fuel: no. 5 (0.03 gal)	Oil, fuel: no. 5 (0.03 gal)	Oil, fuel: no. 5 (unknown)	Oil: diesel (0.03 gal)	Oil, fuel: no. 5 (0.03 gal)
49	Hwy 17 at MM 163, Emelle, Sumter	Mixed waste solvents, possibly contaminated (20 gal)	D008 (0.5 gal)	Waste material D001, D004, F001, F004 (10 gal)	Fuel waste (15 gal)	Incinerator debris (0.5 gal)
28	MP 791 (Norris Yard), Birmingham, Jefferson	Oil: diesel (50 gal)	Asphalt (unknown)	Potassium hydroxide (2 gal)	Oil, misc: lubricating (1 gal)	Sodium chlorite (0.13 gal)
80	Tenn-Tom Waterway MM 216 (Demopolis Yacht Basin), Hwy 43 N, Demopolis, Marengo	Oil: diesel (unknown)	Oil, fuel: no. 2-D (unknown)	Oil, fuel: no 2-D (5 gal)	Bilge slops (2 gal)	Oil, fuel: no. 2-D (1 gal)
44	Alabama State Dock, Mobile, Mobile	Oil, misc: lubricating (40 gal)	Oil: hydraulic oil (4 gal)	Oil, fuel: no. 6 (0.99 gal)	Oil, misc. lubricating (unknown)	Oil, unknown (unknown)
32	MP 401-A (Sheffield Railyard), Sheffield, Colbert	Oil, misc: lubricating (5 gal)	Oil, fuel: no. 2-D (400 gal)	Butyl acetate (2 gal)	Potassium hydroxide (1 gal)	Hydrochloric acid (0.04 gal)
25	Water St S (No. 265) (Bender Shipyard), Mobile, Mobile					
24	Industrial Pkwy (LL&E Dock), Saraland, Mobile					
23	Blakelev Island (Atlantic Marine), Mobile, Mobile					
21	Blakeley Island (Coastal Fuel Docks), Mobile, Mobile					
20	MP 363.4A (Railyard), Decatur, Morgan					
17	MP 149 MB (Mobile railyard), Mobile, Mobile					
91	Mobile (unknown location), Mobile					
15	Blakeley Island (Midstream Fuel), Mobile, Mobile					
13	Tennessee River MM 407.5 (Widow's Creek Fossil Plant), Bridgeport, Jackson					
12	ATC USCG Aviation Training Center, Mobile, Mobile					
11	Mobile Bay (unknown location), Mobile					
01	Fairlawn Rd (No. 1855) Barge Dock, Tuscaloosa, Tuscaloosa					
10	Perdido Beach Blvd (No. 26619), Orange Beach, Baldwin					
9 9	Pinto Island, Mobile, Mobile Viaduct Rd (No. 50). Chickasaw. Mobile					
10	Water St S (No. 5) (Radcliff Economy Marine), Mobile, Mobile					
8	Blakeley Island (unknown location), Mobile, Mobile					
8	Dauphin Island Pkwy (No. 7778), Theodore, Mobile					
-	Dunlap Dr Gate B Pinto Island, Mobile, Mobile					
7	Finley Island Rd (BP Amoco Chemical Company), Decatur, Morgan				:	
7	OCSG 5753 Platform A, Mobile					
7	Viaduct Rd (No. 200), Chickasaw, Mobile					
9	Erie St (No. 1327), Birmingham, Jefferson					

Incidents Sorted by Location

No. of Incidents	Incident Location	Incident 31 Chemicals	Incident 32 Chemicals	Incident 33 Chemicals	Incident 34 Chemicals	Incident 35 Chemicals
75	_	Oil: diesel (unknown)	Oil, fuel: no. 5 (0.03 gal)	Oil:Diesel (bunker C #5) (0.13 gal)	Oil, fuel: no. 5 (unknown)	Oil, fuel: no. 6 (0.06 gal)
49	Hwy 17 at MM 163, Emelle, Sumter	0008 (lead) (1 gal)	D008 (3 gal)	Sweeper trash, D007 (10 gal)	Waste oil sludge (1 gal)	Blast furnace slag (2 gal)
88	MP 791 (Norris Yard), Birmingham, Jefferson	Oil, fuel: no. 2-D (1000 gal)	Flammable liquid, waste (0.5 gal)	Oil: diesel (25 gal)	Potassium hydroxide (1 gal)	Oil: diesel (5 gal)
90	Tenn-Tom Waterway MM 216 (Demopolis Yacht Basin), Hwy 43 N, Demopolis, Marengo	Oil, fuel: no. 2-D (2 gal)	Bilge slops (30 gal)	Oil, unknown (unknown)	Oil, unknown (unknown)	Oil: diesel (1 gal)
44	Alabama State Dock, Mobile, Mobile	Oil: diesel (5 gal)	Oil: hydraulic oil (unknown)	Oil: diesel (unknown)	Oil: hydraulic oil (0.13 gal)	Oil: emulsified (15 gal)
32	MP 401-A (Sheffield Railyard), Sheffield, Colbert	Phosphoric acid (0.1 gal)	Oil, fuel: no. 2-D (500 gal)			
25	Water St S (No. 265) (Bender Shipyard), Mobile, Mobile					
24	Industrial Pkwy (LL&E Dock), Saraland, Mobile					
23	Blakelev Island (Atlantic Marine). Mobile. Mobile					
21	Blakeley Island (Coastal Fuel Docks), Mobile, Mobile					
20	MP 363.4A (Railyard), Decatur, Morgan					
17	MP 149 MB (Mobile railyard), Mobile, Mobile					
16	Mobile (unknown location), Mobile					
15	Blakeley Island (Midstream Fuel), Mobile, Mobile					
13	Tennessee River MM 407.5 (Widow's Creek Fossil Plant), Bridgeport, Jackson					
71	ATC USCG Aviation Training Center, Mobile, Mobile					
ii	Mobile Bay (unknown location), Mobile					
01	Fairlawn Rd (No. 1855) Barge Dock, Tuscaloosa, Tuscaloosa					
10	Perdido Beach Blvd (No. 26619), Orange Beach, Baldwin					
10	Pinto Island, Mobile, Mobile					
2 9	viaduct rd (No. 50), Chickasaw, Mooile Water St S (No. 5) (Radcliff Economy Marine), Mobile, Mobile					
œ	Blakeley Island (unknown location), Mobile, Mobile					
œ	Dauphin Island Pkwy (No. 7778), Theodore, Mobile					
*	Dunlap Dr Gate B Pinto Island, Mobile, Mobile					
7	Finley Island Rd (BP Amoco Chemical Company), Decatur, Morgan					
4	OCSG 5753 Platform A, Mobile					
7	Viaduct Rd (No. 200), Chickasaw, Mobile					
9	Erie St (No. 1327), Birmingham, Jefferson					

Incidents Sorted by Location

No. of Incidents	Incident Incation	Incident 61 Chemicals	Incident 62 Chemicals	Incident 63 Chemicals	Incident 64 Chemicals	Incident 65 Chemicals
7.		Oil find: no 5 (5 mal)		Oil: discal (1 col)	Oil fied no Studensum	Other oil (2 mel)
2	Hwy 17 at MM 163, Emelle, Sumter	Polychlorinated biphenyls (1 lbs)	nyls (1300 ppm)	Polychlorinated biphenyls (unknown)	Fly ash with water (10 lbs)	(1.00 - 1.
88	MP 791 (Norris Yard), Birmingham, Jefferson					
20	Tenn-Tom Waterway MM 216 (Demopolis Yacht Basin), Hwy 43 N, Demopolis, Marengo					
44	Alabama State Dock, Mobile, Mobile					
32	MP 401-A (Sheffield Railyard), Sheffield, Colbert					
25	Water St S (No. 265) (Bender Shipyard), Mobile, Mobile					
24	Industrial Pkwy (LL&E Dock), Saraland, Mobile					
ន	Blakeley Island (Atlantic Marine), Mobile, Mobile					
21	Blakeley Island (Coastal Fuel Docks), Mobile, Mobile					
20	MP 363.4A (Railyard), Decatur, Morgan					
17	MP 149 MB (Mobile railyard), Mobile, Mobile					
91	Mobile (unknown location), Mobile					
15	Blakeley Island (Midstream Fuel), Mobile, Mobile					
13	Tennessee River MM 407.5 (Widow's Creek Fossil Plant), Bridgeport, Jackson					
12	ATC USCG Aviation Training Center, Mobile, Mobile					
11	Mobile Bay (unknown location), Mobile					
10	Fairlawn Rd (No. 1855) Barge Dock, Tuscaloosa, Tuscaloosa					
01	Perdido Beach Blvd (No. 26619), Orange Beach, Baldwin					
01	Pinto Island, Mobile, Mobile					
9 9	Viaduct Ku (No. 50), Chickasaw, Monite Water St S (No. 5) (Radcliff Economy Marine), Mobile, Mobile					
∞	Blakeley Island (unknown location), Mobile, Mobile					
∞	Dauphin Island Pkwy (No. 7778), Theodore, Mobile					
80	Dunlap Dr Gate B Pinto Island, Mobile, Mobile					
7	Finley Island Rd (BP Amoco Chemical Company), Decatur, Morgan					
7	OCSG 5753 Platform A, Mobile	-				
7	Viaduct Rd (No. 200), Chickasaw, Mobile				Ē	
٥	Erie St (No. 1327), Birmingham, Jefferson					

Incidents Sorted by Location

Incidents Sorted by Location

No. of Incidents	Incident Location	Incident 71 Chemicals	Incident 72 Chemicals	Incident 73 Chemicals	Incident 74 Chemicals	Incident 75 Chemicals
25	Battleship Pkwy (No. 2703), Mobile, Mobile	Oil, fuel: no. 5 (unknown)	Oil: diesel, bunker C (3 gal)	Oil, fuel: no. 5 (unknown)	Other oil (unknown)	Oil: diesel (bunker C & #5 fuel oil) (50 gal)
64	Hwy 17 at MM 163, Emelle, Sumter					
85	MP 791 (Norris Yard), Birmingham, Jefferson					
90	Tenn-Tom Waterway MM 216 (Demopolis Yacht Basin), Hwy 43 N. Demopolis, Marengo					
44	Alabama State Dock, Mobile, Mobile					
32	MP 401-A (Shefffeld Railyard), Shefffeld, Colbert					
25	Water St S (No. 265) (Bender Shipyard), Mobile, Mobile					
24	Industrial Pkwy (LL&E Dock), Saraland, Mobile					
23	Blakeley Island (Atlantic Marine), Mobile, Mobile					
21	Blakeley Island (Coastal Fuel Docks), Mobile, Mobile					
20	MP 363.4A (Railyard), Decatur, Morgan					
17	MP 149 MB (Mobile railyard), Mobile, Mobile					
16	Mobile (unknown location), Mobile					
15	Blakeley Island (Midstream Fuel), Mobile, Mobile					
13	Tennessee River MM 407.5 (Widow's Creek Fossil Plant), Bridgeport, Jackson					
12	ATC USCG Aviation Training Center, Mobile, Mobile					
=	Mobile Bay (unknown location), Mobile					
10	Fairlawn Rd (No. 1855) Barge Dock, Tuscaloosa, Tuscaloosa					
10	Perdido Beach Blvd (No. 26619), Orange Beach, Baldwin					
9 9	Pinto Island, Mobile, Mobile Viaduct Rd (No. 50), Chickasaw, Mobile					
01	Water St S (No. 5) (Radcliff Economy Marine), Mobile, Mobile					
80	Blakeley Island (unknown location), Mobile, Mobile					
90	Dauphin Island Pkwy (No. 7778), Theodore, Mobile					
∞	Dunlap Dr Gate B Pinto Island, Mobile, Mobile					
7	Finley Island Rd (BP Amoco Chemical Company), Decatur, Morgan					
7	OCSG 5753 Platform A, Mobile					
_	Viaduct Rd (No. 200), Chickasaw, Mobile					
9	Erie St (No. 1327), Birmingham, Jefferson					

Incidents Sorted by Location

Incidents Sorted by Location

Incidents Sorted by Location

Incidents Sorted by Location

# **Appendix C. Data for Toxic Substances**

The following tables provide the data needed to carry out the calculations for toxic substances using the methods presented in the previous sections. Table C-1 presents data for toxic gases, Table C-2 presents data for toxic liquids, and Table C-3 presents data for several toxic substances commonly found in water solutions and for oleum. The data used to develop the factors in tables C-1 and C-2 are primarily from Design Institute for Physical Property Data (DIPPR), American Institute of Chemical Engineers, *Physical and Thermodynamic Properties of Pure Chemicals, Data Compilation.* Other sources, including the National Library of Medicine's Hazardous Substances Databank (HSDB) and the *Kirk-Othmer Encyclopedia of Chemical Technology*, were used for Tables C-1 and C-2 if data were not available from the DIPPR compilation. The factors in Table C-3 were developed using data primarily from *Perry's Chemical Engineers' Handbook* and the *Kirk-Othmer Encyclopedia of Chemical Technology*.

Table C-1: Data for Toxic Gases (EPA 1999)

CAS	Chemical Name	Molecular	Ratio of		Toxic E	nd point <sup>a</sup>	Liquid Factor		Gas	Vapor	Worst-
Number		Weight	Specific Heats	mg/L	ppm	Basis	Boiling (LFB)	Factor (DF) (Boiling)	Factor (GF) <sup>k</sup>	Pressure @25 °C (psia)	
7664-41-7	Ammonia (anhydrous) <sup>c</sup>	17.03	1.31	0.14	200	ERPG-2	0.073	0.71	14	145	Buoyant <sup>d</sup>
7784-42-1	Arsine	77.95	1.28	0.0019	0.6	EHS-LOC (IDLH)	0.23	0.30	30	239	Dense
10294-34-5	Boron trichloride	117.17	1.15	0.010	2	EHS-LOC (Tox <sup>e</sup> )	0.22	0.36	36	22.7	Dense
7637-07-2	Boron trifluoride	67.81	1.20	0.028	10	EHS-LOC (IDLH)	0.25	0.31	28	f	Dense
7782-50-5	Chlorine	70.91	1.32	0.0087	3	ERPG-2	0.19	0.31	29	113	Dense
10049-04-4	Chlorine dioxide	67.45	1.25	0.0028	1	EHS-LOC equivalent (IDLH)I	0.15	0.30	28	24.3	Dense
506-77-4	Cyanogen chloride	61.47	1.22	0.030	12	EHS-LOC equivalent (Tox) <sup>h</sup>	0.14	0.41	26	23.7	Dense
19287-45-7	Diborane	27.67	1.17	0.0011	1	ERPG-2	0.13	1.13	17	f	Buoyant⁴
75-21-8	Ethylene oxide	44.05	1.21	0.090	50	ERPG-2	0.12	0.55	22	25.4	Dense
7782-41-4	Fluorine	38.00	1.36	0.0039	2.5	EHS-LOC (IDLH)	0.35	0.32	22	f	Dense
50-00-0	Formaldehyde (anhydrous)	30.03	1.31	0.012	10	ERPG-2	0.10	0.59	19	75.2	Dense
74-90-8	Hydrocyanic acid	27.03	1.30	0.011	10	ERPG-2	0.079	0.72	18	14.8	Buoyant <sup>d</sup>
7647-01-0	Hydrogen chloride (anhydrous) <sup>c</sup>	36.46	1.40	0.030	20	ERPG-2	0.15	0.41	21	684	Dense
7664-39-3	Hydrogen fluoride (anhydrous) <sup>c</sup>	20.01	1.40	0.016	20	ERPG-2	0.066	0.51	16	17.7	Buoyant <sup>1</sup>
7783-07-5	Hydrogen selenide	80.98	1.32	0.00066	0.2	EHS-LOC (IDLH)	0.21	0.25	31	151	Dense
7783-06-4	Hydrogen sulfide	34.08	1.32	0.042	30	ERPG-2	0.13	0.51	20	302	Dense
74-87-3	Methyl chloride	50.49	1.26	0.82	400	ERPG-2	0.14	0.48	24	83.2	Dense
74-93-1	Methyl mercy tan	48.11	1.20	0.049	25	ERPG-2	0.12	0.55	23	29.2	Dense
10102-43-9	Nitric oxide	30.01	1.38	0.031	25	EHS-LOC (TLV1)	0.21	0.38	19	f	Dense
7-44-5	Phosgene	98.92	1.17	0.00081	0.2	ERPG-2	0.20	0.35	33	27.4	Dense
7803-51-2	Phosphine	34.00	1.29	0.0035	2.5	ERPG-2	0.15	0.66	20	567	Dense
7446-09-5	Sulfur dioxide (anhydrous)	64.07	1.26	0.0078	3	ERPG-2	0.16	0.33	27	58.0	Dense
7783-60-0	Sulfur tetrafluoride	108.06	1.30	0.0092	2	EHS-LOC (Tox <sup>e</sup> )	0.25	0.25 (at -73°C)	36	293	Dense

<sup>&</sup>lt;sup>a</sup> Toxic endpoints are specified in Appendix A to 40 CFR part 68 in units of mg/L. To convert from units of mg/L to mg/m³, multiply by 1,000.

b "Buoyant" refers to the figures for neutrally buoyant gases and vapors; "Dense" refers to the figures for dense gases and vapors.

<sup>&</sup>lt;sup>c</sup> See Table C-3 of this appendix for data on water solutions.

d Gases that are lighter than air may behave as dense gases upon release if liquefied under pressure or cold; consider the conditions of release when choosing the appropriate figure.

<sup>&</sup>lt;sup>e</sup> LOC is based on the IDLH-equivalent level estimated from toxicity data.

f Cannot be liquefied at 25 °C.

<sup>&</sup>lt;sup>9</sup> Not an EHS; LOC-equivalent value was estimated from one-tenth of the IDLH.

h Not an EHS; LOC-equivalent value was estimated from one-tenth of the IDLH-equivalent level estimated from toxicity data.

Hydrogen fluoride is lighter than air, but may behave as a dense gas upon release under some circumstances (e.g., release under pressure, high concentration in the released cloud) because of hydrogen bonding; consider the conditions of release when choosing the appropriate figure.

LOC based on Threshold Limit Value (TLV) - Time-weighted average (TWA) developed by the American Conference of Governmental Industrial Hygienists (ACGIH).

<sup>&</sup>lt;sup>k</sup> Use GF for gas leaks under choked (maximum) flow conditions.

Table C-2: Data for Toxic Liquids (EPA 1999)

CAS	Chemical Name	Molecular	Vapor	Toxic En	dpoint		Liquid Fac	tors	Density	Liquid	
Number		Weight	Pressure at 25 °C (mm Hg)	mg/L	ppm	Basis	Ambient (LFA)	Boiling (LFB)	Factor (DF)	Leak Factor (LLF)	Worst Case Condition <sup>b</sup>
107-02-8	Acrolein	56.06	274		0.5	ERPG-2	0.047	0.12	0.58	40	Dense
107-13-1	Acrylonitrile	53.06	108	0.076	35	ERPG-2	0.018	0.11	0.61	39	Dense
814-68-6	Acrylyl chloride	90.51	110	0.00090	0.2	EHS-LOC (Tox <sup>c</sup> )	0.026	0.15	0.44	54	Dense
107-18-6	Allyl alcohol	58.08	26.1	0.036	15	EHS-LOC (IDLH)	0.0046	0.11	0.58	41	Dense
107-11-9	Allyamine	57.10	242	0.0032	1	EHS-LOC (Tox <sup>c</sup> )	0.042	0.12	0.64	36	Dense
7784-34-1	Arsenous trichloride	181.28	10	0.01	1	EHS-LOC (Tox <sup>c</sup> )	0.0037	0.21	0.23	100	Dense
353-42-4	Boron trifluoride compound with methyl ether 1:1	113.89	11	0.023	5	EHS-LOC (Tox°)	0.0030	0.16	0.49	48	Dense
7726-95-6	Bromine	159.81	212	0.0065	1	ERPG-2	0.073	0.23	0.16	150	Dense
75-15-0	Carbon disulfide	76.14	359	0.16	50	ERPG-2	0.075	0.15	0.39	60	Dense
67-66-3	Chloroform	119.38	196	0.49	100	EHS-LOC (IDLH)	0.055	0.19	0.33	71	Dense
542-88-1	Chloromethyl ether	114.96	29.4	0.00025	0.05	EHS-LOC (Tox <sup>c</sup> )	0.0080	0.17	0.37	63	Dense
107-30-2	Chloromethyl methyl ether	80.51	199	0.0018	0.6	EHS-LOC (Tox <sup>c</sup> )	0.043	0.15	0.46	51	Dense
4170-30-3	Crotonaldehyde	70.09	33.1	0.029	10	ERPG-2	0.0066	0.12	0.58	41	Dense
123-73-9	Crotonaldehyde, (E) -	70.09	33.1	0.029	10	ERPG-2	0.0066	0.12	0.58	41	Dense
108-91-8	Cyclohexylamine	99.18	10.1	0.16	39	EHS-LOC (Tox <sup>c</sup> )	0.0025	0.14	0.56	41	Dense
75-78-5	Dimethyldichlorosilane	129.06	141	0.026	5	ERPG-2	0.042	0.20	0.46	51	Dense
57-14-7	1,1-Dimethylhydrazine	60.10	157	0.012	5	EHS-LOC (IDLH)	0.028	0.12	0.62	38	Dense
106-89-8	Epichlorohydrin	92.53	17.0	0.076	20	ERPG-2	0.0040	0.14	0.42	57	Dense
107-15-3	Ethylenediamine	60.10	12.2	0.49	200	EHS-LOC (IDLH)	0.0022	0.13	0.54	43	Dense
151-56-4	Ethyleneimine	43.07	211	0.018	10	EHS-LOC (IDLH)	0.030	0.10	0.58	40	Dense
110-00-9	Furan	68.08	600	0.0012	0.4	EHS-LOC (Tox <sup>c</sup> )	0.12	0.14	0.52	45	Dense
302-01-2	Hydrazine	32.05	14.4	0.011	8	EHS-LOC (IDLH)	0.0017	0.069	0.48	48	Buoyant <sup>d</sup>
13463-40-6	Iron, pentacarbonyl-	195.90	40	0.00044	0.05	EHS-LOC (Tox <sup>c</sup> )	0.016	0.24	0.33	70	Dense
78-82-0	Isobutyronitrile	69.11	32.7	0.14	50	ERPG-2	0.0064	0.12	0.63	37	Dense
108-23-6	Isopropyl chloroformate	122.55	28	0.10	20	EHS-LOC (Tox <sup>c</sup> )	0.0080	0.17	0.45	52	Dense
126-98-7	Methacrylonitrile	67.09	71.2	0.0027	1	EHS-LOC (TLV°)	0.014	0.12	0.61	38	Dense
79-22-1	Methyl chloroformate	94.50	108	0.0019	0.5	EHS-LOC (Tox <sup>c</sup> )	0.026	0.16	0.40	58	Dense
60-34-4	Methyl hydrazine	46.07	49.6	0.0094	5	EHS-LOC (IDLH)	0.0074	0.094	0.56	42	Dense
624-83-9	Methyl isocyanate	57.05	457	0.0012	0.5	E1tPG-2	0.079	0.13	0.52	45	Dense
556-64-9	Methyl thiocyanate	73.12	10	0.085	29	EHS-LOC (Tox <sup>c</sup> )	0.0020	0.11	0.45	51	Dense
75-79-6	Methyltrichlorosilane	149.48	173	0.018	3	ERPG-2	0.057	0.22	0.38	61	Dense
13463-39-3	Nickel carbonyl	170.73	400	0.00067	0.1	EHS-LOC (Tox <sup>c</sup> )	0.14	0.26	0.37	63	Dense
7697-37-2	Nitric acid (100%) <sup>†</sup>	63.01	63.0	0.026	10	EHS-LOC (Tox <sup>c</sup> )	0.012	0.12	0.32	73	Dense
79-21-0	Peracetic acid	76.05	13.9	0.0045	1.5	EHS-LOC (Tox <sup>c</sup> )	0.0029	0.12	0.40	58	Dense
594-42-3	Perchloromethylmercaptan	185.87	6	0.0076	1	EHS-LOC (IDLH)	0.0023	0.20	0.29	81	Dense
10025-87-3	Phosphorus oxychloride	153.33	35.8	0.0030	0.5	EHS-LOC (Tox <sup>c</sup> )	0.012	0.20	0.29	80	Dense
7719-12-2	Phosphorus trichloride	137.33	120	0.028	5	EHS-LOC (IDLH)	0.037	0.20	0.31	75	Dense

Table C-2: Data for Toxic Liquids (EPA 1999) (continued)

CAS	Chemical Name	Molecular	Vapor	Toxic En	dpoint		Liquid Fac	tors	Density	Liquid	
Number		Weight	Pressure at 25 °C (mm Hg)	mg/L	ppm	Basis	Ambient (LFA)	Boiling (LFB)	Factor (DF)	Leak Factor (LLF)	Worst Case Condition <sup>b</sup>
110-89-4	Pipridine	85.15	32.1	0.022	6		0.0072	0.13	0.57	41	Dense
107-12-0	Propionitrile	55.08	47.3	0.0037	1.6	EHS-LOC (Tox <sup>c</sup> )	0.0080	0.10	0.63	37	Dense
109-61-5	Propyl chloroformate	122.56	20.0	0.010	2	EHS-LOC (Tox <sup>c</sup> )	0.0058	0.17	0.45	52	Dense
75-55-8	Propyleneimine	57.10	187	0.12	50	EHS-LOC (IDLH)	0.032	0.12	0.61	39	Dense
75-56-9	Propylene oxide	58.08	533	0.59	250	ERPG-2	0.093	0.13	0.59	40	Dense
7446-11-9	Sulfur trioxide	80.06	263	0.010	3	ERPG-2	0.057	0.15	0.26	91	Dense
75-74-1	Tetramethyllead	267.33	22.5	0.0040	0.4	EHS-LOC (IDLH)	0.011	0.29	0.24	96	Dense
509-14-R	Tetranitromethane	196.04	11.4	0.0040	0.5	EHS-LOC (IDLH)	0.0045	0.22	0.30	78	Dense
7550-45-0	Titanium tetrachloride	189.69	12.4	0.020	2.6	ERPG-2	0.0048	0.21	0.28	82	Dense
584-84-9	Toluene 2 4-diisocyanate	174.16	0.017	0.0070	1	EHS-LOC (IDLH)	0.000006	0.16	0.40	59	Buoyant <sup>d</sup>
91-08-7	Toluene 2 6-diisocyanate	174.16	0.05	0.0070	1	EHS-LOC (IDLH <sup>9</sup> )	0.000018	0.16	0.40	59	Buoyant <sup>d</sup>
26471-62-5	Toluene diisocyanate (unspecified isomer)	174.16	0.017	0.0070	1	EHS-LOC equivalent (IDLH)	0.000006	0.16	0.40	59	Buoyant⁴
75-77-4	Trimethylchlorosilane	108.64	231	0.050	11	EHS-LOC (Tox <sup>c</sup> )	0.061	0.18	0.57	41	Dense
108-05-4	Vinyl acetate monomer	86.09	113	0.26	75	ERPG-2	0.026	0.15	0.53	45	Dense

<sup>&</sup>lt;sup>a</sup> Toxic endpoints are specified in the Appendix A to 40 CFR part 68 in units of mg/L. To convert from units of mg/L to mg/m³, multiply by 1,000.

<sup>b</sup> "Buoyant" in the column refers to the figures for neutrally buoyant gases and vapors; "Dense" refers to the figures for dense gases and vapors.

<sup>c</sup> LOC is based on IDLH-equivalent level estimated from toxicity data.

d Use dense gas figure if substance is at an elevated temperature.

E LOC based on Threshold Limit Value (TLV) - Time-weighted average (TWA) developed by the American Conference of Governmental Industrial Hygienists (ACGIH).

See Table C-3 of this appendix for data on water solutions.

LOC for this isomer is based on IDLH for toluene 2,4-diisocyanate.

<sup>&</sup>lt;sup>h</sup> Not an EHS; LOC-equivalent value is based on IDLH for toluene 2,4-diisocyanate.

Use the LLF only for leaks from tanks at atmospheric pressure.

Table C-3: Data for Water Solutions of Toxic Substances and for Oleum For Wind Speeds of 1.5 and 3.0 Meters per Second (m/s) (EPA 1999)

CAS Number	Regulated Substance in Solution	Molecular Weight			dpoint <sup>a</sup>	Initial Concentration (Wt %)	Va	Average por e (mm H)	25	actor at ° C A)	Density Factor (DF)	Leak Factor	
			mg/L	ppm	Basis		1.5 m/s	3.0 m/s	1.5 m/s	3.0 m/s		(LLF)	Worst- Case Condition <sup>b</sup>
7664-41-7	Ammonia	17.03	0.14	200	ERPG-2	30	332	248	0.026	0.019	0.55	43	Buoyant
						24	241	184	0.019	0.014	0.54	44	Buoyant
						20	190	148	0.015	0.011	0.53	44	Buoyant
50-00-0	Formaldehyde	30.027	0.012	10	ERPG-2	37	1.5	1.4	0.0002	0.0002	0.44	53	Buoyant
7647-01-0	Hydrochloric	36.46	0.030	20	ERPG-2	38	78	55	0.010	0.0070	0.41	57	Dense
	acid					37	67		0.0085	0.0062	0.42	57	Dense
						36`	56	42	0.0072	0.0053	0.42	57	Dense
						34`	38	29	0.0048	0.0037	0.42	56	Dense
						30`	13	12	0.0016	0.0015	0.42	55	Buoyant <sup>a</sup>
7664-39-3	Hydrofluoric	20.01	0.016	20	ERPG-2	70	124	107	0.011	0.010	0.39	61	Buoyant
	acid					50	16	15	0.0014	0.0013	0.41	58	Buoyant
7697-37-2	Nitric acid	63.01	0.026	10	EHS-	90	25	22	0.0046	0.0040	0.33	71	Dense
					LOC	85	17	16	0.0032	0.0029	0.33	70	Dense
					(IDLH)	80	10.2	10	0.0019	0.0018	0.33	70	Dense
8014-95-7	Oleum - based on SO <sub>3</sub>	80.06 (SO <sub>3</sub> )	0.010	3	ERPG-2	30 (SO <sub>3</sub> )	3.5 (SO <sub>3</sub> )	3.4 (SO <sub>3</sub> )	0.0008	0.0007	0.25	93	Buoyant <sup>d</sup>

Toxic endpoints are specified in the Appendix A to 40 CFR part 68 in units of mg/L.
 "Buoyant" refers to the figures for neutrally buoyant gases and vapors; "Dense" refers to the figures for dense gases and vapors.
 Hydrochloric acid in concentrations below 37 percent is not regulated.
 Use dense gas figure if substance is at an elevated temperature.

## Appendix D. Data for Flammable Substances

These tables provide the data needed to carry out the calculations for flammable substances using the methods presented in this section. Table D-1 presents heat of combustion data for all regulated flammable substances, Table D-2 presents additional data for flammable gases, and Table D-3 presents additional data for flammable liquids. The heats of combustion in Table D-1 and the data used to develop the factors in Tables D-2 and D-3 are primarily from Design Institute for Physical Property Data, American Institute of Chemical Engineers, *Physical and Thermodynamic Properties of Pure Chemicals, Data Compilation*.

Table D-1: Heats of Combustion for Flammable Substances (EPA 1999)

CAS No.	<u>Chemical Name</u>	Physical State at 25° C	Heat of Combustion (kjoule/k)
75-07-0	Acetaldehyde	Gas	25,072
74-86-2	Acetylene [Ethyne]	Gas	48,222
598-73-2	Bromotrifluoroethylene [Ethene, bromotrifluoro-]	Gas	1,967
106-99-0	1,3-Butadiene	Gas	44,548
106-97-8	Butane	Gas	45,719
25167-67-3	Butene	Gas	45,200*
590-18-1	2-Butene-cis	Gas	45,171
624-64-6	2-Butene-trans [2-Butene, (E)]	Gas	45,069
106-98-9	1-Butene	Gas	45,292
107-01-7	2-Butene	Gas	45,100*
463-58-1	Carbon oxysulfide [Carbon oxide sulfide (COS)]	Gas	9,126
7791-21-1	Chlorine monoxide [Chlorine oxide]	Gas	1,011*
590-21-6	1-Chloropropylene [1-Propene, 1-chloro-]	Liquid	23,000*
557-98-2	2-Chloropropylene [1-Propene, 2-chloro-]	Gas	22,999
460-19-5	Cyanogen [Ethanedinitrile]	Gas	21,064
75-19-4	Cyclopropane	Gas	46,560
4109-96-0	Dichlorosilane [Silane, dichloro-]	Gas	8,225
75-37-6	Difluoroethane [Ethane, 1,1-difluoro-]	Gas	11,484
124-40-3	Dimethylamine [Methanamine, N-methy1-]	Gas	35,813
463-82-1	2,2-Dimethy1propane [Propane, 2,2-dimethyl-]	Gas	45,051
74-84-0	Ethane	Gas	47,509
107-00-6	Ethyl acetylene [1-Butyne]	Gas	45,565
75-04-7	Ethylamine [Ethanamine]	Gas	35,210
75-00-3	Ethyl chloride [Ethane, chloro-]	Gas	19,917
74-85-1	Ethylene [Ethene]	Gas	47,145
60-29-7	Ethyl ether [Ethane, 1,1'-oxybis-]	Liquid	33,775
75-08-1	Ethyl mercaptan [Ethanethiol]	Liquid	27,948
109-95-5	Ethyl nitrite [Nitrous acid, ethyl ester]	Gas	18,000
1333-74-0	Hydrogen	Gas	119,950
75-28-5	Isobutane [Propane, 2-methyl]	Gas	45,576
78-78-4	Isopentane [Butane, 2-methyl-]	Liquid	44,911
78-79-5	Isoprene [1,3-Butadiene, 2-methyl-]	Liquid	43,809
75-31-0	Isopropylamine [2-Propanamine]	Liquid	36,484
75-29-6	Isopropyl chloride [Propane, 2-chloro-]	Liquid	23,720
74-82-8	Methane	Gas	50,029
74-89-5	Methylamine [Methanamine]	Gas	31,396
563-45-1	3-Methyl-1-butene	Gas	44,559
563-46-2	2-Methyl-1-butene	Liquid	44,414
115-10-6	Methyl ether [Methane, oxybis-]	Gas	28,835
107-31-3	Methyl formate [Formic acid, methyl ester]	Liquid	15,335
115-11-7	2-Methylpropene 1-Propene, 2-meth 1-]	Gas	44,985
504-60-9	1,3-Pentadiene	Liquid	43,834
109-66-0	Pentane	Liquid	44,697
109-67-1	1-Pentene	Liquid	44,625
646-04-8	2-Pentene, (E) -	Liquid	44,458
627-20-3	2-Pentene, (Z) -	Liquid	44,520
463-49-0	Propadiene [1,2-Propadiene]	Gas	46,332
74-98-6	Propane	Gas	46,333
115-07-1	Propylene [1-Propene]	Gas	45,762
74-99-7	Propyne [1-Propyne]	Gas	46,165

Table D-1: Heats of Combustion for Flammable Substances (EPA 1999) (continued)

CAS No.	<u>Chemical Name</u>	Physical State at 25° C	Heat of Combustion (kjoule/k)
7803-62-5	Silane	Gas	44,307
116-14-3	Tetrafluoroethylene [Ethene, tetrafluoro-]	Gas	1,284
75-76-3	Tetramethylsilane [Silane, tetramethyl-]	Liquid	41,712
10025-78-2	Trichlorosilane [Silane, trichloro-]	Liquid	3,754
79-38-9	Trifluorochloroethylene [Ethene, chlorotrifluoro-]	Gas	1,837
75-50-3	Trimethylamine [Methanamine, N,N-dimethyl-]	Gas	37,978
689-97-4	Vinyl acetylene [1-Buten-3- yne]	Gas	45,357
75-01-4	Vinyl chloride [Ethene, chloro-]	Gas	18,848
109-92-2	Vinyl ethyl ether [Ethene, ethoxy-]	Liquid	32,909
75-02-5	Vinyl fluoride [Ethene, fluoro-]	Gas	2,195
75-35-4	Vinylidene chloride [Ethene, 1,1-dichloro-]	Liquid	10,354
75-38-7	Vinylidene fluoride [Ethene, 1,1-difluoro-]	Gas	10,807
107-25-5	Vinyl methyl ether [Ethene, methoxy-]	Gas	30,549

<sup>\*</sup> Estimated heat of combustion

Table D-2: Data for Flammable Gases (EPA 1999)

CAS Number	Chemical Name	Molecular Weight	Ratio of Specific		nability (Vol%)	LFL (mg/L)	Gas Factor	Liquid Factor	Density Factor	Worst-Case Conditions <sup>a</sup>	Pool Fire Factor	Flash Fraction
Number		weight	Heats	Lower	Upper	(mg/L)	(GF) <sup>g</sup>	Boiling	(Boiling)	Conditions	(PFF)	Factor
			ricats	(LFL)	(UFL)		(61)	(LFB)	(DOIIIII) (DF)		(111)	(FFF) <sup>f</sup>
75-07-0	Acetaldehyde	44.05	1.18	4.0	60.0	72	22	0.11	0.62	Dense	2.7	0.018
74-86-2	Acetylene	26.04	1.23	2.5	80.0	27	17	0.12	0.78		4.8	0.23
598-73-2	Bromotrifluoroethylene	160.92	1.11	С	37.0	С	41 <sup>c</sup>	0.25°	0.29°	Dense	0.42 <sup>c</sup>	0.15°
106-99-0	13-Butadiene	54.09	1.12	2.0	11.5	44	24	0.14	0.75	Dense	5.5	0.15
106-97-8	Butane	58.12	1.09	1.5	9.0	36		0.14	0.81	Dense	5.9	0.15
25167-67-3	Butene	56.11	1.10	1.7	9.5	39	24	0.14	0.77	Dense	5.6	0.14
590-18-1	2-Butene-cis	56.11	1.12	1.6	9.7	37	24	0.14	0.76	Dense	5.6	0.11
624-64-6	2-Butene-trans	56.11	1.11	1.8	9.7	41	24	0.14	0.77	Dense	5.6	0.12
106-98-9	1-Butene	56.11	1.11	1.6	9.3	37	24	0.14	0.78	Dense	5.7	0.17
107-01-7	2-Butene	56.11	1.10	1.7	9.7	39	24	0.14	0.77	Dense	5.6	0.12
463-58-1	Carbon oxysulfide	60.08	1.25	12.0	29.0	290	26	0.18	0.41	Dense	1.3	0.29
7791-21-1	Chlorine monoxide	86.91	1.21	23.5	NA	830	31	0.19	NA	Dense	0.15	NA
557-98-2	2-Chloropropylene	76.53	1.12	4.5	16.0	140	29	0.16	0.54	Dense	3.3	0.011
460-19-5	Cyanogen	52.04	1.17	6.0	32.0	130	24	0.15	0.51	Dense	2.5	0.40
75-19-4	Cyclopropane	42.08	1.18	2.4	10.4	41	22	0.13	0.72	Dense	5.4	0.23
4109-96-0	Dichlorosilane	101.01	1.16	4.0	96.0	160	33	0.20	0.40	Dense	1.3	0.084
75-37-6	Difluoroethane	66.05	1.14	3.7	18.0	100	27	0.17	0.48	Dense	1.6	0.23
124-40-3	Dimethylamine	45.08	1.14	2.8	14.4	52		0.12	0.73	Dense	3.7	0.090
463-82-1	2,2-Dimethylpropane	72.15	1.07	1.4	7.5	41	27	0.16	0.80	Dense	6.4	0.11
74-84-0	Ethane	30.07	1.19	2.9	13.0	36	18	0.14	0.89	Dense	5.4	0.75
107-00-6	Ethyl acetylene	54.09	1.11	2.0	32.9	44	24	0.13	0.73	Dense	5.4	0.091
75-04-7	Ethylamine	45.08	1.13	3.5	14.0	64	22	0.12	0.71	Dense	3.6	0.040
75-00-3	Ethyl chloride	64.51	1.15	3.8	15.4	100	27	0.15	0.53	Dense	2.6	0.053
74-85-1	Ethylene	28.05	1.24	2.7	36.0	31	18	0.14	0.85	Buoyant⁵	5.4	0.63
109-95-5	Ethyl nitrite	75.07	1.30	4.0	50.0	120	30	0.16	0.54	Dense	2.0	NA
1333-74-0	Hydrogen	2.02	1.41	4.0	75.0	3.3	5.0	е	е	d	е	NA
75-28-5	Isobutane	58.12	1.09	1.8	8.4	43	25	0.15	0.82	Dense	6.0	0.23
74-82-8	Methane	16.04	1.30	5.0	15.0	33	14	0.15	1.1	Buoyant	5.6	0.87
74-89-5	Methylamine	31.06	1.19	4.9	20.7	62	19	0.10	0.70	Dense	2.7	0.12
563-45-1	3-Methyl-1-butene	70.13	1.08	1.5	9.1	43	26	0.15	0.77	Dense	6.0	0.030
115-10-6	Methyl ether	46.07	1.15	3.3	27.3	64	22	0.14	0.66	Dense	3.4	0.22
115-11-7	2-Methylpropne	56.11	1.10	1.8	8.8	41	24	0.14	0.77	Dense	5.7	0.18
463-49-0	Propadiene	40.07	1.16	2.1	2.1	34	21	0.13	0.73	Dense	5.2	0.20
74-98-6	Propane	44.10	1.13	2.0	9.5	36		0.14	0.83	Dense	5.7	0.38
115-07-1	Propylene	42.08	1.15	2.0	11.0	34	21	0.14	0.79	Dense	5.5	0.35
74-99-7	Propyne	40.07	1.16	1.7	39.9	28	21	0.12	0.72	Dense	4.9	0.18

Table D-2: Data for Flammable Gases (EPA 1999) (continued)

CAS Number	Chemical Name	Molecular Weight	Ratio of Specific		Flammability Limits (Vol%)		Gas Factor	Liquid Factor	Density Factor	Worst-Case Conditions <sup>a</sup>	Pool Fire Factor	Flash Fraction
		_	Heats	Lower (LFL)	Upper (UFL)		(GF) <sup>g</sup>	Boiling (LFB)	(Boiling) (DF)		(PFF)	Factor (FFF) <sup>f</sup>
7803-62-5	Silane	32.12	1.24	С	С	С	19 <sup>c</sup>	е	е	Dense	е	0.41
116-14-3	Tetrafluoroethylene	100.02	1.12	11.0	60.0	450	33	0.29	0.32	Dense	0.25	0.69
79-38-9	Trifluorochloroethylene	116.47	1.11	8.4	38.7	400	35	0.26	0.33	Dense	0.34	0.27
75-50-3	Trimethylamine	59.11	1.10	2.0	11.6	48	25	0.14	0.74	Dense	4.8	0.12
689-97 1	Vinyl acetylene	52.08	1.13	2.2	31.7	47	24	0.13	0.69	Dense	5.4	0.086
75-01-4	Vinyl chloride	62.50	1.18	3.6	33.0	92	26	0.16	0.50	Dense	2.4	0.14
75-02-5	Vinyl fluoride	46.04	1.20	2.6	21.7	49	23	0.17	0.57	Dense	0.28	0.37
75-38-7	Vinylidene fluoride	64.04	1.16	5.5	21.3	140	27	0.22	0.42	Dense	1.8	0.50
107-25-5	Vinyl methy1 ether	58.08	1.12	2.6	39.0	62	25	0.17	0.57	Dense	3.7	0.093

### NA: Data not available

- <sup>a</sup> "Buoyant" refers to neutrally buoyant gases and vapors; "Dense" refers to dense gases and vapors.
- <sup>b</sup> Gases that are lighter than air may behave as dense gases upon release if liquefied under pressure or cold; consider the conditions of release when choosing the appropriate
- <sup>c</sup> Reported to be spontaneously combustible.

  <sup>d</sup> Much lighter than air; table of distances for neutrally buoyant gases not appropriate.
- Pool formation unlikely.

  Calculated at 298 K (25 °C) with the following exceptions:

Acetylene factor at 250 K as reported in TNO, Methods for the Calculation of the Physical Effects of the Escape of Dangerous Material (1980).

Ethylene factor calculated at critical temperature, 282 K.

Methane factor calculated at critical temperature, 191 K.

Silane factor calculated at critical temperature, 270 K.

<sup>&</sup>lt;sup>g</sup> Use GF for gas leaks under choked (maximum) flow conditions.

Table D-3: Data for Flammable Liquids (EPA 1999)

CAS Number	Chemical Name	Molecular Weight		Flammability Limit (Vol%)		Liquid F	actors	Density Factor	Liquid Leak Factor	Worst - Case Condition <sup>b</sup>	Pool Fire Factor
			Lower	Upper		Ambient	Boiling		(LLF) <sup>a</sup>		(PFF)
			(LFL)	(UFL)		(LFA)	(LFB)				
590-21-6	1-Chloropropylene	76.53	4.5	16.0	140	0.11	0.15	0.52	45	Dense	3.2
60-29-7	Ethyl ether	74.12	1.9	48.0	57	0.11	0.15	0.69	34	Dense	4.3
75-08-1	Ethyl mercaptan	62.14	2.8	18.0	71	0.10	0.13	0.58	40	Dense	3.3
78-78-1	Isopentane	72.15	1.4	7.6	41	0.14	0.15	0.79	30	Dense	6.1
78-79-5	Isoprene	68.12	2.0	9.0	56	0.11	0.14	0.72	32	Dense	5.5
75-31-0	Isopropylamine	59.11	2.0	10.4	48	0.10	0.13	0.71	33	Dense	4.1
75-29-6	Isopropyl chloride	78.54	2.8	10.7	90	0.11	0.16	0.57	41	Dense	3.1
563-46-2	2-Methyl-1-butene	70.13	1.4	9.6	40	0.12	0.15	0.75	31	Dense	5.8
107-31-3	Methyl formate	60.05	5.9	20.0	140	0.10	0.13	0.50	46	Dense	1.8
504-60-9	1,3-Pentadiene	68.12	1.6	13.1	44	0.077	0.14	0.72	33	Dense	5.3
109-66-0	Pentane	72.15	1.3	8.0	38	0.10	0.15	0.78	30	Dense	5.8
109-67-1	1-Pentene	70.13	1.5	8.7	43	0.13	0.15	0.77	31	Dense	5.8
646-04-8	2-Pentene, (E) -	70.13	1.4	10.6	40	0.10	0.15	0.76	31	Dense	5.6
627-20-3	2-Pentene, (Z) -	70.13	1.4	10.6	40	0.10	0.15	0.75	31	Dense	5.6
75-76-3	Tetramethylsilane	88.23	1.5	NA	54	0.17	0.17	0.59	40	Dense	6.3
10025-78-2	Trichlorosilane	135.45	1.2	90.5	66	0.18	0.23	0.37	64	Dense	0.68
109-92-2	Vinyl ethyl ether	72.11	1.7	28.0	50	0.10	0.15	0.65	36	Dense	4.2
75-35-4	Vinylidene chloride	96.94	7.3	NA	290	0.15	0.18	0.44	54	Dense	1.6

NA: Data not available.

<sup>a</sup> Use the LLF only for leaks from tanks at atmospheric pressure.

<sup>b</sup> "Dense" refers to the tables for dense gases and vapors.